

# Knowledge Organisers Spring Term – Year 10

Name:\_\_\_\_\_

Please remember:

- It is to be kept inside your knowledge organiser book
  - It is to be brought into school every day

Regular retrieval throughout a scheme of learning (daily, weekly and monthly) has been proven to **reduce the rate of forgetting**, supporting you to **retain more** in long term memory- making assessments/ exams way easier! The challenge for you as a student is to make sure you use your knowledge organiser for each subject properly to help you to know more and remember more over time. We've created this walk through to support you in using your knowledge organiser- for more support speak to your subject teachers.

# Using your Knowledge Organiser



1	2	3	4	5
Look	Cover	Write	Check	Repeat
Start with a small section of knowledge	Now <b>cover up this section</b> of your	Self quiz- what can you remember and	Remove the post it and check for	After a short break away from your
that you want to remember e.g Henry	knowledge organiser with a post it note	rewrite? Make sure you do this without	<b>accuracy</b> - did you get the key	knowledge organiser repeat the look,
VIII's wives in History. Read through this	or scrap paper.	looking back at your knowledge	terminology? Was it spelt correctly?	cover, write, check <b>until you can recall</b>
section of the knowledge organiser (a		organiser.	Was the order correct? If you drew a	all of the facts correctly without
couple of times if it helps)			diagram, how much of this did you get	prompts.
			correct?	
				This process can be used for any new
			Most importantly-what did you miss	knowledge that you want to acquire. It
			out?	is good idea to do this on a regular
				basis, once a week.

Strategy 1- Look, cover, write, check – A really simple but effective way to use your knowledge organiser. Focus on a specific area of your knowledge organiser.

1	2	3	4	5
Focus	Big ideas	Explain it	Link it	Record it
Make it manageable by selecting an	Pick out the main points or the <b>big</b>	Explain what you know about the main	Now, see how it links to other areas	Write down as many 'think it, link it'
area of your KO where your learning is	ideas in this section.	points (this could be written or shared	within the subject. E.g Eating meat –	ideas as you can in your book. See if
not secure. Don't waste time going off		verbally – a friend, a family member.	causes global warming. Cows produce	you can beat others in you class!
something you can already do!			methane which is a greenhouse gas.	

Strategy 2- Think it, link it – Great for connecting the big ideas in your subject. How does 'x' relate to 'y'. What are the key factors which make an equation/ experiment/ process work? Challenge yourself to see how many links you can make!

1	2	3	4	5
Select topic	Prepare quiz	Answer it	Self check	Repeat
Decide which area you want to be	Get someone else to prepare 10	Set a time limit (depending on the	Now look at your KO to self check-	Return to this section in 2/3 weeks-see
quizzed on (this might build up over	random questions on that topic to	number of questions) and answer the	make a note of your score. Celebrate	if you can improve your score! Re-do
time)	challenge you.	questions without looking at your KO.	your successes and make a note of	those questions that you missed or got
			anything you missed or got incorrect.	incorrect.

Strategy 3- Knowledge quiz – You might try this after a few weeks of using your knowledge organiser. Get someone to set you 10 questions using your knowledge organiser. These could be spellings, key words, processes, equations etc to see how much you can remember! Record your score and see if you can beat your personal best each half term!

# **Contents Page**

Pages	Subject
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## KS4 – English – A Christmas Carol Plot and Key Quotations

- The narrator informs the reader that Jacob Marley (former business partner) died 7 years ago.
- 2. Scrooge is described as mean and *miserly* (d).
- 3. The weather is harsh and Bob Cratchit, Scooge's worker, is treated harshly..
- 4. Scrooge is mean towards Fred, his nephew.
- 5. He refuses to donate money to the portly gentleman.
- 6. He sees the face of Jacob Marley in his door knocker.
- He is visited by Jacob Marley (bound in chains) who informs him that he will be visited by three spirits.

The Ghost of Christmas Past visits Scrooge.

1.

- 2. Scrooge is taken back to his childhood and sees his former self alone at school.
- 3. We are introduced to his sister Fan and the poor relationship with his father.
- 4. Scrooge is then taken to **Fezziwig** where we see him enjoying himself at a party. He interacts with people.
- 5. Scrooge is then shown his former fiancée **Belle** and how they split. He becomes upset when he sees **Belle** and her family in another vision.
- 6. He demands to be taken home.

- 1. The Ghost of Christmas Present visits Scrooge.
- 2. He takes him to see Bob Cratchit and his family. Scrooge is surprised at how little the family have.
- He is then taken to his nephew, Fred's house. Fred and his guests mock Scrooge and his *miserly (d)* ways.
- 4. He is then taken to a poor part of the city and introduced to *Ignorance (f)* and *Want*.
- 5. The spirit becomes frustrated and leaves him there.

- 1. The Ghost of Yet To Come is shrouded in darkness: silent and scary.
- 2. He is taken to the city and instructed to listen to a group of businessmen who are discussing the recent death of a man who was not very well-liked.
- 3. He is taken to Old Joe's shop where a dead man's belongings are being pawned.
- 4. He is then taken to a bedroom where he sees a corpse (h).
- 5. Scrooge demands to see <u>tenderness</u> surrounding the death. He is then taken to the Cratchit's and learns the death of <u>Tiny Tim</u>.
- 6. Scrooge wants to identify the dead man and he is taken to his grave.

- Scrooge awakes disorientated and worries he has missed Christmas.
- 2. He is joyful when he learns he hasn't.
- 3. He orders and pays for the biggest turkey to be delivered to the Cratchit family.
- 4. He apologises and **donates** a healthy sum of money to the portly gentleman, to give to the poor.
- 5. He visits Fred and his wife and asks to join them for dinner.
- 6. He is **jovial the** next day, playing a trick on a terrified Bob, who arrives late to work.
- 7. He gives Bob a pay rise and pays for treatment <u>for Tiny Tim</u>.
- 8. Scrooge is **reformed** and is a complete contrast to Stave One!







KS4 – Englis	h – Language Paper 1 Section B		Sec	tion B: Question 5
<ul> <li>Start of the exam</li> <li>S minute plan with question in mind.</li> <li>Top Tips</li> <li>Keep your tone consistent</li> <li>(g) throughout: do not use words which suggest a light and playful</li> </ul>	5: The 'Writing for a purpose' question. Question 5: Writing a narrative or description AO5/AO6 You will produce a story or description based on a picture or a sentence detailing your task. Your local newspaper is running a creative writing competition and the best entries will be published. <u>Either</u> write a story about time travel as suggested by this picture or write a story entitled 'Stranded'. <u>Simile</u> Where you compare two	Key skills: AO5: You should:         Ensure the story or descript is clear.         Image: Structure of the story or descript is clear.         AO6: You should:         Use varied and accurate sentence structures.         Metaphor         Pate         Where you compare two	tion te hetic fallacy	Q       S
<ul> <li>atmosphere (b) after you have just spent 15 minutes making the scene sound scary.</li> <li>✓ Use a variety of structural (e) features: flashbacks (f),</li> <li>✓ Keep to one or two</li> </ul>	Key       things by saying they're like       things by saying they're like         Ianguage       something else:       things by saying they're like         devices to       "He was as timid as an       things by saying they're like         "He was as timid as an       "He was as timid as an       things by saying they're like         Overview of each paragraph       "."	Where you something else when it learly is not. She was a night owl."	atmosphere (b). ecame cloudy ness fell." Key terms A Cyclical	to describe the character's five senses. "I could taste blood streaming from my lip." Definition Returning back to a previous
<u>Characters and</u>	scene or setting – decide if it's positive or negative.		B Atmosphere	The tone or mood.
SPaG, including	P2: Character focus – introduce character – show	Connective Unless, although.	C Vice versa	The two items can be
Punctuation to use       Question     Exclamation	symbol, item or even to trigger the shift in time. P3: Come back to present moment, developing the	Adverb     Regretfully, sadly.       Simile     Like a mouse	D Motif	An object which is repeated and has importance to the events. E.g. a raven for death.
Mark ? Mark !	character in more detail. Keep something withheld!	Metaphor Brave lions, they	E Structural	The way a text is put together,
	P4: Describe the setting - zoom out to change focus. Include a motif (d).	Feeling Jealous, she tore up his clothes.		subheading or flashbacks (f) etc.
Semi- ; Speech " Colon Marks "	P5: Cyclical (a) development – back to weather/scene/setting – change from positive to	Verb 'ing'Giggling and laughing,clausethey ran to school.	F Flashback	When the text goes back in time.
Colon : Apostrophe '	negative or vice versa (c).		G Consistent	Keeping something <b>7</b> the same.

KS4 – Englis	<u>h – Language Paper 1 Section B</u>		Sec	tion B: Question 5
<ul> <li>✓ 5 minute plan with question in mind.</li> </ul>	5: The 'Writing for a purpose' question. Question 5: Writing a narrative or description AO5/AO6 You will produce a story or description based on a picture or a sentence detailing your task.	Key skills: AO5: You should: Ensure the story or descripti is clear.	ion	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
<ul> <li>Top Tips</li> <li>✓ Keep your tone consistent</li> <li>(g) throughout: do not use</li> </ul>	Your local newspaper is running a creative writing competition and the best entries will be published. <u>Either</u> write a story about time travel as suggested by this picture or write a story entitled 'Stranded'.	AO6: You should: Use varied and accurate sentence structures.		6 16 20 10
light and playful atmosphere (b) after you have just spent 15 minutes making the scene sound scary.	Key language devices to use:	re you compare two s by saying something nething else when it ly is not. Where you to set the d "The sky be and darkne	use the weather atmosphere (b). ecame cloudy ess fell."	Where you use vocabulary to describe the character's five senses. "I could taste blood streaming from my lip."
<ul> <li>✓ Use a variety of structural (e) features: flashbacks (f),</li> </ul>	Overview of each paragraph		Key terms	
✓ Keep to one or two characters and	P1: Always begin with the weather and describe the	Great sentence openers	A Cyclical	Returning back to a previous point like a cycle.
✓ 5 minutes' of checking		Connectiv Unless, although.	B Atmosphere	The tone or mood.
SPaG, including paragraphing.	P2: Character focus – introduce character – show but not tell then lead in to a flashback. Use a	e	C Vice versa	The two items can be switched around or reversed.
Punctuation to useQuestionExclamation	symbol, item or even to trigger the shift in time. P3: Come back to present moment, developing the	Adverb     Regretfully, sadly.       Simile     Like a mouse	D Motif	An object which is repeated and has importance to the events. E.g. a raven for death.
Mark ? Mark !	character in more detail. Keep something withheld!	Metaphor Brave lions, they	E Structural	The way a text is put together,
Comma , Full stop .	P4: Describe the setting - zoom out to change focus. Include a motif (d).	<b>Feeling</b> Jealous, she tore up his clothes.		wnether through paragraphs, subheading or flashbacks (f) etc.
Semi- ; Speech " Colon Marks "	P5: Cyclical (a) development – back to weather/scene/setting – change from positive to	Verb 'ing' Giggling and clause laughing, they ran to	F Flashback	When the text goes back in time.
Colon : Apostrophe '	negative or vice versa (c).	school.	G Consistent	Keeping something <b>8</b> the same.





# **KS4 – Macbeth – Plot and Key Quotations**

1.

2.

- 1. The witches meet on the heath.
- 2. Macbeth and Banquo have fought and won a battle. They are praised for their bravery by the Captain.
- 3. The witches meet Macbeth and Banquo and they make three predictions. They predict that Macbeth will become Thane of Cawdor and eventually king. Banquo is told that he will not become king but that his children will be. Shortly after, Macbeth is told he is the new Thane of Cawdor!
- 4. Macbeth plans for Duncan to stay at his castle
- 5. Lady Macbeth convinces Macbeth to kill King Duncan.
- 6. Duncan arrives at Macbeth's castle.
- Macbeth tells Lady Macbeth he will not murder Duncan. However, she convinces him to go ahead with the murder.

- Macbeth sees a dagger in front of him on his way to go and kill King Duncan.
- Macbeth murders King Duncan. Macbeth's guilt is apparent as he appears covered in blood. Lady
- Macbeth feels no guilt and places the daggers on Duncan's guards. The dead body is discovered
- The dead body is discovered by Macduff. Duncan's sons, <u>Malcolm and Donalbain, run</u> <u>away</u>: one to England and one to Ireland.
- 4. Macbeth and his wife become king and queen of Scotland.

- Macbeth questions Banquo. He is suspicious of Banquo. As a result, he plans his murder with the help or murderers.
- 2. Banquo is murdered by the hired murderers but his son, Fleance, manages to escape.
- 3. At the coronation (a) banquet, <u>Macbeth sees the</u> ghost of Banquo at the table. Lady Macbeth persuades everyone that Macbeth is ill and they must leave and is furious at Macbeth.
- 4. Hecate, the witches's leader, is angry that the witches meddled with Macbeth without her permission.
- 5. Lennox shares his suspicions about Macbeth.

The witches share three more prophecies (b):

1.

- Macbeth is told that: he must fear Macduff; that he will not be harmed by anyone born of a woman and that he will never be defeated until Birnam Wood moves to 'High Dunsinane Hill .' They also share a vision of Banquo.
- 3. Macbeth has Macduff's wife and children murdered.
- 4. In England, Malcolm tests Macduff's loyalty and checks they are on the same side, wanting the same things.

- 1. Lady Macbeth sleepwalks: she is overcome with guilt and loneliness.
- 2. The rebels discuss the impending battle.
- 3. Macbeth declares he has nothing to fear from the battle. He appears confident.
- 4. The doctor reports on Lady Macbeth's condition. Macbeth orders him to cure her.
- Malcom and Macduff's forces march to Dunsinane castle, disguising themselves with branches from the trees of Birnam Wood.
  - Lady Macbeth dies off stage.

6.

- 7. Macbeth realises that he is not going to win but decides to at least die fighting.
- 8. Malcolm prepares for battle.
- 9. Macbeth kills young Siward.
- 10. Both Macduff and Macbeth fight and Macbeth is killed . His head displayed on a spike, as he was a traitor. Macduff is able to kill him as he reveals he was born by caesarean (k) section.





SWB KS4 – Englis	h – Power and Co	onflict Poetry Co	ontext			Keyword	Definition
	AC A			ČKČ SA	A	Tyrannical	Being cruel with power.
				<b>Global</b>	В	Radical	A person who wants to change society.
Romantic era (g)	Imperialism (d)	Global Conflict	Global Conflict	Displacement (e)	С	Oppressed	Being treated harshly, often by rulers.
1792	1854	1914-18	1939-45	onwards	D	Imperialism	Pushing your country's ideas on others.
Poems: 1. London 2. Extracts from the	Poems: 4. My Last Duchess	Poems: 6. Exposure	Poems: 11. Kamikaze	Poems: 8 Storm on the Island	E	Displace	Taking over the place or role.
Prelude 3. Ozymandias	5. Charge of the Light Brigade Key ideas:	Charge Key ideas:	Key ideas: • In the Second World War:	9. War Photographer 10. The Emigree 12. Tissue	F	Industrial Revolution	The use of machines in factories.
<ul> <li>Key ideas:</li> <li>Poems are not about love.</li> <li>Poems are about personal growth and</li> </ul>	<ul> <li>1 in 5 people in the world called Queen Victoria of Great Britain their governor (j).</li> </ul>	<ul> <li>In the First World War:</li> <li>20 million peopl<u>e died.</u></li> </ul>	<ul> <li>75 million people died.</li> <li>It was fought across the globe.</li> <li>Pearl Harbour in the</li> </ul>	<ol> <li>13 Checkin' Out Me History</li> <li>14. Remains</li> <li>15. Poppies</li> </ol>	G	Romantic era	Poems about nature and its impact in a changing industrial (f) world
<ul> <li>appreciating nature during the Industrial Revolution (f).</li> <li>Poets sometimes</li> </ul>	<ul> <li>People were very poor.</li> <li>People lacked rights.</li> <li>The social</li> </ul>	<ul> <li>It was fought from trenches (holes in the ground) in</li> </ul>	USA was attacked in December 1941. Japanese pilots were trained to bomb ships by flying into them. This	Wars were often     fought against smaller     countries to stop     tyrants (a) or terrorists	H	Hierarchy	A system where people or items are ranked by status or power.
fought back against what they thought were tyrannical (a) rulers who oppressed (c)normal people.	<mark>hierarchy</mark> (h) still defined who had <mark>power</mark> .	Europe.	<ul> <li>meant they killed themselves for their country.</li> <li>Following this, America came into the war on</li> </ul>	<ul> <li>(i).</li> <li>Locals thought Western countries invaded for oil and other resources.</li> </ul>	I	Terrorists	A person or group who uses violence to achieve their goals to change world.
• This made them considered radical (b).			Britain's side, changing which side had the upper hand in the war.	• Many locals were still loyal to the tyrants (a).	J	Governor	The person in charge. <b>13</b>





# KS4 – English – Unseen Poetry Knowledge Organiser





Year 10 Maths – Unit A12 – Quadratic Graphs & Equations				Definition/tip
Before progressing through this section of work	Before progressing through this section of work, you may find it useful to look back at <b>Crossover Unit 19- Expand and</b>			
Simplify, Unit 20 – Factorising, Unit 21-Solving Equations knowledge organisers.			Quadratic	An expression/equation involving x <sup>2</sup>
Solving Quadratic Equations by Factorisation	Solving Quadratic Equations Using the Quadratic Fo	ormula	(expression/ equation)	
You must be able to factorise quadratics in order to solve quadratic equations using this method	Solve $3x^2 + 8x + 2 = 0$ Give your solutions correct to 3 significant figures.	This is a hint for you to use the formula as you will not be able to solve	Factorise	An expression written as a product of it's factors.
<b>Example 1</b> Solve $x^2 + 6x + 5 = 0$ This factorises into $(x + 5)(x + 1) = 0$	$x = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$ You need to learn this!	by factorising. You will need to use a calculator.	Quadratic formula	A formula for solving quadratic equations. The solution of the equation $ax^2 + bx + c = 0$ is given by
Each bracket needs to equal 0 x + 5 = 0 or $x + 1 = 0x = -5$ or $x = -1Example 2$	Figure out a, b and c a = 3 $b = 8$ $c = 2Using brackets here foravoid calculation error$	r a, b and see will help to rs.		$x = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$
Solve $x^2 + 3x - 10 = 0$ This factorises into $(x + 5)(x - 2) = 0$ x + 5 = 0 or $x - 2 = 0x = -5$ or $x = 2$	$2(3)$ You will gain 1 mark for $= \frac{-8 \pm \sqrt{64 - 24}}{6}$ 1 mo	or substituting in a, b and c ark for simplifying	Completing the square	A method of solving quadratic equations which involves rewriting the equation $x^2 + px + q$ in the form $(x + a)^2 + b$
<b>Example 3</b> Solve $x^2 - 6x + 9 = 0$ This factorises into $(x - 3)(x - 3) = 0$ This equation has repeated roots	$-8 + \sqrt{40} = \frac{-8 - \sqrt{40}}{1 + \sqrt{40}}$	/40	Solution	A value or values that we can put in place of a variable (such as x) that makes the equation true.
$(x - 3)^2 = 0$ This means there is only one solution, $x = 3$	$= \frac{-6}{6}$ x = -0.279 x = -2.39 x = -2.39		Coefficient	A number used to multiply a variable. Eg, 3x (3 is the coefficient)
In order to solve quadratic equations, you need to be able to recognise when you can solve by factorising or when you need to use the quadratic formula or complete the square. Sometimes it will tell you which method to use in the question.	1 mark for both		<u>Other top</u> <u>in</u> : A-level	pics/Units this could appear
You will also need to be able to solve problems which involve solving quadratic equations too. (See 2 <sup>nd</sup> slide)	Exam Tips Quadratic formula questions will appear of calculator papers (2 or ) often worth 3 mc	on one of the rks.	Pure 1 -Quadratic -Equations	s and Inequalities <b>18</b>

Year 10 Maths – Unit A12 – Quadratic Graphs & Equations			Keyword/	Definition/tip		
Before progressing through the	Before progressing through this section of work, you may find it useful to look back at Crossover Unit 19 - Expand and				3KIII	
Simplify, Unif 20 – Factorising,	, Unif 21 - Solving Equ	Jations kn	owleage organisers.		Quadratic	An expression/equation involving x <sup>2</sup>
Solving quadratic equations	by Completing the S	quare	<b>Example 2</b> Rewrite $x^2 + 4x - 7$ in the form	$m(x + a)^2$ , b. Hence solve the	equation)	
specified number of decime answers in surd form.	al places or to leave		equation $x^2 + 4x - 7 = 0$ givin places. (Calculator paper)	ng your answer to 2 decimal	Factorise	An expression written as a product of it's factors.
$(x + a)^2 = x^2 + 2ax + a^2$	may remember mar,		Using the same method as	example 1	Quadratic	A formula for solving quadratic
which can be rearranged to $X^2 + 2ax = (x + a)^2 - a^2$	o give:		$x^2 + 4x = (x + 2)^2 - 4$	Ignore the -7 to begin with	formula	equations. The solution of the equation $ax^2 + bx + c = 0$ is given by
This is the basic principal beh	nind completing the	square.	Bring back the -7 so	Divide the coefficient by 2 Put it into your bracket and		$-h + \sqrt{h^2 - 4ac}$
Rewrite the following in the fo	form (x± a) ± b		(x + 2) <sup>2</sup> – 4 -7 giving	square it.		$x = \frac{b \pm vb}{2a}$
$x^{2} + 6x - 7$ Rewrite $x^{2} + 6x - 6x + 3x^{2} - 9$	$x^2 + 6x - 7$ Ignore the -7 to begin with		$(x + 2)^2 - 11$ Combine the constant terms to get the final answer. (-9 - 7 = -16)		Completing	A method of solving quadratic
Bring back the -7 so $(x + 3)^2 - 9 - 7$	Divide the coefficie 2 Put it into your brac	nt by ket	We can now rewrite the equation in the question as; $(x + 2)^2 - 11 = 0$ Rearrange		the square	equations which involves rewriting the equation $x^2 + px + q$ in the form $(x + a)^2 + b$
(x + 3) <sup>2</sup> -16 Conget	and square it. ombine the constant terms t the final answer. (-9 – 7 =	to -16)	(x + 2) <sup>2</sup> = 11	Take the square root of both sides	Solution	A value or values that we can put in place of a variable (such as x) that makes the equation true.
			$x + 2 = \pm \sqrt{11}$	This answer are in surd form and	Coefficient	A number used to multiply a
<b>Example 3</b> Solve $x^2 - 6x - 1 = 0$ by compl	letina the sauare.		$x = -2 \pm \sqrt{11}$ sometimes you may be asked to leave like this. This answer requires two decimal places			Eg, 3x (3 is the coefficient)
Leave your answer in the form a $\pm \sqrt{b}$ . x <sup>2</sup> - 6x = (x - 3) <sup>2</sup> - 9		Use calculator		Other ten	ice // Ireite this could grap car	
$(x - 3)^2 - 9 - 1$ $(x - 3)^2 - 10 = 0$	Rearrange		x = 1.32 or x = -5.32	(to 2 decimal places)	<u>oineriop</u> in:	ics/units this could appear
$(x - 3)^2 = 10$					A-level	
x-3 = $\pm \sqrt{10}$ Take the square root of both sides		lips leting the square question	ns can appear on non-	-Quadratics -Equations a	nd Inequalities	
$x = 3 \pm \sqrt{10}$ calculation would		like you to give your answ	ver.		19	













STEADEMY Year 10 Maths – Unit G	GM8 – Surface Area & V	olume	Volume		
Volume Using Unit Cubes	Exa	mple:		Keyword /Skill	Definition/Tips
This is a unit cube. It has a volume o You may get shapes made from uni	f 1cm <sup>3</sup> . $\longrightarrow$		The volume of this cuboid	Prism	A 3D shape that has two identical ends and flat sides.
and you will need to find the volume shape by counting how many cube	e of that		would be 24cm <sup>3</sup> as there are 24 unit cubes altogether.	Section	cutting through it.
				Diameter	The distance from one point of a circle to another passing through the centre. It is twice the radius.
Volume of Prisms	A prism has a constant cr	oss-section. To find	d the volume of it you first need	Radius	The distance from the centre of the circle to the circumference. It is half the diameter.
	to find the area of the cro the shape.	oss section and mu	ultiply it by the height/depth of	Formula/ Formulae	A rule or fact written with mathematical symbols. $(V = I \times w \times h)$ .
	Formula for Volume of Prir	ms: Area of Cross-	Section x Height	Compound Shape	A compound shape is made up of two or more basic shapes.
Example:				Face	A flat surface of any object.
You are given th section so you no area by the dep	e area of the cross- eed to multiply the		trea of cross-section = $\pi \times r^2$ = $\pi \times 3^2$ = 28.27cm <sup>2</sup>	Perimeter	The distance around the outside of a shape.
Area $12 \text{ cm}^2$ $12 \text{ cm}^2 \text{ x 7} = 84 \text{ cm}^2$	n <sup>3</sup>	↓ <sup>/cm</sup> ∨	olume = Area of Cross-Section x Height = 28.27 x 7 = 197.89cm <sup>3</sup>	Area	The space inside a 2D shape.
7cm				Surface Area	The total area of the outside of a 3D shape.
Volume of Spheres Formula for volume	e of a Sphere: $\frac{4}{3} \times \pi \times r^3$	<u>Exam</u> ! Make sure you u	se the correct units with your	Volume	The amount of 3-dimensional space something takes up.
3cm, You are given this	formula in the exam, you	answer. Area uses square	e units and volume uses cubic	Other to	pics/units this may appear in:
		units. Examples:		Perime	eter and Area
Example: Volume = $\frac{4}{3} \times \pi \times 3$	<sup>3</sup> = 113.1cm <sup>3</sup> (1d.p.)	Area units: cm <sup>2</sup> , Volume: cm <sup>3</sup> , m <sup>2</sup>	m², mm², km², etc. ³, mm³, km³, etc.	<ul><li>3D for</li><li>Mensu</li></ul>	ms vration <b>26</b>







# Year 10 Maths – Unit GM8 – Surface Area & Volume

			Keyword/	Definition/tip
Sphere Surface Ar	$ea = 4 \times \pi \times r^2$	Formula you need to remember for the	Skill	
	= $4 \times \pi \times 5^2$ = $100\pi$ cm <sup>2</sup> or $314$ cm <sup>2</sup> (3 sig figs)	Volume of a Pyramid: $\frac{1}{2} \times area \ of \ base \times height$	Pyramid	A solid object (3D) where the sides are triangles that meet at the top. (Apex)
5cm Volume	$= \frac{4}{3} \times \pi \times r^3$	The formulas you need for the surface area & volume of a cone and a sphere	Cone	A solid object (3D) that has a circular base joined to a point by a curved side. The point is called a vertex.
The radius of this Sphere is 5cm.	$= \frac{4}{3} \times \pi \times 5^{3}$ = 524cm <sup>3</sup> (3 sig figs)	not need to memorise them, just make sure you can use them!	Sphere	A solid (3D object) shaped like a ball. Every point on the surface is the same distance from the centre
Frustum (Shaded bit)	Scale Factor 10÷5	= 2	Frustum	Usually a pyramid or cone with the top cut off flat.
<b>A</b>	Radius of top circle $4 \div 2 =$	= 2 <i>cm</i>	Surface Area	Total area of the surface of a 3d shape
	Volume of big cone $\frac{1}{3} \times \pi \times$	$4^2 \times 10 = 167.6 cm^3$	Volume	The amount of 3D space that something takes up
/ `` 5cm	Volume of small cone $\frac{1}{3} \times \pi \times$	$x 2^2 \times 5 = 20.9 cm^3$		
	Volume of Frustum 167.6 –	$-20.9 = 147 cm^3$ 3 sig figs	<u>Other topics/</u> A-Level – Pur	/ <b>Units this could appear in</b> : e 1 – Forming and Solving
	Length of big cone $\sqrt{4^2}$ +	$10^2 = 10.77cm$	Differential E	quations
4cm	Length of small cone $\sqrt{5^2}$ +	$-2^2 = 5.39cm$		
	Area of curved surface $(\pi \times 4)$	$\times 10.77) - (\pi \times 2 \times 5.39) = 101.5 cm^2$	Exam Tip:	
Remember: The formulae for cones are on the previous page	Surface Area of Frustum 101.5	$+\pi \times 4^2 + \pi \times 2^2 = 164cm^2$	You will sometim 'Similar Shapes' lengths that are	nes need to use the properties of and/or Pythagoras to calculate some required <b>29</b>

# **Biology Knowledge Organisers**



Keyword	Definition
Magnification	The number of times larger an image is than the initial object that produced it
Objective lens	The part of the microscope that is closest to the specimen
Resolution	The smallest change that can be measured by an instrument
Nucleus	The control centre of the "eukaryotic cell"
Eukaryotic	A cell with a nucleus
Microvilli (us)	A tiny fold in the cell surface membrane of a cell, increase the surface area of the cell
Adaptations	The features of something that enable it to do a certain function
Gametes	A haploid cell produced by meiosis used for sexual reproduction
Haploid	A cell or nucleus that has one set of chromosomes. Gametes are haploid
Epithelial cells	A cell found on the surface of internal organs
Chromosomal DNA	The main bulk of DNA found in a cell. In humans, this DNA is found in chromosomes
Prokaryotic	A cell with no nucleus is prokaryotic
Monomers	A small molecule that can join with other molecules like itself to form a polymer
Polymers	A long-chain molecule made by joining many smaller molecules (monomers)
Biuret test	A test that uses copper sulfate solution and potassium hydroxide solution to test for proteins. It turns from blue to purple in the presence of proteins
Benedicts solution	A solution used to detect the presence of reducing sugars (eg. Glucose) in foods
Calorimeter	Apparatus used to measure the energy content of substances by burning them and measuring the temperature increase
Ethanol emulsion test	A test using ethanol to detect lipids (fats) in food
Active site	The space in an enzyme where the substrate fits during an enzyme-catalysed reaction
Denatured	A denatured enzyme is one where the shape of the active site has changed so much that the substrate no longer fits and the reaction can no longer happen
Optimum temperature	The temperature at which an enzymes rate of reaction is greatest, or at which a population of microorganisms grow most rapidly
Osmosis	The movement of water from a high concentration to a low concentration through a partially permeable membrane
Active Transport	The pumping of particles across a cell membrane from a low concentration to high concentration (requires energy)



The gap in a synapse is only about 20 nm (0.00002 mm) wide.



	ar 9 – Science – B3b.	Keyword	Definition		
1. Evidence for human evolution			2. Darwin's Theory of Evolution	Binomial system	The system of naming organisms using two Latin words
Ardi (Ardipithecus ramids)	Human like female fossil		Evolution is a change in the inherited characteristics of a	Evolution	A change in one or more characteristic of a population over a long period of time
	<ul> <li>Walked upright</li> <li>Long arms and short legs</li> <li>Small skull and brain</li> </ul>		population over time. This occurs through a process called natural selection.	Genetic variation	Differences between organisms caused by differences in the alleles they inherit from their parents, or differences in genes caused by mutation. Also called inherited variation
Lucy (Australopithecus afarensis) • More human like female fossil than • Walked upright better than Ardi • Arm and legs were the length better and human		rdi en ape 	<ul> <li>The differences in a population gives some</li> <li>individuals an advantage.</li> <li>This individual is more likely to survive for longer and</li> </ul>	Natural selection	A process in which certain organisms are more likely to survive and reproduce than other members of the same species because they possess certain genetic variations
Turkana Boy (Homo	<ul> <li>Skull and brain slightly larger than Ardi</li> <li>More human like female fossil than Lucy</li> <li>Walked upright better than Lucy</li> <li>Arm and legs were human length</li> <li>Skull and brain larger than Lucy</li> </ul>		<ul> <li>be able to breed to pass on desirable genes.</li> <li>Nature is selecting the individual with the</li> </ul>	Resistance	When an organism has resistance to something, it is unaffected by it, or not affected very much
Richard Leakey			phenotypes most suited to survival ('survival of the fittest). This is called natural selection.	Pentadactyl limb	A limb that has five digits (fingers and thumbs). Amphibians, reptiles, birds and mammals share this characteristics
Evidence for human evolution can also be gained from looking at stone tools, which become more sophisticated overtime			Our understanding of evolution has also been helped by	Classification	The process of sorting organisms into groups based on their characteristics
			Genetic Mutation Causes Drug Resistance Non-resistant Bacteria Some mutations Drug resistant	Kingdoms	There are five kingdoms into which organisms are usually divided: plants, animals, fungi, protists and prokaryotes
			bacteria multiply by make the bacterium bacteria multiply and thrive.	Selective breeding	When humans choose an organism that has a certain characteristic and breed more of these organisms, making that chosen characteristic more and more obvious
3. Classification The number of organism s in each group gets Class		4. Genetic Engineering and Tissue Culture	Varieties	Groups of plants of the same species that have characteristics that make them different to other members of the species	
		man cell bacterium bacterium plasmid	Genetic engineering	Altering the genome of an organism, usually by adding genes from another species.	
		g the insulin gene the nucleus. The vector DNA and the DNA being	GMOs	An organism that has had its genome genetically altered (genetic modification)	
		vest i cuon enzymest	inserted have sticky ends. When mixed with ligsse enzyme, the pieces of DNA combine. This is alled recombinant	Yield	The amount of useful product that you can get from something
but they have	Order	C	Image: Samples develop       Image: Samples develop <t< td=""><td>Unaffected or less affected by a certain disease</td></t<>		Unaffected or less affected by a certain disease
more features in	Family Genus	8		Stem cells	An unspecialised cell that continues to divide by mitosis to produce more stem cells and other cells that differentiate into specialised cells
common	Spec		Bacteria make human insulin.	Tissue culture	Growing tiny pieces of tissue, or cells, in the lab
	ięs			Pests	Animals that cause problems, such as damaging crops
				Biological control	Using organisms to kill problem organisms, such as pests or weeds
using enzymes. The	genes are then transferred	Insecticides	A substance used to kill insect pests 34		

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## KS4 Biology - Health, Disease and the Development of Medicine

#### **Transmission of Disease** Method of Caused Disease Symptom Control spread of disease by: transmission by: Preventing breeding of Recurrent Malaria mosquitoes or use of a net to Animal vector Protist fever prevent being bitten. Remove infected leaf litter Chalara ash Leaf loss and Airborne Clean all tools, vehicles and Fungus dieback bark lesions footwear. Clean drinking water and Cholera Diarrhoea Waterborne Bacteria good hygiene and sanitation. Vaccination programme. Tuberculosis Lung damage Airborne Treat infection with Bacteria antibiotics. Use of condoms / clean Sexual contact or needles. HIV Flu like illness Virus bodily fluids. Treat infection with Direct contact. antiretroviral drugs. Avoid contact with people Haemorrhagic Bodily fluid -Ebola infected with Ebola. Virus fever direct contact



Communicable diseases:	Non-communicable diseases:
Common cold	Diabetes
Influenza	Cancer
STDs	Cardiovascular disease





#### Cardiovascular Disease



Cardiovascular disease (CVD) is a general term for disease which involve the heart or blood vessels. Atherosclerosis is a cause of coronary heart disease where layers of fatty material build up inside the coronary arteries, narrowing them. This reduces the flow of blood through the coronary arteries, resulting in a lack of oxygen for the heart muscle.

Atherosclerosis can be treated in two main ways by placing a stent in the coronary artery and/or using lifelong medication called statins. Lifestyle changes such as a healthy diet, exercise and no smoking are also vital in



replicating viral DNA at each cell division. Stents are metal cylinder grids which can be inserted into an artery to maintain blood flow by keeping the artery open so that the heart continues to receive enough oxygen to function effectively.

Statins are drugs that lower harmful **cholesterol** in the blood and stop the **liver** producing too much cholesterol and reduce the rate at which it is deposited. Patients should change their **lifestyle** and have a healthy **diet**. This **reduces** the risk of heart disease.

### <u>Obesity</u>

Excess weight (obesity) can make a person at risk of cardiovascular disease, a stroke and Type 2 diabetes. A tool called the Body Mass Indicator (BMI) can be used to calculate whether a persons weight lies within a healthy range.

BMI= mass (kg) (height (m))<sup>2</sup>

The use of **BMI** has **limitations** because it simply shows if a person is carrying too much weight. It does not calculate if this is excess fat, muscle or bone. The **waist to hip ratio** should be considered alongside the BMI figure.

35

	KS	4 Biology – Plant Str	Keyword	Definition				
cuticle upper epidermis palisade upper la lisade upper epidermis palisade upper epidermis upper epidermis upper epidermis upper epidermis upper epidermis upper epidermis upper epidermis upper epidermis upper epidermis				flow is from obts to leaves with rand one way flow with light with light			Active Transport	Active transport moves substances from a more dilute solution to a more concentrated solution (against a concentration gradient). The energy is provided by respiration
mesophyll spongy lower epidermis cuticle guard cell stoma		A nthesis	Plan of li f	ts make use ght energy rom the vironment	Carbon dioxide + Water →	Oxygen + Glucose	Xylem	Form hollow tubes made of dead tissue. Long cells with walls toughened by waterproof lignin. Water and minerals flow from the roots
		Photos	END tor (	OTHERMIC) make food glucose)	$CO_2 + H_2O \rightarrow O_2 + C_6H_{12}O_6$			towards the leaves in one direction in a process called TRANSPIRATION. Xylem vessels also provide support to the stam of the plant
	Waxy cuticle (top layer of the leaf)	Reduces water loss from the leaf		Factor	How the rate is affected	Limiting factors (why the rate stops going up)	Phloem	Form tubes made of living tissue.
Epidermal tissues	Guard cells and stomata	Guard cells open and close the stomata to control water loss and allow for gas exchange (oxygen and carbon dioxide).			As the temperature of the environment the plant is in increases rate of	Photosynthesis is an enzyme controlled reaction. If the temperature increases		them. Sucrose in solution move from the leaves to growing tips and storage
Palisade mesophyll	Palisade cells	Cells near the top surface of the leaf that are packed with chloroplasts that contain chlorophyll. Both adaptations maximize photosynthesis.		Temperature	photosynthesis increases (up to a point) as there is more energy for the chemical reaction.	too much, then the enzymes become denatured and the rate of reaction will decrease and stop	Storeg	tissues in both directions. This process is called TRANSLOCATION. There are no forces causing translocation to occur and so the sucrose is moved along using active uptake which requires energy.
Spongy mesophyll	Air spaces in the leaf between cells	Increased surface area for gas exchange so that carbon dioxide can diffuse into photosynthesising cells			Light intensity increases as the distance between the plant and the light sources increases. As light intensity increases so does the rate of photosynthesis (up to a point) as more energy is available for the chemical reaction.	At point X another factor is limiting the rate of photosynthesis. This could be carbon dioxide concentration, temperature or the amount of chlorophyll		
xylem	Hollow tubes strengthened by lignified dead cells adapted for the transportation of water and mineral ions through the plant in the transpiration	Allows transport of water and mineral ions from the roots to the stem and the leaves.		Light intensity			Guard Cell	Cells either side of the stoma that
	Cell sap moves from one	Transports dissolved sugars from	Fac		Carbon dioxide is needed		s	flowering, increase fruit size.
phloem	phloem cell to the next through pores in the end walls	the leaves to the rest of the plant for immediate use or storage (translocation).		Carbon dioxide	for plants to make glucose. The rate of photosynthesis will	At point X another factor is limiting the rate of photosynthesis. This could be light intensity	Ethene	Control ripening of fruit during storage and transport.
Meristem tissue	New cells (roots and shoot tips) are made here including root hair cells	Root hair cells have an increased surface area for the uptake of water by osmosis, and mineral ions by active transport.		concentration	increase when a plant is given higher concentrations of carbon dioxide (up to a point).	could be light intensity, temperature or the amount of chlorophyll	Auxins	Weed killers, rooting powders, promoting growth in tissue culture. <b>36</b>
KS4 Biology – Animo	I Coordination,	Control c	and Homeostasis (part 1)	Keyword	Definition			
---	--	--	--	-----------------------------	--			
The Human Endocrine System		The Men	strual Cycle	Endocrine system	Collection of organs/glands in the human body that release hormones			
Pineal gland Thalamus Pituitary gland	High levels of OESTROGEN stimulate the production of LH (positive feedback) RELEASES LH CESH STIMULL ripening and Q	ARY GLAND ES F5H ATES egg DESTROGEN High leve OESTRO High leve OESTRO	Is of SEN he of FSH dback)	Hormones	Chemical messengers released from endocrine glands that cause a change in bodily responses			
Thyroid: Thyroid cartilage of the larynx Thyroid gland Parathyroid glands for opsterior site	LH STIMULATES         egg           release and STIMULATES         egg	in ovaries) <u>RELEASES</u> <u>PROGESTERONE</u> Hi	Menstruation Lining of the Lining of the uterus is Lining breaks uterus builds up maintained down		released and effects			
Adrenal glands	PROGESTERONE production in the ovaries	PROGESTERONE maintains the lining of he uterus in preparation for fertilised egg	GESTERONE hibit the hibit the hive feedback) 1 7 14 21 28 Day of mentrual cycle		transferred by all the reactions that take place in your body			
Vierus	Hormone FSH	Produced in Pituitary	Causes	Glycogen	How glucose is stored as a polymer			
Ovaries (female) Testes (male)	Follicle stimulating hormone Oestrogen	Gland Ovaries	Lining of the womb to develop. Stimulates pituitary gland to make LH	Fight or flight response	When the body prepares to fight or run away from a perceived danger. This response is caused by the hormone adrenaline			
Anti-diuretic hormone (ADH) Kidney Controls water levels in the blood	LH Luteinising hormone Progesterone	Pituitary Gland Ovaries	Stimulates egg release and progesterone production in the ovaries Maintains the lining of the womb	Menstrual cvcle	A roughly 28 day cycle of changes that occurs in the female reproductive system			
Thyroid-stimulating hormone (TSH)         Thyroid         Stimulates the thyroid gland to secrete thyroxine           Luteinising hormone (I H)         Ovaries         Stimulates egg release and progesterone production in the ovaries		Contro	aception	Monstruction	When the lining of the uter is brooks down and			
Follicle-stimulating hormone (FSH)         Ovaries         Stimulates egg ripening and oestrogen production (in ovaries)           Prolactin (PRL)         Breasts         Stimulates the breasts to produce milk           Growth hormone         All cells in the stimulates growth and consir	The pill - oral contraceptiv oestrogen to inhibit FSH pr no eggs develop and ma	ves that contain roduction so that iture. After takina	Barrier methods such as condoms and diaphragms	Mensiloalion	passes out through the vagina, Also known as a period			
(GH) body Stimulates growth and repair Low levels of thyroxine detected in the blood The hypothalamus in the brain releases TRH into the blood	for a while egg developm will stop completely.	nent and release	prevent the sperm reaching an egg	Ovulation	When an egg is releases from its follicle in the ovary, happens roughly on day 14 of the menstrual cycle			
TRH acts upon the pituitary gland Pituitary gland releases TSH into the blood	progesterone. High levels cervix to produce a thick stops sperm entering the u	s contain stimulate the mucus which uterus.	The 'coil' or other intrauterine devices which prevent the implantation of	Contraception	A method that prevents fertilisation and therefore pregnancy. Contraception can be hormonal or physical/barrier methods			
TSH acts upon the thyroid gland Thyroxine is released	Implant or skin patches of progesterone inhibit the m release of eggs for a numb	f slow release naturation and ber of months or	an embryo Sterilisation or vasectomy -	ART	Assisted reproductive technology that uses hormones and other techniques to increase the chance of pregnancy in infertile women.			
Normal levels of thyroxine inhibit the release of TRH and production of TSH. This is called negative feedback.	Spermicidal agents which sperm.	n kill or disable	surgical methods of male and female sterilisation.		37			



# KS4 Biology – Animal Coordination, Control and Homeostasis (part 2)



				KS4 –	Bio	logy Exc	hange and	l Tro	ansp	ort			Keyword		Definition	
		13						s	-			=	incy word			
6	ar I	A	alveolus					Inction	Right v	ventricle	Pumps blood to the lungs where g exchange takes place.	15	Diffusion	The spreading of	of the particles of a gas	
bronchioles (air sacs at the end)		02 Alveolar wall	vena cava			ve different fu	Left ventricle		Pumps blood around the rest of th body. Thicker cardiac muscle in the wall.	e		or substances in solution, resulting in a net movement of particles from a region where they are of a higher				
			capillary -	red			pulmonary veins	art ha	Deserved		Controls the natural resting heart	ate.		concentration i	o an area of lower	
lung		- bronchus	CO <sub>2</sub> out	biodd cens			left atrium	the he	right	atrium)	Artificial electrical pacemakers car fitted to correct irregularities.	be	Surface	The surface are	a to volume ratio can	
diaphragm			gas exchange in ar	alveolus	ngnt atriur		left ventricle	tructure in	Coronar	y arteries	Carry oxygenated blood to the care muscle.	liac	areas to volume	surface area (S	an object's A) by its volume	
					right ven	tricle		erent s			Prevent blood in the heart from		Cardiac	Cardiac output	= stroke volume x heart	
Trachea	Carries air t the lun	to/from ngs	Rings of cartilage prot airway.	ect the				Diffe	Heart	valves	flowing in the wrong direction.		Output	rate		
	Carries air t	to/from	Splits into multiple pa	thways			-									
Bronchioles	the air sacs	(alveoli)	to reach all the air sac	s.		Aerobic	respiration	_  -		Anae	robic respiration		Stroke	Cardiac output	is the volume of blood	
	Site of gas e	exchange	Maximises surface are	ea for	Res	piration with ox the mitochonc	kygen. Occurs inside dria continuously	e	Respira <sup>.</sup> Oc	tion whe curs dui	en oxygen is in short supp Ting intensive exercise	ly.	VOIDITIE	units are cm <sup>3</sup> min <sup>-1</sup>		
Aiveoli	in the lu	ungs	efficient gas exchange				· · · ·		During ł	nard exe	ercise, muscle cells are		Heart Rate	Heart rate is the	e number of beats per	
	Allows gas e	exchange	Oxygen diffuses into t	he	Gluc tran:	Jucose is oxidised by oxygen to ransfer the energy the organism needs			respiring so fast that blood cannot transport enough oxygen to meet their					minute (ppm)		
Capillaries	between int	d	diffuses out.	ade	to perform it's functions.				needs. Glucose is partially oxidised to				Stroke	Stroke volume i	s the volume of blood	
					Glu	 ∂lucose + oxygen → carbon dioxide +			muscle tissue causing them to become				Volume	contraction (cn	n <sup>3</sup> )	
muscle layer	cor	mu nnective tissue	iscle layer			W	ater		paintul	and tati	gued.		·			
(	1		4.02	$\bigcirc$						Glucc	se $\rightarrow$ lactic acid		Plasma (55%)	Pale yellow	Transports CO <sub>2</sub> , hormones	
	Iume			endothelium one cell thick		Factors affe	cting rate of diffusio	on (Bic	ology					fluid	and waste.	
Artery			Vein	Capillary	,		only)	D''	e				Red blood cel	s Carries	Large surface area, no	
Carry blood	away	Carry blo	od to the heart	Connect arteries ar	ts nd	Surface area	gradient	dis	tance				(erythrocytes (45%)	oxygen	nucleus, tull of haemoglobin.	
				veins		lu ana si sa si	Diffusion is from	T1	a.a.a. 11 - 12		← plasma 55%		White blood ce	IIs Part of the	Some produce antibodies.	
Thick muscular walls, small lumen, carry		Thin walls, large lumen, carry blood under low One c		One cel		surface area	concentration to	the c	diffusion		buffy coat <1%	*	lymphocytes a	immune system	others surround and engulf	
blood under pressure, c	r high arry	stop flov	w in the wrong	diffusion	, ,	surface	A large difference	fast	ter the		and platelets)		(<1%)			
oxygenated blood (except for the pulmonary artery).		direction, carry deoxygenated blood (except for the button carry carry blood under ve low pressu		od ry ire.	increases diffusion.	in concentration will increase rate f diffusion.	on rate o te f diffusio		1, red blood cells 45% (erythrocytes)		Platelets (<1%	) Fragments of cells	Clump together to form blood clots. <b>39</b>			

# **Chemistry Knowledge Organisers**



# Year 9 – Science – C3a. Purifying substances

	Solids	Liquids	Gases	Chromatogram	The end product in chromatography (paper with separated components).
	888888			Solute	The solid that dissolves.
		XXXXXX		Solvent	The liquid that dissolves the solute.
				Solution	Formed when a solvent dissolves a solute.
Arrangement	Particles are close	Particles touch each	Particles are not in a	Dissolve	The act in which a solution is made (forming a solution).
	together, next to each other. Particles	other and are next to each other. Particles	regular arrangement. The particles are	Saturated	When no more solute can be dissolved in a solvent.
	are in rows . Regular	are not in a regular	spaced out.	Unsaturated	When more solute can be dissolved in a solvent.
Movement Very little movement, particles vibrate in		Particles have some	Particles in gases	Atom	Smallest component of an element.
		movement. The	have lots of	Molecule	A group of atoms chemically bonded together.
	their fixed positions. They do not move	particles are able to rollover each other.	movement and move in all directions.	Compound	Two or more different atoms chemically bonded together.
	from one place to			Evaporation	Change of state where a liquid turns to a gas.
Challenge – energy	Particles have very	Particles have some	Particles have lots of	Condensation	Change of state where a gas turns to a liquid.
and attraction of particles	little energy. The particles are	energy. The particles are attracted to each	energy and there is no or very little	Filtration	Separation technique where insoluble particles are separated from soluble particles and liquid.
	other.		the particles.	Crystallisation	Separation technique where the solvent in a solution is left to evaporate, leaving the solute behind.
		(Reversible)	Chemical change (Irreversible)	Distillation	Separation technique where liquid mixtures or soluble solutions can be separated based on their boiling points.
		For example –	For example – frying	Soluble	Can dissolve in water.
Subliming	Freezing	Freezing water into	an egg - rusting	Insoluble	Cannot dissolve in water.
Depositing	Melting	ice No new substances	One or more new	Baseline	The pencil line drawn at the base of the chromatography paper during chromatography.
Be	biling	or products formed.	substances has been	Mixture	Two or more different atoms not chemically bonded together.
Gas		change of state (solid, liquid, gas)	ioinidu.	<b>Boiling point</b>	The temperature that a liquid turns into a gas. <b>41</b>





# Year 9 – Science – C3b. Atomic Structure and the Periodic Table



### Isotopes

- Versions of an element with same atomic number but different atomic mass.
- Number of protons is the same, but number of neutrons is different.
- Relative Atomic Mass is average of the masses of the isotopes, weighted by their relative abundance

• For example, Neon has three isotopes

Neon Isotope Mass	Relative Abundance (%)
20	90.5
21	0.3
22	9.2

- Relative atomic mass of Neon =  $\frac{20 \times 90.5 + 21 \times 0.3 + 22 \times 9.2}{90.5 + 0.3 + 9.2} = 20.2$
- This is why some atoms have a relative atomic mass with a decimal point.

Sub-atomic p	<b>barticles</b>
--------------	------------------

Atoms are made from smaller particles called subatomic particles. There are three type:

Particle	Relative mass	Relative charge	Found?
Proton	1	Positive, +1	In nucleus
Neutron	1	Neutral, 0	In nucleus
Electron	Neglible $\left(\frac{1}{1840}\right)$	Negative, - 1	In shells orbiting nucleus





• The existence of isotopes results in relative atomic masses not being whole numbers

# Reading the Periodic Table19<br/>F<br/>fluorine<br/>9Relative Atomic Mass<br/>(aka nucleon number):<br/>The total number of protons<br/>and neutrons added<br/>together.9Atomic number (aka proton<br/>number):<br/>The number of protons or<br/>electrons.

**Note:** on some periodic tables, they are the wrong way up, just remember that the smaller number is the proton number.

### What's in my atom?

Protons = atomic number Electrons = atomic number Neutrons = relative atomic mass subtract atomic no.



Atomic number = 9 Relative Atomic mass = 19

Protons = 9 Electrons = 9 Neutrons = 19-9 = 10



Atomic number = 16 Relative Atomic mass = 32

Protons = 16 Electrons = 16 Neutrons = 32-16 = 16





### Year 9 – Science – C3b. Atomic Structure and the Periodic Table

Mendeleev

- Arranged elements by increasing atomic mass <u>but</u>....
- He broke this rule and left some gaps if an element's properties weren't similar to the one above it.
- He thought the gaps were for elements that hadn't been discovered yet and predicted their properties.
- When they were discovered, the properties matched the predictions

#### **Electron Configuration**

- Electrons orbit the nucleus in **shells.**
- First shell holds up to two electrons
- Second shell can hold up to 8 electrons
- Third shell can also hold up to 8 electrons



						PEI	RIODSi	ncreasin	gatomic	mass, di	ffering pr	operties						
	1	2			Kev			1 H hydrogen 1				•	3	4	5	6	7	0 4 He helium 2
	7 Li Ithium 3	9 Be beryflium 4		relati ato atomic	ve atomic mic sym	mass <b>bol</b> number			Ele	ment Type	metal		11 B boron 5	12 C carbon 6	14 N nitrogen 7	16 O okygen 8	19 F fluorine 9	20 <b>Ne</b> neon 10
GROUPS	23 Na <sup>sodium</sup> 11	24 Mg magnesium 12											27 Al aluminium 13	28 Si silicon 14	31 P phosphorus 15	32 <b>S</b> sulfur 16	35.5 CI chlorine 17	40 Ar <sup>argon</sup> 18
ŝsimi	39 K potasskum 19	40 Ca calcium 20	45 Sc scandium 21	48 Ti titanium 22	51 V vanadium 23	52 Cr chromium 24	55 Mn manganese 25	56 Fe <sup>iron</sup> 26	59 Co cobait 27	59 Ni 28	63.5 Cu 29	65 <b>Zn</b> 30	70 Ga <sub>gallium</sub> 31	73 Ge <sub>germanium</sub> 32	75 As arsenic 33	79 <b>Se</b> selenium 34	80 Br bromine 35	84 Kr <sup>krypton</sup> 36
lar prope	85 Rb rubidium 37	88 Sr strontium 38	89 Y yttnum 39	91 Zr zirconium 40	93 Nb <sup>niobium</sup> 41	96 Mo molytxdenum 42	[98] Tc technetium 43	101 Ru ruthenium 44	103 Rh <sup>modium</sup> 45	106 Pd pailadium 46	108 <b>Ag</b> silver 47	112 Cd cadmium 48	115 <b>In</b> 49	119 <b>Sn</b> 50	122 Sb antimony 51	128 Te tellurium 52	127   iodine 53	131 Xe xenon 54
prties	133 Cs caesium 55	137 <b>Ba</b> 56	139 La* <sup>Ianthanum</sup> 57	178 Hf <sup>hafnium</sup> 72	181 Ta tantalum 73	184 W tungsten 74	186 Re rhenium 75	190 Os osmium 76	192 Ir <sup>iridium</sup> 77	195 Pt platinum 78	197 Au <sup>gold</sup> 79	201 Hg <sup>mercury</sup> 80	204 <b>TI</b> 81	207 Pb lead 82	209 Bi bismuth 83	[209] Po <sup>polonium</sup> 84	[210] At astatine 85	[222] <b>Rn</b> radon 86
	[223] Fr francium 87	[226] Ra <sup>radium</sup> 88	[227] Ac* actinium 89	[261] Rf nativerkondium 104	[262] Db <sup>dubnium</sup> 105	[266] Sg seaborgium 106	[264] Bh <sup>bohrium</sup> 107	[277] Hs hassium 108	[268] Mt meitherium 109	[271] Ds damstadium 110	[272] Rg roentgenium 111	Elen	nents with at	omic number	rs 112-116 ha authenticated	ave been rep d	orted but not	t fully

\* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.



This can be written as: **2.8.4**; or drawn as:



### Electron configuration and how it links to the Periodic Table:

Group number: shows the number of electrons in the outer electron shell Period number: shows the number of electron shells In the above example, we can see Silicon belongs to group 4, and is in period 3.

-44



# KS4 Chemistry - Chemical Bonding and Types of Substances (part 1)

ĿĊ	Particlos aro opp	ositoly charged ions	Occurs in compounds formed from metals	Keyword	Definition
10			combined with non metals.	lon	An atom with an electric charge, caused by the loss or gain of
alent	Particles are atoms that share pairs Of of electrons		Occurs in most non metallic elements and	Cation	A positively charged ion.
Cove			in compounds of non metals.	Anion	A negatively charged ion.
	Particles are a	toms which share		Electrostatic force	The attractive or repulsive force between two electrically charged objects.
Meto	delocalis	ed electrons	Occurs in metallic elements and alloys.	Attraction	The electric force that acts between oppositely charged bodies, tending to draw them together.
	Keyword		Definition	Intermolecular force	Forces of attraction which act between molecules.
lonic be	ond	A strong electrostation charged ions.	c force of attraction between oppositely	Atom	The smallest unit into which matter can be divided without the release of electrically charged particles.
Covale	nt bond	The bond formed wr two atoms.	ten a pair of electrons is shared between	Element	An element is a substance whose atoms all have the same number of protons.
Lattice	Structure	'sea' of negatively c An arrangement of r	harged electrons. many particles that are bonded together in a	Compound	A substance formed when two or more chemical elements are chemically bonded together.
		fixed, regular, grid-li	ke pattern	Transfer	Movement of a particle from one place to another.
Melting	point	The temperature at v state to the liquid sto	which a substance changed fro the solid ate when heated, or from the liquid state to	Share	Two bodies having equal portions distributed between the two.
Boiling	point	solid state when cod	bled. which a substance changed from a liquid to	Delocalised electron	An electron that is not associated with a particular atom within a shell, or held in a covalent bond.
Charge		a gas. Also known as elect	ric charge, is a characteristic of a unit of	Proton	A particle found in the nucleus of an atom, having a positive charge and the same mass as a neutron.
Electrical conductivity		electrons than proto	ns. ns. to pass through.	Neutron	A particle found in the nucleus of an atom having zero charge and a mass of 1.
Aqueou	eous solution A mixture that is formed when a substance is dissolved in we		ned when a substance is dissolved in water.	Electron	A tiny particle with a negative charge and very little mass.
Molten	en A substance that has been liquefied by heat.		<u>Chall</u>	Area ground a nucleus that can be accurring by clocking and	
Electror	n pair	Two electrons occup molecule, especially	bying the same orbital in an atom or y forming a nonpolar covalent bond	snell	usually drawn as circles.
		between atoms.	-	Nucleus	The central part of an atom or ion. 45

	Chemistry – Chemical Bo	onding and	Types of Substanc	es (part 2)						
Me	etallic bonding	Ionic bonding								
Giant structure of	Electrons in the outer shell of metal atoms are delocalised and free to move through the	High melting and boiling points	Large amounts of energy needed to break the bonds.	Electrons a	re	Metal atoms lose electrons and become positively charged ions	Group 1 metals form +1 ions Group 2 metals form +2			
a regular pattern	whole structure. This sharing of electrons leads to strong metallic bonds.	Do not conduct electricity when solid	lons are held in a fixed position in the lattice and cannot move.	transferred that all ator have a nob gas configuratio	so ms ble on	Non metals atoms aain electrons to	Group 6 non metals form - 2 ions			
• • • •		Do conduct electricity when molten	Lattice breaks apart and the ions are free to move.	(full outer shells).		become negatively charged ions	y Group 7 non metals form - 1 ions			
Delocalised elect	+ + + + + + Delocalised electrons Metal ions		$(1) \rightarrow [N_{a}]^{+} [(1)]^{-}$	Structur	e	<ul> <li>Lattices consi</li> <li>Held together attraction bet</li> <li>Forces act</li> </ul>	st of a regular arrangement of atoms by strong electrostatic forces of ween oppositely charged ions in all directions in the lattice			
High melting and boiling points Pure metals can be bent and	This is due to the strong metallic bonds. Atoms are arranged in layers that can slide over	diagram (2, 8, 1) (2	• • • • • • • • • • • • • • • • • • •	-ide	lf a c er usuc t	compound name ads in –ide, it ally contains only wo elements.	For example: calcium + oxygen → calcium oxide			
shaped Good conductors of electricity and heat	each other. Delocalised electrons transfer energy.	Giant structure	a+ Cl.	-ate	lf a c en usua or mo	compound name ds in -ate, it lly contains three ore elements one which is always oxygen.	For example: Calcium + carbon + oxygen → calcium carbonate			



# KS4 Chemistry – Chemical Bonding and Types of Substances (part 3)

					Covale	nt bonding						
	Simple molecular compounds Giant covalent structures											
	Low melting and Small amounts of energy					Diamond			Gr	aphene an	d fulleren	es
	boiling po	ints	intermolecular forces.	Fach	t	Very hard.	Rigid structure.			E	xcellent	Contains
P	oor conduc electrici	tors of ty	No free electrons to transfer energy.	carbon atom is		Very high melting point.	Strong covalent bonds.	hene			nductor.	electrons.
Siz and	e of atoms d molecules	ms Joined by strong covalent bonds. This means that atoms are smaller than simple molecules.		to four others	·	Does not conduct electricity.	No delocalised electrons.	Grap	Single lay graphite atom th	ver of one hick	ry strong.	Contains strong covalent bonds.
			Dot and cross :	Used for cu	utting tools due	e to being very ho	ard.	][				
	H N H + Show which atom the			Graphite								Hexagonal
suo	Can be sma	п	bonds come from - All electrons are identical	Each carbon atom is		Slippery.	Layers can slide over each other.					rings of carbon atoms with
rs of electr	molecules e.g. ammon	H-N-H 2D with bonds: H + Show which atoms are H bonded together		bonded to three others forming		Very high melting point.	Strong covalent bonds.	ullerenes		Buckm eren First full	buckminsterfull erene, $C_{60}$ irst fullerene to	shapes. Can also have rings of five
oms share pai		À	incorrectly at 90° 3D ball and stick model: + Attempts to show the H-C- H bond angle is 109.5°	layers of hexagonal rings with no		Does	Delocalised electrons					(pentagonal) or seven (heptagonal) carbon atoms.
Ati	Can be gian covalent structures e.g. polyme	rs	$\begin{array}{ccc} H \\ C \\ C \\ H \\ H \\ H \\ n \end{array} \qquad \begin{array}{c} \text{Simple polymers} \\ \text{consist of large} \\ \text{chains of} \\ \text{hydrocarbons.} \end{array}$	covalent bonds between the layers		electricity.	layers.	Di	amond, raphite, silicon	Very high melting points	Lots neec stron	of energy led to break g, covalent <b>47</b>

	KS4 Che	emistry -	- Acids	s and All	Keyword	Definition	
ACADEMY			0 44 44			H <sup>+</sup> ion	A positively charged hydrogen ion
					The pH scale and	OH <sup>-</sup> ion	A negatively charged, diatomic hydroxide ion.
					indicators	Aqueous solution	A mixture that is formed when a substance is dissolved in water.
						Acid	A solution that reacts with alkalis, turns litmus red and has a pH of less that 7.
acidic neutral alkaline					Ded in goid groon in noutral	Alkali	A solution which contains an excess of OH <sup>-</sup> ions, turns litmus blue and
			U	niversal	and blue in alkali	Pres	has a pH greater than 7.
	Acids produce h	vdrogen ion		laicaioi		base	A substance that will react with an acid to form only a salt and water.
Acids	(H <sup>+</sup> ) in aqueou	s solutions.		Litmus	and blue in alkali	Indicator	A subdance which can change colour depending on the plusing
	( ) 1		_			indicator	A substance which can change colour depending on the pH of a solution
A 11 12 -	Aqueous solutio	ons of alkalis	Met	hyl orange	Red in acid, yellow in neutral	Concentration	The amount of a solute dissolved in a certain volume of solvent.
Alkalis	contain hydroxic	de ions (OH-).				Concentrated	Containing a large amount of solute dissolved in a small volume of
				olphthalein	Colourless in acid and in	Diluto	solvent.
Para	A base is any sui	ostance that aid to form o			neorrai ana pirik in aikai	Strong gcid	A low concentration of solute in a solution.
DUSE	salt and wa	salt and water only <b>A neutralisation</b> In neutralisatior			In neutralisation reactions,	Weak acid	An acidic solute that does not dissociate completely into ions when it
	son and water only		reaction is hydrogen ions rec		hydrogen ions react with		dissolves.
Examples of	Alkalis e.g. sodiu	m hydroxide	betw	veen an acid	hydroxide ions to produce	Salt	A compound formed by neutralisation of an acid by a base.
soluble bases	potassium h <sup>.</sup>	ydroxide	ai	nd a base	$H^+ + OH^- \rightarrow H_2O$	Filtration	Using a filter to separate insoluble substances from a liquid.
		Develier		ida		Crystallisation	Separating the solute from a solution by evaporating the solvent.
	Г	Reaction	is with ac			Soluble	A substance that can be dissolved in a certain liquid.
Motals	Metal + ac	id $\rightarrow$ metals	alt +	Magnes	ium + hydrochloric acid $\rightarrow$	Insoluble	A substance that cannot be dissolved in a certain liquid.
MEIGIS	hy	/drogen		magne	sium chloride + hydrogen	Solute	Describes a substance that dissolves in a liquid to make a solution.
				_		Solvent	Describes the liquid in which a substance dissolves to make a
Metal oxides	Metal oxide +	acid $\rightarrow$ met	al salt +	Copper ox	side + sulfuric acid $\rightarrow$ copper	Solution	solution. Formed when a substance has dissolved in a liquid
		water			sulfate + water	Burette	A piece of apparetus used to accurately measure the volume of
	Metal hvdroxid	e+acid → n	netal salt	Sodium hvo	droxide + nitric acid $\rightarrow$ sodium	borene	solution that has been added during a titration.
Metal hydroxides	+	water			nitrate + water	Pipette	A piece of apparatus used in a titration to accurately measure a set
							volume of a solution.
Metal carbonates	Metal carbon	ates + acid -	→ metal	Calcium	carbonate + sulfuric acid $\rightarrow$	End-point	When just enough solution has been added from the burette to react
salt + carbon dioxid		n dioxide + \	vater		rate + carbon aloxide + water	lonic oquation	with all the solution in the flask in a titration experiment.
							spectator ions are not included.
Gas	1	est		Po	sifive result	Half equation	A chemical equation written to describe an oxidation or reduction
Hydrogen	Burnir	ng splint	'squeaky	pop' sound		Consideration	half-reaction.
Carbon dioxide Limewater (		Goes clo	oudy (as a sol	id calcium carbonate forms).	spectator ion	Inese are ions that do not change within a reaction. 40	

	KS4 Chemistry	<ul> <li>Acids and Alkalis (part 2)</li> </ul>	Producing salts from soluble reactants						
	Making pur	e, dry insoluble salts	Soluble cells	Soluk	ble salts can be made from reacting acids with solid insoluble substances				
- d	Add insoluble reactant (e.g. metal oxide) to	Add until there is an excess of insoluble			(e.g. metals, metal oxides, hydroxides and carbonates).				
Ste	acid	reactant.	Production of	Add	the solid to the acid until no more dissolves. Filter off excess solid and then				
ep 2	Filter the solution	Collect the filtrate in a conical flask and							
<u> </u>					Solubility				
Step 3	Crystallisation	Heat the filtrate using a Bunsen burner to evaporate the water from the solution.	Sodium, poto and ammo	nssium nium	All common sodium, potassium and ammonium salts are soluble e.g. sodium chloride and potassium fluoride.				
ep 4	Evaporation	Leave the evaporating basin with the heated filtrate to evaporate any remaining water	Nitrates	5	All nitrates are soluble e.g. potassium nitrate.				
<b>5</b>		and make pure, dry insoluble salts.	Sulfates	5	Common chlorides (e.g. sodium chloride) are soluble, expect those of silver and lead.				
react with ec	ach other to form salt and	water.	Carbonates	and	Common carbonates and hydroxides are insoluble except those				
	Use the pipette to ac	dd 25 cm <sup>3</sup> of alkali to a conical flask and add a	hydroxides		of sodium, potassium and ammonium.				
Ster		tew drops of indicator.	Strong and weak acids (HT ONLY)						
Step 2	Fill the burette with a the acid from the bu	acid and note the starting volume. Slowly add rette to the alkali in the conical flask, swirling to mix.	Concent	rated	High mass of substance in a given volume of solution				
Step 3	Stop adding the acid when the end-point is reached (the appropriate colour change in the indicator happens). Note the final volume reading. Repeat steps 1 to 3 until you get consistent		Dilute	è	Low mass of substance in a given volume of solution				
reddings.		1 Strong a	cids	Completely ionised in aqueous solutions e.g. hydrochloric, nitric and					
State Symbol Meanin		Meaning			sulturic acids.				
	S	Solid	Weak a	cids	Only partially ionised in aqueous solutions e.g. ethanoic acid, citric				
	1	Liquid			acid.				
	g	Gas	Hydroge	n ion	As the pH decreases by one unit (becoming a stronger acid), <b>49</b>				
ad		Aqueous solution		ation	the hydrogen ion concentration increases by a factor of 10.				

	KS4 Ch	nemistry –	Calcula	tions involvi	ina masses	6. Keyword	7. Definition
ACADEMY		,,				atom	The smallest neutral part of an element that can take place in chemical reactions.
1. How to find an Example: 10.0a of	empirical f	ormula:	2. How to from the	o deduce the mole empirical formula	ecular formula a and relative	Avogadro constant*	The number of particles in one mole of a substance (6.02 $\times$ 10 <sup>23</sup> atoms, molecules, formulae or ions).
of chlorine. Find the product that is form	he empirical med.	formula of the	Example:	: The empirical formu	ula for glucose is	closed system	Substances cannot enter or leave such as a precipitation reaction in a stoppered flask.
1. Symbol	Са	CI	CH₂O an Deduce	d its relative formula the molecular formu	ı mass is 180. ıla for glucose.	concentration	The amount of solute dissolved in a stated volume of a solution. Units include g/dm3.
2. Mass (g) 3. A.	10.0         17.8         1. f           40.0         35.5         mc		1. Find the mass by a	1. Find the empirical formulaC + H + H + Omass by adding up the12 + 1 + 1 + 16		conservation of mass	During a chemical reaction, the overall mass of substances does not change so the total mass of reactants is equal to the total mass of products.
4. Divide mass by	10.0 = 40	17.8 = 0.50	relative a of the atc	itomic masses of all oms	= 30	empirical formula	The simplest whole number ratio of atoms or ions of each element in a substance.
A, 5. Divide answers	0.25 0.25 = 1	35.5 <u>0.50</u> = 2	2. Divide mass by t	the relative formula the empirical formula	$\frac{180}{30} = 6$	excess reactant	There is more of this reactant present than is needed so it is not completely used up in a reaction.
by smallest number 6. Empirical formula	0.25	0.25	3. Multipl	y the numbers in the	• CH <sub>2</sub> O x 6	limiting reactant	There is less of this reactant present than is needed so it is completely used up in a reaction. The mass of product formed is controlled by this reactant.
3. How to calcula	ate the mas	ses of reactants	empirica molecula	i formula to get the ir formula	· 50 C <sub>6</sub> H <sub>12</sub> O <sub>6</sub>	mole*	One mole of particles of a substance is defined as: a) the Avogadro constant number of particles (6.02 x 10 <sup>23</sup> atoms, molecules, formulae or ions) of that substance b) a mass of 'relative particle mass' g. The SI unit symbol is mol.
products:					4.	molecular formula	This represents the actual number of atoms of each element in one molecule.
Example: Calculat	e the mass o chloride	f chlorine neede	d to make	r	mass	molecule	A particle consisting of two or more atoms joined together by bonds.
1. Write the balanced	2AI + 3Cl <sub>2</sub> -> 2	2AICI <sub>3</sub>			(g)	open system	Substances can enter or leave such as a reaction in an open flask that takes in or gives out a gas.
equation	• M.C. =	2 x 35.5 = 71		(g/mol)	X moles (mol)	precipitate	An insoluble substance that is formed when two soluble substances react together in solution.
substances in the	• $M_r \operatorname{AlCl}_3 =$	27 + (3 x 35.5) = 133	3.5		$\wedge$	precipitation	A reaction in which a precipitate is formed.
3. Calculate the	•	3Cla makes 2Al			5.	product	A substance formed in a reaction.
ratio of masses	• (3	$(2 \times 71)$ Cl <sub>2</sub> makes (2 x	133.5)AICI <sub>3</sub>	m	ass of	reactant	A substance used up in a reaction.
4 Divide to work	•	213g CL makes 267				reaction	A process in which reactants are converted to different substances called products.
out the mass for 1g of product	• 0.7	<b>267 267</b> 798g Cl <sub>2</sub> makes 1g <i>i</i>	alCl <sub>3</sub>	(g/dm <sup>3</sup> )	tion volume X (dm <sup>3</sup> )	relative atomic mass	$(A_r)$ The mean mass of an atom relative to the mass of an atom of C-12 which is assigned a mass of 12. Unit is g/mol.
5. Multiply to scale up	• (0.798g x ;	<b>53.4</b> ) Cl <sub>2</sub> makes (1g 42.6g Cl <sub>2</sub> makes 53.	x <b>53.4</b> ) AlCl <sub>3</sub> 4g AlCl <sub>3</sub>	<b>Converting un</b> cm <sup>3</sup> to dm <sup>3</sup> div	its: vide by 1000	relative formula mass	$(M_{\rm r})$ The sum of the relative atomic masses of all the atoms or ions in its formula. Unit is g/mol.
				dm <sup>3</sup> to cm <sup>3</sup> mu	ultiply by 1000	stoichiometry*	The ratio of moles of each substance in a reaction. 50
						volume	The amount of space hat a liquid takes up. Units include cm <sup>3</sup> and dm <sup>3</sup> .

	(S4 Chemistry – Electrolytic	c Proc	esses						
Key Word Electrolysis	Definition Decomposition/break down of a compound using electrical energy.	Pro ele	ocess of ctrolysis	Splitting up using electricity	When a water, th to con Passing a	an ionic compound is melted or dissolved in ne ions are free to move. These are then able iduct electricity and are called electrolytes. an electric current though electrolytes causes the ions to move to the electrodes.	5	C.	<u>Oxidation Is Loss</u> , <u>Reduction Is Gain</u> Metals can be extracted from molten compounds using electrolysis.
lons	Charged particle.	Ele	ectrode	Anode Cathode	The The r	e positive electrode is called the anode. negative electrode is called the cathode.		metals u	too reactive to be extracted by reduction with carbon.
Anions	losing electrons. Usually metal form cations.	Where	do the ions go?	Cations Anions	Catio Anior	ons are positive ions and they move to the negative cathode. ns are negative ions and they move to the positive anode		Extracting elec	The process is expensive due to large amounts of energy needed to produce the electrical current. Example: aluminium is extracted in this
	gaining electrons. Usually non-metal form anions.		etrolucia of	Electro	olytic pro	positive dilode.		<u>H</u>	way. ligher tier: You can display what is appening at each electrode using
Electrodes	A rod made of metal or carbon which carries the current in the electrolyte.	Ele	aqueous solutions	KS4 E	EDEXCEL and SC1	Lead ions Pb * Molten lead (II) bromide		h A A	half-equations: At the cathode: $Pb^{2+} + 2e^- \rightarrow Pb$ At the anode: $2Br^- \rightarrow Br_2 + 2e^-$
Cathode	An electrode that is negatively charged.	Att	the negative electrode	Metal will be	e produced	d on the electrode if it is less reactive than hydrogen.			potassium most reactive K sodium Na calcium Ca
Electrolyte	charged.	At	the positive electrode	Oxygen is formed at positive electrode. If you have a halide ion (Cl <sup>-</sup> , l <sup>-</sup> , Br) then you will get chlorine, bromine or iodine formed at that electrode.					magnesium Mg aluminium Al carbon C zinc Zn
Reduction	When a positive ion gains electrons.		The ions discl	narged when an depend on the re	n aqueous elative rec	s solution is electrolysed using inert electrodes activity of the elements involved.			iron Fe tin Sn lead Pb hydrogen H
Oxidation	When a negative ion loses electrons.		Using copper	Copper is a ve electrical co	ery good nductor	Much of the copper available isn't pure enough for this use so it is purified using electrolysis.			opper Cu silver Ag gold Au platinum least reactive Pt
Discharged	When ions convert to elements due to transfer of electrons during electrolysis.		Copper sulfate solution	The anode is impure copy the cathode of pure co	made of per and is made opper	Both electrodes are placed in copper sulfate solution. Copper ions (Cu <sup>2+</sup> ) leave the anode and are attracted to the cathode.			negative ion in solution element given off at positive electrode chloride, Cl <sup>-</sup> chlorine, Cl <sub>2</sub>
equations	electrons in oxidation and reduction.		Electrodes	The cathode copper bui	of pure Ids up	The anode decreases in size. The impurities left behind form a sludge.			iodide, I <sup>-</sup> iodide, I <sup>-</sup> sulfate, SO <sup>2-</sup> oxygen, O <sub>2</sub>



ORMISTON SWB ACADEMY		KS4 Chemistry Dyna	mic Equilik	orium and Fertilisers	Keyword	Definition
		Reversible Read	tions and Equilibria		Le Chatelier's	States that when a system experiences a disturbance (change in condition), it will
Reversible reactions	In some	e chemical reactions, the products can	nical reactions, the products can Changing If the concentration of a reactant is in more products will be formed		Principles	respond to restore a new equilibrium state.
Representing			concentration	If the concentration of a product is decreased, more reactants will react.	Changing	If the concentration of a reactant is increased, more products will be formed.
reversible reactions	A +	B C + D	Changing	If the temperature of a system at equilibrium is increased:	concentration	If the concentration of a product is decreased, more reactants will react.
When although the s	The dire chang	ection of reversible reactions can be ed by changing conditions:	temperature	<ul> <li>Exothermic reaction = products decrease</li> <li>Endothermic reaction = products increase</li> </ul>	Changing	If the temperature of a system at equilibrium
Ine direction	A + E	neaf 3 C + D cool	Changing	For a gaseous system at equilibrium: - Pressure increase = equilibrium position shifts to side of equation with smaller number of	temperature	<ul> <li>Exothermic reaction = products decrease</li> <li>Endothermic reaction = products increase</li> </ul>
Roactants		Graph sketch shows in a reversible reaction, the backward reaction gets faster with time, and the	pressure (gaseous reactions)	<ul> <li>molecules.</li> <li>Pressure decrease = equilibrium position shifts to side of equation with larger number of molecules.</li> </ul>	Changing pressure (gaseous reactions)	<ul> <li>For a gaseous system at equilibrium:</li> <li>Pressure increase = equilibrium position shifts to side of equation with smaller number of molecules.</li> <li>Pressure decrease = equilibrium position</li> </ul>
Products	quilibrium	forward reaction gets lower with time. When they are occurring at the same rate, dynamic equilibrium		methane steam		shifts to side of equation with larger number of molecules.
Time		has been reached.	nd hydrogen from	methane + steam → hydrogen + carbon monoxide	Equilibrium in reversible reactions	When a reversible reaction occurs in apparatus which prevents the escape of reactants and products, equilibrium is reached when the forward and reverse reactions occur exactly at the same rate.
The Haber pro	ocess	natural gas to form ammonia. The reacti and uses optimum conditions and a cat reach dynamic equilibrium.	on is reversible alyst in order to	hydrogen air	Equilibrium in	When a reversible reaction occurs in apparatus which prevents the escape of reactants and products equilibrium is
Optimum tempe	erature	The optimum temperature for the Haber	process is 450°C.	hydrogen + oxygen → water This reaction removes oxygen from the air to leave nitrogen	reactions	reached when the forward and reverse reactions occur exactly at the same rate.
Optimum pres	ssure	atmospheres.				This process uses nitrogen from the air and
Optimum conc	ditions	The optimum temperature for the Haber process is 450 and optimum pressure is 200 atmospheres. These are economically viable conditions as they produce the k yield to cost ratio.		nitrogen hydrogen	The Haber Process	hydrogen from natural gas to form ammonia. The reaction is reversible and uses optimum conditions and a catalyst in order to reach dynamic equilibrium.
The use of a co	atalyst	The Haber process uses an iron catalyst. the position of the equilibrium but it does of the reaction.	This does not alter increase the rate	$\begin{array}{c c} 200 \text{ atmospheres} & \text{Indiggen + Hydrogen \rightarrow animolia} \\ \hline \\ \text{iron catalyst} & \text{N}_2(g) + 3\text{H}_2(g) \rightarrow 2\text{NH}_3(g) \\ \hline \\ \text{The Haber process for making ammonia} \end{array}$	NPK fertilisers	Formulations of various salts containing appropriate percentages of the elements.





# KS4 – Chemistry – Chemical and Fuel Cells

# **Chemical cells**



Keyword

Definition

 $2H_2(g) \rightarrow 4H+(ag) + 4e$ - $O_2(g) + 4H^+(aq) + 4e^- > 2H_2O(I)$ 

ł	SWB ACADEMY			KS4 Chen	nistry	/ Gro	oups	in the	Perio	odic Tabl	e		Keyword	Definition		
	-	Alkali metals			No	ble gases				Group 0						
<u>1</u> Н	2	Transition	metals	Halogens 3 4 5	6 7	0 He	gases	Unreactive	, do not f	orm molecules	Thi ou <sup>-</sup>	is is due to having full ter shells of electrons.		The atoms get larger as you go down, so the single electron in the outermost shell (highest energy level) is attracted less		
Li Na f	Li Be B C N O F Ne Na Mg Al Si P S Cl Ar					Ne Ar	Noble	Boiling po	ints increa group	ase down the	Incre	easing atomic number.	Reactivity of	electrostatic attraction with the nucleus gets weaker because the distance		
K Rb Cs Fr	KCaScTiVCrMnFeCoNiCuZnGaGeAsSeBrKRbSrYZrNbMoTcRuRhPdAgCdInSnSbTeIXCsBaLaHfTaWReOsIrPtAuHgTlPbBiPoAtRFrRaAcRfDbSgBhHsMt????				Kr Xe Rn	eon Helium	Used in bo	illoons	Due to being les bo Glows when	s denso alloons electri	e than air, which means s will float. icity flows through it.	group 1	between the outer electron and the nucleus increases. Also the outer electron experiences a shielding effect from the inner electrons, reducing the attraction between the oppositely			
Ele arra order	Elements arranged in order of atomic Elements with similar properties are in columns elements in the same			Elements in the same same number of outer s elements in the same p	group hav hell electro period (rov	ve the ons and v) have	Argon N	Used in fild light bu	ament Jbs			lament reacting with with unreactive argon		charged outer electron and the nucleus. When Group 7 elements react, the		
n	number called groups the same number				electron sl	nells.		_			111516	edd.		outermost shell. Going down the group,		
		Grou	- 1						Grou	up 7				the outermost shell's electrons get further		
		Grou	рт			Con	sist of m	olecules mad	le of a	Have seven el	ectrons	s in their outer shell. Form	Reactivity of	away from the attractive force of the		
		Reaction		Word equation		Ward a guatien		L	рс	iir of atoms			-1	l ions.	group 7	gain an extra electron. The outer shell will
	weiai	with water	vv	ora equation	ens l	down the group (gas $\rightarrow$ liquid $\rightarrow$			Increasi	ng ato	omic mass number.		also be shielded by more inner shells of			
Li	thium	Fizzing	Lithium hvdro	+ water → lithium xide + hydrogen	Haloge			solid)		Increasing pro	Increasing proton number means an electron			electrons, again reducing the electrostatic attraction of the nucleus for		
		Fizzina moro		,		Rea	ctivity c	lecreases dov	wn the	is less easily g	gained	d as outer shell is further		an incoming electron.		
Sc	odium	vigorously	Sodium + water $\rightarrow$ sodium				group av			force is weaker.				Elements in Group 0 of the periodic table		
		than lithium	nyara	xide · Hydrogen					Motal	+ halogon -> ma	stal	e.g. NaCl		are called the noble gases. They are		
Pot	Potassium Fizzes au burns wit lilac flar		Potassium + water → potassium hydroxide + hydrogen		wi <del>i</del> l	With metals		Forms a metal halide e.g. Sod		halide odium + chlorine odium chloride	+ halogen → metal halide polium + chlorine → polium chloride halogen gains an outer shell electrons			stable arrangements of electrons. The atoms have eight electrons in their outermost shell, apart from helium which		
	Soft ar	nd easily	Low mel	ting and boiling points.		With		Forms a	Hydro	ogen + halogen · vdroaen halide	<b>→</b>	Dissolve in water to form	Reactivity of	has just two but still has a complete outer shell.		
<u>stals</u>	Very re	eactive		one electron in	hy	drogen		nydrogen halide	e.g. Hy hyd	drogen + bromin drogen bromide	e→	acidic solutions.	group ()	The stable electronic structure explains why they exist as single atoms: they have		
kali mé	wate chl	er and the orine	neirouters	hell. Form +1 ions.		With	A m ho	A more reactive halogen will Chlorine		orine + potassium	1	(HT) These are redox reactions. The halogen		no tendency to react to form molecules. The boiling points of the noble gases get		
AI	Reactivity increases down the group		Negative outer electron is further away from the positive nucleus so is more easily lost.		solu ha	tion of a lide salt	eousaisplaceon of aless reactle salthalogen fthe sal		bromide → potassi chloride + bromir		sium ine compound loses electrons.			higher going down the group. For example, helium boils at -269 °C and <b>56</b> radon boils at -62°C.		

ORMISTON SWB ACADEMY	KS4	Chemi	stry	Rates of React	ior	n an	d Energy Change	s in Reactions	Keyword	Definition
	R	ates of Re	eacti	on	7		Energy Cho	inges		
Rate of chemical reaction	Rate of chemical reactionThis can be calculated by measuring the quantity of reactant used or product formed in a given time.Rate = gRate of chemical reactionRate = g		e = <u>quantity of reactant used</u> time taken e = <u>quantity of product formed</u> time taken		Idothermic	Activation energy Products	Products are at a higher energy level than the reactants. As the reactants form products, energy is transferred from the surroundings to the reaction mixture. The temperature of	Collision theory	Chemical reactions can only occur when reacting particles collide with each other with sufficient energy.	
Tempe	Factors c erature	The high	erate of the root	of reaction temperature, the quicker		Ъ	Reactants Time	the surroundings decreases because energy is taken in during the reaction.		
Concentration		The highe	r the c the ro	oncentration, the quicker ate of reaction.		0	Activation energy	Products are at a lower energy level than the	Activation energy	This is the minimum amount of energy colliding particles in a reaction need in order to react.
Surface area		The larger the surface area of a reactant solid, the quicker the rate of reaction.			thermic	Reactants	reactants form products, energy is transferred to the			
Pressure (of gases) When ga		When gas upon t	ses react, the higher the pressure them, the quicker the rate of reaction.			Exo	Time	surroundings. The temperature of the surroundings increases because energy is released		Occur in the following:
lume/cm <sup>3</sup> Slop	the of tangent $= \frac{25 \text{ cm}^3}{60 \text{ s}}$ $\approx 0.42 \text{ cm}^3 \text{ s}^{-1}$	Quantity Unit			during the reaction.				- Salts dissolving in water	
	25 cm <sup>3</sup>	Mass		Grams (g)			Calculate the overall energy	gy change for the forward	chanaes	<ul> <li>Neutralisation reactions</li> <li>Displacement reactions</li> </ul>
0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	رد. ر	Volum	e	Cm <sup>3</sup>			react	ion │		- Precipitation reactions
20 40 60 80	(b) 100 120 140 Time/s	Rate of reaction		Grams per cm <sup>3</sup> (g/cm <sup>3</sup> ) HT: moles per second (mol/s)		lation	Bond energies (in kJ/mol):	H-H 436, H-N 391, N≡N		
Catalyst	Catalyst       A catalyst changes the rate of a chemical reaction but is not used in the reaction.         Enzymes       These are biological optical reaction to the reaction tot the reactin to the reaction to the reactinet to the reaction to			ergy calcu	Bond breaking: 945 + (3 2253 k	x 436) = 945 + 1308 = J/mol	Exothermic reactions	Heat energy is given out as bonds are being formed.		
Enzymes				id en	Bond making: 6 x 3	91 = 2346 kJ/mol				
How do they work?		vide a tion fre not require		REACTANTS		Bon	Overall energy chan -93kJ,	ge = 2253 - 2346 = /mol	Endothermic reactions	Heat energy is taken in as bonds are being broken.
	as much energy when they co	gy to react llide.		TIME			Therefore reaction is	exothermic overall.		57

# **Physics Knowledge Organisers**

	Motion		Speed	Scalar measurement that shows how fast an object is moving. Measure in m/s
<u>Scalar</u> <u>Vector</u>	Calculating speed/velocity			(meters per second).
Distance Displacement		$\wedge$	Velocity	Vector measurement that shows how
Speed Velocity	$\int \mathbf{Speed} \ (m/s) = \mathbf{disfance} \ (m) \div \mathbf{fime} \ (s)$	$D = S \times T$		fast an object is moving in a specific direction. Measured in m/s (meters per
Speed velocity	How to remember the equation?	$\searrow$ S = D ÷ T		second).
Power Momentum	"Don't Step on Turtles"		Distance	Measurement of how far an object is
Mass Acceleration				moving/has moved. Measured in m
Volume Weight				(meters).
Temperature			Time	Measurement of time. Measured in s
Eorce	Distance-time graph     Key features:		Acceleration	(seconds). When an objects speed increases over
			Acceleration	time.
Pressure		You can calculate speed	Conversion	Changing a measurement to another
Calculating a gradient	fast, steady getting	trom this distance-time araph.		form.
24	w speed. faster	9.00.0	Deceleration	When an objects speed decreases over
Gradient = $\frac{Change in y}{Change in x}$		Steeper gradient=	Soglar	time.
Change in X	Speed stationary	idsierspeed.		only.
	returning		Vector	A measurement that shows magnitude
Calculating acceleration	to start			and direction.
			Plateau	A straight horizontal line on a graph.
Acceleration is the <u>rate of change</u> of	TIME		Gradient	Difference between two values, shown
Velocity	Velocity-time graph		Constant	by a incline or decline on a line graph.
Acceleration $(m/s/s) = \frac{Change in velocity}{m/s}$	Key features:	ou can calculate acceleration	Constant	Shown by a straight line on a line graph
time taken (s)	Constant	from this velocity-time	Maanitude	Another term used for size.
$a = \frac{v-u}{t}$	speed/velocity	graph.	Direction	The course which an object is moving.
a= acceleration	Acceleration Deceleration	louidting the grag beneath the		We show North, West, East, South or a
u= initial velocity	Welocity m/s	lines, is the same as the		combination of two.
t= time		overall distance travelled	Initial	The beginning.
		Stopper product	Final Displacement	Ine end.
a t		faster acceleration		shortest distance to the final 50
	0 10 20 30 40 50 Time/s			place an object ends up.



# KS4 Physics – Forces and Motion

so jus resista weigh make



Newton's First Law of motion can be written as:

- a moving object will continue to move at the same speed and direction unless an external force acts on it
- a stationary object will remain at rest unless an external force acts on it.

The acceleration of an object is a measure of how much its velocity changes in a certain time. Sir Isaac Newton's Second Law of Motion describes the factors that affect the acceleration of an object.

The acceleration in the direction of a resultant force depends on:

- the size of the force (for the same mass, the bigger the force the bigger the acceleration)
- the mass of the object (for the same force, the more massive the object the smaller the acceleration).

Newton's Third Law is about the forces on two different objects when they interact with each other. This interaction can happen:

- when objects touch, such as when you sit on a chair
- at a distance, such as the gravitational attraction between the Earth and the Moon.

On Earth the **gravitational field strength** has a value of about 10 newtons per kilogram (N/kg). This means that each kilogram is pulled down with a force of 10 N. The gravitational field strength is different on other planets and moons.

The weight of an object can be calculated using the following equation:

weight = mass × gravitational field strength (N) (kg) (N/kg)

This is often written as:  $W = m \times g$ 

		лі —	acc
5 seconds a	after jumping, 3 seconds after jumping	<ol> <li>12 seconds after jumping,</li> <li>speed = 55 m/s</li> </ol>	
speeu -	- 5 m/s speeu - 25 m/s	speed - 55 m/s	bala
			force
			resu
			scale
		sultant = 0	spee
istance incr	reases with speed, Her air resistance is larger but	out her She is moving so fast that the air	unbo
after jumpi ince is much The large	ing the air weight stays the same. The is smaller than her resultant force is smaller, so resultant force still accelerating, but not as n	resistance balances her weight. she is She continues to fall at the same much speed.	force
her accele	and doollar to to a rate downwards.	пист. органа.	vect
	Momentum is calculate	ed using this equation:	<u> </u>
tion	momentum – mass	x velocity	velo
	(kg m/s) (kg)	(m/s)	cent
on it		(, 5)	force
JII IL.	This can also be written	n as $p = m \times v$ ,	mas
nges	where <i>p</i> stands for more	mentum.	weig
<u>j</u>	Momentum and acceleration		
	Table C shows two equations involving acce mass × change in velocity	eleration. These can be combined to give: m(v - u)	grav
	$rorce = \frac{1}{time}$	$\frac{1}{t}$	field
	As mass × velocity is the momentum of	f an object, this equation can also be	inert
ger	written as:		
	force = change in momentum time or	<u>mv – mu</u> t	actio
ect	H Momentum and collision	26	reac
	When moving objects collide the tota	al momentum of both objects is the	bala
they	same before the collision as it is after no external forces acting. This is know Remember, momentum is a vector so	r the collision, as long as there are wn as <b>conservation of momentum</b> . b you need to consider direction when	force
	you add the quantities together. If two directions, we give the momentum of	o objects are moving in opposite f one object a positive sign and the	equi
and	other a negative sign.		cons
i anu			of m
			mon
s per	before collision	1	
ce of			kine

after collision

oloured balls will be the same as the

Keyword	Definition
acceleration	A measure of how quickly the velocity of something is changing. It can be positive if the object is speeding
	up or negative if it is slowing down.
balanced	When the forces in opposite directions on an object are the same size so that there is a zero resultant force.
forces	
resultant force	The total force that results from two or more forces acting upon a single object. It is found by adding
	together the forces, taking into account their directions.
scalar quantity	A quantity that has a magnitude (size) but not a direction. Examples include mass, distance, energy and
speed	speed. How fast something is moving. Often measured in metres per second (m/s), miles per hour (mph) or
speed	kilometres per hour (km/h)
unbalanced	When the forces in opposite directions on an object do not cancel out, to there is a non-zero resultant
forces	force.
vector avantity	A auantity that has both a size and a direction. Examples include force, velocity, displacement.
. sere: deaminy	momentum and acceleration.
velocity	The speed of an object in a particular direction. Usually measured in metres per second (m/s).
centripetal	A force that causes objects to follow a circular path. The force acts towards the centre of the circle
force	
mass	A measure of the amount of material there is in an object. The units are kilograms (kg).
weight	The force pulling an object downwards. It depends upon the mass of the object and the gravitational field
	strength. The units are newtons (N).
gravitational	A measure of how strong the force of gravity is somewhere. It is the force on a 1 kilogram mass, so the units
field strength	are newtons per kilogram (N/kg).
inertial mass	The mass of an object found from the ratio of force divided by acceleration. The value is the same as the
	mass calculated from the weight of an object and gravitational field strength.
action-	Pairs of forces on interacting objects. Action-reaction forces are always the same size, in opposite
reaction forces	directions, and acting on different objects. They are not the same as balanced forces.
balanced	Forces acting on the same object. Balanced forces are always equal, in opposite directions, and always
forces	act on the same object. They do not have to be the same type of force An object acted on by balanced
	forces will not change the way it is moving
equilibrium	When a situation is not changing because all the things affecting it balance out.
conservation	The total momentum of moving objects before a collision is the same as the total momentum afterwards,
of momentum	as long as no external forces are acting.
momentum	The mass of an object multiplied by its velocity. Momentum is a vector quantity, with units kilogram metres
	per second (kg m/s).
kinetic energy	A name used to describe energy when it is stored in moving things. The amount of energy stored depends
	on the mass of the object and on its speed (or velocity) squared.
work done	The energy transferred when a force acts through a distance to move an object or change its speed. It is
	calculated using the size of the force and the distance moved in the direction of the force. The unit for
	work done is the joule (J).
acceleration	A measure of how quickly the velocity of something is changing. It can be positive if the object
	is speeding up or negative if it is slowing down.



actrical appliances. Turn off devices that are on standby									
ashing clothes	Air dry clothes, wash o	on a lower temperatu	ire						
at lost from home	Install insulation – dou	ble glazing, loff/floor	insulation						
		Renewable Energy	Quickly replenishes its energy used. Infinite	Wind power, solar power, hydroelectric power, tidal power, geothermal power, biomass					
cy = Useful nergy input		Non-renewable Energy	Is finite (will run out). Does not quickly replace energy used	Fossil fuels – coal, oil and natural gas Nuclear power					
Keyword	Definition		•	•					
Chemical	Energy si transferre	tore that is emptied of to the surrounding	during chemical reactions v 1s.	when energy is					
Conduction	The trans	sfer of heat by passin	g on energy (or electrical o	charge) to nearby					
Convection	The proc	cess by which heat tro	avels through fluids (gases	and liquids).					
Elastic potential	An energ	gy store that is filled v	e that is filled when a material is stretched or compressed.						
Electrical	Energy s	tore resulting from the	e movement of electrical of	charge (electrons).					
Energy	This is the	e ability to make som	ething happen when it is tr	ransterred.					
Gravitational pote	Linit of a	iore inat is tilled whe	an object is raised.						
Kinetic		neigy, represented b	moving object speeds up						
Light	A form o	fradiation that can	ransfer energy in a wave	•					
Non-renewable	An energ	av resource that will h	be used up, and not repler	ished in our lifetime.					
Nuclear	An energ	gy store associated v	vith nuclear interactions.						
Radiation	Radiatio waves. T spectrun vącuum	Radiation is the transfer of internal energy in the form of electromagnetic waves. This radiation lies in the infrared region of the electromagnetic spectrum. It does not require particles to move, it can travel through a							
Renewable	An energ	y resource that can	be readily replenished in a	our lifetime.					
Sound	A form o	f energy transferred	by sound waves.						
Thermal	An energ	gy store that is filled v	vhen an object is heated.	<b>61</b>					
Transformation	Energy tr	ransformation is the p	Internal         An energy store that is filled when an object is heated.         61           Transformation         Energy transformation is the process of changing one form of energy to changing one f						

Keyword

heating

Hot water

Definition

Put more jumpers on and turn off central heating

Take showers, only boil the amount of water you need

	KS4 Physics – Waves		Frequency	The number of waves	Higher frequency =	Increasing frequency	Longitudinal Wave		
Wave	Vibrations that transfer energy from	place to place.		pass a point in a second.	more energy transferred	= higher pitch	Wavelength Amplitude	creat waxlength	
Transverse	A wave where the vibrations are at a direction in which the wave is travel	right angles to the ling.	Amplitude	de Maximum Increasing Inc disturbance amplitude am from its increases – in	Increasing amplitude	Compression Rarefaction Compression	S deputition		
Longitudinal	A wave where the vibrations are pa direction in which the wave is travel	rallel to the ling.	undisturbed energy position. transferred		in volume	Cisplacement of air molecules	toogh		
Frequency	The number of vibrations (or the nun second, measured in hertz.	nber of waves) per	Wavelength	The distance between a	Increasing wavelength		Longitudinal Wave	vs. Transverse Wave	
Period	The time taken for one complete work It is measured in seconds.	ave to pass a point.		point on one wave and	= decrease energy		Longitudinal Particles oscillate (vibrate) in the direction	Transverse Particles oscillate (vibrate) at right angles	
Wavelength	The distance between a point on or same point on the next wave.	ne wave and the		point on the	transterrea		of the wave's movement	to the direction of the wave movement	
Amplitude	The size of vibrations or the maximum particle moves away from it resting provide waves passes.	n distance a position when a	Transverse Wave Wavelength			<b>]</b>	Sound waves, ultrasound	light), water waves Diagram	
Refraction	The change in direction when a way medium to another.	ve goes from one	Amplitude		le	Reflection light	n = Light waves reflect from surfaces. When		
Normal	An imaginary line drawn at right ang of a mirror or lens where a ray of ligh	gles to the surface It hits it.	undisturbed position (equilibrium) Wavelength			bouncing a surface	off waves reflect, they obey the law of reflection: the angle of	angle of incidence angle of reflection	
						-	incidence equals the	plane mimor	
wave formula	Wave	speed = waveleng	gth x frequer	ю		Refraction	n = Waves change speed when they pass		
Example	Wave speed is measured in meters per second (m/s)Wavelength is measured meters (m)		sured in	sured in Frequency is measured in Hertz (Hz)			across the boundary between two	Incident Ray Normal	
Dylan is standir 1.3m. He count Wave speed =	ig on the end of a pier. He measures s 2 waves every second. Find the wave frequency x wavelength	ng past him. Th	ne wavelength c	f each wave is		different densities, such as air and glass. This causes them to change	i Angle of incident Refracted Ray		
Wave speed = $2.6 \text{ m/s}$	= 2 x 1.3						direction and this effect is called refraction.	62	



KS4 Physics (C and EM Spect	CP5/SP5) – Light	hortest wavelength ighest frequency	longest wavelength lowest frequency	Separates o	nly.	the middle than a	t the edges. It	makes	
Separates only. Specular reflection		10 <sup>-12</sup> m 10 <sup>-9</sup> m 10 <sup>-6</sup> m 10 <sup>-3</sup> m 1 m 10 <sup>3</sup> m			parallel rays of light converge (come together) at the <b>focal point</b> . The <b>focal length</b> is the distance between the focal point and the centre of the lens. A <b>diverging lens</b> is thinner in the middle than at the edges. The focal point is the point from which the rays seem to be coming after passing through the lens.				
Long wavelength, low frequency	Uses		Dangers	converging lens	focal length		focal le	ngth diverging lens	
Visible light	Light bulbs, our eyes detect it	From a laser can dar	nage the retina in the eye	Separates	only.				
Infrared	Communication – TV remote, grills, toasters	Felt as heat, and car	a cause skin to burn	Heat transfe	r by radiatio iation. Unlike	n - Heat can k e conduction	be transferre	əd by <b>ction</b> -	
Microwaves	Communications, mobile phone microwave for food	es, Can cause internal h	Can cause internal heating of body tissue			n or movemer pe of electron is absorbed b	nt of <b>particle</b> nagnetic ra v an obiect	<b>es</b> - Idiation. Lit is	
Radio waves	Radio broadcast, communications	Very large doses can	cause cancer	heated and	its tempera	ture rises.			
Short wavelength, high frequency	Uses		Dangers	Separates •Dark matt s	•Dark matt surfaces are better at absorbing heat energy			eneray	
Ultraviolet	aviolet Used to kill microorganisms in water, detecting forge bank notes		can lead to skin cancer	than light sh •Dark matt s	iny surfaces. urfaces are l	better at radic	ating heat e	ergy	
X-rays Hospitals – to check for broken		High frequency, trans	sfer a lot of energy and can			Separates or	nly.		
	bones	penetrate the body.	Excessive exposure may	Surface	Abs	orption	Emi	ission	
Gamma rays	Cancer treatment, sterilising	High frequency, trans	High frequency, transfer a lot of energy and can			rber of heat	Good emit radiation	ter of heat	
hospital equipment		penetrate the body. cause DNA mutation	Excessive exposure may , possibly leading to cancer.	Shiny	Poor absorber of heat radiation		Poor emitter of <b>6</b> heat radiation		

	KS4 Physics (CP	6/SP6) – Radioad	ctivity
Dalton's model	Plum Pudding - Thomson	Rutherford	Bohr/Chadswick
000			e protection e de terminales de la constante e de terminales de terminales de la constante e de terminales de la constante e de terminales de termin
John Dalton thought that all matter was made of tiny particles called atoms, which he imagined as tiny spheres that could not be divided.	Thomson carried out experiments and discovered the electron. This led him to suggest the plum pudding model of the atom. In this model, the atom is a ball of positive charge with negative electrons embedded in it.	Rutherford suggested a new model for the atom, called the nuclear model. In the nuclear model: the mass of an atom is concentrated at its centre, the nucleus the nucleus is positively charged	Bohr did calculations that led him to suggest that electrons orbit the nucleus in shells. The shells are at certain distances from the nucleus. Chadwick found evidence that the nucleus contains no charged particles called the neutron.
Atomic Mass - The number of protons å neutrons in the nucleus. Atomic Number - The number of just protons in the nucleus. Number of elect	Carbon r of = number trons	e elec e elec e neu	Atomic structure – protons and neutrons found in the nucleus. Electrons orbit the nucleus on electron shells.
rolativa	barga relativa maas	Atoms of a single Three	ee Isotopes of Hydrogen
proton +1	1	have different	
neutron 0 neu	tral 1	numbers of neutrons, but	
electron -1	1 / 1840	same number of Protium	<sup>2</sup> H <sup>3</sup> H n Deuterium Tritium

Atomic number (also called proton number)	Number of protons in an atom		
Background radiation	Naturally radioactive substances in the environment that produce radiation.		
Becquerels (Bq)	Unit to measure radioactivity. One Bq is one nuclear decay each second.		
Count rate	Number of clicks on a GM tube when radiation is detected. It is the amount of radiation per second or minute.		
Elements	Substances that contain the same type of atoms		
Geiger-Muller (GM) tube	An instrument to measure radioactivity.		
Half-life	Time taken for half the unstable nuclei in a sample of a radioactive isotope to decay.		
Isotopes	Atoms of a single element that have different numbers of neutrons, but same number of protons.		
Kinetic theory	Model that helps explains the properties of solids, liquids and gases.		
Mass number (also called nucleon number)	Total number of protons and neutrons.		
Nucleons	Smaller particles that make up the nucleus.		
Neutrons	Sub-atomic particle found in the nucleus, with no charge.		
Particle theory	Model that helps explains the properties of solids, liquids and gases.		
Protons	Positively charged sub-atomic particle found i the nucleus.		
Subatomic particles	Particles smaller than atom, and make up an atom. Protons, neutrons and electrons.		

# KS4 Physics (CP6/SP6) – Radioactivity



electrons is called an ion. Radiation that causes electrons to escape is called

ionising radiation.

1							
	Mass	Relative 4		Relative 0 (1/1840)		None	
	Range in air	3-5cm		15cm		Long range	
	Penetration ability	Low, stopped by paper		Increased, stopped by aluminium or lead		Great slo concret	wed by e, lead
	lonising ability	Highly ionisinig			Fairly	Least ionising	
	Effects of a magnetic field	Deflected     Attracted to       Attracted to     A       negative     posi       electrode     posi		Deflected		Unaffe	ected
	Effects of an electric field			Att positiv	racted to /e electrode	Unaffe	ected
ſ	Particle	Symbol		Dangers of radioactivity			
ľ	Alpha	α	4 2	He can damage inside a cell.		age the DNA ell. This	
ľ	Beta	B⁻	0	<sub>-1</sub> e	damage is called mutation. Gene mutation		utation
Ī	Positron	β+	0,	n matrices		assed on to the eration. Some	
Ī	neutron					s can cau	se
	[Separates] Using - Killing microor - Radioactive d	g radioactiv ganisms letecting	vity				[Sep - Ro

Checking thickness of paper

Smoke alarms – contains a source of alpha particles

Type of nuclear radiation

Alpha

Nucleus of a

helium atom

+2

Property

Charge

Beta

electron

-1

Gamma

EM waves

None

The penetration power of the three types of adiation.



[Separates] Radioactivity in medicine

- Radioactive materials are used to diagnose medical conditions without having to cut into a patient's body.
- Tracers that emit positrons used to detect medical problems.
- Treating cancer external radioactivity which uses 66 a beams of gamma rays, x-rays or protons directed at the tumour.



#### Year 10 Art - GCSE DEVELOPMENT BOARD AO1: Developing ideas, artist research. AO2: Using resources, testing out ideas and media. AO3: Recording ideas, photos and drawings.

### What needs to be included in a GCSE Development Board?

- Drawings /photographs to develop your own ideas tonal and colour. (Linking to your artist's work.)
- X3 small experiments using artist's style/techniques.
- Artist response.
- Developed response..

### How do I develop my ideas inspired by my artist's work?

- Decide what objects might link to your artist's work. Take photographs (Primary)find images on the internet of them (Secondary).
- Draw them in pencil, pen, pencil crayon and paint.

### How do I create experimental pieces?

Take some of your drawings and try out your artist's materials and techniques on them to develop them further.

### What is an artist's response and developed response?

- An <u>artist's response</u> is <u>your own work</u> developed further using the style/techniques/materials/ideas of your artist.
- A developed response makes you look again at what other materials and techniques your artist uses to further develop your own ideas.

A Good development Board should include drawings using a range of materials and techniques and show a clear journey towards your final piece.

	Wider Thinking:
GCSE Art	and Design - BBC Bitesize
www.stud	entartguide.com

### Expert modelling example...



**Development Board.** 



Response/Developed response

### Stretch and Challenge:

Use materials and techniques with a high level of skill and control. Record finer surface textures and details.

Keyword	Definition		
Tone	Refers to light and dark values.		
Texture	Refers to the surface quality in a work of art. (How things look and feel.)		
Colour	Colour has the strongest effect on our emotions. It is the element we use to create the mood or atmosphere of an artwork.		
Record	To capture visual elements like proportion and form. For recording your ideas. It is important to observe your sources closely.		
Artist Response	Your own work developed using your artist's work as inspiration.		
Contextual	Connections made to the work of other artists from different and similar times, places and cultures.		
Composition	Ways of arranging, organising and laying out elements in a piece of art and design.		
Develop	Bring out potential.		
Explore	Try out the qualities of materials, techniques or processes through practical investigation.		
Imaginative	Develop ideas and concepts in new, engaging and inventive ways.		
Refine	Improve initial work taking into account feedback and aims. 67		



DEVELOP

**EXPERIMENT** 

AO3

# **Y10 TEXTILES** KNOWLEDGE ORGANISER ORGANIC STRUCTURES



	How do Lidentify the formal elements of my major project: Cakes Biscuits & Sweets?	Keyword	Definition
	Observation	The action or process of closely observing or monitoring something or someone.	
ARTISTS and other sources, showing analytical and cultural UNDERSTANDING	<ul> <li>within your chosen genre.</li> <li>Define Form/shape/pattern/experiment.</li> <li>Apply numerous techniques during development.</li> </ul>	Silk Printing	A design is cut out of paper or another thin, strong material and then printed by rubbing,
REFINE ideas through EXPERIMENTING and	How do I create a response to chosen Artists work?		through the cut out areas.
SELECTING appropriate resources, media, techniques and processes	<ul> <li>Use the ladas behind an arrists work to inspire your own designs.</li> <li>Watch a demonstration by your teacher.</li> <li>Use decorative/dyeing/printing/experimental</li> </ul>	Fabric Manipulation	Experimenting with the <b>fabric</b> to change its appearance, drape or shape.
The second ideas, observation and insights RELEVANT to your INTENTIONS	<ul> <li>Textiles techniques with skill and control.</li> <li>Create a response to your chosen artists work using influence from their work.</li> </ul>	Influence	Something or someone that influences a person or thing, then, has an influence on that person or thing.
Present a <b>PERSONAL</b> response, showing analytical understanding and realising <b>INTENTIONS</b> for your project, making connections in your work	What needs to be included to ensure a successful final piece?		
189 the work of	<ul> <li>Commit to design throughout project.</li> <li>Use shape, scale and proportion accurately.</li> <li>Make your work as detailed as possible using the Textiles techniques explored.</li> </ul>	Moodboard	An arrangement of images, materials, pieces of text, etc. intended to evoke or project a particular style or concept.
great way to . Starting with a showing their the ideas and tist.	<ul> <li>Take inspiration from your chosen artist and show clear development in response to their work.</li> <li>Create a mock-up of a final product.</li> <li>Create final product signifying the conclusion to the journey you have created throughout your sketchbook.</li> </ul>	Batik	A method (originally used in Java) of producing coloured designs on textiles by dyeing them, having first applied wax to the parts to be left undyed.
		Applique	Layering pieces of fabric that are sewn or stuck on to a larger piece to form a picture or pattern.
		Toile	An early version of a finished garment made up in cheap material so that the design can be tested and perfecte

PRESENT

RECORD

# Artist Response

Responding to the work of other artists is a great way to generate ideas. ... Starting with a direct response showing their understanding of the ideas and aesthetic of an artist.





### Year 10 Business LEARNING OUTCOME 4: Attracting and retaining customers



fashions.

R064 Learning Aims 4.1+4.2: Pricing a product, strategies and appropriateness of each **Pricing strategies** Factors to consider when pricing a product to attract and retain customers Type of strategy Advantages Disadvantages Competitive pricing: this is Will attract new Businesses must think of Target market Cost of production Price of when a business sets the customers to the business other ways to attract competitor 6 price of the product which is and retain existing customers, not just price. products similar to their direct • Profit could be low as the customers Organiser competitors Competitors sales could price may only cover the • A business neds to When decidina decrease as they may production costs of the make sure that the on a price, a not be able to compete product costs involved when business will with the low prices making the product Research of need to are covered when understand: similar products pricing the product Psychological pricing: Could attract new • If customers are • The income currently on the for sale. Otherwise, aimed at customers who like customers which would purchasing online, the levels of market will they will make a loss. a bargain by setting a price increase revenue and price may appear to be targeted enable a for a product that appeals profit for the business. aood value, but when business to sell Example, if it costs customers to customers because of the Items could be sold for a shipping costs are added, 49p to make a their products at Whether or value for money that it little less than their actual the purchase could be product and you not the a lower price appears to show value abandoned. charge 99p, a profit customers and attract Attracts customers as the Most people are aware of of 50p will be made Knowledge Example: £299 instead of can afford customers away price appears to be a this strategy so may not each time the £300 (₽) good deal be convinced that it is a the product from product is sold. good deal. competitors. Example: Remember the Competitors can produce Price skimming: A business • High profits can be family car price needs to can charge a higher price achieved by a business a similar products for a be high enough versus sports due to a lack of when there is no cheaper price to make a profit cars competitors. Aimed at competition in the same Customers may be put off customers who like to have market by a higher price, as not the lasted product • High prices give an value for money. Key Term • Some customers may not impression of a good Example: Apple Iphone quality product. be able to afford the new product. HIGH PF Pricing Strategies - various methods that a Price penetration: often Achieves high sales for Profit margins will be business uses to attract customers to a **†∕∂** ★ the duration of the offer. reduced during the offer used when a product particular product by changing the regular launches, where a low price Attracts customers to the period. price to one that is often cheaper to gain an product because of the Not all products will is set first to attract new N edge against their competitors customers and they suit this pricing method, price such as short-term gradually increased. Increase market share








ORMISTON SWB ACADEMY	RO64 – Enterprise and Ma	arketing Concepts	
<u>)</u>	<ul> <li>LO5: Understanding factors for up a business</li> </ul>	or consideration when starting	OWNERSHIP FOR BUSINESS     START-UPS
Q	Sole trade	Partnership	Franchise
	An individual person owns a business. unlimited liability	Two or more people own a business. unlimited liability	A franchise is when one business, the franchisor.
lise	They can employ other people. Some examples include – hairdressers, plumbers or electricians.	Jointly responsible for the business. Some examples are solicitors, small restaurants, dentists, vets	Gives permission to an entrepreneur, the franchisee, to set up a business using its brand name and its products.
QU			
je Org	<ul> <li>Advantages</li> <li>Shared responsibility</li> <li>Shared workload</li> <li>Partner – brings different skills and attributes</li> <li>More capital – investment from different partners</li> </ul>	Advantages • Easy to set up with low cost • Decide what happens to the profit • Choose their own working hours and holidays • Limited legal requirements in	ranchisee: ays a start-up cost – this covers equipment and ometimes a premises. ays a royalty each year – this is a percentage of the evenue earned. Aust follow the rules laid down by the franchisor.
		relation to accounting	r <b>anchisor:</b> rovides equipment and resources rovides training
	Disadvantages	Disadvantages Adv	antages <b>a</b>
Mom	<ul> <li>Growth slow – due to amount of money available</li> <li>Limited amount of work one person can do</li> </ul>	<ul> <li>Cannot make a decision without all partners agreeing         <ul> <li>time consuming</li> <li>Problems agreeing can lead</li> </ul> </li> </ul>	asier to raise finance as the brand is well known upport and training is offered asier to gain customers as brand is already established
	<ul> <li>Long working hours (hard to take holidays</li> <li>Unlimited liability – must pay for all debts</li> </ul>	to disagreements and conflict • Share profits • Partners reputation affects business	dvantages igid rules to be followed tart-up costs and operating fees are expensive

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• LO5: U up a b	Understanding fa	ctors for consideratio	on when starting	SOURCE OF CAPTIAL FOR BUSINESS START-UPS
Ē	When starting and set business, it is vital tha able to secure the cc or money) required businesse	ting up a new t the owner is upital (finance to fund the s.	The appropriatene number of factors: • Purpose – what • Time period – ho • Amount – how r • Type of business affect the type	ss of each source of finance will depend on a does the business require the money for? wo long does the business need the money for much money is required? s – the size and ownership of the business will of finance that is available.
Sour	ce of Finance		Descrip	otion
Bank Loan		An amount of money i period of time	s borrowed from the bar	nk, then repaid (with interest) over a set
Crowd fund	ding	Groups of investors tha	t join together to offer fu	unding to a business
Small Busin	ess Grant	Money given to the bu projects – especially th	isiness by the governme ose that create new job	nt or charity Used to help finance new os
Business An		Finance invested in sm wealthy entrepreneurs	all, risky business e.g. ne	w business start-ups by experienced and

ORMISTON SWB ACADEMY	RO64 – Enterprise	and Marketing Concepts	
9	<ul> <li>LO5: Understanding up a business</li> </ul>	factors for consideration when starting	SOURCE OF CAPTIAL FOR     BUSINESS START-UPS
<b>D</b>	Source of Finance	Advantages	Disadvantages
	Borrow money from friends / family	<ul> <li>May charge little or no interest</li> <li>Do not have to provide security</li> </ul>	Could lead to disputes between family members and friends • May not be able to lend large amounts
າມໂຮແ	Own savings	Does not have to be paid back	There may be a limit to the amount of money which can be invested
Orgo	Crowd funding	<ul> <li>It can be a fast way to raise finance with no upfront fees</li> <li>Can be a valuable form of marketing and result in media attention</li> </ul>	If the target is not reached, all the money will have to be paid back • Someone can steal your idea if it is not protected
ත්ලීම	Small business grants	<ul> <li>Does not have to be paid back</li> <li>Do not need to lose control of your</li> <li>Business</li> </ul>	there may be strict rules and not every business is eligible
Jowle	Bank Loans	<ul> <li>The repayments can be spread out</li> <li>over a long period of time</li> <li>Access to larger amounts of</li> <li>finance</li> </ul>	Must be paid back with interest • Lender may require security for the loan
Kr	Business angels	<ul> <li>BA have lots of experience which</li> <li>helps with decision making</li> <li>Can bring in large amounts of</li> <li>finance</li> <li>No repayments or interest</li> </ul>	the owners lose a certain percentage of the business





R064

0

Organiser

Knowledge

#### **LEARNING OUTCOME 6:**

Learning Aim 6.1 The purpose of each of the functional activities that may be needed in a new business and the main activities of each functional area

#### **Functional Areas**

Marketing Identifies the needs and wants of their customers. Develop products that customers would want to purchase

Finance Manage all the finances for the business and record and report on financial performance.



Operations Organise and cost how the products will be processed, made and delivered to their customers.









#### Finance

- Responsible forall things finance related. • Paying employees
- Making and receiving payments
- Recording financial transactions
- Preparing annual financial accounts including statement of financial position
- and income statementCashflow forecasts and financial

performance

#### Human Resources (HR)

HR recruit and select employees what will work for the business. They manage the performance of the employees and ensure that health and safety and employment legislation is complied with. Focus: Training, Health, Employment (legislation) Safety, Recruitment, Performance

#### Operations Responsible for managing the production processes (logistics). They plan this process making sure that they have:

- All the necessary raw materials
  - Working machinery
  - Staff to operate equipment
- Quality control (ensuring the finished products meets the standards that are expected)

#### Key Terms

Functional activities-The range of tasks that each functional area will complete within their area of specialism.

#### Functional areas

The different sections of a business which are divided into different areas of expertise.

4P's-stands for Product, Price The different sections of a business which are divided into different areas of expertise.

#### Logistics

How a business manages the production of their product from manufacture to point of sale

#### Quality control

Ensuring the finished product meets the standards that are expected

In a small business, come of the tasks may be completed by one or two people. In a larger business, they will have many people working within each **functional area** due to the size of the business.





# Year 10 – Computer Science – Programming Techniques

## Constant

Value STORED IN A **MEMORY LOCATION** that **never changes WITHIN A PROGRAM** 

# Variable

Value STORED IN **MEMORY LOCATION** that can change WITHIN IN A PROGRAM

**Sequence:** Completing steps in the order which they must happen

**Selection:** Where a choice is made in a program depending on a condition or outcome

**Iteration:** Act of repeating or lopping specific sections of code

Count controlled Iteration: Repeats a set number of times Condition controlled: Repeats until a condition is met or something in the program changes

## Careers

- Software development
- Programing
- Software Engineering

### Syntax Error

An error in the rules/grammar of the language Eg missing colon / spelling mistake Logic Error

The program is written to do something other than what the programmer intended Eg Resetting only the first 9 elements in an array instead of all 10.

### **Run Time Error:**

More difficult to spot as it can run a program without reporting an error. E.g. runs but Doesn't give an output. Or the program hangs or Becomes inactive

# Data Types

### Real /Float

Number with decimal Point

### Integer

Number without a decimal Point

## String

A series of characters/TEXT

### Character

A single letter or symbol

### Date/Time

Date and Time in any format

Boolean Yes no, true false value

# **Other Info**

## **Concatenate** To join different data types together

# # Comments

Use these to add comments in to your code to explain what you have done

Validation: An computer check to ensure that the data entered is sensible and reasonable. It does not check the accuracy of data.

# **Wear 10 – Computer Science – Programming Techniques**

Comparison Ope	rators	Aritmetic Opera	rators								
==	Equal to	+	Addition	n eg x=€	6+5 gives 11	Ca	reers				
!=	Not equal to	-	Subtracti	ion eg >	=6-5 gives 1	•	Software deve	lopme	nt		
<	Less than	•	Multiplicati	ion eg x	=12*2 gives 24	•	Programing				
<=	Less than or equal to	/	Division	n eg x=1	12/2 gives 6	•	Software Engin	neering	5		
>	Greater than	MOD	Modulus	eg 12N	NOD5 gives 2						
>=	Greater than or equal to	DIV	Quotien	nt eg 17	DIV5 gives 3						
		٨	Exponentia	iation e	g 3^4 gives 81						
ТҮРЕ	INFO	SYI	ΝΤΑΧ		• Two-dimen	sional (2	2D) arrays ai	re ind	exed	by tw	/0
TYPE LIST	INFO MUTABLE	SYI	NTAX []		<ul> <li>Two-dimen subscripts,</li> </ul>	sional (2 one for	2D) arrays ar the row and	re ind one f	exed or the	by tw colu	/o Imn.
TYPE LIST	INFO MUTABLE DIFFERENT DATA TYPES	SYI E.G. [1,"H	NTAX [ ] HELLO", 3.4]		<ul> <li>Two-dimen subscripts,</li> <li>Example:</li> </ul>	sional ( one for	2D) arrays ar the row and	re ind one f	exed or the	by tw colu	vo imn.
TYPE LIST	INFO MUTABLE DIFFERENT DATA TYPES	SYI E.G. [1,"H	NTAX [ ] HELLO", 3.4]		<ul> <li>Two-dimen subscripts,</li> <li>Example:</li> </ul>	sional ( one for ra	2D) arrays an the row and ting	re ind one f	exed or the	by tw e colu	/o Imn.
TYPE LIST TUPLE	INFO MUTABLE DIFFERENT DATA TYPES IMMUTABLE	SYI E.G. [1,"H	NTAX [ ] HELLO", 3.4] ( )		<ul> <li>Two-dimen subscripts,</li> <li>Example:</li> </ul>	sional (2 one for ra	2D) arrays an the row and ting	re inde one fe <i>movie</i>	exed or the	by tw colu	/O Imn. dex)
TYPE LIST TUPLE	INFO MUTABLE DIFFERENT DATA TYPES IMMUTABLE DIFFERENT DATA TYPES	E.G. [1,"H	NTAX [ ] HELLO", 3.4] ( ) "Hello", 4.3)		<ul> <li>Two-dimen subscripts,</li> <li>Example:</li> </ul>	sional ( one for ra <i>col</i>	2D) arrays an the row and ting	re ind one f <i>movie</i> 0	exed or the (seco 1	by tw colu	/O Imn. dex) 3
TYPE LIST TUPLE	INFO MUTABLE DIFFERENT DATA TYPES IMMUTABLE DIFFERENT DATA TYPES	SYI E.G. [1,"H E.G. (1,2, 1	NTAX [ ] HELLO", 3.4] ( ) "Hello", 4.3)		<ul> <li>Two-dimensubscripts,</li> <li>Example:</li> <li>row</li> <li>rating[0][2</li> </ul>	sional (2 one for ra <i>col</i> ] = 2	2D) arrays an the row and ting <i>reviewer</i> 0	re ind one fo <i>movie</i> 0	exed or the (secc 1 6	by tw colu and inc 2 2	/0 imn. <i>dex)</i> 3
TYPE LIST TUPLE ARRAY	INFO MUTABLE DIFFERENT DATA TYPES IMMUTABLE DIFFERENT DATA TYPES IMMUTABLE	SYI E.G. [1,"H E.G. (1,2, 1	NTAX [ ] HELLO", 3.4] ( ) "Hello", 4.3)		<ul> <li>Two-dimensubscripts,</li> <li>Example:</li> <li>row</li> <li>rating[0][2</li> <li>rating[1][3</li> </ul>	sional (2 one for ra <i>col</i> ] = 2 ] = 8	2D) arrays an the row and ting <i>reviewer 0</i> (first 1	re inde one fe <i>movie</i> 0 4 7	exed or the (secc 1 6 9	by tw colu ond inc 2 2 4	/0 imn. /ex) 3 5 8

# **Year 10 – Computer Science – Data Representation**

### Careers

- Software development
- Programing
- Software Engineering

Binary to denary

 $1 = On \quad 0 = Off$ 

Data Representation

128	64	32	16	8	4	2	1
0	0	1	1	0	1	0	0

32 + 16 + 4 = **52** 

## Binary 00110100 = 48 Denary

Denary to Binary

24 =	128	64	32	16	8	4	2	1
24 -	0	0	0	1	1	0	0	0

# **Year 10 – Computer Science – Data Representation**



# Year 10 – Computer Science – 1.1 System Architecture

## **VON NEUMANN ARCHITECTURE**

describes a system where the CPU runs programs stored in memory. Programs consist of instructions and data which are stored in memory addresses as binary digits

In short this is the internal, logical structure and the organisation of the computer hardware



#### **Possible Careers**

- Computer hardware engineer
- Computer developer
- System Engineer

#### - summary

- Fetches instructions (from memory)
- Fetches data (from memory)
- Decodes instructions

CPU

Executes instructions

Fetch- Decode – Execute cycle – How the CPU processes instructions: FETCH:

The processor checks the program counter to see which instruction to run next.

The program counter gives an **address value** in the memory of where the next instruction is.

The processor fetches the instruction value from this memory location.

## **DECODE:**

Decoding the instructions in the the <u>ALU</u>, storing the result of this in the CIR.

### EXECUTE:

The instruction is performed. Once this is complete, the processor goes back to the program counter to find the next instruction.

This cycle is repeated until the program ends.

### **PROGRAM COUNTER (PC)**

STORES THE LOCATION OF THE NEXT INSTRUCTION IN A PROGRAM WAITING TO BE FETCHED

MEMORY ADDRESS REGISTER (MAR) STORES THE LOCATION FOR DATA TO BE FETCHED FROM OR SENT TO MEMORY

#### MEMORY DATA REGISTER (MDR)

STORES THE DATA THAT HAS BEEN FETCHED FROM OR IS WAITING TO BE SENT TO MEMORY

ACCUMULATOR

STORES THE RESULT OF THE CALCULATION PERFORMED BY THE ALU

### CURRENT INSTRUCTION REGISTER

STORES THE INSTRUCTION READY TO BE DECODED BY THE ALU

### **ARITHMETIC LOGIC UNIT (ALU)**

part of a (CPU) that carries out arithmetic and logic operations in computer instruction

### **CONTROL UNIT (CU)**

WORKS WITH THE CPU TO **CONTROL THE FLOW OF DATA WITHIN THE SYSTEM** AND TO DECODE INSTRUCTIONS

### CACHE

SMALL TEMPORARY VOLATILE MEMORY, STORES FREQUENTLY USED INSTRUCTIONS. QUICKER FOR CPU TO ACCESS THAN MAIN MEMORY

### MAIN MEMORY (RAM)

This the volatile memory that stores data and programs **currently in use.** 

# Year 10 – Computer Science – 1.1 System Architecture



# Year 10 – Computer Science – 1.1 System Architecture

#### What is the purpose of the CPU?

It Processes Data by fetching, decoding and executing instructions.

HINT: when you answer this type of exam question – you need to EXPLAIN WHY - putting more cores = better performance isn't enough!

### FACTORS AFFECTING PERFORMANCE OF CPU:

# **Clock Speed (measured in Hertz)**

- Represents the number of fetch execute cycles / instructions the CPU can process in a given time
- The higher the clock speed the faster the CPU will run WHY? Because it will be doing more Fetch-Decode and Execute cycles per second which means more instructions are being processed.

## **Cache Size**

 The holding area for data from the RAM – stores frequently used instructions. More cache then the better the performance. WHY? The more cache the CPU has the less time is spent accessing memory (RAM) this means it can retrieve instructions quicker and programs can run faster.

Level 1 Cache – Quicker to access, doesn't store as much, Level 2 = slightly slower to access, holds slightly more than L1, Level 3 = Even slower to access than L2, but can hold

#### even more

### **Number of Cores**

- Number of Independent processors within the CPU.
- Multiple Instructions able to be processed simultaneously in the same cycle
- The more cores the quicker the performance WHY? Quad Core = 4 cores. Can perform 4 instructions at same time in same cycle, 8 cores can perform 8 instructions simultaneously so more cycles/instructions are being processed per second

## **EMBEDDED SYSTEMS:**

They are dedicated systems that are designed for a fixed purpose. They are a system within a larger system e.g. Washing machines, car park barriers, microwaves, car engines, MP3 etc

#### **GENERAL PURPOSE SYSTEM:**

A machine that is capable of carrying out some general data processing under program control. Your PC /laptop is an example of this but also it could be something far more basic too.

# ✓ Year 10 – Computer Science – 1.7 System Software

## **Operating Systems**

#### Manages hardware and software in a computer system

#### **Memory Management**

Controls where the programs go in memory when being run.

#### **User Interface**

Provides a method of interaction with the user.

#### MultiTasking

Allows more than one program to run at once by sharing CPU time between programs.

#### Peripheral Management & driver software

Manages all Input, Output and Storage devices. Allows the OS and the external hardware such as printers, USB's etc to talk to each other

#### Security

Protects the machine is free from harmful viruses or unwanted access.

#### File and Disk Management

Helps to store files (images, music, documents etc) and their file extensions, helps you organise and search for files

Disk management such as space on hard drives, and utility software such as disk defragmentation software.

#### User management

The OS can deal with User accounts – single or multi users – eg. More than one person can us a computer at once. It also allows for use access, e.g. logging in to a system and having access to certain files and permissions etc. OS is also responsible for things like biometric anti-theft measures, fingerprint, and retina scanners.

#### System Cleanup (Utility)

(Searches for and) deletes files/programs which are no longer used

- ... eg deletes temporary files / installation files
- ... deletes settings / registry values which are no longer used

### System Information Utility

displays important data about the current state of the computer · e.g. temperature, free memory, network speed, % processor used

#### Possible Careers

- Software Developer
- Data Analyst
- System Analysist
- Teaching

Systems software – Designed to run and maintain a computer system Examples

Linux

No. redhat

Windows

Sun Cobal

solaris

# **UTILITY SOFTWARE**

Helps to maintain or configure a computer. Most are installed in the OS but you can add others.

**Disk Defragmenter** 

Moves (parts of) of files around so that all parts of a file are stored together (allowing files to be accessed more quickly)

Free space is collected together (allowing large files to be saved easily) Disk Defragmenter

D(C:)         NTFS         74.45 GB         4.65 GB         6 %           Elements (G:)         Defragmenting         FAT32         466 GB         161 GB         34 %           Estimated disk usage before defragmentation:         Estimated disk usage before defragmentation:	· ·
Elements (G:) Defragmenting FAT32 466 GB 161 GB 34 %	-
Estimated disk usage before defragmentation:	
	1

# 

# **UTILITY SOFTWARE**

Helps to maintain or configure a computer. Most are installed in the OS but you can add others

### Back-ups

Back up data – copy of a sysytem files and settings stored somewhere externally

Full back up – copy of EVERY file is taken. Faster to restore from Uses A LOT of storage space, can take a long time to create

Incremental Back up – only files created and edited since the last back up are copied. Uses LESS storage and much QUICKER to create. BUT full system restore is SLOW

### **Data Compression**

Reduces the size of files so they take up less disk space. Can help upload and download files quicker or send them across email. Standard formats include .zip and .rar to compress the files

#### Encryption Scrambles (encrypts) data – this stops people from accessing it. Encryption happens by scrambling the message, you can only decrypt this if you have a special 'key' Encrypted text is called – Cipher text Decrypted is called Plain text Encryption is essential for sending data over a network e.g. internet so it is kept secure





# Year 10 – Cambridge National IT - Mindmap







You must be able to understand the purpose and use the content of different pre-production documents

# Storyboards

- ⇒ A storyboard is used by many people to illustrate a sequence of *moving* images.
- ⇒ A storyboard shows the flow of scenes that occur in a timeline, a succession of events.
- $\Rightarrow~$  This is different to a visualisation diagram which are used for a single of events.
- $\Rightarrow$  Each scene of the story is placed in chronological order (in the order







Moment of clarity. "Ahal" Ding or chimes; lightbulb moment.



Submitting via Coursework. Fade out as if ending.



Pan

L



Working in a dark dorm room. Sounds of clock ticking and pencil scratching on paper.



Back to the classroom. Keep as similar as possible to original. "Elaborate on your storyboards!"



Proudly shows off finished storyboard. Wipes sweat off brow. Victory music. Zoom in on storyboard



Fade out.

# Why use storyboards?

#### Best way to share your vision for the project

- ⇒ A visual aid makes it much easier for you to share and explain your vision for your video with others.
- ⇒ When you have a storyboard, you can show people exactly how your video is going to be mapped out and what it will look like. This makes it much easier for other people to understand your idea.

#### Makes production much easier

- ⇒ When you storyboard a video you're setting up a plan for production, including all the shots you'll need, the order that they'll be laid out, and how the visuals will interact with the script.
- ⇒ The storyboard is a starting point or suggested storyline around which you can plan your story (all the angles you will shoot of a scene). This really comes in handy when you're making your video, as it ensures you won't forget any scenes and helps you piece together the video according to your vision.

#### Saves you time

- ⇒ While it may take you a little while to put your storyboard together, in the long run it will save you time in revisions later.
- ⇒ Not only will it help you explain your vision to your team, it will also make the creation process go more smoothly.



#### You must be able to understand the purpose and use the content of different pre-production documents

# Storyboard and camera angles

- $\Rightarrow$  The storyboard could be used by several people who could be involved in the production process.
- $\Rightarrow$  Camera shots a angles are important aspects to a storyboard
- $\Rightarrow$  The camera operator or animator will use the storyboard to decide how to create each scene.
- $\Rightarrow$  Each scene is usually defined by changes to the camera use for each shot.



#### Purpose of a storyboard

- $\Rightarrow$  To provide a visual representation of how a media project will look along a timeline
- $\Rightarrow$  To provide a graphical representation of wat a sequence of movements will look like
- $\Rightarrow$  To provide guidance on what scenes to film or create
- $\Rightarrow$  To provide guidance on how to edit the scenes into a story

#### Uses of a storyboard

- ⇒ Any project where movement or a sequence is required, especially along a timeline, for example
  - ◊ Video projects
  - ◊ Digital animations
  - Ocomic books to illustrate the story
  - ◊ Computer games, to illustrate game flow, narrative or story
  - 0 Multimedia projects, to illustrate the sequence between scenes

#### Content of a storyboard

- $\Rightarrow$  Images, for content of each scene
- ⇒ Locations
- ⇒ Camera shot type and angles
- ⇒ Cameral movement
- ⇒ Shot length and timings

# Year 10 – Cambridge National IT - Visualisation Diagrams

isplaced

plete First Series

The Co



You must be able to understand the purpose and use the content of different pre-production documents

down and 17224 Complete first as ally boadca BBC ON hatirean 12/0/200 50 2/02/2010

DVD

These episodes were originally

roadcast on BBC One between 12/01/2010 sill 12/02/2010

# Visualisation Diagrams

Visualisation diagrams are a rough drawing or sketch of what the final static image product is intended to look like. They will have annotations to describe the design ideas. Typically, a visualisation diagram is hand drawn, but it does not need any artistic skills to communicate ideas.

It is intended to demonstrate the layout and content of the product that is being illustrated

You might produce several drafts to demonstrate ideas to your client. Your client might choose the draft they like the most. There must be sufficient information in the visualisation diagram for the client to make a decision about their preferred design.

Visualisation diagrams are valid for static designs, that is an image that does not move. It is, therefore, relevant for designs such as a magazine cover, a DVD cover, or an image for a website. It would not be suitable for a video or an animation.

Look closely at the detail in the example visualisation diagram. Compare the concepts in the visualisation diagram and compare them to the final product that was produced. Do you notice the similarities and the differences.

Notice how the visualisation
diagram was not modified as
ideas developed in the pro-

Purpose:	
----------	--

- Plan the layout of a static or still image in a visual manner
- ⇒ To show how a finished item might look like
- Uses :
  - $\Rightarrow$  CD/DVD cover design
  - ⇒ Poster, such as for a film, event, leaflet or advertisement
  - ⇒ A single game scene of display of a single scene

$\Rightarrow$	Comic	book	page	layout

- $\Rightarrow$  Web page layout
- ⇒ Magazine front cover

Content:

- ⇒ Multiple images, layout and positions of items.
- ⇒ Colours and colour schemes
- $\Rightarrow$  Position and styles of text
- ⇒ Fonts, font colours and size



# Year 10 – Cambridge National IT - Flowcharts

# **Key Vocabulary**

Algorithms	A set of rules or instructions to be followed.
Flowcharts	A graphical way of showing an algorithm.
Selection	Deciding what code to run based on a decision or answer to a question. E.g an IF statement.
Sequence	A set of instructions that are completed in the exact order that they are written.
Iteration	Where a set of instructions is repeated. E.g a while loop, for loop and repeat until loop.
Input	Data that is given to the computer or program to then use.
Output	Information that is provided by the computer or program.
Procedure	A group of instructions grouped together that can be used by the main program.
Variable	A name given to a value in a program that can change when the program is running.

# **Data Types** Real /Float Number with decimal Point Integer Number without a decimal Point String A series of characters/TEXT **Character** A single letter or symbol Date/Time Date and Time in any format **Boolean** Yes no, true false value **Flowchart Symbols** Start/Stop Used at the start and end of a flowchart. Input/Output Controls all the inputs and outputs. General instructions and calculations car-Process ried out by the computer. Where a question/decision is asked. Must Decision have a 'Yes' and 'No' output. Used to connect flowchart symbols to show

the direction of flow in the program. 92



# Year 10 – Cambridge National IT - Wireframe

# What is a wireframe?

A sketch or a guide before you build an actual form/document

A 'skeleton graph' that shows how the contents will be laid out.

Frames with no colours, images or any graphics.



# Sime Year 10 – Cambridge National IT – Introduction to Augmented Reality

Augmented reality is a new technology where companies have developed software on a higher scale than ever for mobile phones and proprietary devices. This is in order to allow users to see the real world overlaying additional information onto the scene. For example, augmented reality allows users to see the real world with trees blowing in the breeze, dogs chasing after a ball, or children playing football, dogs playing with a digital version of themselves or children playing with an alien goalkeeper when scoring a goal.



With the latest developments in augmented reality, the technology is used regularly in shopping apps such as Argos and IKEA, as well as with product apps such trainers including Nike. This software allows the customer to see items of furniture in their rooms before making a purchase.

The most well-known use of augmented reality in the 2010s is *Pokémon Go*, where users searched for *Pokémon* in the real world by throwing *Pokéballs* at *Pokémon* in an augmented reality world. Between 2016 and 2020, the app has been downloaded 1.1 billion times making \$4.34 billion in revenue during the same period.

#### Examples of augmented reality:

- Navigation systems use augmented reality to superimpose routes over a live view of the road.
- During football games, broadcasters use augmented reality to draw lines on the pitch to show and analyse movements and how the goal was scored.
- Military fighter pilots see an augmented reality projection of their height, speed, and other data on their helmet visor, which means that they do not need to waste focus looking down to see the information on their data panels.
- Sometimes, neurosurgeons use augmented reality projection of a 3D brain to assist them in performing surgery.
- At historical sites such as Rome in Italy, AR can be used to project scenes of ancient civilizations over today's ruins, bringing the past to life.
- Airport ground crew at Singapore's Changi airport wear AR glasses to see information on the goods being loaded into the hold, speeding up loading times.

Advanta	ges and di	sadvantag	jes of AR
---------	------------	-----------	-----------

Advantages	Disadvantages
he best of both worlds: combining the ligital world with the real world.	Privacy and security concerns: collecting private and personal information about the users.
/irtual communication: improves digital communication by making it more mmersive with virtual information.	Intrusiveness issues: the software records the world around the user which raises questions regarding taking photos of the public.
Supports business activities: retailers can use augmented reality to display their products.	Promotes dangerous behaviour: by focusing on augmented reality information accidents could happen in the real world.
extends smart devices so they play an additional role in users' lives.	It can be expensive to install and maintain augmented reality software and hardware.

# **Sevent** Year 10 – Cambridge National IT – Purpose of AR

What does reality mean?

natural

real.

٠

What does augment		
mean?		
•	enhance	
•	alter	
•	make more	

- What does real-world environment mean?
  - our surroundings
  - our planetour world.

# The sectors where AR can be used in

• architecture

retail

education

effective.

• lifestyle.

• entertainment

# The uses of AR:

- training
- virtual tours
- marketing
- to cre to cre to bro to bro
- visualisation of designs, interiors,
  - and concepts

# What is AR and what is its purpose?

- Technology:
  - overlay of computer-generated inputs:
    - text
    - graphics
    - photographs
    - audio
    - video.



# Forms of reality

- Augmented reality:
  - the overlaying of computer-generated inputs onto a real-world environment.
- Mixed reality:
  - the interaction of computer-generated inputs with physical objects in the real-world environment.
- Extended reality:
  - a term used for all forms of real and virtual environments.
- Virtual reality:
  - a computer-generated simulated environment.

#### SWB ACADEMY Year 10 – Cambridge National IT – Types of AR

## Types of AR:

- object recognition/marker-based ٠
- location-based/markerless ٠
- superimposed (sometimes referred to as superimposition).

## Types of augmented reality applications







Location-based



Marker-based

**Projection-based** 

Outlining

# **Consideration when Designing Types of AR**

- Questions to ask:
  - What content do we want to display on the live camera view?
  - Where should we place the content within the user's view?
- Answers dependent on:
  - AR software application being used each require different user interaction.

# **Markerless**

- More versatile than marker-based.
- More adaptable to a wider variety of functions/activities.
- The user decides where to place virtual object.
- It relies on device's hardware to gather information.
- An accelerometer detects the orientation of a smart device.
- A digital compass:
  - is a sensor that finds direction
  - uses Earth's magnetic fields
- always finds North.
- GPS (global positioning system) uses satellites to establish the location of an object.

# **Object recognition/marker-based**

This is the use of specifically designed markers to trigger augmented experience.

- Markers:
  - visual prompts
  - trigger virtual object/information
  - created using distinct patterns
  - act as anchors.
- Triggers:
  - activate AR experience.
- Anchors:
  - objects recognised by AR software. •

## **Superimposed**

- Uses object recognition.
- Replaces original image partially or fully.
- Used a lot in healthcare sector e.g. to superimpose x-ray onto patient's body.



Markerless



## Location-based

- Digital content tied to specific location.
- Often used for navigation.



#### Year 10 – Cambridge National IT – AR interaction

Types of user interaction/layers

- static
- interactive.

## Types of user interaction

- User interaction:
  - reflects the real user experience after an interaction with the user interface
  - experiences are generated by the user.
- Design consideration:
  - think about how the user is going to interact with the AR app
  - depends on the devices they are going to use.

### Layers

- Layers contain information and/or objects laid on top the real world viewed by the user.
- Static:
  - digital content such as text, 3D models and visual cues
  - appearance does not change during user interaction
  - no display of continuous movement.
- Interactive:
  - digital content such as animation and videos
  - changes appearance during user interaction
  - changes continuously with time
  - displays flow of continuous movement.

# User interaction for AR headsets

- Types of interaction:
  - gaze
  - hand gestures.
- Gaze:
  - gaze-based navigation
  - tracks where the user is looking
  - targets items in the environment.
- Hand gestures:
  - hand gestures used to interact with targeted item.

# User interaction for AR headsets

Important considerations:

- user's hands must remain in the viewing area of the headset
- will not function correctly if user's hands out of view of headset
- user's should be notified if hands reach the boundaries of headset view
- design must use hand gestures accepted by the headset
- keep interactions simple.

### **AR** headset



## User interaction for mobile devices

- Type of interaction:
  - hand gestures by touching the screen
  - interaction occurs in 3D (real-world) space.
- Hand gestures are based on touching the screen:
  - swipe
- pinch
- tap
- rotate.
- Important considerations:
  - interactions should be kept simple
  - user should receive feedback when placing or interacting with an object.

## Voice interaction

- Type of interaction:
  - voice control.
- Voice control considerations:
  - Commands should be kept simple and concise.
  - Commands must be able to be stopped, reversed undone.
  - Do not use similar sounding commands.
  - Do not use commands that are pre-set system commands.

## Voice interaction

- Additional important considerations:
  - User should receive feedback as with any other form of interaction.
  - Consider that users can have different dialects and access.
  - Test with different people.
  - Give the user prompts to use a command.

# Sear 10 – Cambridge National IT – Using AR to Present Information

Types of devices AR can be used on:

- mobile devices
- smart devices
- laptop/PC.

# **Categories of devices**

Device	Mobile device	Smart device	Laptop/PC
Smartphone	Х	Х	Х
Tablets/phablets	х	Х	Х
AR headsets	х	Х	Х
AR glasses	х	Х	Х
Kiosk systems/AR installations		Х	Х
Desktop computer			Х
Laptops	Х		Х

# **Category: smart device**

- A smart device is an interactive, electronic gadget.
- It can connect to, share with and be interactive with other devices and with users.
- Includes:
  - smartphones
  - tablets/phablets
  - AR headsets
  - AR glasses
  - AR kiosks
  - AR installations.

# **Category: smart device**

- AR kiosks and installations:
  - greater computer power
  - higher graphic quality
  - user does not require any additional device, e.g. smartphone.

# Category: laptop/PC

- Desktop AR creates an illusion of 3D objects floating in front of the screen.
- Users wear anaglyph glasses (red and blue) when using the AR app.
- They may use 3D monitors.
- There can be a better user experience using a laptop/PC.

# **Category: smart device**

- AR headsets:
  - users can see images superimposed on real-world environment
  - are more immersive than AR glasses (cover the eyes completely)
  - work in a way that is similar to AR glasses.
- AR glasses:
  - use a front-facing camera
  - superimpose 3D images, animation and video on realworld environment
  - can retrieve information from other devices
  - support Wi-Fi, Bluetooth and GPS.



### **Category: mobile device**

- A mobile device can be easily moved from one location to another.
- It can also be a smart device.
- Includes:
  - smartphones
- tablets/phablets
- laptops.
- Design considerations:
  - AR apps can be used anytime, anywhere (depending on their purpose)
  - tablets/phablets have greater computer power and larger screens (phablet: combination of a phone and a tablet – Phone + Table).

Year 10 – Cambridge National IT – AR – Content, Triggers, Assets and layers

## Layers and user interaction

- Static:
  - Will the digital content change during the user interaction or remain the same?
  - Digital content that does not change is referred to as static.
- Interactive:
  - Will the digital content change during the user interaction or remain the same?
  - Digital content that changes is referred to as interactive

## Layers and user interaction

- Layers:
  - information and/or objects laid on top the real world viewed by the user.
- User-interaction:
  - reflects the real user experience after an interaction with the user interface
  - experiences are generated by the user.
- Questions to ask
  - What content do you want to display on the user's live camera view?
  - Where do you want the content to be placed within the user's view?

# Layers and user interaction

- Action flow:
  - How will the user interact with the AR app to move from one area to another?
  - Will there be multiple layers?
  - How will the action flow?



# Triggers

- Triggers:
  - sometimes referred to as marker image
  - scanned by the device using the AR App
  - the scanned image 'triggers' the device to show the augmented digital content.

#### Markers:

- visual prompts
- trigger virtual object/information
- created using distinct patterns
- act as anchors.

# Triggers

- Location-based:
  - digital content tied to specific location
  - often used for navigation.
- Superimposed:
  - uses object recognition
  - replaces original image partially or fully
  - used a lot in healthcare sector, e.g. to superimpose x-ray onto patient's body.

## Triggers

- Object recognition/marker-based:
  - use of specifically designed markers to trigger augmented experience.
- Markerless:
  - more versatile than marker-based
  - more adaptable to a wider variety of functions/activities
  - user decides where to place virtual object **100**
  - relies on device's hardware to gather information.

# Sime Year 10 – Cambridge National IT – AR – Content, Triggers, Assets and layers

# Planning and design considerations

- Four key principles
  - How will you represent the physical setting?
  - How will you anchor the content?
  - How will you format the content?
  - How will you source the content?

# Planning and design considerations

- Physical setting
  - What is the physical environment like?
  - How will the computer vision detect flat surfaces, walls or surface planes?
- Anchoring the content
  - Devices and AR software must be able to recognise something in the real world.
  - Where and how will you place the augmented content (the anchor)?
  - Use anchors to trigger the AR experience.

# **Content and assets**

- Considerations for audio
  - What is its purpose?
  - It must be clear and easily understood.
  - It can include sound effects, but must also be of good quality.

## **Content and assets**

- Considerations for photographs/images:
  - must have high level of detail and contrast
  - must be unique
  - do not use logos as marker images
  - avoid using same images, symbols in the same design (use them only once)
  - must not be blurred.
- Considerations for text:
  - must remain legible even when user moves
  - must be legible even with different background colours
  - Helvetica is a popular font for AR.

# **Content and assets**

- Considerations for video:
  - may or may not include sound (depends on the purpose)
  - if audio used with video, must be clear and easy to hear.
- Take your time!
- Think carefully about what you are trying to achieve.
- Spend time selecting the most suitable content.
- Obtain good quality sources.

# **Content and assets**

- Content:
  - can be displayed in many formats, e.g. text, images, video, sound, etc.
  - can come from a variety of sources
  - sources can be digital or paper-based
  - must comply with legislation, e.g. copyright and licensing laws.
- Assets:
  - digital content referred to as assets
  - can be: audio, video, graphics, photographs, text.

## **Content and assets**

- Considerations for charts and graphs:
  - used to help explain and analyse data
  - must be appropriate for the data being presented
  - must be clearly and accurately labelled
  - must be of good quality and easy to read.
- Considerations for hyperlinks/weblinks:
  - Will the links be available for the lifetime of the AR app?
  - How will the user interact with the links (touch screen, voice control)?

# **Sweet Year 10 – Cambridge National IT – AR Prototype**



# AR model prototype

- High-fidelity prototype:
  - looks real and simulates how the design will work
  - used to present ideas to clients and end-users
  - used to gather feedback from clients and end-users
  - allows user interaction
  - looks as close as possible to the final product.
- Live-data prototype:
  - has access to live data
  - should not have an impact on any process using the live data outside of the design.

### AR model prototype

• High-fidelity prototype:



## AR model prototype

- Feasibility prototype:
  - try out new technological/software-related ideas
  - aids understanding of technical risks associated with performance
  - used to test specific design elements/features.
- Low-fidelity prototype:
  - basically an interactive wireframe and does not look real
  - used to simulate processes to identify potential user issues
  - does not include user interaction, or includes minimal user interaction.

## AR model prototype

Characteristics of prototypes:

- A prototype is not a full product.
- It confirms the functionality of, e.g., the HCI and how triggers and layers work.
- It shows how the product looks and feels to the user.
- It has access to real data.
- If the purpose of the app is to access and use real data (think of a SatNav), then the prototype must show that this can happen.

## AR model prototype

- Feasibility prototype:
- Low-fidelity user prototype:





SWB Year 10 – Ca	mbridge Nation	nal IT – Triggers 🏼	Types
Characteristics Triggers:     sometimes referred to as a ma     scanned by the device using the the AR experience and display content     one of the basic building block app.	arker image ne AR app to 'trigger' the augmented digital ks when creating an AR		<ul> <li>Markerless: <ul> <li>scans a marker which triggers the AR experience.</li> </ul> </li> <li>Superimposition: <ul> <li>partial or full replacement of the original view of an object with an augmented view of the same object</li> <li>object recognition is used to identify the original object; once identified, the superimposition trigger takes over and replaces partially or in full the object with an augmented object.</li> </ul> </li> </ul>
	A trigger image: Should be unique (different trigger images to trigger the different AR content) contain plenty of detail have good image size (to provide a recognisable view)	Should not         contain anything reflective as it reduces the quality of the image, e.g. water, mirrors, white paper         be moving         be something that is not flat         contain blurred images (will not show	
<ul> <li>Types</li> <li>Remember the context!</li> <li>The type of trigger you use will defor using the trigger so ask yoursel</li> <li>Do I want to provide information</li> <li>Do I want to overlay onto an observation of the street of the select whe virtual object?</li> </ul>	pend on your reason f: on on an object? bject (e.g. street here to place the	contain too much text contain too much white space	<ul> <li>Types</li> <li>Marker: <ul> <li>an image or an object that can be recognised by an AR enabled app, used to trigger the augmented reality features.</li> </ul> </li> <li>Object recognition/marker-based <ul> <li>a specifically designed marker, used to trigger the augmented experience.</li> </ul> </li> <li>Location-based: 103</li> </ul>

# Similar 10 – Cambridge National IT – Layers

# Interactive access for the user

- Swipe:
  - think about the way you use a smartphone or tablet
  - dragging the finger across the screen to change views.
- Click/select:
  - click on an image or menu to select an action to be carried out.

# Interactive access for the user

- Voice:
  - simple voice commands to tell the AR app the action to be carried out
  - voice commands must be able to be stopped or reversed
  - do not work well in noisy environments
  - must provide feedback to the user.

#### Layers

- Single layer:
  - One trigger is created and an AR asset is overlaid on the top of it.
  - Assets can be images, audio, video, photographs, charts/graphs, etc.
  - The user activates the trigger.
- Multiple layers:
  - One or more triggers are used to overlay multiple assets.
  - For example: <u>https://youtu.be/w\_IDI4ag5aQ</u>

## Access to layers

- Static:
  - has no view of continuous movement
  - does not change appearance with user interaction.
- Interactive:
  - shows continuous movement
  - appearance changes with user interaction
  - the user physically controls the layers of the AR app by using a variety of controls, e.g. swipe, click/select/voice.

### Layers

#### Points to consider:

- Do not overcomplicate the layers keep them simple.
- Think about the user interaction with the AR app when triggering the layers.
- Merge multiple layers together (link them into layers).
- Give layers names that indicate what they include, what they do and how they are organised.
- Layers use the triggers, so must be set up to use them.

# Sim Year 10 – Cambridge National IT – Testing

# Types of test plans

- User testing:
  - sometimes referred to as usability testing
  - user testing carried out with several users from target audience
  - tests must include the different features, e.g., user interaction, action flow, etc.
  - tests are set on the test plan by the technical tester
  - results and feedback analysed to identify issues and potential improvements.

# Types of test plans

- Technical testing:
  - used to test prototypes (and eventually full products)
  - tests the functionality of the prototype/product, e.g., user interaction, action flow, triggers, layers, information output
  - analysis of results and report written to confirm the issues that arose, that they were addressed and any potential improvements
  - tests are identified by the technical tester and a test plan developed.

# Using a test plan

Content and use of a test plan:

- test number all tests should have a test number
- what is being tested brief statement of what is being tested (tests should **not** be grouped, e.g., testing triggers) and how it is being tested
- expected results -what you want to happen
- actual results a short description of what actually happened (particularly important if there are issues)
- remedial action if there was an issue, what you did to address it
- re-test!
- must be carried out iteratively (as you are developing the model prototype)

# **Test plans:**

Purpose of a test plan:

- ensures that the AR model prototype (and eventually the final product), works as intended
- should be carried out iteratively (as you are creating the model prototype)
- allows users to physically use the prototype (easier to understand something that can be seen and used)
- showing prototypes of potential fully functional products and proving they work can attract investors

105

 a check against the original design to ensure it meets design priorities.

# Year 10 – Cambridge National IT - Excel

# Type of Functions

There are many types of Functions that can be used in spreadsheets helping to make calculations a lot easier. Some common uses are **MIN** of Minimum, **MAX** for Maximum, **AVG** for Average and **SUM** for Sum total



This is the correct way to structure a function – Remember always start with a =

# Absolute Cell Referencing

# **Absolute Reference in Excel**

with GST
E\$33

Select a cell which you need to permanently look at and press **F4** on you keyboard to make the absolute cell referencing \$ to appear around the selected cells.

# Type of Graphs

Bar chart	Visual tool, uses bars easy to
	see difference, long bar
	means greater value
Pie chart	size of portion represents the
	quantity, visually simple to
	flow, good for summaries
Scatter	show relationship between 2
graph	variables, maximum and
	minimum values are easy to
	work out





Keywords	Definition			
Absolute Cell Referencing	When you want a formula to consistently refer to a particular cell.			
Function	is a predefined <b>formula</b> that performs calculations in a particular order			
Formula	is an expression which calculates the value of a <b>cell</b>			
Conditional formatting	is a feature which allows you to apply a format to a cell or a range of cells based on certain criteria			

# IF Statement...

C2	2 <b>*</b> : × •	=IF(B2<=5	0,"Fai	I","Pass") ←		
	А	В	С	D	E	F
1	Student Name Scores		Result			
2	BRUCE GEYER	37	Fail	-		
3	ELIZABETH STERN	73	Pass		Criteria	Result
4	MASATOSHI HENDERSON	62	Pass		Below or Equal to 50	Fail
5	CHRISTINE YOSHIMURA	43	Fail		Above 50	Pass
6	JOHN ADAMSON 35		Fail			
7	IRVING PIANKA 86		Pass			
8	EILEEN HAAS	81	Pass			
9	VINCENZO KWAN 50		Fail			

The IF function can perform a logical test and return one value for a TRUE result, and another for a FALSE result. For example, a "Fail" is scores below 50: =IF(B2<=50, "Fail", "Pass") 106





# Year 10 – Cambridge National IT - Excel

Key Words	
Absolute	A cell reference within a spreadsheet which remains
referending	the same when copied (replicated) to another cell.
Alignment.	The way that text is set out, for example, right, left,
	centred or justified.
AutoSum	This feature will add up the numbers you have
	entered in your sheet and displays the total in a cell
	of your choosing.
Cell	A box on a spreadsheet that can contain text
	numbers or a formulae.
Cell attributes	The way the spreadsheet cell displays data such as
	numbers or dates.
Cell format	The way the spreadsheet cell looks, for example
	changing the column width or alignment.
Cell protection	Locking the contents of the cell to prevent them
	from being changed.
Charting	Drawing a graph from a set of numerical data.
	usually from a spreadsheet program.
Columns	Vertical parts of a spreadsheet.
Complex formulae	Complicated formulae's used in a spreadsheet.
COUNTA	Count the number of non-blank cells
COUNTBLANK	Count cells that are blank
Data	A general term used for numbers, characters,
	symbols, graphics and sound which are accepted
	and processed by a computer system.
Delete	To remove data
File	Information held on backing storage or in memory.
Formulae	A calculation involving one or more cell references in
	a spreadsheet.
Hard Copy	A printed copy of your work, usually on paper.
Insert	To put Information.
Insert column	To add a column .
Insert row	To add a row.
Integer	A whole number, with no fractional part or decimal
	point.
Merged Cell	When two or more cells are combined, it's become
	what is known as a merged cell.

Kaul Inada	
key words	
Print file	To obtain hard copy.
Relative	When a formulae in a spreadsheet is changed relative to
Referencing	its position, during copying or replication.
Replicate	To duplicate
Rows	Horizontal parts of a spreadsheet.
Ribbon	Above the workbook is a section of command tabs
	called the Ribbon.
Spreadsheet	A program used mainly for dealing with number data. It
	divides the screen into rows and columns. The cells can
	contain text, numbers and formulae.
Table	A table is made up of data arranged in rows and
	columns like a spreadsheet and data is placed in the cells.
Workbook	The workbook refers to an Excel spreadsheet file.
Worksheet	Within the workbook is where you will find documents
	called worksheets.

Functions			
SUM	adds values in selected cells		
MEINEMUM() or MEIN()	A function used to find a minimum value in a range.		
MAXIMUM() or MAX()	A function used to find a maximum value in a range.		
AVERAGE()	A function used to find the average value in a range.		
COUNT	counts how many of the selected cells have numbers in them		
IF	change the value of a cell if something is true, eg if a customer's total bill is over £100, deduct 10% from their bill.		
COUNTIF	adds up cells that meet a certain rule, eg count the number of students that achieved level 6.		
VLOOKUP	is an Excel function to lookup and retrieve data from a specific column in table.		
Symb	Symbols used in formulas		
+ 0	dd 🛛 🔹 multiply		

divide

.

subtract
# Year 10 – Cambridge National IT - Excel

TRe	Hone Ins	ent. Page Leyout	Formulas Deta	Review View	r Add-Ins Ablel	Sits.com Acrobat		
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C	olumn	<i>I</i> ∐ · ⊞	- <u>≫</u> - <u>A</u> - ≡	まま 休休	Merge & Center *	- % · 3	Conditional Format Formatting * as Table *	Cell Insert Deleti Styles
		Font		Aligne	ent is	Number	To Styles	Cells
_	16	- (* J.						Pow
1	A	В	C	D	E	F	G	
1	Surname	Forename	Year of Birth	House	Form	Birth Place	<b>Behaviour Points</b>	P.ward Points
2	Anderson	Brian	2001	Mason	M1	Hull	0	30
3	Banks	Peter	2000	Mason	M2	York	1	67
4	Delta	Simon	2000	Hockney	H7	Hull	0	54
5	Carrington	Frank	2000	Palin	P3	Hull	2 1	30
6	Ellis	James	2000	Mason	M3	Beverley	0	10
7	James	Martin	2001	Hockney	H3	Selby	3	25
8	Hughes	Jake	2000	Garrett	G2	Hull	0	30
9	Reed	Ben	2001	Garrett	G1	Hull	5	15
10	Bassett	James	2002	Palin	P3	York	0	30
11	Williams	Jordan	2003	Hockney	H3	Hull		35
12	Robinson	Sam	2004	Hockney		York	E Ce	30
13	Green	Alex	2005	Mason	M4	Selby		67
14	Dean	Mark	2006	Mason	M9	Hull	0	30
15	Griffiths	Tom	2007	Hockney	H7	Hull	1	23
16	Rose	Thomas	2008	Garrett	G8	Selby	0	30
17	Senior	William	2009	Garrett	G3	Hull	10	24

Charts and gra	phs
line graph	to show a change over time
ple chart	show the individual parts that make up a whole
bar chart	compare things that aren't directly related
scatter graph	look for a pattern or link between two sets of data

# Scatter



# Every cell has a unique name called a Cell Reference. The cell reference for this cell is D12

1	A	В	C	D	
1	Surname	Forename	Year of Birth	House	Form
2	Anderson	Brian	2001	Mas	M1
3	Banks	Peter	2000	Masor	M2
4	Delta	Simon	2000	Hockn y	H7
5	Carrington	Frank	2000	Palin	P3
6	Ellis	James	2000	Masor	M3
7	James	Martin	2001	Hockn y	H3
8	Hughes	Jake	2000	Garret	G2
9	Reed	Ben	2001	Garret	G1
10	Bassett	James	2002	Palin	P3
11	Williams	Jordan	2003	Hockn y	H3
12	R moon	0am	200.0	Hockney	H9
13	Green	Alex	2005	Mason	M4





# DT Knowledge Organiser: Year 10



Design Brief: A Design Brief is a short paragraph explaining the situation you have been given and the problem you need to solve.

Purpose:

- Identify a Problem
- Identify the client
- How to go about solving the problem
- Solutions

Client Needs/Brief: What the client requires of a product, here are some examples:

- Function
- Disabilities
- Social
- Anthropometrics/Measurements
- Material/Finishes
- Health and Safety
- Costing

Design Specification: A list of points to state what the product must have to meet the needs:

Possible Sections: Material, Safety, Ergonomics, Environmental, Costing, Manufacture, Finishes, Age Range, Functions, Secondary Function and Quality Assurance. Sustainability:

What does it mean? To preserve resources, materials and processes for future generations.

Examples:

.

- Recycled wood allows afforestation to occur
- If you have to incinerate a material it causes Co2 emissions
- Re-using electronic components saves on the processing of materials and saves on wastage and energy.

# Click to add text

- **Recycle** Take an existing product that has become waste and re-process the material for use in a new product.
- <u>Reuse</u> Take an existing product that's become waste and use the material or parts for another purpose, without processing it.
- **Reduce** Minimise the amount of material and energy used during the whole of a products life cycle.
- **<u>Refuse</u>** Don't accept a product at all if you don't need it or if its environmentally or socially unsustainable.
- <u>**Rethink**</u> Our current lifestyles and the way we design and make.
- **<u>Repair</u>** When a product breaks down or doesn't function properly, fix it.

# DT Knowledge Organiser: Year 10



## <u>Timbers</u>

#### Softwoods:

- Coniferous trees
- Pale in Colour
- Soft/Easy to Cut
- Used for Furniture and Construction Trade
- More sustainable

#### Hardwoods:

- Deciduous trees
- Durable
- Hard to shape and cut
- Long time to Mature
- Furniture
- Some are rare to find

#### Manufactured Boards:

- made through human intervention
- consist of recycled woods part
- mixed with a liquid for example Glue or resin
- Flat Pack Furniture
- Cheap to manufacture





#### **Finger Joint**

If the joint is cut accurately the 'fingers should fit together without any gaps and the glue ensures that they are virtually indestructible. They are used for a wide range of products including jewellery boxes

Non Ferrous: A metal that does not

Ferrous: A metal that does contain

Alloy: a metal that is made of 2 or

more metals combined with

improved properties.

Metals

iron.

contain Iron.



#### Lap Joint

The shoulder can be seen clearly and has been pushed into the shoulder and this means it is level with the sides. It is very common and is used for furniture and box constructions such as jewellery boxes.



#### Housing Joint

A housing joint is a very simple, very strong woodworking joint that joins two work-pieces at right angles. It is done by cutting a slot into one piece and fitting the other piece into it.

## **Polymers**

Thermo-plastics: A plastic that can be heated and changed shape without breaking. It can also be reheated and remoulded many times as it has a memory structure, once set becomes rigid.

Thermo-Setting Plastics: A plastic that once set it cant be remoulded as it has a very rigid structure once set. If heated it will become flammable.

# DT Knowledge Organiser: Year 10







# BTEC Engineering Component 3 Learning Aim A (Year 10 Spring)

Production plan



Prototyping Prototyping is an important part of the development of most products, not just those in engineering. Some products are only ever produced in prototype form, as only one is needed. Prototypes can be: The main reason for prototyping is to physical models prevent expensive mistakes from computer simulations happening. The use of prototypes allows an engineer to find any faults and problems with a design before it is made on a large scale. Using gauges The go/no-go gauge is used for checking the diameter of holes. The 'go' gauge, coloured green, must be able to fit through the hole. If the 'go' gauge does not fit, the hole is too small. 40 • -ength (mm) Anomalous results 30 An anomalous result is one 20 that does not fit in with the expected pattern or trend 10 0 100 200 300 400 0 Mass (g)



Pictograph



# Year 10 Hospitality & Catering KO - LO2 – Understand how hospitality and catering provisions operate

Kitchen Workflow	Small Equipment	. [	Types of Custom	er		Good Customer Service
1. Delivery 2. Storage	Chef's Knife	Leisure	Local Residents	Business/ Corporate	<ul> <li>✓ Respect</li> <li>✓ Helpful c</li> <li>✓ Smart ar</li> </ul>	ful and polite and attentive ad professional
<ul> <li>3. Food preparation</li> <li>4. Cooking</li> <li>5. Holding</li> <li>6. Food service</li> <li>7. Wash up</li> <li>8. Waste disposal</li> </ul>	Bread Knife Boning Knife	Customers who visit the establishments in their leisure time e.g. a meal with friends, a family day out, tourists.	Customers who live in the local area who visit the establishment often e.g. regular Sunday	E.g. business lunches. Use business facilities in establishment for meetings or presentations, courses and	<ul> <li>Friendly</li> <li>Patient</li> <li>Deals wi effective</li> <li>Knowled</li> <li>Makes th want to</li> </ul>	th customer problems ely dgeable ne customer feel welcome, to return.
	Filleting Knife		lonch	conierences	Keyword	Definition
<ul> <li>(fridge, freezer)</li> <li>Time sheets (staff working hours)</li> </ul>	Paring Knife	All businesses shou	Safety and Secur	ar health and safety	Perishable Foods	Foods with a short shelf life such as meat, fish and dairy
<ul> <li>Accident report form (to log first cid/accidents)</li> </ul>	Cleaver Knife	your business and affect staff or ment should tell	Staple Foods	Foods with a longer shelf life such as canned products		
Equipment fault reports	Large Equipment	you whether you or risks. Ways to achie	are doing enough to eve this is through:	Stock Control	Using the First In First Out (FIFO) rule to rotate food stock	
Kitchen Dress Code (PPE)	All large scale equipment such as a floor standing mixer, walk in fridge/ freezer and a large oven must be: Turned off before cleaning	CCTV Staf	f Lockers Safes fo	or Money Lighting	Consumer Rights Act 2015	A legal <b>right</b> to reject goods that are of unsatisfactory quality, unfit for purpose or not as described
Neckerchief Jacket	<ul> <li>Use correct cleaning materials</li> <li>Any attachments should be stored correctly</li> <li>If equipment is not working correctly it must be reported</li> </ul>				Food Safety Act 1990	Businesses do not include, remove or treat <b>food</b> in any way that would be damaging to the health of people.
Apron	<ul> <li>Ensure there is no food left on the equipment as it will contaminate</li> </ul>	↓ The right to be	protected (against h	nazardous goods)	Equality Act 2010	All must be treated equally regard to age, gender, race
Chequered Trousers Non-slip shoes	future use	<ul> <li>The right to be</li> <li>The right to have</li> <li>The right to see</li> <li>The right to reconstruct description</li> </ul>	informed (quality, qu ve their complaints b k compensation eive satisfactory goo ption	pantity, allergies) he heard hat match their	Disability Discriminat ion Act 2010	Unlawful to discriminate against mental and physical disabilities <b>115</b>
0.1000			I			

# Year 10 Hospitality & Catering KO - LO3 – Understand how hospitality and catering provision meets health and safety requirements

#### HASAWA

- □ All employers must take care of their own health and safety and not endanger others.
- □ The HSE exists to protect peoples health and safety by ensuring risks are properly controlled.
- HASWA also protects employees from risks to their health and safety arising out of the activities of people at work.
- □ The law applies to everyone at work and anyone can be prosecuted if they do not act safely.

## RIDDOR

- The law requires employers and other people in control of work premises (known as the 'responsible person') to report to the Health and Safety Executive (HSE) and keep records of the following:
- ✓ Work related fatalities
- ✓ Work related accidents causing certain serious injuries
- ✓ Certain work related diagnosed occupational diseases such as severe cramp of the hand due to work related issues

# COSHH

- □ These Regulations require employers to control exposure to hazardous substances to prevent ill health.
- □ Substances covered by COSHH:
- ✓ Chemicals including cleaning chemicals
- ✓ Micro-organisms
- ✓ Dusts
- ✓ Medicines, pesticides, gases
- COSHH Symbols:

Oxidising

Gas under pressure Repro









A. Comment est-o	ce que tu u	tilises des	s réseaux s	ociaux ? How do y	ou use soo	cial media	Ś			J.	
Normalement Normally D'habitude Usually Quelquefois Sometimes Souvent Often	j'utilise   use   don't us   don't us j'aime ut   like to us	<b>e pas</b> Se <b>iliser</b> Se	eBay Faceboo Fitbit Instagran Netflix Snapchar Spotify TikTok WhatsAp	k sur mon ordin on my comp n sur mon porto on my mobile sur ma tablet on my tablet	ateur uter ible r e	pour to to tél	atter avec des amis. cha ontrôler mon activité phys voyer des messages. to s re des achats. to do shop Jer aux jeux-vidéos. to pl urtager des vidéos. to sha garder des séries. to wate ffer sur Internet. to go onli écharger de la musique.	t with friends. <b>ique</b> . control my physical activity. end messages. pping. ay video games. re videos. ch series. ne. to download music.		f }	
B. Quels sont les a	vantages	et les inco	onvénients	des réseaux sociau	<b>Ix ?</b> What	t are the ac	dvantages and disadvan	tages of social media?			
<b>D'un côté, les rése sociaux sont</b> On the one hand media are	e <b>aux</b> , social	divertiss entertai gratuits free of c pratique practicc populai populai	ants ning charge es al res	<b>de l'autre côté</b> on the other hand <b>en revanche</b> on the other hand	un inca c'est q a disaa that un dés c'est q a disaa that	<b>convénient</b> qu' advantage i <b>savantage</b> qu' advantage i	is <b>il y a le danger</b> there is the danger is	d'y devenir accro. of becoming ad d'être victime du harcèlement. of being the victim of bullying. d'être victime de la cybercriminalité of being the victim of cybercrime. de rencontrer des gens avec de mo of meeting people with bad intentio de voir du contenu inapproprié. of seeing inappropriate content.	dicted. é. auvaises intentions. ons.		

$11 \times 13$	
'(- X -)	$\times$ -)

C. C'est comment ta routine	quotidienne ? What is your daily routine like?		
		à une heure at 1 o'clock	
Avant d'aller à l'école	je me brosse les dents I brush my teeth	à deux heures at 2 o'clock	
Befoing going to school	je me couche I go to bed	à trois heures at 3 o'clock	
	je me détends l relax	à quatre heures at 4 o'clock	
Après l'école	<b>ie me douche</b> I have a shower	à cina heures at 5 o'clock	
After school	ie m'habille   aet dressed	à six heures at 6 o'clock	et auart auarter past
	ie me lève l'act up	à sept heures at 7 o'clock	<b>et demi</b> half past
Le matin In the morning	ie fais la arasse matinée (iusau')   have a lie in (until)	à huit heures at 8 o'clock	moins le quart quarter to
L'après-midi	ie fais mes devoirs I do my homework	à neuf heures at 9 o'clock	
In the afternoon	ie prends le petit-déjeuner   have breakfast	à dix heures at 10 o'clock	
Le soir	je prends le diner l have dinner	à onze heures at 11 o'clock	
In the evening	je rentre à la maison Lao back home	à midi at midday	1
	, · · · · · · · · · · · · · · · · · · ·	à minuit at midniaht	
			117



D. Décris-moi ur	<u>ne fête que t</u>	u as fêt	<b>ée/tu aimerais fêter.</b> Descrit	pe a fes	estival th	<u>at you celebrated/you wor</u>	<u>uld like to celebra</u>	ate.	•		
L'année dernière Last year <b>Récemment</b> Recently	année         srnière         ist year         cemment         cently         j'ai fêté         l celebrated         nous avons fêté         we celebrated         le Ramadan Ramad         la Noël Christmas		l'Aïd Eid le Dipavali Diwali le Hanouka Hanukkah le Pâques Easter le Ramadan Ramadan la Noël Christmas	a et c'était and it was ir p		agréable pleasant divertissant entertaining formidable terrific impressionnant impressive incroyable incredible passionnant exciting	<b>car</b> because	j'a j'a j'a j'a j'a il y il y	ii chanté. I sang. ii dansé. I danced. ii fait de nouveaux amis. ii mangé des repas délic ii passé du temps avec r ii reçu des cadeaux. I re y avait des feux d'artifice y avait de grands repas.	I made new friends. <b>Sieux.</b> I ate delicious meals. <b>na famille.</b> I spent time with my family. ceived gifts. <b>A.</b> there were fireworks. there were big meals.	
Quand je serai vieux/vieille When I'm older Dans le futur In t	he future	une fê a festi celeb une fê c'est a festi	ival which I would like to orate is ival which I would like to orate is ival to which I would like to g	t le N Ic r, lc	e Carna Mardi Gi a Fête n a Chana a Saint-3	aval Carnival ras nationale Bastille Day deleur Sylvestre New Year's Eve	parce que ce serait because it would be		<b>une opportunité de/d'</b> an opportunity to	s'amuser. have fun. apprendre de la culture. learn about cult décompresser. relax. découvrir de cultures différentes. discove manger de la cuisine typique. eat typica passer du temps avec les autres. spend t	rure. er different cultures. I food. ime with others.
<b>À l'avenir</b> In the	future	is			la Saint-Valentin Valentine's Day					voir des défilés. watch parades.	

	E. Qu'est-ce que tu vas	aire pour erre en bonne sante ? what are you going to do to keep nealthy?							
		<b>je vais</b> I am going to	avoir un régime équilibré have a balanced diet boire de l'eau drink water bien dormir sleep will éviter la melheuffe queid fact food	<b>sinon je risque d'être</b> if not, I risk being	accro. addicted. malade. ill. obèse. obese. stressé. stressed.				
	<b>Pour être en bonne santé</b> To be in good health	l must	eviter la malboutte avoid tast food         faire de l'exercice do exercise         faire du sport do sport         manger sainement eat healthily         me reposer rest	<b>sinon je risque d'avoir</b> if not, I risk having	<b>un cancer (des poumons).</b> (lung) cancer. <b>une crise cardiaque.</b> a heart attack.				
		<b>je peux</b> I can		<b>puisque c'est bon pour</b> as it is good for	le corps. the body. le mental. the mind. la santé. the health.				
	<b>Pour rester en forme</b> To stay in shape	<b>je ne vais pas</b> I am not going to <b>je ne dois jamais</b> I must never	boire trop de l'alcool drink too much alcohol me droguer take drugs m'enivrer get drunk fumer smoke	<b>étant donné que c'est</b> given that it is	dangereux. dangerous. illégal. illegal. inquiétant. worrying. mauvais pour la santé. bad for your health. une perte d'argent. a waste of money. une perte de temps. a waste of time. 118				



# Year 10 French – Topic 3 – Where I live and social issues

Le weekend dernier

Last weekend

Récemment

Recently

j'ai

we

nous avons

Theme 2: Local, national, international and global areas of interest

A. Qu'est-ce qu'il y a là	où tu habite	es ? What is there where	you live?		-	-				
Dans ma région In my region Dans ma ville In my town Dans mon village In my village Dans mon quartier In my neighbourhood	il y a there is/th are il n'y a pa de* there isn'	un aquarium an un bowling a bo un centre comm a shopping cent un centre sportif un cinéma a cin un stade a stadiu un stade a stadiu une bibliothèque une cathédrale a une église a chu une gare a train t une mosquée a une piscine a sw une poste a posi des galeries gall des musées mus des parcs parks des restaurants r	aquarium wling alley ercial re a leisure centre ema arket um atre a library a cathedral rch station mosque rimming pool toffice eries eums monuments estaurants	<b>par contre</b> on the other hand <b>pourtant</b> however	<b>dans le passé</b> in the past	<b>il y avait</b> there was/the were <b>il n'y avait pa</b> there wasn't	ere un aqu un boy un cer un cer un cer un cer un cer un stac un stac un stac un thé une ce une bil une ce une bil une ce une de une de une de une de une de une pil des ge des me des res	ving ving the commercial the sportif éma rché de âtre bliothèque thédrale glise tre osquée scine oste uleries usées onuments trcs staurants		
		B. Qu'est-ce que tu as	fait récemment	dans ta région ? Wha	at have you done re	ecently in your r	egion?	-		
		Hier Yesterday Hier soir Yesterday evening La semaine dernière Last week	je suis allé[e] I went nous sommes we went	à l'aquarium à la piscine tr au centre co au cinéma to à un concert à un match d prendre un vo appris de l'hi appris de la o fait du shopp fait du sport c	to the aquarium o the swimming poor mmercial to the sho o the cinema to a concert le foot to a football erre for a drink stoire learnt about th culture learnt about ing did shopping did sport	ol opping centre match nistory culture	<b>à mon avis</b> in my opinion	<b>c'était</b> it was	absolument absolutely extrêmement extremely	divertissant. entertaining fascinant. fascinating. génial. great. sensass. sensational.

fait du tourisme did sightseeing

fait des promenades did walks

fait des randonnées did hikes

fait de la natation did swimming

ate Chinese/Indian/Italian meals

visité le musée visited the museum visité la galerie visited the gallery

joué au basket/tennis played basketball/tennis mangé des repas chinois/indiens/italiens

sans doute

without doubt it wasn't

ce n'était pas

tellement

so

plutôt

rather

affreux. awful. barbant. boring. casse-pieds. Irritating. décevant. disappointing.

119





ШҮ
-Ò-

C. Où vas-tu habiter à l'avenir	– en ville ou à la campag	ne ? Where are you going	to live in the future	e – in the city or in the countryside?	i fil	Шŀ
	je vais habiter	en banlieve		il y a beaucoup de magasins. there are lots of shops.		1111:
Quand is serai vieux /vieille	I am going to live	in the suburbs		il y a plus de distractions. there are lots of things to do.		ШŸ
When I'm older				il y a plus de possibilités d'emploi. there are more employment opportunities.		<b></b> !!
	je ne vais pas habiter	en ville	parce que/qu'	il y a tellement de bruit. there is so much noise.	_  •• ii	117
Quand i'aurai x ans	I am not going to live	in town	because	il y a trop de circulation. there is too much traffic.		<u>п</u>
When I'm x years old				il y a trop de pollution. there is too much pollution.		
when my years or	je vais vivre	au centre-ville	vu que/qu'	il y a trop de déchets par terre. there is too much rubbish on the ground.	_	
À l'avonir	I am going to live	in the town/city centre	seeing as	il y a plus d'air frais. there is more fresh air.		
In the future				il y a plus d'espaces vertes. there are more green spaces.		5
	j'aimerais habiter	à la campagne	néanmoins	il y a plus de pistes cyclables. there are more cycle paths.		۲
Dans la futur	I would like to live	in the countryside	nevertheless	il y a de beaux paysages. there are beautiful landscapes.		╏┍╴
In the future				<b>il n'y a rien pour les jeunes.</b> there is nothing for young people.		
	je voudrais vivre	dans un petit village		<b>il n'y a rien à faire.</b> there is nothing to do.		$\overline{\prime}$
	I would like to live	in a small village		il y a plus de chômage. there is more unemployment.		٩ı



D. Quels sont les problèmes pour les SDF? What are the problems faced by homeless people?				
Je considère que	les gens au chômage, unemployed people les gens vivant dans la pauvreté,	<ul> <li>ils ont faim. they are hungry.</li> <li>ils ont froid. they are cold.</li> <li>ils ont soif. they are thirsty.</li> <li>ils n'ont pas d'argent. they don't have any money.</li> <li>ils n'ont pas d'amis. they don't have any friends.</li> <li>ils n'ont pas de médicaments. they don't have any medicine.</li> </ul>		
I consider that <b>J'estime que</b> I feel that	people living in poverty les sans-abris, homeless people les SDF, homeless people	ils se sentent they feel	abandonnés. abandoned. déprimés. depressed. inutiles. useless. socialement exclus. socially excluded. seuls. alone. vulnérables. vulnerable.	
	nomeless people	<b>ils peuvent être</b> they can be	victime d'une attaque. victims of an attack. en danger. in danger.	



E. Qu'est-ce qui te préoc	ccupe le plus dans le monde aujourd'hui ? Wha					
Ce qui me préoccupe le plus, c'est What concerns me the most is Ce qui m'inquiète le plus, c'est What worries me the most is Le plus grand problème pour la planète, c'est The biggest problem for the planet is	le changement climatique climate change le chômage unemployment le déboisement deforestation le racisme racism le sexisme sexism la corruption corruption la faim hunger/famine la guerre war la pauvreté poverty la pollution de l'air air pollution la surpopulation overpopulation l'environnement environment l'inégalité inequality l'injustice injustice les catastrophes naturelles natural disasters	car c'est because it is puisque c'est as/since it is	catastrophique. catastrophic. effrayant. scary. inacceptable. unacceptable. inquiétant. worrying. préoccupant. concerning.	À mon avis In my opinion Selon moi According to me Sans doute Without doubt	on a besoin de/d' we need il faut plus de/d' we must have more	campagnes publicitaires.         advertising campaigns.         coopération internationale.         international cooperation.         investissement dans l'éducation.         investment in education.         investissement dans les énergies renouvelables.         investment in renewable energies.         lois plus strictes.         stricter laws.         paix dans le monde.         peace in the world.         projets d'aide.         aid projects.

	F. Qu'est-ce qu'il fa	ut faire pour protéger l	'environnement ? W	nat must we do to protect the environment?		
りつ	Bien que ce soit difficile, Even though it is difficult, Bien que ce soit embêtant, Even though it is	<pre>pour protéger l'environnement to protect the environment pour protéger la planète to protect the planet</pre>	<b>il faut</b> you must <b>on doit</b> we must <b>on peut</b> we can	consommer moins d'énergie. consume less energy. créer d'espaces verts. create green spaces. économiser de l'eau. save water. faire du compost. compost. faire des achats responsables. shop responsibly. installer des panneaux solaires. Install solar panels. recycler. recycle utiliser les transports en commun. use public transport.	<b>Personnellement,</b> <b>je vais</b> Personally, I'm going	aller à l'école à vélo. to go to school by bike. baisser le chauffage. to turn down the heating. éteindre la lumière. to turn off the lights. recycler. to recycle. prendre une douche. to take a shower. réduire mon empreinte carbone.
-	annoying,	<b>pour lutter contre la</b> <b>pollution</b> to fight against pollution	<b>il ne faut pas</b> you mustn't <b>on ne doit jamais</b> we must never	gaspiller de l'eau. waste water. manger de la viande. eat meat. prendre un bain. take a bath. utiliser les sacs en plastique. use plastic bags. voyager en voiture. travel by car.		trier les déchets. to separate rubbish.

Geography – Topic	: 1 Changir	ng Physical Lands	capes	Keyword	Definition
• The study of rocks and different rock types	<u>C</u>	oastal Landforms	Formation of a Wave-Cut Platform	Erosion	The wearing away and removal of material by a moving force e.g. rivers or waves.
	Deposition	al Erosional	the cliff. 2. The sea attacks the base of	Abrasion	Erosion caused by sediments rubbing against the river bed, bank or cliff.
layers (e.g. chalk) • Igneous = formed from cooled	Beach	Headlands and Bays	the cliff forming a wave-cut notch.	Attrition	Erosion where particles in rivers or the sea are
<ul> <li>Iava (e.g. granite)</li> <li>Metamorphic = formed under intense heat and pressure</li> </ul>	Spit	Coves Coves 4. The backwash carries the		Hydraulic	Erosion caused by the sheer force of water
(e.g. marble)	Bar	Stacks and Stumps	rubble towards the sea forming a wave-cut platform. 5. The process repeats and the	Solution	Erosion where some rock minerals slowly
Erosion	Tombolo	Wave cut platforms	cliff continues to retreat.	:	
<ul> <li>Inere are four types of erosional processes that occur in both coastal and</li> </ul>		Strategies to Redu	uce Coastal Erosion	Transportation	The movement of sediment by rivers or waves.
river landscapes. <ul> <li>Abrasion</li> <li>Attrition</li> <li>Hydraulic Action</li> <li>Solution</li> </ul>	Stratogy Type	Stratogy	Advantages and Disadvantages	Traction	Transport of sediment through a rolling action along the river bed or sea floor.
	Hard	Sea Wall – concrete walls	+ Effective at stopping the seg	Saltation	Transport of sediment being bounced along the river bed or sea floor.
Weathering         • There are three main types of weathering processes that can affect rocks.         • Biological Weathering (e.g. plant roots)         • Chemical Weathering (e.g. acid rain)         • Mechanical/Physical Weathering (e.g. freeze-thaw)	Engineering	built at the top of a beach	- Very expensive to build and maintain	Suspension	Transport of sediment carried within the water flow.
	Hard Engineering	d <b>Rip Rap</b> – large boulders piled at top of a beach	<ul> <li>+ Relatively cheap and easy to maintain</li> <li>- Restrict access to beach</li> <li>- Do not fit in with local geology</li> </ul>	Solution	Transport of sediment particles that have been dissolved in the water.
	Hard Engineering	<b>Groynes</b> – wooden or rock structures built along the beach at right angles	+ Quick to construct + Trap sediment and widens the beach reducing wave energy - Stopping movement of sediment can affect elsewhere on coast	Longshore Drift	The movement of material along a beach transported by wave action.
				Weathering	The breakdown and decay of rock by natural processes acting on rocks, on cliffs and valley sides.
	Cott	Beach Nourishment –	- Can be ugly + Can absorb more wave energy	Mass Movement	The movement of material down a slope due to gravity.
There are four main ways that sediment is transported in river and coastal environments. Traction Saltation Suspension	Engineering	adding sediment to a beach	+ Easy and cheap to maintain - Needs constant maintenance	Deposition	A process where sediments are dropped by the river or waves that carried them.
	Soft Engineering	Sand Dune Regeneration – grasses and bushes are planted to stabilise dunes	+ Maintains a natural coastal environment - Areas of beach have to be fenced off	Hard Engineering	Strategies using artificial structures (e.g. concrete) to prevent river or coastal flooding
<ul> <li>Solution</li> <li>Longshore drift is also a method of transportation in coastal environments.</li> </ul>				Soft Engineering	Flood defences that work with natural processes to reduce the risk of river or coastal flooding.



# Geography – Topic 1 Changing Physical Landscapes

#### Formation of a Spit

- Prevailing wind blows at an angle to the coastline.
- Waves come in at an angle, resulting in 2. lonashore drift.
- Longshore drift moves sediment along the 3. coastline.
- A spit is formed when the material is deposited. 4.
- Over time, the spit grows and can develop a 5. hook if wind direction changes further out.
- 6. Where the spit meets a river, or where there is faster water, the spit stops forming as deposition no longer occurs.
- There is a sheltered area behind the spit. Silt (fine sediment) is deposited and mud flats or salt marshes form.



Formation of Caves, Arches, Stacks and Stumps

- 1. Caves occur when waves force their way into cracks in the cliff face. The water contains sand and other materials that grind away at the rock until the cracks become a cave. Hydraulic action is the predominant process.
- 2. If the cave is formed in a headland, it may eventually break through to the other side forming an arch.
- The arch will gradually become bigger until it 3. can no longer support the top of the arch. When the arch collapses, it leaves the headland on one side and a stack (a tall column of rock) on the other.
- The stack will be attacked at the base in the same way that a wave-cut notch is formed. This weakens the structure and it will eventually collapse to form a stump.

How does a river channel change as it travels downstream?

- The Bradshaw Model is used by geographers to describe how the characteristics vary between the upper course and lower course of a river.
- Key characteristics are that width, depth, velocity and discharge all increase further downstream whereas particle size and gradient of the channel decrease further downstream.

**Causes of River Flooding** 

- Intense rainfall
- Long duration of rainfall
- Impermeable rocks



Urbanisation

#### Strategies to Reduce River Flooding

Strategy Type	Strategy	Advantages and Disadvantages
Hard Engineering	<b>Levees</b> – high banks on/near riverbanks	+ Stop water spreading into areas where it could be problematic - Can burst under pressure
Hard Engineering	Channelisation – deepening or straightening the river	+ Allows water to run through channel more quickly - Water taken downstream may put other places at risk
Soft Engineering	Washlands – areas on floodplain allowed to flood	+ Give a safe place for floodwater to go - May limit the use of the land (e.g. for recreation)
Soft Engineering	Flood-plain Zoning – allocate areas to different uses	+ Prevents using high risk zones for businesses and housing - May cause accessibility issues for the public

#### Formation of a Meander

As the river makes its way to the middle course, it gains more water and therefore more energy. Lateral erosion starts to widen the river. When the river flows over flatter land they develop large bends called meanders.

- 1. As a river goes around a bend, most of the water is pushed towards the outside. This causes increased speed and therefore increased erosion (through hydraulic action and abrasion).
- 2. The lateral erosion on the outside bend causes undercutting of the bank to form a river cliff.
- 3. Water on the inner bend is slower, causing the water to slow down and deposit the eroded material, creating a gentle slope.
- 4. The build-up of deposited sediment is known as a slip-off slope (or sometimes river beach).



## Formation of an Oxbow Lake

- 1. Due to erosion on the outside of a bend and deposition on the inside, the shape of a meander will change over a period of time.
- 2. Erosion narrows the neck of the land within the meander and as the process continues, the meanders move closer together.
- 3. When there is a very high discharge (usually during a flood), the river cuts across the neck, taking a new, straighter and shorter route.
- 4. Deposition will occur to cut off the original meander, leaving a horseshoe-shaped oxbow lake.

## Formation of a Waterfall

Waterfalls form when there are horizontal bands of resistant rock (hard rock) positioned over exposed, less resistant rock (soft rock).

- 1. The soft rock is eroded quicker than the hard rock and this creates a step.
- 2. As erosion continues, the hard rock is undercut forming an overhang.
- 3. Abrasion and hydraulic action erode to create a plunge pool.
- 4. Over time this gets bigger, increasing the size of the overhang until the hard rock is no longer supported and it collapses.

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- 5. This process continues and the waterfall retreats upstream.
- 6. A steep-sided valley is left where the waterfall once was. This is called a gorge.



Downstream

Discharge

Upstream

# **SWB** Geography – Topic 2 Weather Hazards and Climate Change

## **Global Atmospheric Circulation**

Three atmospheric cells (Hadley, Ferrell, Polar) in which heat circulates (moves) alobally.

- Hadley cells: warm, moist air rises at equator creating rainforests. Tropical air flows north and south creating areas of desert.
- Ferrel cells: air sinks over deserts creating dry conditions (warm deserts).
- Polar cells: air sinks over deserts creating dry conditions (cold deserts).

## Movements of air

- Rising air = low pressure = moist conditions.
- Sinking air = high pressure = dry conditions.
- Ocean currents transfer heat energy across the globe

# **Global Climate Change**

We live in the quaternary period (the last 2 million years).

## The climate is changing naturally:

- Eccentricity/orbit: Our orbit changes shape every 100,000 years. Circular orbit = interalacial period (warmer). When the Spherical = alacial period (colder)
- Precession: the earth wobbles on its axis creating warmer summers/colder winters as it wobbles towards/away from the sun.
- Large volcanic eruptions can block out solar radiation and cause glacial periods.

## Humans are causing climate change.

- Transport: cars become more affordable, people's disposable income increases. Burning fossil fuels release greenhouse gases.
- Industry: more disposable income means more goods need to be made by factories. More fossil fuels are burnt.

# **Negatives of Climate Change**

- Sea level rise: melting ice in Antarctica adds to the amount of water in oceans.
- Flooding: money lost as tourism reduces. Beaches close, coastal businesses close. Eq: The Maldives.
- Coral reefs: are bleaching which affects animal habitats.
- Food production: will be lower in some areas. Can cause malnutrition in developing countries.

#### **UK Climate**

- Climate is temperate (mild temperatures, steady rainfall).
- Temperature is warmer in the south of England (nearer the equator).
- Precipitation is higher in the north and west of the UK because of the mountainous relief.
- The UK has aone through 2 major changes: medieval warm period (the year 1,000) and the little ice age (the year 1,700).

## The UK's climate is affected by its location:

- Maritime influence: the UK is • surrounded by sea, meaning air over the UK is moist.
- Altitude: if air has to rise over mountain. • it has to drop water as rainfall first.
- North Atlantic Drift: this warm ocean current from Mexico is driven by the prevailing wind. This makes our winter colder than expected.

Keyword		Definition	
7	Altitude	The height of the land from sea level	
     	Bleach(ing)	When the water is too warm, coral (below) loses its colour and is more likely to die	
	Circulates	Continuously moves	
	Coral	A hard stony material underwater in warm locations that is living	
	Disposable income	The money people have to spend on what they choose, not what they need	
     	Fossil fuels	Coal, oil, gas. Are non renewable	
_i _i	Glacial	Colder periods of time when ice spreads on the land	
Greenhouse gases		Gases such as carbon dioxide which cause global warming	
	Interglacial	Warmer periods of time where there ice is melting on the land	
	Malnutrition	A serious lack of food	
	Maritime	Describes anything related or connected to the sea	
   	Orbit	The movement of the earth around the sun	
	Precipitation	Rainfall from clouds	
     	Solarradiation	Energy that the sun gives out	
	Tourism	Travel for pleasure. For example, holidays	
   	Transfer	The movement of one thing to another <b>124</b>	













Geography – Topic 2 Weather	Hazards and Climate Chanae	Keyword	Definition
	-,,	Arid	Little or no rain, dry
<ul> <li>Iropical Cyclones</li> <li>Large rotating storms that start over oceans.</li> <li>Features: eve (center calm) Eve wall</li> </ul>	<ul> <li>Causes of Drought</li> <li>Arid areas are based on dry climatic conditions</li> <li>whereas drought is an extreme weather condition</li> </ul>	Climatic	The weather over a long period of time
(heavy clouds).	<ul> <li>Causes of drought can be meteorological, hydrological or human.</li> </ul>	Coriolis	A force that causes rotation
<ul> <li>High temperatures cause air to rise over oceans.</li> <li>This evaporation of the ocean creates heavy rain clouds.</li> <li>Cool air sinks towards the ocean surface</li> </ul>	<ul> <li>Meteorological:</li> <li>A lack of precipitation (rainfall).</li> <li>Climate change can affect and change patterns of rainfall.</li> <li>Hydrological:</li> </ul>	Cyclone	A storm created by the weather. Also called hurricanes (around America) and typhoons (around south east Asia)
<ul> <li>which is then re – heated.</li> <li>The Coriolis effect causes the rapidly rising evaporation</li> </ul>	<ul> <li>A lack of water stored. If water isn't stored and kept for when a country is in drought, more</li> </ul>	Dams	A barrier built on rivers to hold back water
Storms need:	Human:	Deforestation	The cutting down of trees
<ul> <li>Winds at the surface of the ocean.</li> <li>30 degrees north and south from the</li> </ul>	Deforestation: means tree roots can't absorb     water from underground, nor can the water be	Endangered	At serious risk of being lost forever (extinct)
equator.	evaporated through transpiration.	Evaporation	The sun heating up water and it turning into a gas (water vapour)
Impacts of Tropical Storms	<ul> <li>Impacts of Drought</li> <li>Social: people become ill due to drinking poor</li> </ul>	Hydrological	Water and the way it moves around the land
<ul> <li>Intense rainfall – can cause flooding.</li> <li>Storm surges – these tall waves can destroy coastal areas.</li> <li>Coastal flooding – can affect farming and tourism.</li> </ul>	<ul> <li>quality water. Migration as people move away from affected areas.</li> <li>Economic: unemployment rises as businesses close. Food prices increase as there is a lack of food.</li> <li>Environmental: river habitats are destroyed. Endangered species become extinct.</li> <li>The impacts are worse in developing countries because:</li> <li>They don't have water stored.</li> <li>There aren't enough hospitals to help the ill people. The government can't predict when the drought will start.</li> <li>People are too poor so can't afford the higher</li> </ul>	Meteorological	The weather high in the atmosphere
		Migration	Permanent movement of people or animals from one place to another
<ul> <li>Ine Impacts are worse in developing countries</li> <li>because:</li> <li>They are slower to respond.</li> </ul>		Rotating	Spinning
<ul> <li>There are no well trained response teams.</li> <li>Buildings are weak and easily collapse.</li> <li>The government can't track the tropical storm so they don't know when it will hit.</li> <li>People don't own vehicles so find it difficult to evacuate.</li> </ul>		Storm Surge	A rising of the sea and waves at approximately 3m higher than usual
		Transpiration	Water evaporated from leaves on plants/trees
└		Unemployment	The loss of a job



# SWB Geography – Topic 3 Ecosystems. Biodiversity & Management

#### The World's Ecosystems

- Biomes are large ecosystems spread across the world.
- Each biome has a different climate and type of vegetation.

#### **Examples of Biomes**

- Tropical Rainforest: hot & wet all year, tall trees.
  - Deserts: very hot all year, very dry all year, plants have deep routes to find water.
- Tundra: freezing temperatures for most of the year, low precipitation. Very few plants grow.

## The Biosphere (The Earth's Surface)

- Humans use the biosphere for water, for fossil fuels and minerals like gold, silver and metal.
- Humans have exploited the biosphere.

#### The Nutrient Cycle

- Nutrients move around the biosphere.
- Nutrients move between soil, biomass and litter.



- As animals die the nutrients fall into the litter store. As the animal body decomposes the nutrients move into the soil. As plants take nutrients from the soil the nutrients move back into the biomass (it's a cycle).
- The climate affects how quickly this cycle happens.

## **UK Ecosystems**

- UK terrestrial ecosystems are: woodland (deciduous trees), moorland (upland areas), wetlands (areas around rivers and lakes), heathland (sandy soil, can't be farmed).
- UK marine ecosystems are used for tourism and leisure activities.

## The Benefits of UK Ecosystems

Bring in £3bn into the economy and provide 200,000+ jobs.

## Marine Ecosystems – Human Exploitation

- Chemicals used by farmers are being washed into rivers/lakes/seas.
- Large windfarms change bird migration routes.



- High temperatures & rainfall = high biodiversity (lots of plant and animal species).
- Trees arow tall but have shallow routes because only the top layer of the soil is fertile.
- Nutrient cycle: biomass is the biggest store of nutrients. Small litter store because of decomposition.

#### Tropical Rainforests – Importance

- They provide goods: timber, oxygen, medicine.
- They provide services: store carbon dioxide, provide animal habitats, bring in tourists.

#### Adaptations To The Rainforest

- Buttress roots think roots that spread across the soil to give tallest trees stability.
- Poison dart frog its poison skin helps protect it from predators.

#### Tropical Rainforests Are Chanaina

- Climate change: creating drier conditions which is increasing the risk of forest fires.
- Climate change: drier conditions are reducing biodiversity and causing animals to migrate away.

#### Deforestation

- Trees are cut down for social and economic reasons.
- Social: population is increasing which means we need more space for housing and farming (growing food).
- Economic: more money can be made from farming animals, trees need to be removed so we humans can extract minerals.

#### **Deforestation Affects Animals**

- Habitats are destroyed which might cause extinction of plant and animal species.
- Food chains are affected which means the number of animals is affected.



















# Geography – Topic 3 Ecosystems. Biodiversity & Management





Swa Geography – Iopic 4 Chang	ging Cifies – Mumbai, India	Keyword	Definition
(Major City in an Emerging	Country)	British Colony	An area of land, or a country, that was owned by Britain
Birmingham's Location History	<ul> <li>Migration</li> <li>Young people move into the city for work or for</li> </ul>	Brownfield site	An area of land that has been built on before
<ul> <li>1700s went from a small market town to creating industries in jewellery and guns.</li> </ul>	Birmingham).	CBD	Central business district (the centre of the city, full of businesses and offices)
<ul> <li>1830s – canals and railways were built.</li> <li>This connected Birmingham to the</li> </ul>	India + Pakistan to work in transport and steel.	Connections	Where two or more things meet
country.	<ul> <li>2000 – migrants came from eastern Europe to</li> <li>work</li> </ul>	Decentralisation	Shops moving out of the city center (CBD)
<ul> <li>Was an industrial city – secondary industry.</li> </ul>		Deindustrialisation	Factories and businesses moving abroad
Today • CBD – is redeveloped. Expensive land.	<ul> <li>Industry in Birmingham</li> <li>Secondary industries are now in decline.</li> <li>Eactories are moving abroad because of cheaper (1)</li> </ul>	Deprived	An area without basic things, eg good schools, good housing
Lots of offices, theatres, hotels and shops • Housing – tower blocks and terraced housing in the inner city. Larger semi detached housing in the suburbs.	<ul> <li>A raciones die moving abload because of cheaper wages, land and better global transport.</li> <li>This has created unemployment and brownfield sites.</li> </ul>	Housing	Terraced – rows of smaller houses Semi detached – 2 houses joined together Detached – houses that aren't joined together Apartments – expensive flats
<ul> <li>Birmingham - Changes in the 1900s</li> <li>*Urbanisation: happened in the 1800s</li> <li>because of the jobs in the city.</li> <li>1920s/30s</li> <li>Suburbanisation happened.</li> <li>Semi detached homes built with greenery.</li> <li>1970s - 1990</li> <li>Counter urbanisation happened.</li> <li>People left the city for a quieter lifestyle in</li> </ul>	<ul> <li>Areas are deprived because of deindustrialisation. Eg: South West of the city.</li> <li>Sparkbrook: deprived area, high unemployment, poor housing, health issues.</li> <li>Sutton Coldfield: wealthy area.</li> </ul> Retail (Shopping) in Birmingham <ul> <li>1970s: decentralisation. Shops moved to the suburbs for space and cheaper land.</li> <li>1980s onwards: CBD is redeveloped. The Mailbox has expensive shops and restaurants in</li> </ul>	Industry	A type of business
		International	From one country to another
		Mailbox	A building in Birmingham that was turned from a Royal Mail building into expensive apartments and shops/restaurants
		Migrants	People who have moved to a new area, usually looking for work, for at least 6 months
rural areas. People could afford cars so they could		Re – urbanisation	The movement of people back into the city center
commute back into the city for work.	The Bullring shopping center is built in 2003 to	Redeveloped	When something is improved
<ul> <li>Re – urbanisation happened.</li> <li>New apartments were built in the city center.</li> <li>Old factories were changed into apartments.</li> </ul>	attract people back into the city.	Secondary Industry	Jobs that were in manufacturing/making things using materials, eg turning steel into guns.
	<ul> <li>The city is recycling 30% of waste. This is increasing.</li> <li>Bus + cycle lanes are being built to reduce</li> </ul>	Suburbanisation	The movement of people into the suburbs (edge of the city)
<ul> <li>Population</li> <li>1.1m people live in Birmingham.</li> </ul>	greenhouse emissions. • More schools are being built to improve education.	Sustainable	Improving something for the future, eg a city
<ul> <li>Birmingnam is ethnically alverse. 42% from non – white backgrounds.</li> </ul>	Homes are built more energy efficient with     insulation	Urbanisation	The movement of people into cities <b>129</b>

🛟 🐝 Geography – Topic 5 Glo	bal Development	Keyword	Definition
academy		Capita	Person
<ul> <li>Development – an Infroduction</li> <li>About improving people's quality of life.</li> <li>Examples:</li> </ul>	<ul> <li>Measuring Development</li> <li>Gross Domestic Produce (GDP): the total value of what is made in a country. Ea: if 2 cars are made and they cost</li> </ul>	Colonialism	Taking over another country and taking its resources (gold, diamonds, oil etc)
Levels of literacy improve because teachers     are trained.	<ul> <li>£20,000 each. GDP = £40,000.</li> <li>GDP per capita: divide the GDP by the population.</li> </ul>	Corruption	People using their power for themselves
Rural areas get electricity because there are more power lines.	<ul> <li>Human development index is a score between 1 – 0. 1 = most</li> <li>developed. It looks at:</li> </ul>	Composite	Made up of more than 1 thing
Economic – average income, the	<ul> <li>Life expectancy</li> <li>Liferacy rate</li> <li>Gross national income</li> </ul>	Demography	Population and its birth/death rate
<ul> <li>Social – life expectancy, literacy rate.</li> <li>Technological – % of people with internet access, % of people with electricity.</li> </ul>	<ul> <li>Single indicators are not as reliable as composite indicators. Composite indicator = HDI.</li> <li>Political corruption: the corruption index grades countries from </li> </ul>	De – industrialisation	Factories leaving to go abroad
Food Security	highly correct to very clean.	Development (gap)	The difference between rich and poor
800 million people live without enough food because:	<ul> <li>The Brandt Line – shows the developed north and the developing south. Now outdated, Made in 1980s.</li> <li>Countries have variation in development. Urban areas are more developed than rural areas.</li> <li>Gini coefficient: shows income inequality in countries.</li> <li>Reasons why there is a gap in development:</li> <li>Physical environment: landlocked countries can't trade, some countries have natural disasters.</li> <li>Demography: a lower birth rate = economic growth.</li> <li>Politics: open economies (eg UK) encourage foreign investment. Closed economies (eg Russia) don't.</li> <li>History: European countries exploited the world's resources through colonialism</li> </ul>	Domestic	Produced in that country. The home country
<ul> <li>people need it.</li> <li>People don't have the money to buy it.</li> </ul>		Exploited	Taken advantage of
<ul> <li>People in developed countries are consuming (eating) too much.</li> <li>Water Security</li> <li>People don't have access to clean water because:</li> <li>Water is being wasted in the developed world for industry and food production.</li> </ul>		Industry	Businesses and/or factories
		Landlocked	Surrounded by land/other countries
		Life expectancy	The age you are predicted to live until
Water in oceans, rivers, lakes and reservoir     is being polluted.		Literacy rate	The amount of people that can read + write
	Social investment: countries that invest in education and health develop fastest.	Rural	Countryside, open green fields
 		Undernourished	Without proper food
<ul> <li>The UK: has a north Vs south divide. London has the high per house. The south east has the highest average hous Why:</li> </ul>	<ul> <li>ine impact of uneven Development</li> <li>30% of the world's population live in slums</li> <li>6.6m children under 5 die every year</li> <li>775m poople cannot read or write</li> </ul>	Unemployment rate	The number of people without a job
<ul> <li>Mountainous areas in northern Scotland, Southern Eng flat (easy to build).</li> </ul>	and is quite 805m people are undernourished 111111111111111111111111111111111111	Urban	City
De – industrialisation happened in the north as factorie abroad for cheap labour.	smoved	Variation	Differences 130



# SWB Geography – Topic 6 Resource Management

#### **Types of Resources**

- Abiotic: found from things that can't reproduce. Eg: soil, water.
- Biotic: found from things that can reproduce. Eg animals and plants.
- Non renewable: resources that either cannot be remade or would take millions of years to make again. Eg: fossil fuels.
- Renewable: resources than can be used again and again or re-created in a short amount of time. Eg: wind, solar, hydro electric power.

#### The Location of Natural Resources Around the world

- Gold and diamonds are found near volcanoes.
- Fossil fuels are found where there is sedimentary rock.
- Oil is found in countries such as America, Iraq, United Arab Emirates.

#### In the UK

- Iron and coal helped Britain in the industrial revolution but there is little left in Britain.
- Oil and gas are found in the north sea but these supplies are running out.

#### Problems in the UK

- Rainfall is higher in the north and west of the UK but this isn't where people live.
- The water is in the wrong place. The supply of water cannot meet the demand.

#### Consumption of Resources Around the world

- People are using more resources everywhere in the world, but the biggest increase is in Asia.
- America is eating up the most calories. This leads to undernourishment in other countries.

#### Why are we using more resources?

- People are richer and can afford more cars.
- The population is increasing.
- People are buying more technology.

All of this means we need to build/make more which takes up resources.

#### Where do we get our electricity from?

- In the past, we have always burnt fossil fuels (non renewable energy sources).
- Recently, we have started to use renewable energy sources.
- All energy sources have positives & negatives

# The Energy Mix (where countries get there electricity from)

- 80% of the world's energy comes from non renewable sources.
- 60% of the UK's energy comes from non renewable sources.
- Factors that affect a country's energy mix: size of population, wealth of the country, what energy resources are nearby.

# Coal – a fossil fuel

## Advantages

- ✓ Should last for another 200 years.
- $\checkmark$  Cheap and easy to mine.
- ✓ Creates large amounts of electricity.

#### Disadvantages

- Releases greenhouse gases, polluting the air.
- Destroys animal habitats.

#### Wind energy – renewable energy Advantages

- ✓ Does not pollute greenhouse gases.
- ✓ Creates cheap electricity for customers.

#### Disadvantages

- Can ruin the look of the landscape.
- Doesn't work when it isn't windy.
- Can be expensive to build.

Keyword	Definition	
Calories	Energy that comes from eating food	
Expensive Costs a lot of money		
Fossil fuels	Coal, oil, natural gas. All non renewable	
Greenhouse gases	Harmful gases that are released into the air. For example when burning fossil fuels	
Habitats	Places where something lives	
Hydro electric power	Electricity that is made from water powering machines	
Industrial Happened in the 1700s where people have revolution jobs in factories		
Landscape	The land, the environment. Eg fields	
Population	The total number of people in an area	
Reproduce	Make more of	
Sedimentary rock	Layers of rock that have fossils (dead plants/animals) in them	
Undernrouish ment	People that do not have enough food to eat	

























# SWB Geography – Topic 6 Resource Management

## Nuclear Power Plants

#### Advantages

- $\checkmark$  Produce electricity all year round.
- ✓ Produces huge amounts of electricity.
- $\checkmark\,$  Produce less carbon dioxide than fossil fuels.

# Disadvantages

- Expensive to build.
- Can be dangerous if they explode.

# Fracking Natural Gas

- This is a new way of finding natural gas.
- Involves drilling down to shale rock.
- Blasting water, sand and chemicals into the rock.
- This breaks the rock and allows the gas to come out.

# Advantages

- ✓ Produces large amounts of gas.
- Has made natural gas cheaper to use/buy.
   Produces less greenhouse gases than coal.

# Produces less Disadvantages

- Can be dangerous as gas enters the water supply and can come through kitchen taps.
- The use of chemicals can damage animal habitats.

## Managing & Protecting Our Energy Individuals can:

- Measure their carbon footprint.
  Use public transport like buses, not private

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transport like cars.Insulate homes and use solar panels.

## Companies can:

- Restaurants can re use cooking oil as fuel for lorries.
- Use LED lights which use less electricity.

# Governments can:

- Use smart meters in homes.
- Build renewable energy sources.
- Improve renewable energy sources by improving technology.

## Case Study: China, an emerging country

- 70% of its energy comes from burning coal.
- 7 cities in China are classed as 'heavily polluted'.

# Why does China have pollution problems?

- More people are owning cars.
- China's roads are heavily congested.
- Poor quality petrol causes air pollution.

## What has China done to solve these problems?

- 2006: China introduced the 'China Renewable Energy Law'. Which aims to minimise the use of coal and use more nuclear and wind power.
- China built the 'Three Gorges Dam' to create hydro electric power.
- Solar farms (rows of solar panels) have been built in the Gobi desert.

# Case Study: Germany, a developed country

- Germany wants to use less nuclear power because of the Fukushima disaster in Japan, 2012.
- By the end of 2020 Germany plans to reduce greenhouse emissions by 40%, and by 80% in 2050.

# How will it meet these targets?

- Germany has built the Bavaria solar park with 60,000 panels. This will reduce carbon emissions by 100,000 tonnes.
- Germany has invested in new wind turbines in the north sea, called Norsee Ost farm. This will increase how much energy is created.





Keyword	Definition	
Carbon dioxide	A gas that is given off when burning fossil fuels	
Carbon footprint	The amount of carbon dioxide each person emits (gives out) because of their lifestyle. Eg, a person who drives will have a larger carbon footprint than someone who walks	
Congested	Heavy traffic	
Dam	A barrier built on rivers to hold back water	
Emissions	Gases released into the air	
Fossil fuels	Coal, oil, natural gas. All non renewable	
Fukushima	An area of Japan where a nuclear power plants exploded because of a powerful earthquake	
Greenhouse Harmful gases that are released into the c gases For example when burning fossil fuels		
Hydro electric powerElectricity that is made from water pow machines		
Invested	Money spent in a new area to improve something	
LED lights	The best lights for the environment	
Minimise	Reduce, make less	
Nuclear	Electricity created by using atoms	
Shalerock	A type of rock where natural gas can be found	
Smart meters	A small computer that tells you how much electricity you are using each hour	
Wind turbines Large wheel that is turned by wind which creates electricity		









# **SWB** Geography – Topic 8 UK Challenges

#### The UK's Population

• Is increasing. 2020: 67m. 2050: 77m people.

#### This increase will put pressure on resources:

- More housing means more greenfield sites are needed to be built on.
- More food will mean natural habitats are destroyed to make space for farms.



 More energy might mean more greenhouse gases if fossil fuels are burnt.

#### To solve these problems:

- Use brownfield sites for building.
- Encourage people to limit their food intake and have a healthy lifestyle.



• Use sustainable, renewable energy sources for electricity.

## Transport in the UK

their cars at home.

 Global transport releases emits huge amounts of greenhouse gases.

The UK is trying to reduce greenhouse gases:

Improve public transport so people leave

- London has created cycle routes and Boris bikes for people to rent for the day or longer.
- Congestion charges: charging people for driving through city centers has reduced traffic and pollution.

#### **Greenfield Sites** Advantages:

- Cheaper and quicker to build on.
- Environment is usually cleaner and more pleasant to look at.

#### Disadvantages:

- Valuable farms or open spaces are lost.
- Animal habitats are destroyed.



# Advantages:

- Less countryside is lost.
- Old dis used urban areas are cleaned up.

#### Disadvantages:

- Often more expensive because old buildings need to be destroyed/rebuilt.
- Higher levels of pollution in these areas.

## The UK's Economy

The 2 speed economy: London and the south east's economy grows faster than the north of England.

#### Why is this?

- More government money is invested in the south to attract more businesses.
- The big businesses in London attract the brightest people from the north of England. These people move south to work.

How is the government trying to stop the north Vs south divide?

- A high speed railway (HS2) is being built from London to northern cities. This will create jobs in the north.
- Airports in the north are improving to encourage travel and businesses to set up in the north.

#### Miaration

- Immigrants: mainly from India + China.
- Emigrants: mainly leave to Australia.
- Reasons for migration: better jobs, better education, family reasons or to retire.

#### Advantages of migration nationally:

- Brings more workers so more taxes are paid.
- Different cultures bring different music and

## food.

#### Disadvantages of migration locally:

- More people means we need more hospitals, schools and doctors. This costs money.
- Some people think migrants take jobs from British people.





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# **SWB** Geography – Topic 8 UK Challenges

# UK's Landscape Challenaes

## How can we protect national parks?

- Employ young people to work in the park so they grow up seeing its importance.
- Encourage public transport into the national park.

#### The UK is at risk from river flooding because:

- The population is increasing so more people have to live nearer rivers.
- More urban areas means more impermeable surfaces.
- More extreme rainfall because of climate change.

#### The UK government is planning to:

- Reduce the number of buildings near rivers.
- Build flood defences hard/soft engineering.
- Help people prepare through warnings.

#### Specific example: Somerset floods, 2013/14.

- Heavy rainfall flooded 600 homes.
- Flood defences failed to protect people.
- Rivers were blocked with silt which meant rivers couldn't hold much water.

## UK's Coastal Challenaes

- Storm surges cause severe flooding because of the wave height and wind power.
- Storm surges can destroy some coastal defences.

# Reducing coastal flooding in the UK:

- Build sea walls to reduce flooding and erosion.
- Use managed retreat in certain areas.

## Specific example: 2013, across east England.

- The Thames flood barrier protected London . from flooding.
- Flood barriers were quickly built in Norfolk.
- Advice was passed out through social media which meant that 800,000 homes were protected.

#### UK's Climate Change Challenges The UK will become:

- Warmer in winter and summer by 3-5dearees.
- 30% wetter in winter, 30% drier in summer Extreme rainfall will be more severe and
- often.

#### Climate change will impact the UK:

- More flooding near rivers and the coast.
- More drought and less water in rivers may affect animal habitats and numbers of species.
- More heatwaves could mean more illnesses like heatstroke. This puts pressure on the NHS.

#### Responding to climate change Individual people can:

- Walk or cycle to reduce areenhouse emissions.
- Recycle waste to reduce resource consumption.
- Build solar panels and insulate homes to conserve heat.

#### Governments can:

- Place limits on carbon emissions from businesses.
- Create adverts which encourage recycling.
- Sign agreements like the Paris agreement to work with other governments.

## Problems aovernments face:

- Encouraging economic growth and development usually means using more resources.
- Reducing climate change can damage economic arowth.





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# Patronaae

During Elizabethan times power came from the top down by a system of **patronage**. God had chosen the king or, in Elizabeth's case, the queen and as such she had the right to appoint whoever she wished to help her rule the country. She usually chose from the **nobility** and could take their power away if they offended her.



# How did Elizabeth control Parliament?

- Parliament could only meet if Elizabeth called it, and she set the agenda.
- Parliament could only talk about what Elizabeth allowed it to discuss.
- Elizabeth mainly used Parliament to grant her taxes - this was her main income.
- Elizabeth could close (prorogue) Parliament at any time.
- Parliament only met 13 times in her 45 year reign for an average of three weeks per year.
- Elizabeth made sure her **Privy Councillors** sat in Parliament to help control proceedings.
- Both Walsingham and Cecil used Parliament to introduce key issues such as marriage, the succession. and religion. Elizabeth fought back by stating that these were issues for her to decide just like any other monarch.



# Year 10 History - KT1: Elizabeth and Government

How did Elizabeth use her power?

# Elizabeth's Secretaries of State

- 1. William Cecil
- Elizabeth appointed Cecil as the **Secretary of State** in 1558.
- He was her most important **minister** and guided her wisely for 40 years.
- Cecil was a moderate Protestant and was concerned about the threat of English Catholics.
- Whilst he shared her caution as he did not want to see England embroiled in expensive foreign wars he was very much his own man.
- He was clear on the fact that he was an advisor to the aueen and as such he would challenge her by using parliament and her courtiers to encourage her to change her mind about various issues.
- After the execution of Mary, Queen of Scots in 1587 Elizabeth refused to see Cecil. She was angry with him and blamed him

# 2. Sir Francis Walsingham

- He was in charge of Elizabeth's secret service and advised on foreign affairs.
- Walsingham, like Cecil was also worried about the Catholic threat
- Walsingham was a cold and calculating man and whilst Elizabeth admired his straight talking nature she could be angered by him.
- In the 1580s when Walsingham wanted to repress the Catholics she was angry with him and even threw a slipper at him!
- In 1586, he uncovered the plot that led to Mary, Queen of Scots' execution.
- When Walsingham died Cecil took back the role of Secretary of State.

# **Privy Council**

Elizabeth surrounded herself with trusted advisers, a group called the **Privy Council**. Privy councillors could be dismissed by Elizabeth if they displeased her.

# **Role of the Privy Council**



# How did Elizabeth control her Privy Council?

- She limited the council to 19 members and of these only eight or nine met regularly
- She appointed councillors with different viewpoints
- She used flattery and played the councillors off against each other
- She used rewards and the threat of removing those rewards in order to maintain control
- She also discussed policies and the business of state with courtiers who were not members of the Privy Council
- In this way her reign differed from that of her predecessors where factions or groups at court had caused trouble for the monarch

# Rebellion of the Earl of Essex

# Causes of the rebellion

- Elizabeth promoted Robert Cecil as **Secretary of State** in 1596 as his father, William Cecil was too ill
- Essex was angry she had not chosen him
- Cecil and Essex each began to build up rival factions at court. This was something Elizabeth had sought to avoid for most of her reign.
- Essex was sent to Ireland to defeat a rebellion, but instead made peace with the rebels against Elizabeth's orders. Upon his return from Ireland Essex broke into Elizabeth's bedchamber hoping to explain his actions which alarmed and angered her.
- As punishment he was banned from court and financially ruined.



# The Rebellion

- Rather than accept his punishment Essex turned to desperate measures and set out to seize power.
- He gathered his allies such as the Duke of Southampton, another **noble** who was out of favour and heavily in debt.
- Key Catholics and Puritans also joined him along with former soldiers who were impressed by his military skill.
- Essex wanted to get rid of the influence of Cecil at court and to make Elizabeth declare James VI of Scotland as her successor.
- The rebellion failed as the Privy Council heard of the impending attack and offered to be merciful to those who had initially supported Essex as long as they deserted him.
- Essex was executed for treason on 25 February 1601.

# Elizabeth and her people

Elizabeth set up two new roles to tackle the problems in England. Lord Lieutenant - one for every county. A wealthy **noble** who acted as a direct link to the queen and **Privy Council**. He was expected to inform the Privy Council of any potential local unrest. The Lord Lieutenant was also responsible for providing part time soldiers to serve the queen should she need it.

**Justices of the Peace (JPs)** - 40 in each county. These were unpaid officials selected by the queen to oversee law and order. It was a great honour to be picked. They performed the following duties:

# How did Elizabeth control her people?

- Summer progresses allowed Elizabeth to be seen by her subjects and also to keep an eye on her nobles in the provinces. Nobles were expected to provide lavish entertainment for the queen.
- Elizabeth also used the Church to control her people. Church was compulsory and you could be fined for not attending. There was even a prayer for the 'Queen's Majesty' that was included at each service.
- Each year a service of thanksgiving was held on Ascension day. At this service people had to thank God for their wise and **Protestant** queen and they had to pray to keep her free from the threat of the **Catholics**.
- Elizabeth controlled printing and as there were only 60 printing presses in the country it was fairly easy for the Privy Council to censor publications.
- Elizabeth's speeches were published to show how much she loved her people.
- Elizabeth and her ministers also controlled the portraits that were painted of the queen. She was made to look as beautiful and ageless as possible in keeping with the 'Gloriana' myth.



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	KEY TERMS		
Patronage	Favours given in exchange for loyalty		
Democracy	A system of government where people vote for the leader		
Parliament	Those who run the country alongside the monarch		
Nobles	A person of high rank		
Gentry	Rich, highly regarded people, below the nobility		
Sermon	A religious speech, usually given in Church		
Jousting	A medieval sporting contest in which two opponents on horseback, typically knights, fight with lances.		
Privy Council	A body of advisors appointed by the monarch		
Secretary of State	The head of Elizabeth's government		
Rebellion	an act of armed resistance against the monarch/leader		
House of Lords	one of the two parts of the UK parliament, whose members are not elected but have a high social position		
House of Commons	one of the two parts of parliament in the UK, whose members are each elected to represent a particular official area of the country		
Puritan	An extreme Protestant		
Monopolies	When someone has exclusive rights to produce something – reduces competition		
Pageant	a public entertainment consisting of a procession of people in elaborate, colourful costumes		



# The Religious Settlement

- Elizabeth sought a 'middle way' between Catholics and Protestants.
- Elizabethan Religious Settlement of 1559 was deliberately vague and gave little mention of liturgy (the way religion should be practised).
- As long as the people recognised Elizabeth's supremacy (authority) over the Church and at least followed the new prayer book they were safe.
- It held the same structure as the Catholic Church in that bishops retained responsibility for the organisation, administration and supervision of the church and its clergy.

# Year 10 History – KT2: Catholic Threats

# Why were there so few Catholics in England by 1603?



# <u>Catholic Plots</u>

Date	Plot	Elizabeth's action
1583 - The Throckmorton Plot	A young Catholic man, Francis Throckmorton, organised a plan for a French army to invade England and replace Elizabeth with Mary, Queen of Scots, paid for by the Pope and King Philip II of Spain.	Throckmorton was executed and Mary was moved to Tutbury Castle in Staffordshire, where she was held in isolation and allowed no visitors.
1586 - The Babington Plot	Sir Anthony Babington planned to rescue Mary, Queen of Scots from jail and murder Elizabeth. Secret letters between the plotters and Mary were discovered which gave the evidence needed to prove Mary's guilt.	This finally led to the execution of Mary, Babington and six other plotters.

# **Increased Threat**

- From 1568 Mary, Queen of Scots was imprisoned in England. She had been overthrown by the **nobles** of Scotland and had escaped to England. **Catholics** saw Mary as the rightful queen of England.
- In 1570 the Pope produced a **Papal Bull of Excommunication** that said that Elizabeth was excommunicated (thrown out) of the Catholic Church and he ordered Catholics not to obey her. This meant that by the 1580s Elizabeth was under threat from the Catholic Church.
- From the mid 1570s newly trained Catholic priests began arriving in England and from 1580 onwards the Pope sent specially trained priests called Jesuits to aid them.
- In 1571 new Treason Acts were passed which made it an offence to deny Elizabeth was the queen of England.
- In 1581 Parliament passed a new law against Catholics. Recusants (those who refused to attend church) had to pay a bigger fine of £20 per month and those who tried to encourage people to become Catholic could be accused of treason.
- In 1584 William of Orange, the leader of the Dutch Protestants was murdered by a Catholic. Parliament responded by passing the Bond of Association. This stated that if Elizabeth was murdered, Parliament would make sure that the murderers were punished along with anyone who had benefitted from Elizabeth's death.
- In 1585 war broke out between England and Spain. Parliament ordered all Catholic priests to leave the country within 40 days. If they were found in England after that time they would be executed.





# Why was Mary Queen of Scots a threat?

- Mary, Queen of Scots was a threat to Elizabeth's rule because she had two claims to the English throne:
- Many people believed Elizabeth to be **illegitimate** and so felt she had no right to be on the throne. (Her father, Henry VIII, had divorced his first wife. Catholics didn't recognise divorce and so viewed his second marriage to Elizabeth's mother as illegal.)
- Elizabeth had converted England's official religion to **Protestantism**, leaving many Catholics disgruntled. Mary was a Catholic and many viewed her as their figurehead and a rightful replacement to the throne.

# Mary Queen of Scots

# What should Elizabeth do with Mary?

- Getting Mary out of the country sending Mary back to Scotland or France could backfire, allowing her to gather more support to challenge Elizabeth.
- Keeping Mary imprisoned this would reduce the risk of Mary gathering more support back in Scotland or France and allowed Elizabeth to keep tight control. But unlawfully imprisoning Mary could provoke international outcry and attacks anyway.
- **Executing Mary** whilst this would remove the figurehead for **Catholic** challengers, it could also provoke international attacks and set a precedent for killing a queen.
- In the short term Elizabeth decided to keep Mary as a closely guarded royal 'guest', moving her from castle to castle so she would not be the focus of plots.



# Consequences of Mary's death

- As it turned out the consequences were not as serious as Elizabeth had feared:
- France fearing the might of Spain, they wanted to maintain their alliance with England.
- Spain were already at war with England due to the action of the 'Sea Dogs' and events in the Netherlands.
- Scotland King James VI was on the throne of Scotland, he was Elizabeth's heir and so he took no action.
- **English Catholics** remained loyal to Elizabeth.

# <u>Rivalry with Spain</u>

Issue	Explanation	
Religious differences	Spain was a Catholic country and England a Protestant country – meaning that the two rulers had conflicting spiritual outlooks.	
Marriage rejection	King Philip of Spain had been married to Elizabeth's sister, Mary I. When Mary died he offered to marry Elizabeth but she rejected him.	
Piracy	English Sailors like Hawkins and Drake attacked and stole treasure from Spanish ships in the New World. King Philip was furious but Elizabeth encouraged and rewarded adventurers.	
The civil war in France	France was the traditional enemy of both England and Spain, meaning that they united together against the country. Now France was in civil war it was preoccupied with its own issues and no longer posed a threat – so the alliance between Spain and England was not necessary anymore.	
Spain supported Catholic plots	There was evidence of Spanish support for plots to restore Catholicism to England, particularly involving getting Mary, Queen of Scots on the throne and Elizabeth off.	
The Dutch Revolt	Protestants in the Netherlands began a revolt against Spanish rule in 1572. Elizabeth secretly supported the Dutch rebels because she knew the Dutch revolt would keep the Spanish too busy to threaten England.	
Elizabeth's army joined the Dutch rebels	Elizabeth sent an army to help the Dutch rebels fight Spain. For the first time English and Spanish armies were fighting each other. <b>England and Spain were now at war</b> .	

# <u>Spain</u>

France at civil war

Religious differences

Piracy

Spain supports plots against Elizabeth

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# Failure of the Armada

Factor	Explanation
Leaders	The Duke of Medina Sidonia led the Spanish fleet, but he was inexperienced in naval battle and so made some fatal errors in his planning and tactics.
Planning	The strength of the Spanish fleet came from its crescent formation plan – but when the English broke this up with their fireships, the Spanish became vulnerable and exposed to attack.
No reinforce ments	The Spanish plan relied on stopping to pick up the Duke of Parma's army to boost their numbers, but the fleet was unable to anchor and so never picked them up.
Tactics	Spanish tactics were to get close enough to English ships to board them, whereas the English tactic was to attack from a safe distance.
Ships	Spanish ships were slower and less equipped for the bad weather than the English ships.
Guns	The English ships had more long-range guns (497 to the Spanish's 172) able to fire at a safe distance. The design of the Spanish gun carriages meant that they could not be re-loaded quickly.
Support	The Spanish overestimated the level of support there would be in England for Spanish control and a return to Roman Catholicism.
Weather	The lack of a secure port where the Spanish could take shelter meant that the Spanish ships were buffeted by the wind. The thinking was that God intervened and the windy weather was a sign that God was on Elizabeth's side.

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# Spanish Armada Events

- The Armada leaves Lisbon on 28 May 1588. It's delayed by storms and repairs, and finally sets sail for England on 21 July.
- The Armada is sighted by the English from Lizard Point in Cornwall. Warning beacons are lit along the English coastline.
- The Armada continues through the English Channel now chased by English ships, but suffers little damage.
- The Armada anchors near Calais where more troops are meant to join, led by the Duke of Parma, but the English send burning fire ships into the fleet. The Spanish ships panic and are scattered out of formation.
- The Spanish ships are blown towards dangerous sandbanks and the English attack again in the Battle of Gravelines. This time they battle at close range and significant damage is inflicted to the Spanish fleet.
- The Armada is forced north around the east coast of Britain and the English fleet turn back after food and ammunition supplies run low.
- The Armada sets sail for home but are forced around Scotland and Ireland. Many ships are wrecked in storms and thousands of sailors drown.
- The surviving Spanish ships arrive back in Spain, but almost half of their fleet is lost.
- Victory for England.



5. Strong storms/wind push the Spanish fleet towards Gravelines where 'The Battle of Gravelines' takes place. The English inflict havy damage here. 6. The wind/ storm forces the Spanish fleet north and the English turn back. 7. The Armada sails home but has to go around Scotland and Ireland - and many ships are wrecked. 8. The Spanish fleet arrives back home bu most ships are lost.

	KEY TERMS	
Recusant	Catholics who refused to attend Protestant services	
Church Papist	Someone who attended Protestant Church but remained a Catholic in private	
Conformer	Someone who dropped their Catholic beliefs and became Protestant	
Gentry	Rich landowning men in Tudor England	
Mass	The most important Catholic church service	
Treason	A crime against the King, Queen or Country	
Excommunicate	To ban someone from the Catholic religion	
Seminary Priest	Priests from abroad who came to support Catholics	
Jesuit Priest	Priests from abroad who came to convert Protestants to Catholics	
Priest hole	A place priests hid from the authorities	
Traitor	A person who betrays his King, Queen or Country	
Convert	To change to something different e.g. religion	
Rivalry	Dislike/competition between people	
Armada	A large fleet of ships from Spain 142	

# The Labourina Poor

Half of the population in Elizabethan England were the labouring poor.

- The houses of the labouring poor were very simple one storey dwellings. They did not have chimneys. There were few windows and they did not have glass in them. A labourer would have very few possessions perhaps a bed a table and a chair.
- Work most of the labouring poor worked in the countryside on the farms of yeomen and husbandmen. Although farmers needed lots of labour at harvest time for the rest of the year labourers could struggle to find work. They often found it difficult to pay their rent and buy food or fuel.
- Land some labourers might be lucky enough to have an acre or two of land and they might enjoy common rights (this would allow them to araze their animals on the common land). Two thirds of labourers had only their cottages and gardens. Food - barley and rye bread was the staple diet of the labouring poor. It was supplemented by a soup called pottage made from vegetables grown in the labourer's aardens.
- During good times labourers might have eggs, cheese, fish or bacon but during hard times labourers struggled to put food on the table.





# Year 10 History – KT3: Daily Life What mattered to the Elizabethans?

# The Middling Sort

The term 'middling sort' was used by Elizabethans to describe those people who had more than the poor but who had not reached the dizzy heights of the gentry.

- The houses of the yeomen were much bigger than those of the labourers who worked for them. They might have chimneys, glass in the windows and a number of rooms (including two stories) thereby denoting their relative wealth and status.
- Work in the town the 'middling sort' was made up by the tradesmen and craftsmen who ran their own businesses. In the countryside the 'middling sort' was the **yeoman** and the **husbandmen** who farmed some land of their own. Although they did not have the vast amounts of money of the gentry they lived a more comfortable existence than the labouring poor.
- Food Whilst yeomen families may have eaten meat it would be simple such as beef, pork and mutton.
- Bread was an important part of their diet but unlike the gentry, who ate white bread regularly, the 'middling sort' would save this for cakes and pastry for guests. 'Yeoman' bread had more bran left in it.
- They would drink beer and mead instead of wine.
- Like the gentry they were able to grow fruit and vegetables in their gardens and orchards but they did not have exotic fruit.

# The Gentry

The gentry were landowners and during this period they began to build and extend their homes to reflect their status. They had the following:

- Multiple storied houses with lots of alass in the windows.
- Orchards and kitchen gardens these would provide the family with fruit and vegetables
- Lavish food a feast in a gentleman's house would consist of a variety of different types of meat including venison and swan and these would be accompanied by fine wines imported from France and Italy. After the meal there might be a sweet course consisting of sweetmeats and marzipan confections.
- Land About two percent of the population were gentlemen and they owned over half the land in England at the time. The rents from their estates meant that the gentry did not have to work and they were free to enjoy leisure pursuits such as hunting.
- Political power they acted as Justices of the Peace and helped the monarch to maintain law and order in the provinces. Some of them were Members of Parliament and helped the queen to govern the country.



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# **Husbands and Wives**

- In Elizabethan England all marriages were recorded in the parish register of the church where the marriage took place.
- Men were usually in their late-twenties and women were usually in their mid-twenties when they married.
- Most newly married couples needed to save up before they could get married in order to set up their own home. Those from gentry families tended to marry earlier as money was less of an issue.
- Parish registers also reveal that many babies were baptised a few months after their parents married which suggests that they were conceived prior to their decision to get married.
- There were few **illegitimate** children as single mothers would be brought before the church courts and punished.
- In wealthy families, parents would expect to have a say over who their children married although this did not mean that they arranged marriages.
- In 'middling' families children would often wish to gain the blessing of their parents before they married as parents often gave sizeable gifts of money, land and furniture to their children. In most cases it would seem that children from 'middling' and labouring families were free to marry whom they wished.
- Marriages rarely ended in divorce as this needed an Act of Parliament. Marriages did break down and this could lead to separation.
- More frequently marriages ended due to the death of the husband or wife in which case the surviving spouse would normally remarry

# How were women treated?

Despite being ruled by a queen Elizabethan England was a patriarchal society which meant that men were in control. However this did not mean that women were second-class citizens. Many women had some freedoms and certainly those from the 'middling' sort helped run farms and workshops. Although domestic violence did occur it was punished harshly. Wife beating was disapproved off and equally a nagging housewife could be accused of being a scold.

# **Families**

# Parents and Children

- The nobility and the gentry could afford to have large families
- High infant mortality especially in the poor.
- In fact, around one quarter of children died before they were ten.
- Most Elizabethan women looked after their children and it was only in the wealthiest families that people used wet nurses (a woman who would breast feed and care for someone else's child) to care for their children.
- Children whose parents had money might be sent to school from the age of seven.
- In poorer families, children would work at home or on the farm from the age of seven. At the age of twelve or thirteen boys would leave home to become apprentices or to work as farm servants.
- Girls often left to become servants at the same age. Most young people lived with another family to learn the skills they would need in adult life.
- Physical punishment was used more readily, particularly in the grammar schools. At home there does not seem to be a great deal of evidence that Elizabethan parents were violent towards their children.

# <u>Kinship</u>

- There is little evidence to suggest that people lived with their wider family and most Elizabethans lived in a nuclear family with parents and children.
  There were occasions where wider family members were taken in as a result of them being unable to care for themselves but this was not the norm.
- Wider kin rarely lived in the same village.
- The practice of sending children away to be apprentices often resulted in them marrying and settling down away from home. Thus Elizabethan families were scattered although perhaps not as far as families today.
- Kinship did matter to some, particularly the gentry and sometimes those of the 'middling' sort, for most people neighbours were as important.
- Studies of wills have shown that people were far more concerned about their own immediate family than their wider kin.













WILL

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Why Poverty Increased

# <u>Poverty</u>

Reason	Explanation		
Rising population	The population rose by a million during the Elizabethan period. More people meant there was more demand for goods, and so prices rose.		
Inflation	rices for goods rose, but wages fell as there were more people around to do the ork.		
Cloth trade collapse	Woollen cloth was England's main export. There was a decline in demand and this led to unemployment.		
Wars	Taxes increased to compensate for the price of waging war.		
Bad harvests	Harvests were particularly bad in the 1590s leading to even higher demand and more rising prices.		
Changes in farming	Many landlords decided to enclose their fields and keep sheep instead of crops which led to unemployment.		

## What actions were taken – national level

They brought in:

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- A compulsory nationwide Poor Rate system
- everyone had to contribute and those who refused would go to jail
- begging was banned and anyone caught was whipped and sent back to their place of birth
  - **almshouses** were established for people in genuine need to live in

# Why bring in the Poor Law 1601?

- Fears that the 'social order' might be threatened if the growing number of poor people ganged together and started a rebellion
- The risk that **vagabonds** and beggars might turn to crime
- Fears that the poor might spread disease
- Landowners were giving more to the poor

# <u>What actions were taken – local level</u>

Unpaid local officials (called Justices of the Peace) were made responsible for poverty related issues. For much of the century the authorities grouped people into three groups:

- **Impotent poor** people unable to work due to age, disability or other infirmity. Limited relief was provided by the community in which they lived.
- **Able-bodied poor** these were people who were physically able to work but were often unemployed. The Poor Law provided no relief for these people except employment in the workhouse.
- **Vagabonds** it was thought vagabonds were lazy, idle and threatened the established social order. Anyone over 18 who became a vagrant having been caught before could be hanged. In 1589 the government went so far as to ban individuals from sheltering people in their own homes.





	KEY TERMS		
Gable	The triangular upper part of a wall at the end of a ridged roof		
Coat of Arms	A special design in the form of a shield that they use as a symbol of their identity.		
Orchard	A piece of land with fruit trees		
Great Chamber	The second most important room in a Tudor house		
Estate	An extensive area of land in the country, usually with a large house, owned by one person, family, or organization.		
Merchant	A person involved in trading goods		
Husbandmen	A farmer		
Yeomen	Someone who owned a small piece of land		
Parlour	The sitting room in a Tudor house		
Labourer	Someone who does unskilled manual labour		
Contraception	Something used to prevent pregnancy		
Wet Nurse	A woman employed to breastfeed another woman's child.		
Kin	Wider family		
Vagrant	A person without a settled home or regular work who wanders from place to place and lives by begging		
Impotent Poor	Physically unable to work due to illness or age		
Able Bodied Poor	Those were able to work but could not find a job		
Vagabond	Those who avoided work		
Gaol	Prison 146		

## A 'golden age' of culture

- For the wealthy at court, there were banquets and feasts, <u>masques</u> and tournaments (competitions of sword fighting, wrestling or jousting) to enjoy almost every day of the week.
- The rest of society visited fairs and celebrated religious festivals.
- Dancing was very popular formal dances like the <u>galliard</u> at court, and <u>jigs</u> among the poor and middling sort.
- Watching animals fight was also popular with rich and poor. Queen Elizabeth herself enjoyed bear and bullbaiting (where the bull or bear was pitted against a pack of dogs). Among the poor, dogfights and cock fighting were common pastimes, and people often gambled on the outcome.
- Gambling at cards, dice games, backgammon or draughts was popular among men of all social groups it was considered inappropriate for women to gamble (although that didn't stop Elizabeth!)
- Among the nobility, the invention of the printing press and spread of education meant that intellectual pursuits were popular, such as reading the classics, studying music, debating theology and writing poetry
- Hunting and <u>hawking</u> were also very popular among the well-to-do, as were physical pastimes such as <u>bowls</u>, fencing and archery.
- For the rest of society, inns and taverns were an important part of every social ritual. People could drink, gamble, play cards or <u>skittles</u> and watch plays and animal baiting
- The poor and middling sort enjoyed physical games as well, such as wrestling and stick fighting. An early form of football was also played in Elizabethan times. I



# Year 10 History – KT4: Merry England

What lay behind changes in popular culture?



#### Rural past times

- Around 85 per cent of people lived in the countryside and many people never left their villages - and others only went as far as the next town.
- Everyone had to work extremely hard for a living. This meant that they would relish any chance they got for a celebration.

Celebrations were largely tied to the calendar and the religious calendar marked:

- Christmas
- Easter
- Shrove Tuesday
- Lammas Day
- Saint's days
- The farming calendar was another big influence. Lords of the manor would sometimes hold feasts at harvest time, and there was usually a celebration on Plough Monday (just after Christmas, when the ploughing began). A big event in the year was May Day, which was celebrated with dancing, fairs or markets, and the selection of a village girl to be Queen of the May.
- Although there were no theatres, travelling entertainers would visit the villages - especially on holidays - and perform plays, sing songs, perform acrobatics and tell stories.

There would also be 'Mummers' plays', which were simple plays or mimes on popular themes, such as:

- two men competing for the same woman
- folk tales, such as Robin Hood
- religious stories, such as St George and the Dragon

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<ul> <li><u>Puritans</u></li> <li>The main opponents of popular entertainment were the <u>Puritans</u>. The Puritans were extreme Protestants who wanted to 'purify' the English church of any Catholic influence.</li> <li>They also believed that God intended that society should be a single unit - so any bad behaviour or immorality affected everyone.</li> <li>The Puritans wanted people to live devout and godly lives, and put God first. T</li> <li>hey opposed anything that distracted people from God - and that, of</li> </ul>	<ul> <li><u>Witchcraft</u></li> <li>Dark magic was the domain of witches – people who had supernatural powers because they had made a pact with the Devil.</li> <li>Witchcraft was blamed for all sorts of bad fortune</li> <li>They believed that witches had 'familiars' - an animal they could turn into so they could spy on people and move around without being seen. The most popular familiars were cats and toads, which is why witches are often depicted with these animals.</li> </ul>
<ul> <li>course, included most forms of fun.</li> <li>They wanted people to spend their time praying, listening to sermons and reading the Bible.</li> <li>They disapproved of gambling, holidays, dancing, and popular songs and most of all, they disapproved of the theatre.</li> <li>The theatre attracted huge crowds - sometimes up to 3000 people - and those people were not at work</li> <li>Puritans thought the things that playwrights wrote about were unsuitable and disapproved of the rude jokes</li> <li>Also, the Puritans worried about the less desirable types who went to the theatres: gamblers, thieves, prostitutes and beggars</li> <li>To appease the Puritans, Elizabeth banned theatres within the London city boundary. However that didn't stop several large playhouses such as the Globe, being built just outside London, within easy reach of the public.</li> </ul>	<ul> <li>The majority of people who were accused of witchcraft were women - usually poor, single women who had nobody to protect them.</li> <li>Many people also believed that women were spiritually 'weaker' than men, and therefore easier for the Devil to win over.</li> <li>Illustrations from the time show us what people imagined 'witches Sabbaths' to look like. They usually involve potions bubbling in cauldrons, witches flying on broomsticks or on the backs of goats and sometimes worshipping the Devil.</li> <li>Once someone was accused of witchcraft, they were tested, usually by 'ordeal'. One of the common ways was to 'duck' the accused under water. They believed that water was purifying, so if the person floated, it was because the water had rejected them, and they were guilty. If they sank, they were innocent. Of course, this meant that many 'innocent' people drowned!</li> <li>The Witchcraft Act of 1563, which made witchcraft a felony. This meant that once a 'witch' was found she would be killed.</li> </ul>

Opposition to Past times

## <u>Authorties</u>

- Elizabeth too, although she loved the theatre, was not prepared to risk her opponents using it to spread their ideas.
- As a result the Queen appointed the Master of the Revels to <u>censor</u> plays for <u>blasphemy</u> or dubious politics.
- The Lord Mayor of London and his councillors were responsible for law and order and they were worried about the effect of so many people in one place.

They thought that the crowds and the theatre could cause problems like:

- Disorder as many people drank at the theatre and fights and quarrels were common
- Crime pickpockets and thieves were attracted to large crowds
- Idleness they thought people should be at work instead
- Because the theatres were outside London, the Lord Mayor couldn't deal with them himself he had to get the <u>Privy Council</u> to agree with him, and lots of Privy Councillors liked the theatre!
- The authorities also worried that the large crowds could spread disease and here the government agreed with them. The theatres were often closed during outbreaks of the Plague to stop the disease spreading.



<u>Theatre</u>		<u><u> </u></u>	eatures of a theatre
What was the theatre like in Elizabethan England? Plays were performed: • at court for the nobility • in inpugged for the page	No curtain to show the the play	e start and end of	Plays often had a dramatic start (a shipwreck, an argument) to grab the audience's attention. They finished with music and dancing.
<ul> <li>in purpose-built theatres with tiered seating, which were open to all</li> <li>At the theatre, the rich could sit in the best seats, with cushions, and the poor</li> </ul>	No lighting to show nig	iht and day	The words of the play had to tell the audience what time of day it was.
<ul> <li>could stand in front of the stage for a penny. The atmosphere in Elizabethan theatres was very different to how the atmosphere is in theatres today:</li> <li>The audience would wander about, talk among themselves, shout at and heckle the actors (and throw things if they were really unimpressed).</li> </ul>	Female roles were played by men		Female characters wore gorgeous costumes, and the other characters talked about how beautiful they were.
<ul> <li>There was no curtain, and the audience often surrounded the stage on three sides.</li> <li>There were no intervals, so vendors would come round during the play selling food and drink</li> </ul>	There were no interval performed without a b	s, so the play was oreak	There were side plots and comic scenes to give the main actors a break and a chance to change their costumes.
<ul> <li>Women were not allowed to be actors. Female parts were played by teenage boys.</li> </ul>			Gallery above the slage - oxed for actions or monitoria
Why were theatres popular?	83	P	

- Elizabeth encouraged the theatre, which made it fashionable.
- It was an affordable entertainment (costing as little as a penny for a two hour play).
- The theatre was a good place to socialise.
- The plays dealt with topics that appealed to Elizabethan audiences: love and romance, magic, patriotism, exploration and travel - and often had dirty jokes and fight scenes to keep people entertained.





	KEY TERMS			
Golden Age	The period when a specified art or activity is at its peak.			
Orpharian	A musical instrument			
Popular culture	Types of activities/preferences that are popular at the time			
Parish	A small area that usually belongs to a Church			
Merry making	Having fun			
Mumming play	A type of folk play			
Alehouse	A pub			
Festivities	Celebrations			
Sabbath	Sunday			
Familiars	Animals that are supposed to be the pets of witches			
Prosecution	Charging someone with a crime			
Opposition	Going against something			
Miracle Play	A religious play			
Jig	A lively dance			
Cut-purse/pick pocket	Someone one steals			
	150			

## <u>John Dee</u>

- He was the brain behind Elizabethan exploration
- He was interested in maths, astronomy, astrology, medicine and navigation
- He wanted to Britain to rival Spain's advancements and suggested Britain should expand its empire and discover new territory
- He coined the phrase 'British Empire'
- Dee produced maps and books and helped sailors understand navigation to make the mission a success

# Humphrey Gilbert

- He was sent to try and establish the first colony in North America
- He wanted to find a passage around America to China
- He was also driven by hatred of the Spanish
- He explained to Elizabeth that a base in North America would help attack Spanish ships in the New World
- His first attempt was a disaster and only one ship managed to cross the Atlantic
- He then planned to claim vast amounts of land in America and sell it to Englishmen for a huge profit
- 5 ships set sail to achieve this in 1583, he arrived to find English, Spanish and Portuguese fishermen there who celebrated his 'achievement' – putting a flag into a piece of land
- They shared fish, wine, marmalade and biscuits
- However, the land was barren and cold and Gilbert's men became ill – trying to sail South for supplies failed and they attempted to return home
- The ships sank in violent storms along with Gilbert
- No colony was established



# Year 10 History – KT5: Going Global

What did the Elizabethan adventurers achieve?

# Why did exploration occur?

In the 15th century came the first long sea voyages. These were made possible by:

- New navigational instruments like the compass
- Fast, light ships that could be manoeuvred to catch the wind
- The invention of guns



## Francis Drake

- Drake was the first Englishman to circumnavigate the globe – this means go all the way around the world!
- He had made journeys to the Caribbean several times to plunder Spanish ships
- His plan was to sail around South America to plunder Spanish ships but it did not go to plan!

## His Journey:

- In November 1577 he set off and began by plundering Spanish and Portuguese ships in West Africa
- By Spring/Summer of 1578, he arrived in Brazil and entered the Strait of Magellan at the tip of South America, claiming several islands for Elizabeth
- By the winter of 1578, he had travelled north to Chile and Peru, had attacked several native settlements and attacked Spanish ships carrying gold and silver
- In June 1589, he landed on the coast of California and claimed it, calling it New Albion
- By the summer of 1579, Drake realised if he sailed home the same way he would be attacked by Spanish ships waiting for his return. He decided to sail West
- In Moluccas (Spice Islands) he traded linen cloth for cloves, ginger, and pimento
- He then travelled home via the Cape of Good Hope
- He arrived home very rich indeed but most went to his investors and then to the Queen's treasury!
- Drake was knighted aboard his ship 'The Golden Hind'.







#### The Reconnaissance Mission

In 1584, Elizabeth gave Walter Raleigh a charter, allowing him to form a colony in America. Raleigh and Elizabeth hoped that the colony would:

- provide riches from the New World
- act as a base for disrupting Spanish treasure ships

Raleigh sent Philip Amadas and Arthur Barlowe to find a suitable site for a colony.

- They arrived on Roanoke Island in July 1584 and met the native tribes, Barlowe came back to England, bringing two tribesmen, who told Raleigh all about the area.
- Raleigh thought it sounded good and sent a second expedition of five ships, led by Richard Grenville.

#### What happened to everyone at Roanoke?

- It was some time before help could be sent to Roanoke - the weather was too bad to cross in the winter
- Then there were further delays caused by the Spanish Armada and the war with Spain.
- It was three years before they could try again.
- When White returned in August 1590, he found the settlement deserted.
- The 90 men, 17 women and 11 children including his granddaughter, Virginia had vanished.
- There was no sign of a battle.
- The weather was worsening, so White was unable to find out what happened.
- Nobody went back to Roanoke until 1609.

# Walter Raleigh

#### July 1584

Sent by Walter Raleigh, Philip Amadas and Arthur Barlowe arrive on Roanoke Island and meet the native tribes, before returning to England

#### August 1585

Richard Grenville and 107 colonists land and construct a fort on Roanoke Island, before Grenville departs



# 1587

Raleigh sends another group to set up a colony on Chesapeke Bay and asked them to visit Roanoke and check on the settlers - when they arrive, the settlement is deserted

The fleet's commander decides to have the settlers stay on Roanoke to re-establish the colony, and John White returns to England for help

# August 1590 White returns to Roanoke, finding the settlement deserted again. The 90 men, 17 women and 11 children - including his granddaughter, Virginia - had vanished

The Lost Color

#### Establishing a Colony

- Bad weather separated the ships, but eventually a group of settlers made their way to Roanoke.
- In August 1585, 107 colonists landed and constructed a fort on the island.
- Grenville promised to come back in April 1586 with more colonists and supplies, but the reinforcements didn't arrive.
- Many of the colonists returned to England and brought back tobacco and potatoes with them, but some of them stayed behind.
- In 1587, Raleigh sent another group of colonists to set up a colony on Chesapeke Bay and asked them to visit Roanoke and check on the settlers there.
- When they arrived, the settlement was deserted. They found a skeleton, but no sign of anyone.
- Instead of going on to Chesapeke, the fleet's commander made the settlers stay on Roanoke to re-establish the colony.
- It soon became clear that the colony's relations with some of the native tribes had not been friendly. Fearing for their lives, they begged the governor, John White, to go back to England and ask for help.
- 115 colonists stayed on the island including White's grand-daughter, Virginia Dare. She was the first English child to be born in the New World.

The search for the North West Passage was dangerous - and unsuccessful.

- Martin Frobisher (1535-1594) tried to reach China by going round North America. He reached the Arctic and was driven back by snow and ice.
- Humphrey Gilbert (1539-1583) also believed that there was a route around North America. His ship was hit by storms and he never returned.
- John Davis (1550-1605) made three separate voyages looking for a northern route to China. Like Frobisher, he was driven back by the cold weather.



#### <u>Success</u>

- The first person who established trade with the Spice Islands was **James Lancaster.**
- He took the southern route around Africa (capturing Portuguese trading ships along the way) and reached the East Indies, where he bought spices and valuable dyes used in the woollen industry.
- When he came back, the East India Company was set up to attract investors and in 1600, Lancaster took the first of the Company's fleets to modern day Indonesia. He met with local leaders and made alliances and set up trading posts.
- The East India Company grew under later monarchs and became very important and prosperous.







## First Contact with India

- Although the East India Company was set up to trade with South East Asia (the 'East Indies'), it eventually traded more with India and Ceylon (now Sri Lanka).
- One of the first English sailors to visit India was **Ralph Fitch** - a merchant who spent eight years travelling around the Middle East, India and Burma.
- He was one of the first English travellers to visit Mesopotamia and Persia (modern day Iraq and Iran), and he travelled to Aleppo in Syria and Tripoli in Libya with the Levant Company on a ship called the Tyger. From there, he went to India, where he saw 'salt, opium, hinge (asafoetida, a spice), lead, carpets and diverse other commodities' being traded.
- He came back in 1597, and in 1598 he published a book about his travels and mentioned how rich India was. The English merchants were keen to get access to this valuable market.
- The East India Company didn't start trading out of India until 1608, after Elizabeth's death, but the foundations for the trade with India were laid during her reign.
- The first Indian trading point was set up in Surat, in the Gujarat region.
- Ralph Fitch remained a valued advisor to Lancaster and the Company about Indian matters.

	KEY TERMS			
Circumnavigate	To go all the way around – e.g. the world			
Plunder	To steal			
Astrology	The study of the stars and their meanings			
Colony	Land taken and ruled by a foreign power			
Empire	When several countries are ruled by a foreign power			
Navigation	Finding a route - e.g. at sea			
Expedition	A journey to investigate/research			
Territory	A piece of land			
Reconnaissance	Looking around, information gathering			
New World	North and South America			
Global	Relating to the whole world			
Astronomy	The study of the universe and its contents outside of Earth's atmosphere			
Commission	An instruction/command given 154			



# Year 10 – PRE – Christian Beliefs: Part 1 – The Nature of God



#### Key Words

Monotheistic: A religion which believes in one God Holy: Separate and set apart for a special purpose by God

**Omnipotent:** All powerful, Almighty

**Omnibenevolent:** all-loving

Just: Fair

Trinity: One God has 3 parts; God the Father, Son and Holy Spirit

Holy Spirit: God's presence in the world

God the Son: Jesus – enables humans to have a special relationship with God

Creation: God bringing the universe into being

The Word: Jesus - as described in the book of John Genesis: The first book in the Bible which has the creation story in it

Incarnation: God in human form – Jesus.

**Resurrection:** comina back from the dead

Blasphemy: saying or doing something which goes against God

Crucifixion: Roman method of execution where a person is nailed to a cross

**Ascension:** 40 days after the resurrection when Jesus returned to God in heaven

Afterlife: What happens when you die

Day of Judgement: God will judge all souls at the end of time

Heaven: Eternal happiness, being in the presence of God

Hell: Eternal suffering, absence of God

Purgatory: Catholic belief in which souls are cleansed in order to enter heaven

Sin: Any action against God

**Original Sin:** First sin in the world committed by Adam and Eve which means all humans are born with this in them

**Salvation:** saving the soul from sin and going to heaven thanks to Jesus' sacrifice

Grace: A quality of God which shows to humans that God loves them, which they don't need to earn

Forgiveness: pardoning someone for their wrong doing **Atonement:** restoring the relationship between people

and God through the life, death and resurrection of

Jesus

#### God as omnipotent, loving and just

- Christians believe God is omnipotent: all-powerful. 'Nothing is impossible with God' 61

- God is omnibenevolent: all-loving. Guidelines are given for us to live the best lives we can. Christians should love each other treating everyone with care and respect. 'God so loved the world he gave his one and only Son...

- God has unlimited power and authority, together with complete love, and therefore gives justice in a fair way. Christians should try and bring about fairness in the world.

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	Different Christian Beliefs about Creation		
sis 1 ∕ ∕	<ul> <li>God created the world in 6 days and rested on day 7. 'In the beginning God created the heavens and the earth'. God created the perfect world in the beginning: 'It was good'</li> <li>Not all Christians believe this story is literally true, but rather believe it is a story which represents the idea that God created life.</li> <li>A further quote from Genesis: 'The Spirit of God hovered over the waters' indicates that the Holy Spirit was present at creation.</li> </ul>	Combined, these ideas suggest that all 3 parts of the Trinity were present at creation.	- ( sa - Firs Trin - Cre - Om
	<ul> <li>'In the beginning was the Word, and the Word was with God, and the Word was Godthrough him all things were made'.</li> <li>'The Word' refers to Jesus and therefore he was present at the beginning of the world and involved in the creation of the world</li> </ul>		om om knc om (ev

#### The Problem of Evil and Suffering: The Inconsistent Triad

- Christians believe God is omnipotent (all-powerful) and omnibenevolent (all-loving). 4
- However, why is there evil and suffering? -
- If God was all powerful, he would be able to stop evil and suffering.
- If he was all loving, he would care enough to stop it. So he would stop it.
- But he doesn't! So God cannot exist, or is at least not omnipotent or omnibenevolent.
- A Christian defence: God is transcendent (beyond our understanding) and therefore we can trust God to know best - there is a reason for this inconsistency, even if we can't understand it.

The Oneness of God and the Trinity				
<ul> <li>The Holy Trinity: God the Father, the Son and the Holy Spirit.</li> <li>God is three in one. There are not three Gods, but different forms of the same thing. Each part is fully God, but also these parts of the Trinity are not the same.</li> </ul>				
God the Father - First person of the Trinity. - Creator - Omnipotent, omnibenevolent, omniscient (all- knowing) and omnipresent (everywhere).	God the Son - Second person of the Trinity. - Jesus, who became God in human flesh (known as the Incarnation). - Performed miracles, healed the sick and showed kindness to outcasts.	God the Holy Spirit         - Third person of the Trinity.         - God sent the Holy Spirit to influence, guide and sustain all life on earth after Jesus ascended.         - The unseen power of God.		

#### Different Christian Beliefs about the Afterlife

- Based on God's judgement Christians believe that people will go to heaven or hell.
- Judgement will happen at death or at the day of judgement (Christians vary on their understanding of this)
- The Parable of the Sheep and the Goats shows how people will be judged by God. The sheep represent those who did good actions (therefore going to heaven) and the goats represent those who did bad actions (therefore going to hell)
- Jesus also said, "I am the way the truth and the life, no-one comes to the Father except through me." \_



- Heaven is seen as being with God and eternal happiness where there is no suffering. Hell is seen as eternal torment or suffering and being absent from God, and where the Devil is.
- Some Christians believe that Heaven is a literal, real place you will go. Other Christians believe it is just being with God, in the same way hell may not be actually real but an absence of God.
- The Bible teaches that there will be a resurrection of the body for all people who as to heaven, though the details of this are debated.
- In the book of Revelation it mentions that people who go to hell will burn in a lake of fire. -
- Roman Catholics believe in a place called purgatory in which your soul goes to be cleansed as not everyone is ready yet to go to heaven. -





# Christian Beliefs: Part 2 – Jesus Christ and Salvation



	Jesus' Life: Key Events	
Event	Key Details	Importance/ Influence on Christians
Incarnation: Jesus is God in human form	<ul> <li>'Incarnate' mean 'In the flesh' – Jesus was God in the flesh.</li> <li>Jesus' birth is explained in the Christmas story: the Nativity.</li> <li>Humble birth (in a manger) shows Jesus was fully human, but he was born through the immaculate conception (Mary was a virgin), so he was fully God</li> <li>This means Jesus does not have Original Sin because he was not conceived through sexual relations.</li> <li>'Before they came together, she was found to be pregnant through the Holy Spirit'.</li> <li>'The Word became flesh and made his dwelling among us'.'</li> </ul>	<ul> <li>Encourages them to celebrate Christmas for its true meaning</li> <li>Helps them in difficult times as they know their role model was human too</li> <li>Allows them to have a personal relationship with God through Jesus</li> </ul>
Crucifixion: Jesus' death on the cross	<ul> <li>Crucifixion is remembered on Good Friday.</li> <li>Jesus was arrested (having been betrayed by Judas) and put to death by Pontius Pilate. He was crucified alongside two criminals.</li> <li>As Jesus was fully human he suffered pain as an ordinary human did. 'Father, into your hands I command my spirit'</li> <li>On the cross Jesus said 'Father forgive them, for they know not what they do'.</li> </ul>	<ul> <li>By accepting Jesus' sacrifice they can be forgiven for sin and go to heaven.</li> <li>Encourages them to follow Jesus' example and forgive others</li> <li>Reminds them to be thankful and remember Jesus (especially on Good Friday)</li> <li>Reminds them that suffering is a part of life and God can understand what it I like for someone to suffer.</li> </ul>
Resurrection: Jesus rose from the dead	<ul> <li>Jesus was buried in a tomb and left there until Sunday. Due to it being the Sabbath (on the Saturday), no-one could touch the body until after this.</li> <li>Mary Magdalene returned to the tomb - it was open and empty.</li> <li>An angel appeared and said Jesus had risen from the dead.</li> <li>Evidence of resurrection: he appeared to people including disciples, they saw him eat, Thomas was encouraged to tough Jesus' palms to prove he was not a ghost.</li> </ul>	<ul> <li>Shows Jesus was divine and not just a human so it may strengthen faith</li> <li>Shows the power of good over evil</li> <li>Encourages them to not fear death</li> </ul>
Ascension: Jesus went back up to heaven to be with God	<ul> <li>Happened 40 days after the resurrection - Jesus ascended to heaven.</li> <li>He gave the disciples the Great Commission: 'Go and make disciples of all nations, baptizing them in the name of the Father, the Son and the Holy Spirit'.</li> <li>The Holy Spirit was left to guide and comfort people.</li> </ul>	<ul> <li>Will encourage them to call on the Holy Spirit for guidance and comfort</li> <li>Will encourage them to spread the message of Christianity (evangelise)</li> </ul>

God (Holy)

#### Sin and Salvation

- Sin: anything that goes against God's laws. Separates humans from God.
- Salvation means to be saved from sin (therefore being able to go to heaven)
- Humans are not perfect impossible not to sin
- All humans are born with Original Sin passed down from Adam and Eve.
- This action separated humans from God and brought about death into the world. They were tempted by the serpent (devil) and Christians believe that Christians are tempted in life to do bad things.
- Salvation through Law: humans have free will but should use this to make the right choices using God and Jesus' teachings to guide them, e.g. 10 Commandments.
- Salvation through Grace: being saved by accepting the sacrifice Jesus made on the cross – this showed God's grace: 'For by grace you have been saved through faith'
- Salvation through Spirit: having the Holy Spirit as a guide to accept God's Grace and follow his Law.

#### The Role of Christ in Salvation: Atonement

Salvation is offered through Jesus, "For the wages of sin is death, but the gift of God is eternal life in Christ Jesus".

 $\checkmark$ 

- Jesus' death makes up for Original Sin. Humans can receive forgiveness for their sins because of Jesus' death and then receive eternal life.
- His sacrifice provides atonement, which means our relationship with God is restored. This removes the effects of sin and allows humans to get back to God. "He is the atoning sacrifice for our sins and for the sins of the whole world".
- Jesus paid the price for the sin of all mankind through his death and Christians believe if you put your trust in him you can receive eternal life with God.
- Links with Salvation through Grace: salvation is a gift people must choose through belief in Jesus atoning for their sins

# Exam Terminology

Influence: The capacity to have an effect on people's character, behaviour or actions Contrasting: To show a difference Contemporary: Occurring in the present time Sacred Writings: Writing that is believed to contain words of God e.g. The Bible Evaluate: Consideration of different viewpoints before arriving at a final judgement Justified Conclusion: A final decision which is based upon a range of evidence.



# Christian Practices: Part 1 – Worship and Festivals



Worship and Festivals			
Practice and Key Words	Details/ Contrasting Views		Importance and Quotations
Worship: Act of religious honour or devotion	Liturgical         - Takes place in a church and is led by a priest         - Formal, set prayers are read out, and the worship follows a set pattern and structure         - A more traditional, and formal form of worship         - E.g. Eucharist in the Catholic Church	Non-Liturgical         - Also takes place in a church but less formal         - No set prayers, instead people take turns to preach and read from the Bible         - Can be modern and appealing to young people         - Service is usually focused around a Bible reading         - E.g. Methodist or Baptist services	<ul> <li>Shows gratitude, love and respect to God</li> <li>Could be a way of asking for forgiveness or asking for help</li> <li>Brings comfort and strength</li> <li>Gives time for reflection</li> <li>'Sing to the Lord, for he has done glorious</li> </ul>
	<ul> <li>Informal</li> <li>Spontaneous prayers or sharing of thoughts</li> <li>Community or house churches might meet to eat together and share their faith</li> <li>Pentecostal Church – 'charismatic' worship. Led by the Holy Spirit and may involve dancing, clapping, calling out, speaking in tongues.</li> </ul>	Private         - Spending time with God alone or with close friends/ family         - May involve prayer, meditation, studying the Bible.	things; let this be known to all the world .
Prayer: Communicating with God	Informal - Prayers that are made up by the individual using his or her own words.	<ul> <li>Set Prayers</li> <li>Prayers which have been written down and said many times by many people</li> <li>E.g. The Lord's Prayer: The prayer that Jesus taught the disciples to pray, which includes thanks, asking for forgiveness and asking for guidance. "Our father who art in heaven".</li> </ul>	<ul> <li>Set prayers can bring a sense of unity</li> <li>Prayer brings comfort and builds relationship with God</li> <li>'Call on me and come and pray to me, and I will listen to you'.</li> </ul>
Sacraments: An outward sign of inward grace. Eucharist/ Holy Communion	<ul> <li>Roman Catholic Church (Mass)</li> <li>Readings from the Bible</li> <li>Offering of bread and wine brought to the alter</li> <li>Priest says the words of Jesus at the Last Supper, says the Lord's Prayer and gives a sign of peace</li> <li>Congregation come to the alter to receive the communion</li> </ul>	<ul> <li>Orthodox Church (Divine Liturgy)</li> <li>Hymns, prayers, readings from Bible.</li> <li>Priest comes through Royal Doors to chant the gospel.</li> <li>Lord's Prayer said, behind Royal Doors words of Jesus said.</li> <li>Bread divided into four – three consecrated as body and blood and fourth broken into small pieces. Priest gives bread and wine together on a spoon.</li> </ul>	<ul> <li>Jesus started the tradition at The Last Supper (which took place the day before he died)</li> <li>Christians now remember Jesus' death – reminds them of Jesus' sacrifice, and reminds them to forgive others.</li> <li>"This is my body which is for you, do this in remembrance of me"</li> </ul>
Sacraments: An outward sign of inward grace Baptism	<ul> <li>Infant Baptism</li> <li>Everyone is a descendent of Adam and Eve and therefore carries Original Sin. Baptism washes this away.</li> <li>It also welcomes them to the church community.</li> <li>Infant wears white, Godparents are chosen, font holds the water</li> <li>E.g. Catholic, Orthodox</li> </ul>	<ul> <li>Believer's Baptism (Adult Baptism)</li> <li>Some Christians think children are too young to understand the meaning and therefore don't baptise infants.</li> <li>The person is old enough to understand the meaning behind what they are doing.</li> <li>This includes a full immersion in a pool to wash away sin and start a new life in Jesus.</li> <li>This is known as being 'born again'</li> <li>E.g. Baptist and Pentecostal churches</li> </ul>	<ul> <li>Brings a person into the Christian family/ community</li> <li>Water symbolises the washing away of sins</li> <li>Jesus was baptised, setting an example for others to follow</li> <li>Jesus also encouraged baptism in the Great Commission: . "Therefore go and make disciples of many nations, baptising them in the name of the father, son and Holy Spirit".</li> </ul>
Pilgrimage: A special religious journey to a holy site.	<ul> <li>Lourdes (France)</li> <li>Dedicated to Mary as Bernadette believed to have seen visions of Mary in the 19<sup>th</sup> Century.</li> <li>A spring of water was discovered which had healing powers. Now millions of people have been to drink from the spring of water in the hope of being healed.</li> </ul>	<ul> <li>Iona (Island off west coast of Scotland)</li> <li>Small community set up by St. Columba, an Irish missionary in the 6<sup>th</sup> Century</li> <li>Pilgrimages happen there in dedication to the virgin Mary.</li> <li>The community in Iona hold daily services in the Church leading a seven mile hike to holy spots – it is a physical and spiritual challenge</li> </ul>	<ul> <li>Pilgrimage shows commitment to God and strengthens faith</li> <li>People may go on pilgrimage for healing</li> <li>It brings a sense of community</li> </ul>
Festivals: Celebrations for religious reasons	<ul> <li>Christmas <ul> <li>Remembers the birth of Jesus – his incarnation.</li> <li>It is celebrated on the 25<sup>th</sup> December.</li> </ul> </li> <li>Trees and homes are decorated with nativity scenes. Lights remember Jesus is the light of the world. Carol services happen in Churches with readings from the Bible. Children act out nativity plays and midnight mass takes place on Christmas Eve.</li> </ul>	<ul> <li>Easter</li> <li>Most important festival which celebrates Jesus' resurrection from the dead leading up from holy week.</li> <li>Jesus was crucified on Good Friday and rose on Easter Sunday.</li> <li>Special services take place and processions led by someone carrying a cross.</li> <li>On Easter Sunday, sunrise services take place with hymns which celebrate the resurrection. Easter Eggs are used as a reminder of new life. Paschal candle is lit.</li> </ul>	<ul> <li>Festivals celebrate the most important events of Jesus' life – his birth, death and resurrection.</li> <li>They are a time for believers to come together and celebrate their faith.</li> <li>"I bring you glad tidings that today a 157 is born"</li> <li>'Christ is risen from the dead'.</li> </ul>



# Christian Practices: Part 2 – The Role of the Church in the Local and Worldwide Community

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Key Words		The Role of the Church in the Local Con	nmunity		Church Growth	
<b>Church:</b> The holy people of God, the body of Christ or a building where Christians worship <b>Agape:</b> compassionate love <b>Mission:</b> A calling where an individual or group go out and spread the word of God. <b>The Great Commission:</b> Jesus instruction to his followers to go and spread his message "Go and make disciples of many	Food Banks	d Banks       -       Provide free food (usually non-perishable items) to those in need, usually through donations       The street of the s	The work of food banks and street pastors supports the key Christian message to show love to all. Key words and quotes to support: Agape – compassionate love 'Love your neighbour as you love yourself' 'Faith, if not accompanied by action, is dead' 'For I was hungry and you gave me something to eat' (Parable of the Sheep and the Goats)	Church Growth	<ul> <li>Estimated to be 2.5 billion Christians in the world</li> <li>Christians are taught to help to grow the church by sharing testimonies (how God has changed their lives), inviting people to meetings, prayer, social events etc.</li> </ul>	<ul> <li>The Great Commission 'Go and make disciples of all nations, baptising them in the name of the Father, the Son and the Holy Spirit' (Jesus' words before he</li> </ul>
	Street Pastors	<ul> <li>Volunteers who patrol streets in urban areas</li> <li>Do not actively preach but show their faith through their actions</li> <li>Give out flip flops, lollipops and water on nights out, help people to get home safely, offer reassurance and support</li> </ul>		Mission and Evangelism	<ul> <li>Evangelism means to spread the message of Christianity through preaching the Gospel (which means 'Good News')</li> <li>Some do this through Mission work, which means evangelism overseas</li> <li>The aim is to tell people that Jesus is the saviour of the world, in the hope of converting them to Christianity.</li> </ul>	ascended to heaven). - Example: The Alpha Course. Anyone is welcome to join in with a meal and conversation about the 'Big Questions' of Christianity.
nations" Missionary: A person sent	The	Role of the Worldwide Church: Reconciliation	and Persecution		The Role of the Worldwide Church: World	Povertv
Missionary: A person seni on a religious mission to promote Christianity in a different country through preaching or charity work Evangelism: Spreading the Christian message through preaching the Christian gospels Alpha course: An example of evangelism -trying to tell others about Christianity Convert: Someone who has decided to become committed to a religion and change his or her	Working for Reconciliation	<ul> <li>Jesus came to earth to restore the relationship between humans and God, so Christians believe they have a responsibility to restore the relationship between themselves and others.</li> <li>Lots of examples of arguing between faiths, even within Christianity e.g. Catholics and Protestants. Irish Churches Peace Project set up to reconcile these denominations</li> <li>Coventry Cathedral – bombed during WW2. Cathedral has now become a centre for reconciliation as Christians wranted to respond with forgiveness.</li> </ul>	<ul> <li>'Therefore, if you are offering your gift at the altar and there remember that your brother or sister has something against you, leave your gift there in front of the altar. First go and be reconciled to them; then come and offer your gift'.</li> <li>'Blessed are the persecuted because of righteousness, for theirs is the Kingdom of Heaven'.</li> </ul>	Responding to World Poverty	<ul> <li>Jesus emphasised helping the poor so Christians follow his example</li> <li>Jesus told a rich man to sell everything he had and give it to the poor.</li> <li>Christian Aid: They aim to stop poverty, encourage sustainable development and provide emergency relief in areas such as Africa and the Middle East.</li> <li>Their slogan is 'We believe in life before death'</li> <li>They provide emergency food, shelter, water, sanitation and run a Christian Aid Week to fundraise every year.</li> </ul>	<ul> <li>'If anyone has material possessions and see a brother or sister in need but has no pity on them, how can the love of God be in that person?'</li> <li>Parable of the Sheep and the Goats</li> <li>'Go, sell everything you have and give to the poor, and you will have treasure in heaven. Then come, follow me.'</li> </ul>
Reconciliation: A sacrament in the Catholic faith, also making up after		not revenge.	suffers with it' (St Paul, likening members of the Church to different parts of the body)		<u>Exam Terminology</u>	
an argument or disagreement Persecution: hostility or ill- treatment, because of race or religious or political beliefs. Poverty: When people live without having basic human rights such as having enough food, water or shelter	Responding to Persecution	<ul> <li>Persecution (ill-treatment) happens all over the world – Christians are tortured or even killed for their faith</li> <li>Christians have a responsibility to help those who are persecuted</li> <li>They might pray for them, donate to charity or get involved with charities who work abroad.</li> <li>The Barnabus Fund: send financial support, raise awareness for those persecuted, send spiritual and material support.</li> </ul>	'Love your neighbour as you love yourself' Agape: Compassionate Love	Influence: T or actions Contrasting Contempor Sacred Writ Evaluate: C judgement Justified Co evidence.	The capacity to have an effect on people (To show a difference (ary: Occurring in the present time (ings: Writing that is believed to contain w consideration of different viewpoints befor (onclusion: A final decision which is based of	e's character, behaviour ords of God e.g. The Bible re arriving at a final upon a range of <b>158</b>



# Component 1 Learning Aim B: Explore factors that affect growth and development

ACADEMY	Year 10 Child Development KO	FACTORS AFFECTING GROWTH AND DEVELOPMENT		
L		PHYSICAL FACTORS		
	KEY WORDS for FACTORS	BEFORE BIRTH: An effect on the foetus.		
Physical	Growth and other physical changes that happen to our body throughout	Prenatal: based on genetics that are passed on through parents.		
development	life	Genetic abnormalities can be caused by:		
Intellectual	The development of language, memory and thinking skills	Maternal nutrition and exercise		
development		Paternal drug or substance abuse		
Emotional	The ability to cope with our feelings about ourselves and others	Premature/low birth weight		
development		Mothers mental health		
Social	The ability to form friendships and relationships and to learn to be	AFTER BIRTH		
development	independent	Health status: chronic or life limiting illness		
Cognition/	Acquiring knowledge and understanding though thoughts, experiences	Diet and dietary deficiency- e.g. not enough calcium		
Communication	and senses.	Amount of exercise: not encouraged to eat healthy tood, risk of		
Communication	Exchanging information through speaking and writing.			
Language	The method of human communication	ENVIRON/VIENTAL FACTORS		
Genes	Inherited characteristics transferred from parents to children	Housing:		
Abnormalities	An abnormal (something unusual or not normal) feature or characteristic	Housing needs- When the local council look at the amount of homeless		
Chromosomes	A threadlike structure inside most living cells that carry genetic information	households, those living in temporary accommodation, and households		
Foetus	An unborn human.	not big enough for their family.		
Spina bifida	A detect in the spine, which causes a gap in the backbone. This causes			
6	paralysis of lower limbs and sometimes-learning attricuities.	Home environment:		
Substances	An intoxicating and stimulating chemical or drug that causes harm.	Living with parental conflict, experiences of abuse and neglect		
Montal hoalth	A baby born before the full ferm of pregnancy.			
Deprivation	A lack or denial of remething that is peasager, a g feed			
Housing need	Considers the households who do not have access to accommodation	Exposure to narmini substance:		
noosing need	that meets the normal requirement			
Abuse	Treated cruelly or with violence regularly or repeatedly.	Discrimination against the obild		
Neglect	Failure to care for someone.	Based on their race, social or cultural arounds. For example, what religion		
Exploitation	Treating another unfairly to benefit yourself	they follow, or the groups of individuals they associate with.		
Mutilation	Inflict serious damage on.			
Prescription	Medicine that is only given with a doctor's prescription	Income and poverty:		
drugs	(recommendation)	Unemployed or workless families		
Illegal drugs	A drug that is forbidden by the law due to the harm it causes.	Whether they had access to early education services, preschool, nursery.		
Socio economic	Social and economic factors.			
Discrimination	Unjust or prejudice treatment of different groups of people based on their	Poor relationships with significant adults:		
	age, sex and race.	Whether they receive warmth and affection from family		
Social exclusion	Removal from the social system and its rights and privileges due to	How they respond to significant adults offering support, attention.		
	poverty or belonging to a social group.			
Poverty	The state of being extremely poor.	J.		

#### Knowledge Organiser

LAB= Learning Aim B: Explore factors that affect growth and development

Child Development BTEC Technical Award - Component 1

#### Key Words

Impact- the effect or influence on something

Development- an acquisition of skills and/or knowledge

Fine motor skills – small precise movements e.g. holding a pencil Gross motor skills- Using large muscles, balance and coordination Socioeconomic – the combination of social and economic (money) factors Environment - a person's immediate surroundings

#### COMPONENT 1- ASSIGNMENT CRITERIA LAB

P3→ Explain the ways that different factors have affected growth and development in selected case studies P4→ Explain the impact of physical, environmental and social factors on growth and development in selected case studies

M2→ Compare the impact of physical, environmental and social factors on growth and development in selected case studies

D2→ Assess the impact of physical, environmental and social factors on growth and development in selected case studies

# Factors that affect growth and development - PHYSICAL

#### PRENATAL:

Genetics and genetic abnormalities. E.g. Downs syndrome cause by a chromosomal abnormality Substance Abuse Maternal nutrition/Exercise Premature/low birth weight Mother's mental health

#### **KEY QUESTIONS FOR ALL FACTORS:**

 What is the factor → Define and Explain
 How does the factor affect development → Look at positive/negative impact on <u>all</u> areas of development





#### SOCIOECONOMIC (Social and Economic)

Experiences of discrimination (social, racial or cultural)
 Income and poverty (unemployed and workless families, access to good early education experiences)
 Poor relationships with significant adults – level of warmth, affection and attention received



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#### **ENVIRONMENTAL:**

Housing – areas of deprivation or experiencing housing needs Home environment – parental conflict, abuse and neglect Effects of exposure to drugs, alcohol and smoking

> Assignment tips: You need to be able to define and explain each factor. You also need to be able to explain how the factor can affect each area of development!

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# Year 10 Health & Social KO - Component 1 Learning Aim B – Life events and types of support

KEY WORDS for LIFE EVENTS					
Life event	A change in an individual's life, which can cause disruption or positive change				
	to their lifestyle and everyday activities				
Expected	Something th	hat is <b>likely</b> to happen	ø		
Unexpected	Not thought	it is likely to happen	8 1°).*		
Physical events	Making changes to your physical health, <u>body</u> or mobility.				
Relationship	A significant change in relationship status such as 🗼 🔺 🛹				
changes	engagement, marriage, divorce, <u>separation</u> or death.				
Life	Impacts on <u>day to day</u> life and the choices you make.				
Reasoning	The action of thinking about something in a logical and sensible way				
Adapt	To adjust to new conditions or circumstances				
Professional	Describes a member of a profession who is trained and skilled in their area of work				
Transition	The process of changing from one state or condition to another.				
Disposition	An individual's attitude or qualities				
Income	Money received on a regular basis from work.				
Long term illness	An illness that cannot be cured by medicine or treatment.				
Restriction	A limitation of someone or something.				
Responsibility	Being accountable, having control over something or being to blame.				
Chronic illness	Another word for a long term illness.				
Lifestyle	The way in which a person lives				
Grief	Intense sorrow, caused by someone's death				
Mutual understanding	A shared feeling or action, in which both people involved have sympathy for.				
Physical life event		Relationship ch	ange	Life circumstance	
III health Accident and		Bereavement New relationshi Marriage Divorce	ps	Moving house Starting or moving school Exclusion from education	
Injury		Parenthood		Redundancy Retirement	

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Informal support	÷	Informal support is given by anyone who you know <b>outside of a professional capacity.</b> This could be <b>family</b> and <b>friends</b> or those you know well. This will involve the supporter offering security and <b>practical help</b> , through <b>emotional</b> <b>support</b> and <b>information and advice</b> .
Formal support		Formal support is offered by <b>statutory care</b> <b>services</b> provided by the government. It could also be from <b>private care services</b> and <b>charitable organizations.</b>
Emotional support	<b>F</b>	This involves showing <b>empathy</b> , <b>compassion</b> and <b>genuine care</b> for others. This can be via informal, <b>formal</b> or voluntary support.
Practical help	×	This involves helping an individual practically by helping them with finances, childcare or transport issues such as finding alternative transport services.
Information and advice	i	Involves providing the individual with information to improve their life event or circumstance. This will help them to understand where to go for <u>help2</u> What services are available? How <u>that will</u> benefit them?
Voluntary support	***	Working for free and offering support, working alongside those providing <u>informal</u> and <u>formal</u> support. This could involve organisations such as The Princes Trust and Relate.
Occupational therapist	in the second se	Helping ill and disabled people to continue with daily activities and tasks at ease. This will include shopping, making meals and walking upstairs.
Counsellor	<b>١</b>	A person trained to give <b>guidance</b> on personal or <b>psychological</b> problems.
Accident and injury	P	Something that happens unexpectedly at an abrupt state. This means the individual will have to adjust quickly to their new life circumstance.
Social worker	¢ Ĥ	Assists individuals handle everyday life problems who have experienced <b>neglect</b> , <b>abuse</b> , <b>mental</b> <u>health</u> and <b>domestic violence</b> . 161

#### New relationship:

New relationships develop qualities such as trust, <u>patience</u> and empathy. Having a new relationship will teach individuals about their own qualities, and how equal compromise is important. Having a relationship can take time to adjust to, especially if you have been used to independence. Mutual understanding is highly important when forming new relationships with others.

This can involve intimate relationships and friendships.



#### Engagement:

Engagement shows commitment to another individual. On acceptance, mutual understanding between both partners is important as they are planning to commit to each other for their remaining life. Engagement will be exciting for both, as planning for a wedding, house and family can be discussed.

#### Marriage/Civil partnership:

Marriage/Civil partnership is a joining of two people together. This is a very happy time, as both individuals can forge a new life together, make plans and goals they want to achieve. This could involve starting a family. Marriage/Civil partnership involves trust, honesty, mutual understanding, respect and empathy.

Couples will need to adapt to living together, changing their lifestyle to suit both partners equally, provides security and safety. It will also involve sexual intimacy.

#### Moving house:

Moving house is an exciting time, it can involve a couple, friends or just the individual. There are opportunities to meet new people and join a new community. However, there can be apprehension moving away from family and friends and starting in a new community of unknown people. There are also pressures from up keeping mortgage fees and household bills.

#### Parenthood:

Parenthood is an exciting time. You are bringing a new life into the world, which involves responsibility, and can cause anxiety especially if there is a single parent. There will be less time for themselves, and more time focused on looking after their child. A change of lifestyle and routine, can cause lack of sleep, adding pressure to the relationship. However, having a child brings positive emotions, excitement and content.

#### Exclusion from education:

Removing a child from education, could eliminate the issues that caused the exclusion, relieving stress and anxiety. However, moving to a new school or educational provision can cause apprehension about new routines and lack of interaction and socialisation. Missing out on valuable learning, can have a negative impact on intellectual development.



#### Divorce/Separation

According to research, divorce is the second life event after death that has the highest emotional impact on an individual. By having strong family ties, support and stability can be offered during the uneasy time of divorce. However, a break down in relationship causes insecurity, which can lead to low self-esteem. With a loss of wider family networks and friendships, social development can be affected.

A positive outcome of divorce, if both people were unhappy, a fresh start could be needed for them to take a new direction in life.

#### Redundancy

Although this can be devastating, it can be an opportunity for a career change. Losing a job, can have an impact of lifestyle and diet, with a loss of earnings food options and socialising may be limited.

#### Imprisonment:

Being imprisoned offers reflection, for the individual to solve issues to change their life. There are options for learning and developing new skills through voluntary roles within the prison. However, a loss of independence and socialisation, will have a negative impact on emotional and social development.

#### Bereavement:

With an expected death, this can be easier to come to terms with, as emotionally the people around have prepared themselves for death. This doesn't make it easier to get over, as death is a gradual process of coming to terms with. If it is unexpected, this will take longer to come to terms with. It is not about getting over the <u>death</u>, <u>but</u> finding a way to cope with the fact it has happened.









Examine equipment and technology required for participants to use when taking part in sport and physical activity / Prepare participants to take part in physical activity • Learning Aim B & C:

# **DIFFERENT TYPES OF SPORTS CLOTHING AND EQUIPMENT FOR SPORTS PARTICIPATION**

- Clothing
- Footwear
- Sports related equipment
- Protection and safety equipment
- Equipment for people with disabilities
- Facilities  $\checkmark$

**Organiser** 

**Knowledge** 

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<u>Component</u>

- Officiating equipment
- Performance analysis



# **DIFFERENT TYPES OF TECHNOLGY** AND BENEFITS TO IMPROVE **PARTICIPATION AND PERFORMANCE**

- Clothing to increase performance and experience
- Footwear
- Sports specific equipment
- Protection and safety equipment
- Equipment for people with disabilities
- Facilities
- Officiating
- Performance analysis

The limitations of using technology in sport and physical actviity

- Time
- Access to technology
- Cost of technology
- Accuracy of data provided by equipment
- Useability

# **PLANNING A WARM UP**

- $\checkmark$  Types of activity
- ✓ Response of the Cardiovascular system to the pulse raiser
- ✓ Response of the musculoskeletal system
- $\checkmark$  Mobiliser activities and the response of the cardiorespiratory and musculoskeletal system
- ✓ Stretching activities and the response of the cardiorespiratory and musculoskeletal system



## **ADAPTING A WARM UP**

TOPIC

- ✓ Adapting warm ups for different categories of participants
- ✓ Adapting warm ups to specific activity or sport
- $\checkmark$  Delivery of warm ups to prepare participants for physical activity (organisation & demonstration and supporting participants that are taking part)



# Explore types and provision of sport and physical activity for different types of participants

## • Learning Aim A:

Organiser

**Knowledge** 

- ✓ Team & individual sport
- $\checkmark$  The benefits of taking part in sport
- Outdoor adventurous activities and its benefits
- ✓ Benefits of taking part in physical activities
- The provision of sport (public, private and voluntary sectors)
- ✓ Advantages and disadvantages of the sectors



# TYPES & NEEDS OF SPORT & PHYSICAL ACTIVITY PARTICIPANTS

- Types of participants
- Age groups of participants (primary school/ adolescents/adults/older adults)
- Participants with disabilities
- Participants with long term health conditions
- Physical activity needs of participants

# BARRIERS TO PARTICIPATION IN SPORT

- ✓ Barriers to participation
- ✓ Cost of clothing, equipment, transport
- ✓ Access to sport and physical activity
- ✓ Time constraints (family, school, work)
- ✓ Personal barriers
- ✓ Cultural barriers

# METHODS TO ADDRESS BARRIERS TO PARTICIPATION IN SPORT

- Cost (discounted pricing, hiring equipment, free parking)
- Access (transport discounts, free parking, taster days, staff training, ramps, assistive technology)
- Time (creche facilities, extended opening time)
- Personal barriers (private changing facilities, parents and child sessions, campaigns to increase participation)
- Cultural barriers (women only sessions, diverse staff body, staff training in cultural awareness)





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# UNDERSTANDING THE BODY AND THE SUPPORTING TECHNOLOGY FOR SPORT AND ACTIVITY/ Component 1



#### Learning Aim B:

<mark>A</mark>SK

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FLEXIBILITY EXERCISES AND YOGA/PILATES  $\checkmark$ 







Knowledge Organiser omponent



# UNDERSTANDING THE BODY AND THE SUPPORTING TECHNOLOGY FOR SPORT AND ACTIVITY/ Component 1



• Learning Aim C

