

Knowledge Organisers Autumn Term – Year 11

Name: _____

Please remember:

- It is to be kept inside your knowledge organiser book
 - It is to be brought into school every day

Regular retrieval throughout a scheme of learning (daily, weekly and monthly) has been proven to **reduce the rate of forgetting**, supporting you to **retain more** in long term memory- making assessments/ exams way easier! The challenge for you as a student is to make sure you use your knowledge organiser for each subject properly to help you to know more and remember more over time. We've created this walk through to support you in using your knowledge organiser- for more support speak to your subject teachers.

Using your Knowledge Organiser



1	2	3	4	5
Look	Cover	Write	Check	Repeat
Start with a small section of knowledge	Now cover up this section of your	Self quiz- what can you remember and	Remove the post it and check for	After a short break away from your
that you want to remember e.g Henry	knowledge organiser with a post it note	rewrite? Make sure you do this without	accuracy- did you get the key	knowledge organiser repeat the look,
VIII's wives in History. Read through this	or scrap paper.	looking back at your knowledge	terminology? Was it spelt correctly?	cover, write, check until you can recall
section of the knowledge organiser (a		organiser.	Was the order correct? If you drew a	all of the facts correctly without
couple of times if it helps)			diagram, how much of this did you get	prompts.
			correct?	
				This process can be used for any new
			Most importantly- what did you miss	knowledge that you want to acquire. It
			out?	is good idea to do this on a regular
				basis, once a week.

Strategy 1- Look, cover, write, check – A really simple but effective way to use your knowledge organiser. Focus on a specific area of your knowledge organiser.

1	2	3	4	5
Focus	Big ideas	Explain it	Link it	Record it
Make it manageable by selecting an	Pick out the main points or the big	Explain what you know about the main	Now, see how it links to other areas	Write down as many 'think it, link it'
area of your KO <u>where your learning is</u>	ideas in this section.	points (this could be written or shared	within the subject. E.g Eating meat –	ideas as you can in your book. See if
not secure. Don't waste time going off		verbally – a friend, a family member.	causes global warming. Cows produce	you can beat others in you class!
something you can already do!			methane which is a greenhouse gas.	

Strategy 2- Think it, link it – Great for connecting the big ideas in your subject. How does 'x' relate to 'y'. What are the key factors which make an equation/ experiment/ process work? Challenge yourself to see how many links you can make!

1	2	3	4	5
Select topic	Prepare quiz	Answer it	Self check	Repeat
Decide which area you want to be	Get someone else to prepare 10	Set a time limit (depending on the	Now look at your KO to self check-	Return to this section in 2/3 weeks- see
quizzed on (this might build up over	random questions on that topic to	number of questions) and answer the	make a note of your score. Celebrate	if you can improve your score! Re-do
time)	challenge you.	questions without looking at your KO.	your successes and make a note of	those questions that you missed or got
			anything you missed or got incorrect.	incorrect.

Strategy 3- Knowledge quiz – You might try this after a few weeks of using your knowledge organiser. Get someone to set you 10 questions using your knowledge organiser. These could be spellings, key words, processes, equations etc to see how much you can remember! Record your score and see if you can beat your personal best each half term!

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SWB GCSE – English – Spoken Language

Your Task:

Prepare a spoken presentation on a specific topic. The duration of the whole assessment should be around **8 minutes**.





- ✓ Ensure you use Standard English(a),
- \checkmark Look at your audience,
- ✓ Don't just read from your notes,
- Don't have too many words on your power point (if you are using one) and
- Present your ideas on something you are passionate about.

Practise until you believe that you can deliver your speech with confidence.

✓ Key Criteria

- Presentations must be formal (b).
- Presentations must be **planned** and organised.
- You must listen to and respond appropriately to questions and feedback.

Assessment Objectives:

- □ AO7: Demonstrate presentation skills in a formal setting
- AO8: Listen and respond appropriately to spoken language, including to questions and feedback on presentations
- AO9: Use spoken Standard English effectively in speeches and presentations.

	Key terms	Definition
A	Standard English	English that is clear and most easily understood by a wide audience. It is the 'correct' form of English.
В	Formal	Language that doesn't use any slang or abbreviations.
C	Sophisticated	Complicated or showing higher level knowledge.



SWB GCSE English – Spoken Language

STEP 1: Choosing your Topic

Your topic should be of interest to you. Think carefully about whether you will be able to speak in a sophisticated and formal way about it though! You must talk for <u>4 minutes</u> so ensure that it's something you are knowledgeable about. Topics that have been successful in previous years:

- Does rap music glorify gang culture?
- Gaming: is it good for young people?
- Technology and its impact on our generation.
- Social Media: Benefits and Flaws
- Should animals be used for testing/ human entertainment?
 OR Choose a topic you're interested in.
 Previous successful topics:
- Is basketball a fair sport given it picks based on height and speed?
- Do we still need to travel abroad when we can see it all on the internet?



Step 2: Getting Started

To get started, answer the following questions: • What is important about this topic? • Who are you presenting

• Who are you presentin to?

• Why have you chosen this topic and why should your audience listen?



Step 4: Practise

Read through your presentation to parents, guardians and friends.



Step 3: Planning your Presentation

You must make sure you plan and practise your presentation. Your organisation and preparation will improve your confidence and this will result in a higher grade.

- 1. Break your topic down into smaller sections:
- Introduction,
- Statistics,Facts,
 - Facis,
- Specific cases/ case studies/ anecdotes or
- Personal opinion.
- 2. Do your research:
- You could look online for appropriate articles, documentaries etc.
- You could read some books/ magazines/ newspapers/ blogs.
- You should discuss your topic with friends and family to gain their ideas and inspiration.

3. Engagement Strategies: methods you could use to enhance your presentation:

- True/False with the audience
- Audience interaction .
- Some students have even played instruments and worn costumes!
- Emotive language, rhetorical questions, shocking facts/ statistics.
- When presenting, adapt your tone of voice, use hand gestures and eye contact!

Plan the questions for the questions you may be asked and how you will respond to them.







SWB KS4 – English – A Christmas Carol Plot and Key Quotations

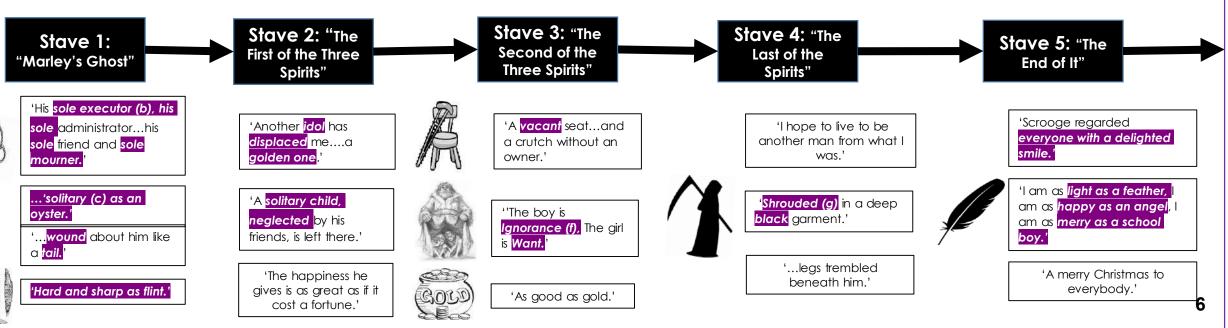
- 1. The **narrator** informs the reader that Jacob Marley (former business partner) died 7 years ago.
- Scrooge described as mean 2 and *miserly* (d).
- 3. Harsh weather and harsh conditions towards Bob Cratchit.
- 4. Scrooge is mean towards Fred, his nephew.
- 5. He refuses to donate money to the Portly gentleman.
- 6. He sees the face of Jacob Marley in his door knocker.
- 7. He is visited by Jacob Marley (bound in chains) who informs him that he will be visited by three spirits.

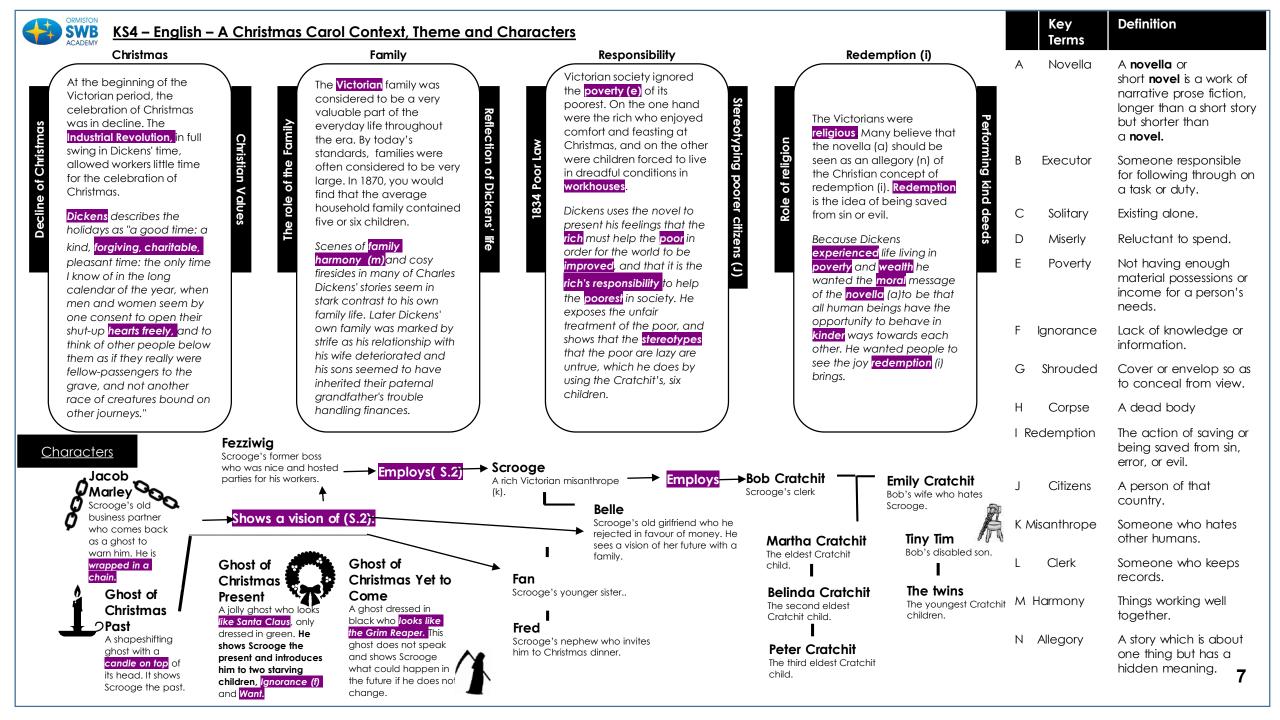
- The Ghost of Christmas Past 1. visits Scrooge.
- 2. Scrooge his taken back to his childhood and sees his former self alone at school.
- 3. We are introduced to his sister -Fan and the poor relationship with his father.
- 4. Scrooge is then taken to Fezziwia where we see him enjoying himself at a party. He interacts with people.
- 5. Scrooge is then shown his former fiancée **Belle** and how they split. He becomes upset when he sees **Belle** and her family in another vision.
- 6. He demands to be taken home.

- The Ghost of Christmas 1. Present visits Scrooge.
- 2 He takes him to see Bob Cratchit and his family. Scrooge is surprised at how little the family have.
- 3 He is then taken to his nephew, Fred's house. Fred and his quests mock Scrooge and his *miserly (d)* ways.
- 4. He is then taken to a poor part of the city and introduced to Ignorance (f) and Want.
- 5. The spirit becomes frustrated and leaves him there.

- The Ghost of Yet To Come is shrouded 1. in darkness : silent and scary.
- 2. He is taken to the city and instructed to listen to a group of businessmen who are discussing the recent death of a man who was not very well-liked.
- 3. He is taken to Old Joe's shop where a dead man's belongings are being pawned.
- 4. He is then taken to a bedroom where he sees a corpse (h).
- 5. Scrooge demands to see tenderness surrounding the death. He is then taken to the Cratchit's and learns the death of Tiny Tim.
- 6. Scrooge wants to identify the dead man and he is taken to his grave.

- Scrooge awakes disorientated and worries he has missed Christmas.
- 2 He is joyful when he learns he hasn't.
- 3. He orders and pays for the biggest turkey to be delivered to the Cratchit family.
- 4. He apologises and **donates** a healthy sum of money to the portly gentleman, to give to the poor.
- 5. He visits Fred and his wife and asks to join them for dinner.
- 6. He is jovial next day, playing a trick on a terrified Bob, who arrives late to work.
- 7. He gives Bob a pay rise and pays for treatment for Tiny Tim.
- 8. Scrooge is reformed! A complete contras to Stave One.







KS4 – Macbeth – Plot and Key Quotations

- 1. The witches meet on the heath.
- 2. <u>Macbeth and Banquo</u> have fought and won a battle. They are praised for their bravery by the Captain.
- 3. The witches meet Macbeth and Banquo and they make three predictions. They predict that Macbeth will become Thane of Cawdor and eventually king. Banquo is told that he will not become king but that his children will be. Shortly after, Macbeth is told he is the new Thane of Cawdor!
- 4. Macbeth plans for Duncan to stay at his castle.
- 5. Lady Macbeth convinces Macbeth to kill King Duncan.
- 6. Duncan arrives at Macbeth's castle.
- Macbeth tells Lady Macbeth he will not murder Duncan. However, she convinces him to go ahead with the murder.

 Macbeth questions Banquo. He is suspicious of Banquo. As a result, he plans his murder with the help or murderers.

Macbeth sees a dagger in

front of him on his way to go

and kill King Duncan.

Macbeth murders King

apparent as he appears

covered in blood. Lady

places the daggers on

Duncan's auards.

to Ireland.

Duncan. Macbeth's guilt is

Macbeth feels no guilt and

The dead body is discovered

by Macduff. Duncan's sons.

Malcolm and Donalbain, run

away: one to England and one

Macbeth and his wife become

king and aueen of Scotland.

1.

2.

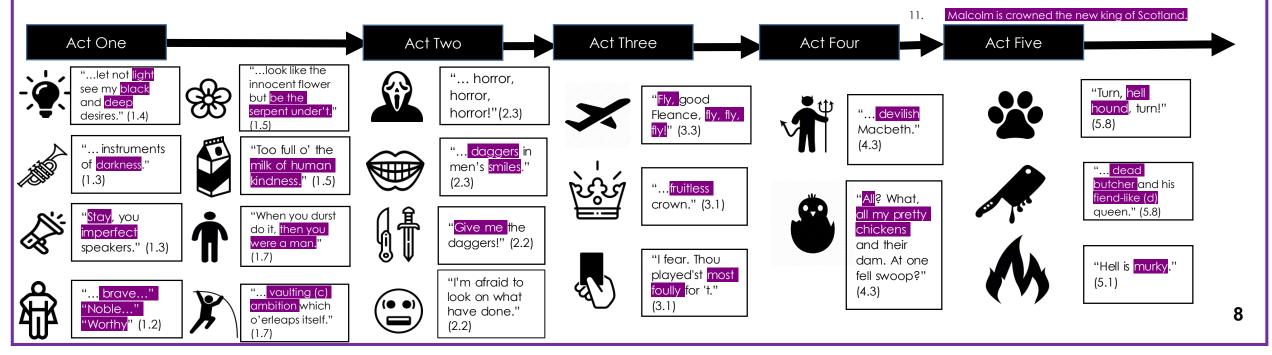
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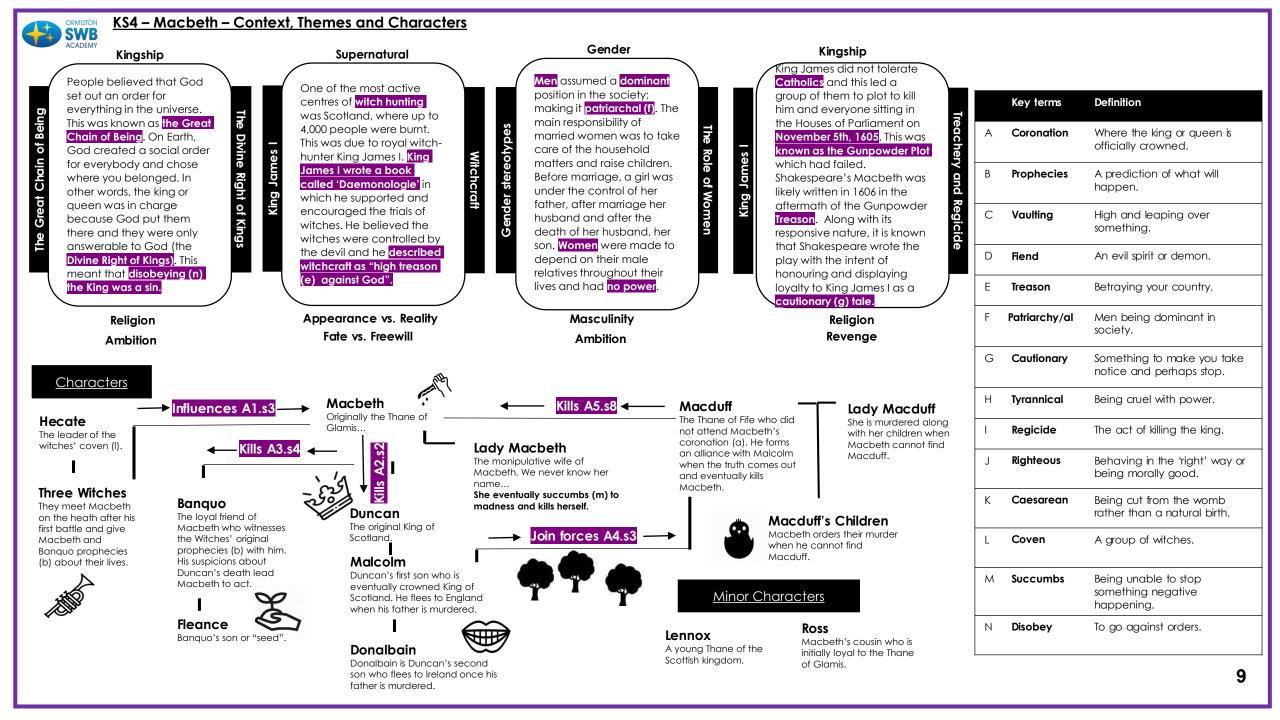
4.

- 2. Banquo is murdered by the hired murderers but his son, Fleance, manages to escape.
- 3. At the coronation (a) banquet, Macbeth sees the ghost of Banquo at the table. Lady Macbeth persuades everyone that Macbeth is ill and they must leave and is furious at Macbeth.
- 4. Hecate, the witches's leader, is angry that the witches meddled with Macbeth without her permission.
- 5. Lennox shares his suspicions about Macbeth.

- 1. The witches share three more prophecies (b):
- 2. Macbeth is told that: he must fear Macduff; that he will not be harmed by anyone born of a woman and that he will never be defeated until Birnam Wood moves to 'High Dunsinane Hill .' They also share a vision of Banquo.
- 3. Macbeth has Macduff's wife and children murdered.
- 4. In England, Malcolm tests Macduff's loyalty and checks they are on the same side, wanting the same things.

- 1. Lady Macbeth sleepwalks: she is overcome with guilt and loneliness.
- 2. The rebels discuss the impending battle.
- 3. Macbeth declares he has nothing to fear from the battle. He appears confident.
- 4. The doctor reports on Lady Macbeth's condition. Macbeth orders him to cure her.
- Malcom and Macduff's forces march to Dunsinane castle, disguising themselves with branches from the trees of Birnam Wood.
- 6. Lady Macbeth dies off stage
- 7. Macbeth realises that he is not going to win but decides to at least die fighting.
- 8. Malcolm prepares for battle.
- 9. Macbeth kills young Siward.
- 10. Both Macduff and Macbeth fight and Macbeth is killed. His head displayed on a spike, as he was a traitor. Macduff is able to kill him as he reveals he was born by caesarean (k) section.







KS4 – English - An Inspector Calls

Act 1:

- 1. The Birling family are celebrating the engagement of Sheila Birling and Gerald Croft.
- 2. An Inspector arrives and tells them that a woman named Eva Smith has committed suicide.
- 3. After Inspector Goole questions Mr Birling, it becomes clear that Mr Birling fired Eva from his factory for asking for higher wages.

Act 2:

- 1. The interrogation (I) moves to Gerald; it comes to light that Gerald had an affair with Eva after she was fired from Milwards - Eva had changed her name to Daisy Renton by now
 - and he had kept her in an apartment for six months before ending their relationship.
- 2. The Inspector then questions Mrs Birling about when Eva came to her charity for help whilst pregnant. She refused to help her as she believed Eva to be "impertinent" (H) for using

Act 3:

- 1. The Inspector then questions Eric as it has become clear that it was Eric who got Eva pregnant. Eric reveals that he forced himself on Eva. He says" 'I was in that state when a chap easily turns nasty." Eric then confesses to stealing money from his father but Eva refused the money.
- 2. Inspector Goole leaves after delivering his final speech; he warns that there is a lesson to be learnt in "fire and blood and anguish" if we do not start taking responsibility for our actions and start treating everyone in society fairly.
- 3. Mr Birling calls up the police department who confirm there is no Inspector Goole so the family believe they've been tricked until they receive a phone call to let them know a girl has committed suicide and an Inspector is on his way to question them.

Definition

Kev terms

4. Inspector Goole then interrogates(I) Sheila	the name (Mrs Pirling)			key terms	Semination
and it is revealed that she used her position as a wealthy local person to get Eva fired from	the name 'Mrs Birling'.			A Socialism	A political system where people work together for society; where wealth is shared and everyone benefits equally.
her job at Milwards. Act 1	Act 2		Act 3	B Capitalism	A system of private ownership that allows the rich to get richer; this system allows the lower classes to be exploited (D).
Narrator: The Inspector:		Gerald:	Eric:	C Dramatic Irony	When the audience has more knowledge of what is happening than a character.
"The dining room is of a fairly large suburban house, belonging to a "A chain of	"A girl died tonight. A pretty, lively sort of girl, who	"she told me she'd been happier	"I was in that state when a chap easily	D Exploitation	The action or fact of treating someone unfairly in order to benefit from their work.
prosperous events."	never did anybody any harm. But <u>she died in misery</u> and agony – hating life – , "	than she'd ever been before."	turns nasty – and I threatened to	E Bourgeoisie	Middle Class
			make a row."	F Proletariat	Working Class
Mr Birling: Mr Birling: unsinkable, unsinkable,	Mrs Birling:		<u>The Inspector:</u>	G Patriarchy	A society ruled by males in which women are valued less and often act in a obedient way.
absolutely absolutely	"Girls of that class."	"Public men, Mr Birling, have responsibilities as well as privileges."	We are members of one body. We are responsible for each other."	H Impertinent	Rude
				I Interrogates	Asks lots of questions
<u>Mr Birling:</u> <u>Sheila:</u>				J Morality	Having morals/ good values
"as if we were all "But these girls	Sheila:	Mrs Birling:	<u>Sheila:</u> "The point is,	K Conservatives	A political party who values more capitalist)b) attitudes.
 mixed up together like bees in a hive – community and all that nonsense;" mixed up together like bees in a hive – community and all that nonsense;" 	Us the rope so that we'll hang ourselves."	ever refuse	you don't seem to have learnt	L Mouthpiece	Someone placed there to speak your own views.
		money!"	anything."	K Credited	Given the praise for. 10

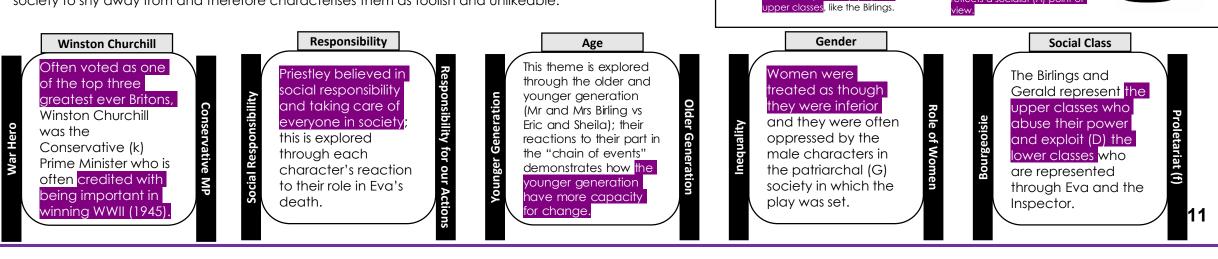


<u>Context</u>

- The play was written in 1945 but set in 1912 during this period, both world wars had happened
- Priestley fought in WWI and saw the horrors. At the end of the WWII he saw the class, age and gender divides all broken down and fought for the country to keep it like that, rather than go back to Edwardian attitudes of 1912 of patriarchy (g) and rich people ruling without considering the poor.
- After WWII, the Labour Party (a socialist (A) party) won the general election in a landslide victory against Churchill's Conservatives (k).
- Priestley held socialist (A) beliefs and wanted to encourage others to behave in a socialist (A) way
- In 1912 women did not have the right to vote and were seen as less than men; by 1945 they could vote and had more value in society after the war

Writer's Intentions

- Priestley wrote the morality (J) play, An Inspector Calls, as a warning for people to treat each other more equally otherwise horrible events could follow like World War One and Two.
- He highlighted the divisions that existed in pre-war Britain by setting the play in 1912 and through the characters.
- He wanted his 1945 audience to see how damaged society was and wanted to encourage them to work together and look after each other.
- He believed these socialist (A) values would help the country to avoid another war. He uses the character Inspector Goole as his mouthpiece for his socialist views and he shows his audience that the younger generation (through Eric and Sheila) have the ability to enforce these changes that society need.
- The arrogant and ignorant attitudes of Mr and Mrs Birling, the bourgeoisie (E), are what Priestley wanted society to shy away from and therefore characterises them as foolish and unlikeable.



Characters

Sybil Birling

suicide.

Eric Birlina

on Eva, got her

preanant, then stole

to support her. He

Edna

their wealth

money from his father

accepts responsibility.

The maid of the Birling

family; used a symbol o

Eva Smith

She never appears on stage; she

has already committed suicide.

She represents the proletariat(F)

who are exploited (D) by the

Arthur's wife of a higher

class. An unsympathetic

woman who represents the

bourgeoise (E) upper class.

She is adamant that she is

blameless in Eva Smith's

Sybil and Arthur's son.

He is adolescent in his

manner and drinks too

much. He forced himsel

Arthur Birlina

Sybil's husband. He represents

ontrols the wealth and means

f production: he is concerned

conventional attitudes. He is a

wealthy factory owner in his mid

Sheila Birlina

Sybil and Arthur's

daughter. She's in her

early twenties; bright,

Unlike her parents and

fiancé, she expresses

va Smith's suicide

deep regret for her role in

lively and optimistic.

ne capitalist (B) class that

with material gain and

50s

Gerald Croft

Inspector Goole

A mysterious figure. His name

have supernatural powers of

evokes the word 'ahoul', meaning

spirit or phantom. He appears to

observation and persuasion. He

reflects a socialist (A) point of

Sheila's fiancé. Gerald Croft

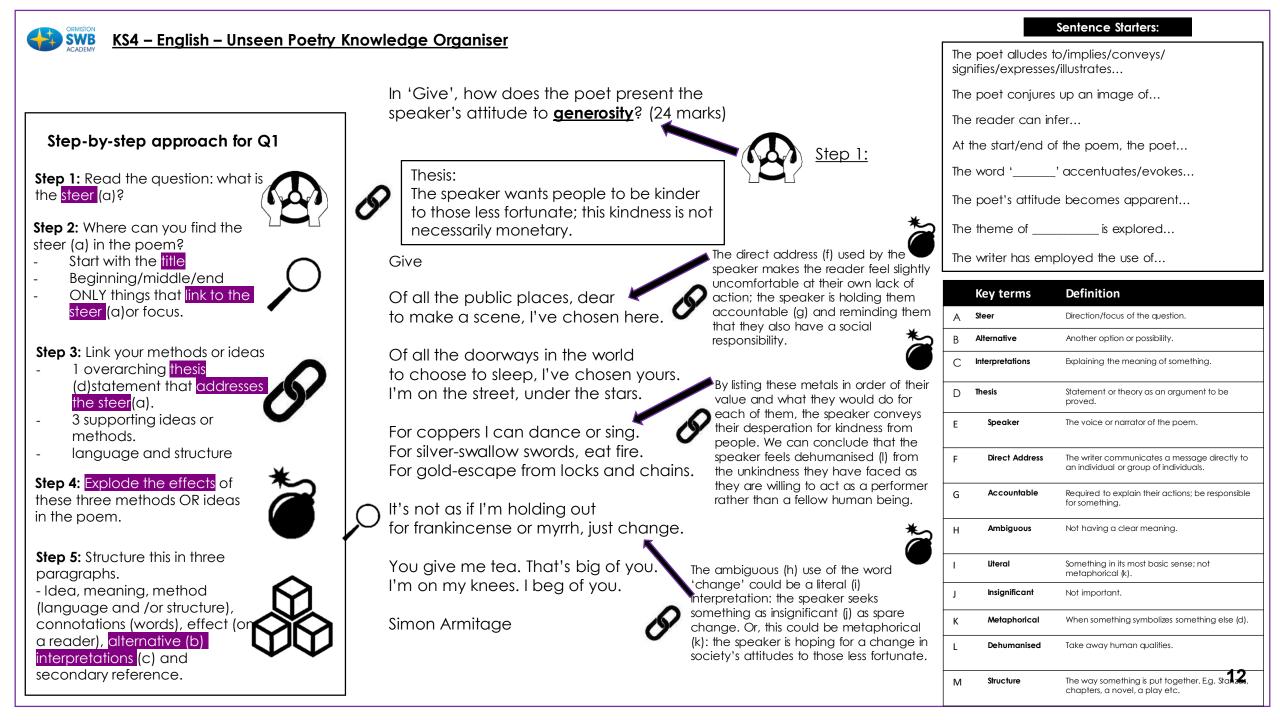
represents the aristocracy: the

of rich land owners and people

who inherit their wealth from their

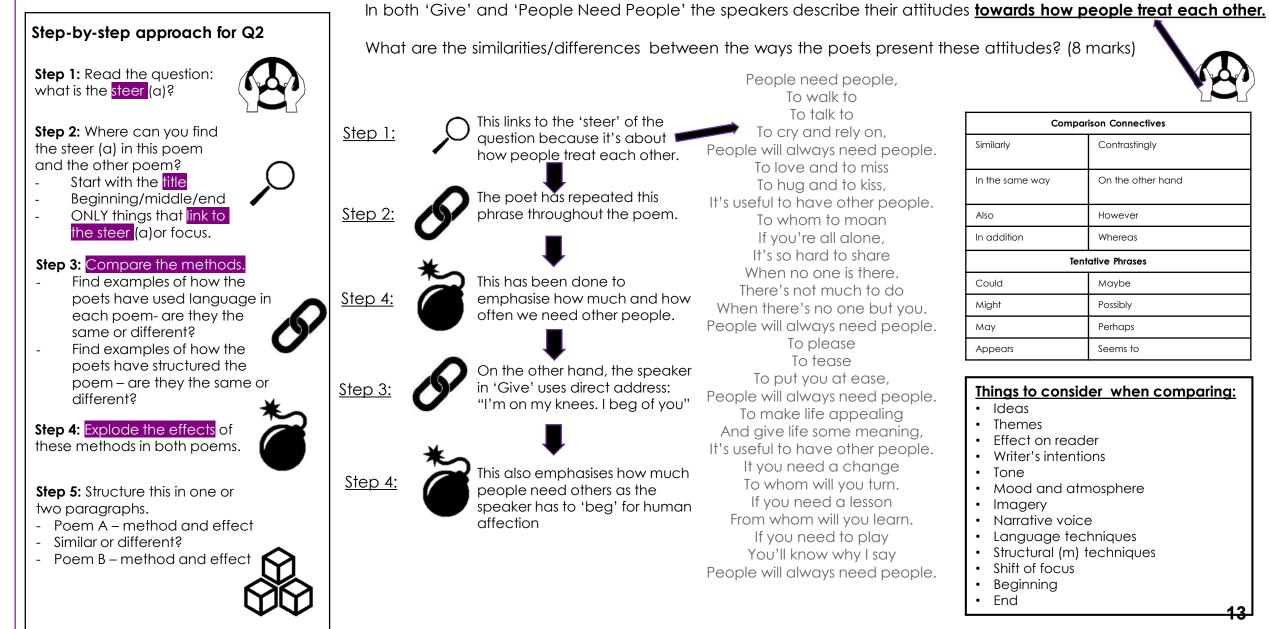
parents. He had an affair with Eva

highest class of society, comprised



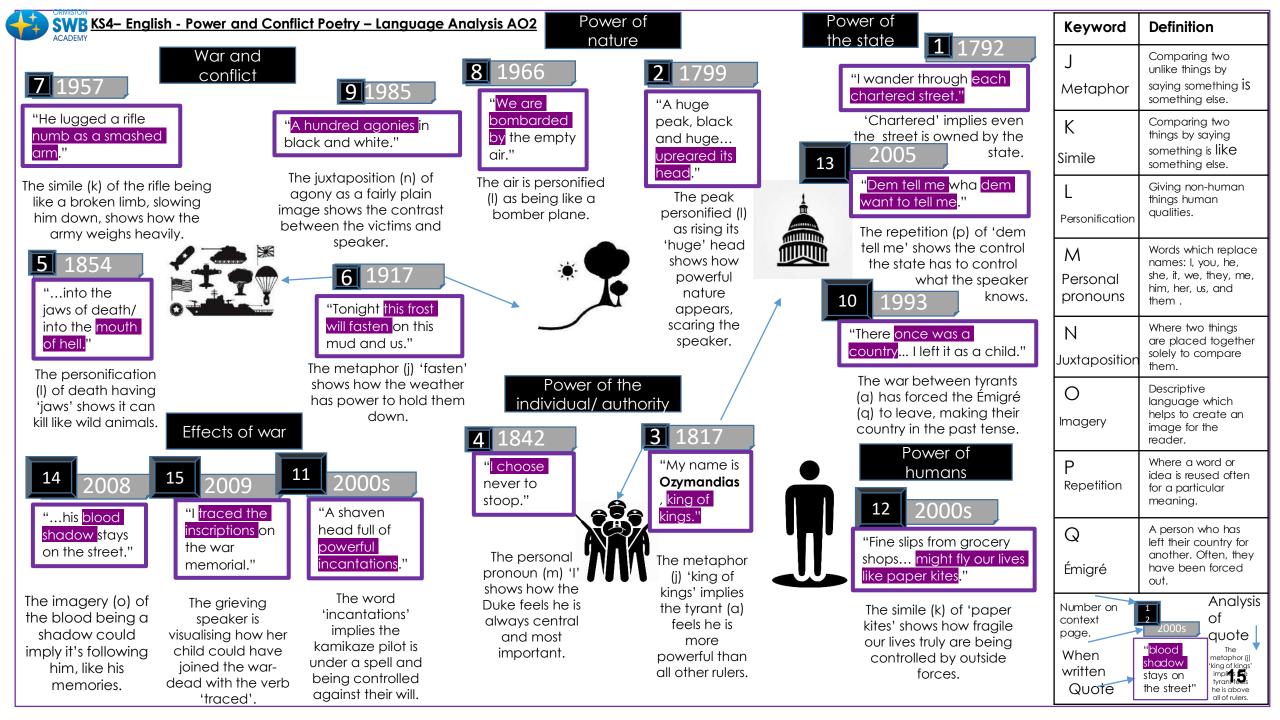


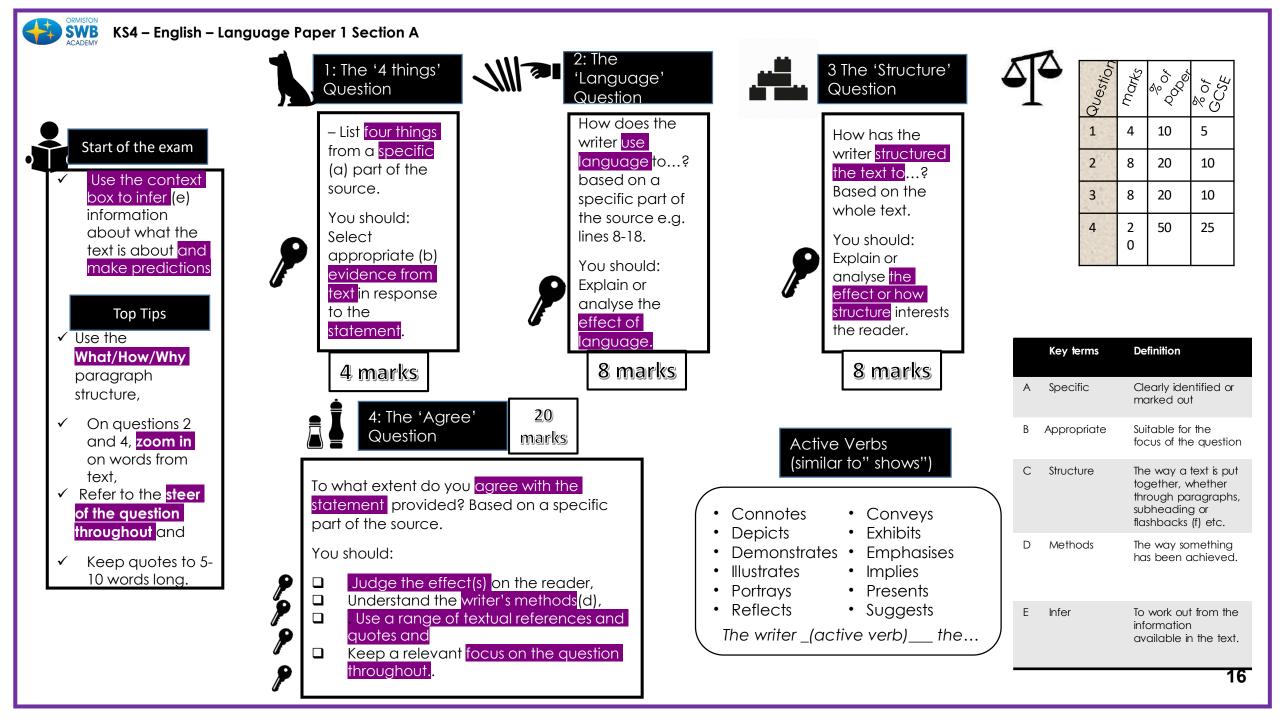
KS4 – English – Unseen Poetry Knowledge Organiser



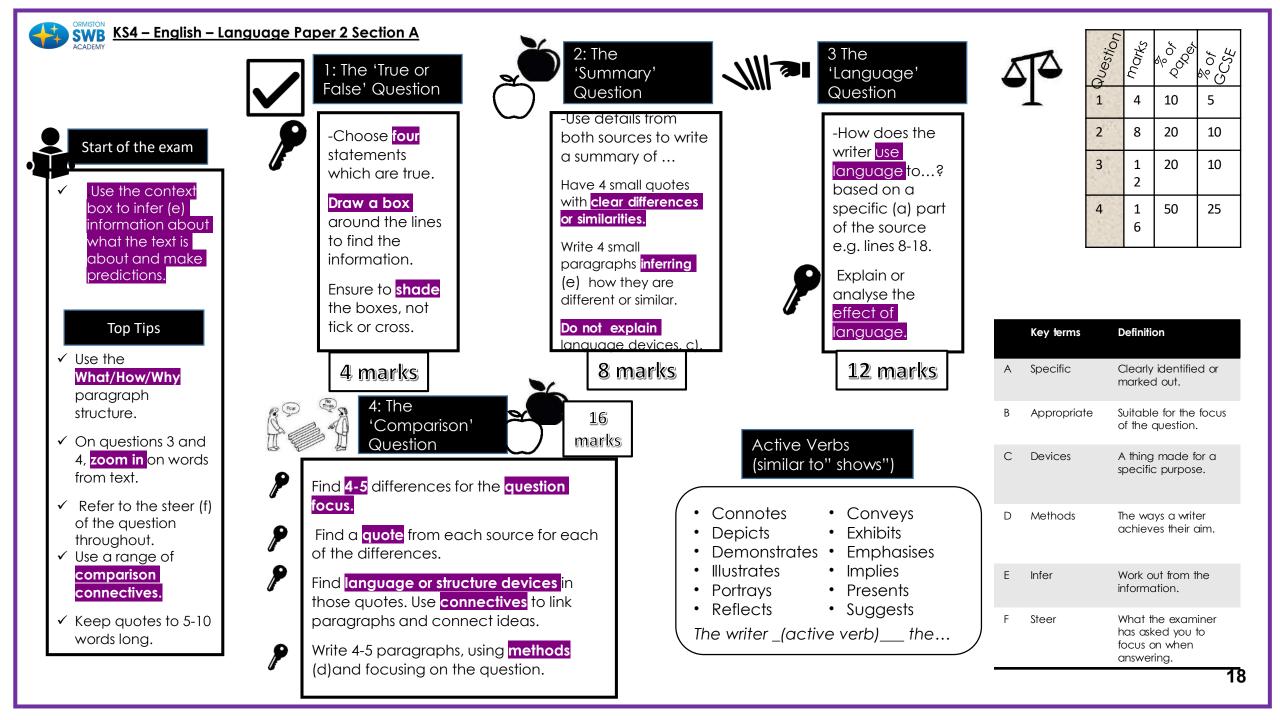
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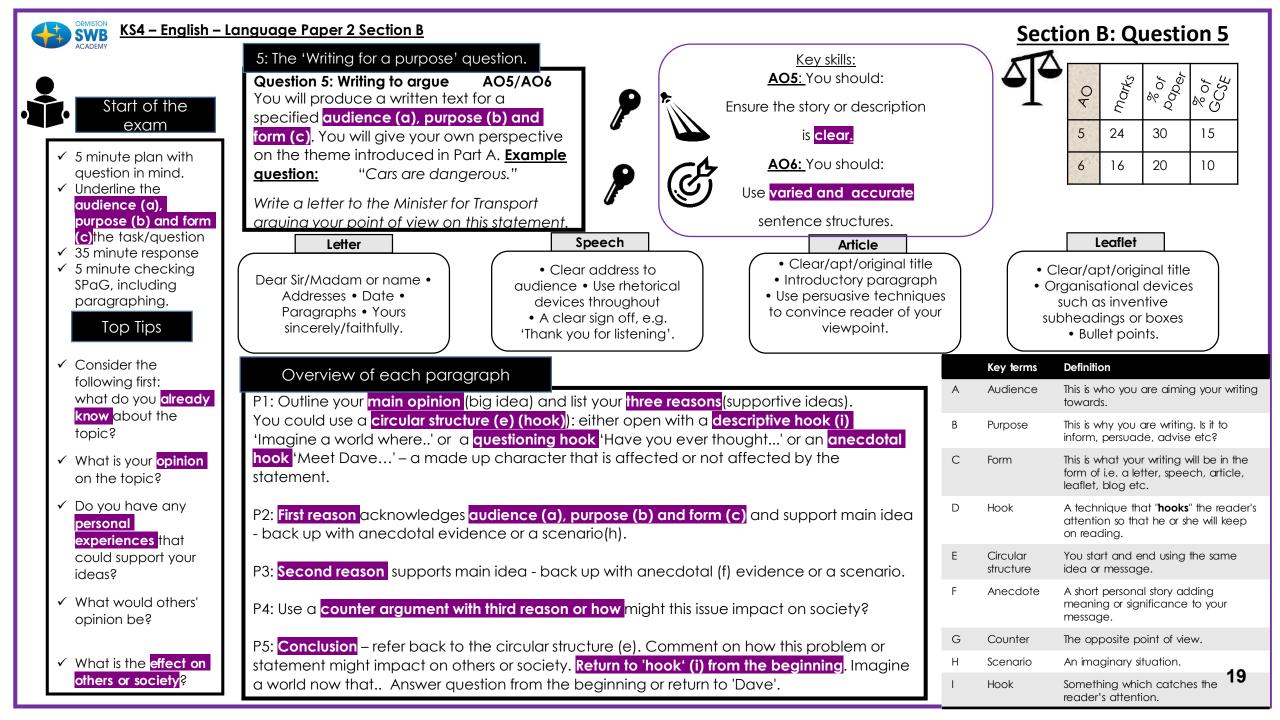
KS4– English – Pc	ower and Conflict Poetry C	<u>Context</u>				Keyword	Definition
		-		ČŘÍ ST	A	Tyrannical	Being cruel with power.
				Global	В	Radical	A person who wants to change society.
Romantic era (g)	Imperialism (d)	Global Conflict	Global Conflict	Displacement (e)	С	Oppressed	Being treated harshly, often by rulers.
1792	1854	1914-18	1939-45	onwards	D	Imperialism	Pushing your country's ideas on others.
Poems: 1. London 2. Extracts from the	Poems: 4. My Last Duchess 5. Charge of the	Poems: 6. Exposure 7. Bayonet	Poems: 11. Kamikaze	Poems: 8. Storm on the Island	E	Displace	Taking over the place or role.
Prelude 3. Ozymandias	Light Brigade	Charge Key ideas:	Key ideas: • In the Second World War:	 9. War Photographer 10. The Emigree 12. Tissue 	F	Industrial Revolution	The use of machines in factories.
 Key ideas: Poems are not about love. Poems are about personal growth and 	 1 in 5 people in the world called Queen Victoria of Great Britain their governor (j). 	 In the First World War: 20 million people died. 	 75 million people died. It was fought across the globe. Pearl Harbour in the 	13 Checkin' Out Me History14. Remains15. PoppiesKey ideas	G	Romantic era	Poems about nature and its impact in a changing industrial (f) word.
 appreciating nature during the industrial revolution (f). Poets sometimes 	 People were very poor. People lacked rights. The social 	 It was fought from trenches (holes in the ground) in 	USA was attacked in December 1941. Japanese pilots were trained to bomb ships by flying into them. This	Wars were often fought against smaller countries to stop tyrants (a) or terrorists	Н	Hierarchy	A system where people or items are ranked by status or power.
fought back against what they thought were tyrannical (a) rulers who oppressed (c)hormal people.	hierarchy (h) still defined who had power <mark>.</mark>	Europe.	 meant they committed suicide for their country. Following this, America came into the war on Britain's side, changing 	 (i). Locals thought Western countries invaded for oil and other resources. 	I	Terrorists	A person or group who uses violence to achieve their goals to change world.
• This made them considered radical (b).			which side had the upper hand in the war.	 Many locals were still loyal to the tyrants (a). 	J	Governor	The person in charge. 14





CRMISTON SWB ACADEMY KS4– English – La	nguage Paper 1 Section B		Section B: Question 5
 Start of the exam 5 minute plan with question in mind. Top Tips Keep your tone consistent (g) throughout: do not use words which suggest a 	5: The 'Writing for a purpose' question. Question 5: Writing a narrative or description AO5/AO6 You will produce a story or description based on a picture or a sentence detailing your task. Your local newspaper is running a creative writing competition and the best entries will be published. <u>Either</u> write a story about time travel as suggested by this picture or write a story entitled 'Stranded'. <u>Simile</u>	Key skills: AO5: You should: Ensure the story or description is Clear. AO6: You should: Use varied and accurate sentence structures.	5 24 30 15 6 16 20 10
 Words which suggest a light and playful atmosphere (b) after you have just spent 15 minutes making the scene sound scary. Use a variety of structural (e) features: flashbacks (f), Keep to one or two characters and 	Key language devices to use: Where you compare two things by saying they're like something else: "He was as timid as an	re vou compare two	Where you use vocabulary to describe the character's five senses.
✓ 5 minutes' of checking	scene or setting-decide if it's positive or negative.	Connective Unless, although.	B Atmosphere The tone or mood.
SPaG, including paragraphing.	P2: Character focus – introduce character – show but not tell then lead in to a flashback. Use a	Adverb Regretfully, sadly.	C Vice versa The two items can be switched around or reversed.
Punctuation to use	symbol, <mark>item or even to trigger the shift in time.</mark>	Simile Like a mouse	 D Motif An object which is repeated and has importance to the
QuestionExclamationMark?Mark!	P3: Come back to present moment, developing the character in more detail. Keep something withheld!	Metaphor Brave lions, they	events. E.g. a raven for death.
Comma , Full stop .	P4: Describe the setting <mark>- zoom out t</mark> o change focus. Include a motif (d).	FeelingJealous, she tore up his clothes.	E Structural The way a text is put together, whether through paragraphs, subheading or flashbacks (f) etc.
Semi- ; Speech " Colon Marks "	P5: Cyclical (a) development – back to	Verb 'ing' Giggling and laughing, clause they ran to school.	F Flashback When the text goes back in time.
Colon : Apostrophe '	weather/scene/setting – change from positive to negative or vice versa (c).		G Consistent Keeping something the same 17





	ogressing th knowledg			y find it useful to look b	back at Crossover Unit 52-Direct and Inve	erse Keyword/ Skill	Definition/Tip	
Reminders: Direct Proportion y is directly proportional to x			Example 2 y is directly proportional to the square of x. When $x = 6$, $y = 27$. Find y when $x = 4$.		Two quantities change in the same way. When one increases or decrease, so does the other one.			
$\propto x$ = kx		ultiplied b is constar	y x nt (of proportionality)	$y \propto x^2$ $y = k \times x^2$	y = 0.75 x 4 ² y = 0.75 x 16	Inverse Proportion	As one quantity increases the other decreases.	
verse Pro	verse Proportion is inversely proportional to x		27 = k x 36	y = 12	Equation	The value of two expressions is equal (Shown by =)		
1.			$\frac{27}{36} = k$ 0.75 = k		Constant of Proportionality	A constant value, written as k, relating to amounts that rise or fall uniformly together.		
X Y	y proportio 36 2 he value c	a 5	$5 = \frac{1}{3} \times \sqrt{\alpha}$	F = 16. Find F wh	ly proportional' with '= k ÷' $16 = \frac{128}{\sqrt{F}}$	Other top • A-level • Pure 1 • Differen • Expone • Pure 2	 Pure 1 Differentiation Exponentials and Logarithms 	
$= k \times \sqrt{36}$ $= k \times 6$			$15 = \sqrt{\alpha}$	$32 = \frac{\kappa}{\sqrt{16}}$	$16 = \frac{128}{\sqrt{F}}$	<u>Exam Tips</u>		
$\frac{2}{6} = k$ 15 ² = a		$32 = \frac{k}{4} \qquad \qquad \sqrt{F} = 8$		When you to" replace	 When you see words, "Inversely Proportio to" replace with "= k ÷" to get you started 			
			a = 225	k = 128	$F = 8^2 = 64$		estions can range from 2-4 marks. ain a mark for finding 'k' in each	

Working Above Unit 10 - Similarity in 2D and 3D Before progressing through this section of work, you may find it useful to look back at Crossover Unit 46 - Congruence & Definition/tip Keyword/ Similar Shapes knowledge organiser. Skill DON'T FORGET **Reminder- Linear Scale Factor** Similar Two shapes are **similar** when one can become the other after a resize, flip, slide or turn. Linear scale factor = 35cm 20cm 22cm A measure of distance. How far Length from end to end, or from one end Area scale factor = $3^2 = 9$ to another. С A 6cm h Scale factor The ratio by which a length or other 4cm In order to find length DF Volume scale factor = $3^3 = 27$ measurement is increased or Write two of the corresponding sides as a ratio = AB : DE decreased 5 : 20 12cm Simplify - 1 A ratio shows the relative sizes of 2 : 4 Ratio With a scale factor of 4 or more values. To find DF- look at the corresponding This is the The size of a surface Area length AC = 6 cm scale Area Scale factor 6 x 4 = **24cm** factor Example Volume The amount of 3D space something The 2 supermarket tickets are mathematically similar takes up. Also called capacity. **Volume Scale Factor** Example A child's rugby ball is 10cm long and has a volume of 200cm3. 2 cm It is similar in shape to a full size rugby ball. A full size rugby ball Other topics/Units this could appear in: 6 cm is 22cm long. Find the volume of the full size rugby ball. Congruence Geometric Proof 111-11 118-610 The area of the smaller ticket is 7cm². Calculate the area of the larger tickets. Corresponding lengths written as a ratio = 2 : 6 10cm Exam Tips 1:322cm • To help you start similarity questions you will Linear Scale factor = 3Area Scale factor = $3^2 = 9$ Corresponding lengths = 10:22need to find the scale factor. 1:2.2 You will gain a mark for recognising and Area of small ticket x area scale factor = Scale factor = 2.2^3 21 using area and volume scale factor. 7 x 9 =63cm² 200 x 2.23 = 2129.6cm³



Working Above Unit 10 - Similarity in 2D and 3D - Problem Solving.

Before progressing through this section of work, you may find it useful to look back at Crossover Unit 46 - Congruence & Similar Shapes

Problem Solving with Area Similarity Example 1: A model yacht is made to a scale of $\frac{1}{20}$ of the size of the real vacht. The area of the sail of the model is 150 cm^2 . What is the area of the sail of the real yacht. $\frac{1}{20}$ can be written as the linear scale factor = 1 : 20 Area scale factor = $20^2 = 400$ Area of real sail = $400 \times \text{area}$ of model sail $= 400 \times 150$ $= 60.000 \text{ cm}^2$

Take care if you are asked to give your answer in m²

To convert from cm ² into m ² you need to divide by 100 ²	
= 10,000	

60.000 ÷ 10.000 = 6m²

Problem Solving with Area and Volume Similarity Example:

Mark has made a clay model. He will now make a clay statue that is mathematically similar to the clay model.

The model has a base area of 6cm² . The statue will have a base area of 253.5cm² Mark used 2kg of clay to make the model.

Clay is sold in **10kg** bags.

Mark has to buy all the clay he needs to make the statue.

How many bags of clay will Mark need to buy?

Area scale factor = $253.5 \div 6 = 42.25$

Linear scale factor = $\sqrt{42.25} = 6.5$

Volume scale factor = $6.5^3 = 274.625$

Problem Solving with Area Similarity Example 2:





A 20 Euro note is a rectangle 133mm long and 72mm wide.

A 50 Euro note is a rectangle 165mm long and 82mm wide.

Show that the two rectangles are not mathematically similar.

The rectangles will be mathematically similar if the scale factors for the lengths and widths are equal. You need to show that they are different.

Lengths Widths 72:82 133:165 1:1.1388... 1 : 1.240606 The scale factors are different therefore the rectangles are not mathematically similar.

> Use the area scale factor to calculate linear scale factor. then the volume scale factor.

274.625 x 2 =549.25kg of clay needed 549.25 ÷ 10 = 54.925kg clay needed = 55 bags

Keyword/ Skill	Definition/tip
Similar	Two shapes are Similar when one can become the other after a resize , flip, slide or turn.
Length	A measure of distance. How far from end to end, or from one end to another.
Scale Factor	The ratio by which a length or other measurement is increased or decreased.
Ratio	A ratio shows the relative sizes of 2 or more values.
Area	The size of a surface. The amount of space in a 2D surface.
Volume	The amount of 3D space something takes up. Also called capacity.

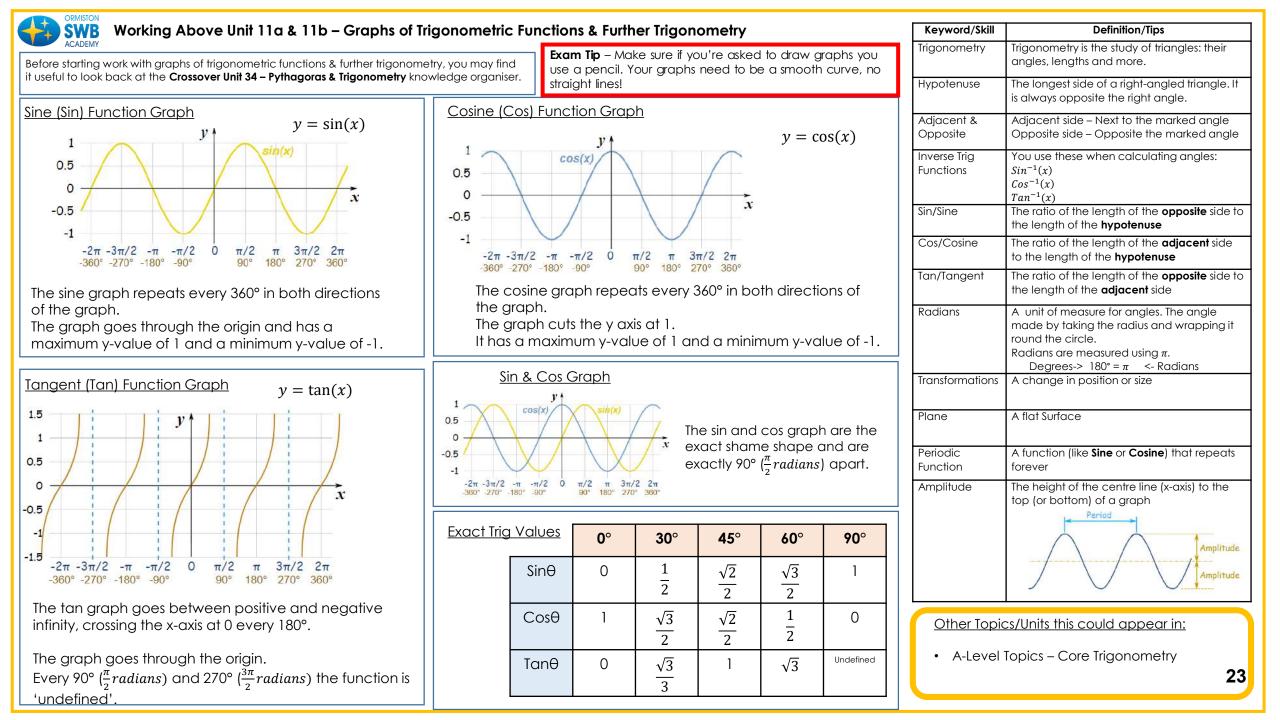
Other topics/Units this could appear in:

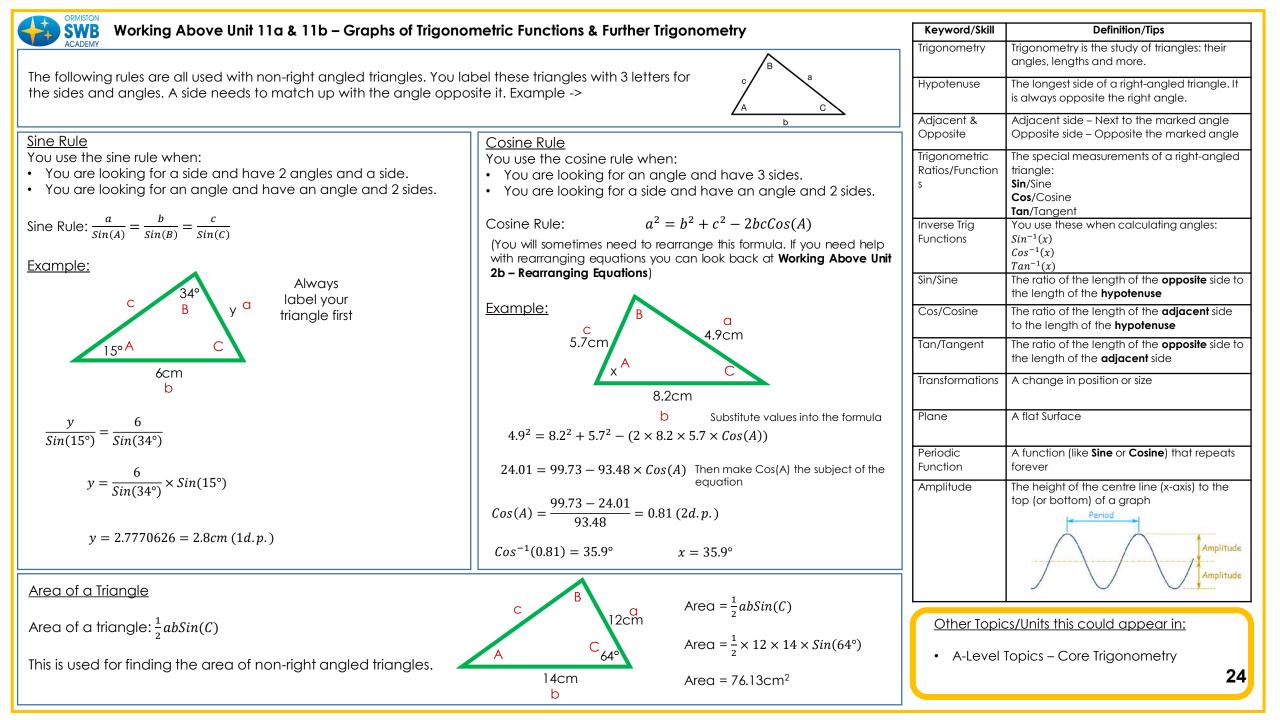
- Congruence
- Geometric Proof

Exam Tips

- To help you start similarity questions you will need to find the scale factor.
- You will gain a mark for recognizing and using area and volume scale factor.

22





	s of Trigonometric Functions & Further Trigonometry	Keyword/Skill	Definition/Tips			
ACADEMY	Calculator Help	Trigonometry	Trigonometry is the study of triangles: their angles, lengths and more.			
CASIO	ones when you are finding a length.					
fx-300ES PLUS DATURAL-UP.A.M. TWO WAY POWER		Adjacent & Opposite	Adjacent side – Next to the marked angle Opposite side – Opposite the marked angle			
	To get the inverse trig functions you need to press the SHIFT button first before you press the function you need. You use these ones when you are finding an angle.	Trigonometric Ratios/Function s Inverse Trig	The special measurements of a right-angled triangle: Sin/Sine Cos/Cosine Tan/Tangent You use these when calculating angles:			
	Which Rule to use	Functions	$Sin^{-1}(x)$ $Cos^{-1}(x)$ $Tan^{-1}(x)$			
HODE SETUP ON		Sin/Sine	The ratio of the length of the opposite side to the length of the hypotenuse			
	Is the triangle right-angled?	Cos/Cosine	The ratio of the length of the adjacent side to the length of the hypotenuse			
Abs (x ³) (x ⁻¹) log _#]	YES	Tan/Tangent	The ratio of the length of the opposite side to the length of the adjacent side			
$ \begin{array}{c} \hline \\ \hline $	Does the question Do you know a side	Transformations	A change in position or size			
(-) (-) (-) (-) (-) (-) (-) (-) (-) (-) (-)	involve any angles? and its opposite angle?	Plane	A flat Surface			
	YES NO YES NO	Periodic Function	A function (like Sine or Cosine) that repeats forever			
7 8 9 DEL AC	Use trig ratios: sin, cos and tanUse Pythagoras's TheoremUse the Sine RuleUse the Cosine Rule	Amplitude	The height of the centre line (x-axis) to the top (or bottom) of a graph			
4 5 6 × ÷ 1 2 3 + -			Amplitude			
	Exam Tip: The sin, cos & area of a triangle formulas will sometimes need to be rearranged. You need to be able to decide which rule to use and when you will need to rearrange it. You also need to learn these rules as you will not get them in the exam!	· · · ·	<u>cs/Units this could appear in:</u> Topics – Core Trigonometry 25			

Work	king Above	Unit 12a – Samp	oling	Before starting work with sampling, you at the Crossover Unit 38 Sampling know		Keyword/Skill Data	Definition/Tips Facts that are collected.
Types of Samp Random Samp				Systematic Sampling		Population	The whole group from where the sample is taken, i.e. a whole year group.
Random Samp person in the g interested in ho	roup you ar as an equal	every		This follows a system. E.G every fourth person is chosen.		Primary Data Secondary Data	Data you collect yourself. Data which other people have collected.
of being chose Names might k	be placed ir	i a nat		Capture/Recapture		Hypothesis	An idea or an assumption that you can test to see of it is true.
and then picke could be chose computer.		mes		You might recognise this from biolo 1) Take a sample of the po	e ,	Quantitative Data	Data that is given in numbers.
Stratified Samp	oling		20%	 Mark each item Put the items back into the population and ensure they are thoroughly mixed 		Qualitative Data	Data that is given in words.
(population) is smaller groups	Where the group of people, (population) is divided into smaller groups so that the same PROPORTION can be		 4) Take a second sample and count how many of your sample are marked 5) The proportion of marked items in your new sample should be the same as the proportion of marked items from the population in your first sample Example: 10 fish are caught in a lake, marked and released back into the lake. A week later, 20 fish are caught and 4 are 		Discrete Data	Data that only takes a certain value. E.G. number of people in class. (Only whole numbers)	
					Continuous Data	Data that has a number of possibilities between two fixed points. E.G. The weight of a new born baby (Can include decimal numbers)	
Year Group	No. of Pupils	How to work out pupils in each group.	No of Pupils in Sample	found to be marked. Estimate the	number of fish in the lake.		
7	180	<u>180</u> 1000 X 50 =9	9	$\frac{10}{10} = \frac{4}{20}$	_	Bias	A built in error that makes the
8 9	200	$\frac{200}{1000} \times 50 = 10$ $\frac{240}{1000} \times 50 = 12$	10	n 20			results wrong or "unfair", i.e. a weighted dice so when rolled it
10	220	$\frac{1000}{1000} \times 50 = 12$	11	$n = \frac{10 \times 2}{4}$			lands on 6 more than the rest.
11	160	$\frac{160}{1000}$ X 50 = 8	8	n = 50		Other Topics	/Units this could appear in:
Check your answers add up to 50 (9+10+12+11+8 = 50).				There are approximately	50 fish in the lake	• A Level S	tatistics - Data Collection 26



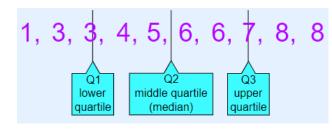
Working Above Unit 12b - Cumulative Frequency & Box Plots

Quartiles

Quartiles are the values that divide a list of numbers into quarters:

- Put the list of numbers in order
- Then cut the list into four equal parts
- The Quartiles are at the "cuts"

Example: 1, 3, 3, 4, 5, 6, 6, 7, 8, 8 The numbers are already in order, cut the list into quarters:



In this case Quartile 2 is half way between 5 and 6: Q2 = (5+6)/2 = 5.5

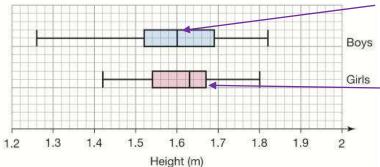
And the result is:

- •Quartile 1 (Q1) = 3 (Lower Quartile)
- •Quartile 2 (Q2) = 5.5 (Median)
- •Quartile 3 (Q3) = 7 (Upper Quartile)

Making Comparisons

The box plot summarise the heights of samples of 14 and 14 year old boys and girls

Heights of boys and girls



Before starting work with quartiles, you may find it useful to look back at the Crossover Unit 23 Averages and Un 24 Averages from a Table knowledge organisers.

= Q3 - Q1

Q3

7

IQR

The IQR "Interguartile Range" is the spread of the middle 50% of data.

As it is only the middle 50% the IQR is less likely to be affected by outliers. Whereas the range (which is the spread of all the data) would be affected by outliers. Q3 Q2

To calculato	it just subtract Quartile 1 from	(21 Q	2 Q	3
	in just submact Quarme T morn			la constante	
Quartile 3.		25%	25%	25%	
Using the exa	mple from the quartiles:				
Q3 = 7			5	~	
Q1 = 3	So the interguartile range is 4	4 I	nterquart	ile Range	е

So the interguartile range is 4

Box Plot

A box plot shows a visual representation of the **median** and **quartiles** of a se of **data**.

Q2

5

6

Q1

4

To draw a box plot, the following information is needed:

Lowest Value

• minimum value



upper quartile

• maximum value

The median is labelled as Q2, use a ruler to read the value

2

Median height of girls = 1.63mMedian height of boys 1.60m

On average, the girls are taller than bo

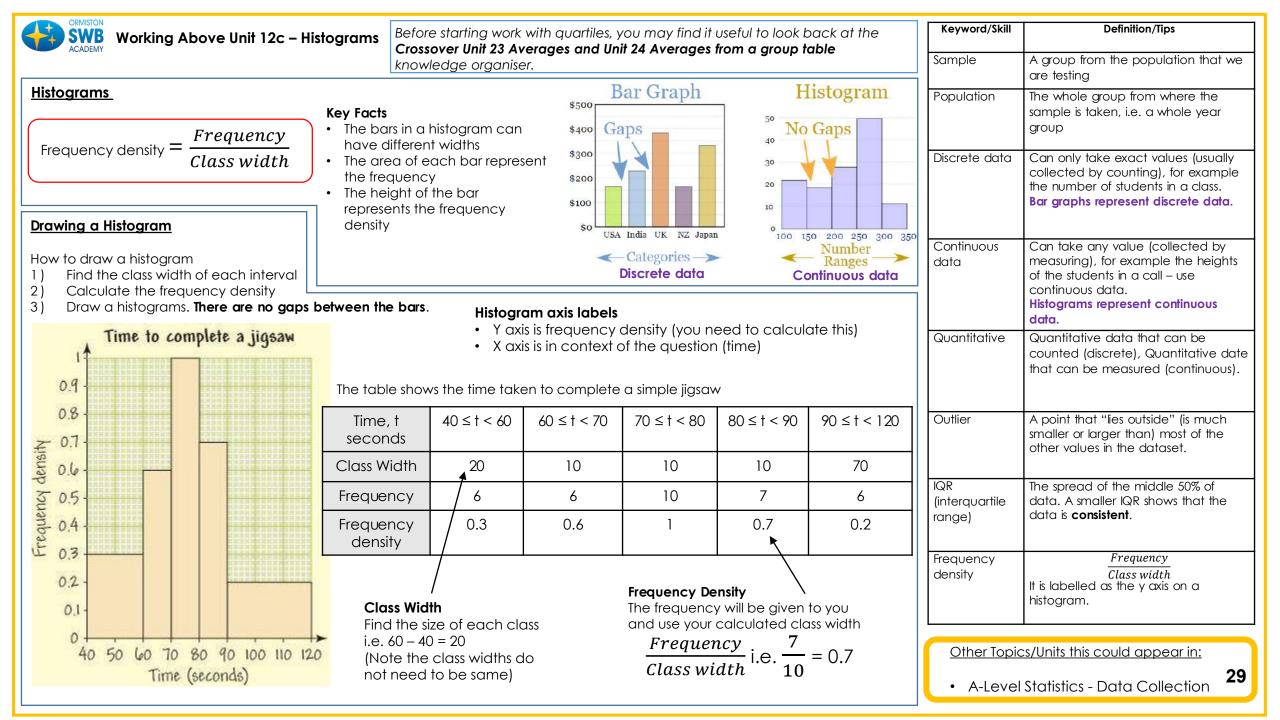
The IQR is the UQ subtract the LQ, read Q3 & Q1 then do Q3 – Q1

3

IQR for girls = 1.67 - 1.54 = 0.13m IQR for boys = 1.69 - 1.52 = 0.17m The IQR for the boys is greater than the girls, the girls heights are more **consistent** than the boys

nay find it useful	Keyword/Skill	Definition/Tips
r erages and Unit rganisers.	Sample	A group from the population that we are testing .
of data.	Population	The whole group from where the sample is taken, i.e. a whole year group.
I by outliers . Id be affected	Discrete	Discrete data can only have a finite or limited number of possible values. (Whole numbers)
Q3 5% 25%	Continuous	Continuous data can have an infinite number of possible values within a selected range. (Can include decimal numbers).
Range Q1	Quantitative	Quantitative data that can be counted (discrete), quantitative date that can be measured (continuous)
uartiles of a set	Mode	The number which appears most often in a set of numbers
Highest Value	Median	Place the numbers in value order and then find the middle number. When there are two numbers in the middle, we find the average them.
<u> </u>	Range	The difference between the highest and lowest values.
8 9 10	Outlier	A point that "lies outside" (is much smaller or larger than) most of the other values in the dataset.
taller than boys.	IQR (interquartile range)	The spread of the middle 50% of data. A smaller IQR shows that the data is consistent .
o Q3 – Q1 s greater than hts are more oys		cs/Units this could appear in: Statistics- Data Collection 27

Working Above Unit 12b – Cur	Before starting work with quartiles, you may find it useful to look back at the Crossover Unit 23		Keyword/Skill	Definition/Tips	
Cumulative Frequency Graphs]	Averages and Unit 24 Averages From a Table knowledge organisers.		Sample	A group from the population that we are testing .
A cumulative frequency diagram creates a running total of the amounts within a table.A cumulative frequency diagram is drawn by	Making Comparisons When making comparisons us	Making Comparisons These cumulative frequency		Population	The whole group from where the sample is taken, i.e. a whole year group.
plotting the upper class boundary with the cumulative frequency.	spread to back up your stater Masses of men ar	ment!	samples of 100 men and 100 women.	Discrete	Discrete data can only have a finite or limited number of possible values. (Whole numbers)
38 36 34 32 30 Range =	100- 	en / t	Finding the median mass for 100 women. Position of the median is $100 \div 2 = 50$ so find the mass of the 50 th person.	Continuous	Continuous data can have an infinite number of possible values within a selected range. (Can include decimal numbers).
Position of the largest length 28 median - smallest 24 $40 \div 2 = 20^{th}$ length $= 55 - 30 = 20^{th}$	70- 60-	Z Y Men	Read from 50 on the cumulative frequency axis to the value on the mass axis. 1) Median mass of women =	Quantitative	Quantitative data that can be counted (discrete), quantitative date that can be measured (continuous)
18 25cm	50 Median 40- 30- 20-		65kg Median mass of men = 73kg	Mode	The number which appears most often in a set of numbers
$\begin{array}{c} 10 \\ 8 \\ 6 \\ 4 \\ 2 \\ 0 \\ 0 \\ 30 \\ 35 \\ 40 \\ 45 \\ 50 \\ (55) \\ 60 \\ \end{array}$	10- 0- 30 40 50 60 70 Mass (kg	0 80 90 100 g)	On average , the women are lighter than men Range of masses = largest	Median	Place the numbers in value order and then find the middle number. When there are two numbers in the middle, we find the average them.
Length (cm) The upper class boundaries for this table	2) Range of women's masses Range of men's masses = 100 -		mass (read from the final point) subtract the mass from where the line starts	Range	The difference between the highest and lowest values.
are 35, 40, 45, 50 and 55.	The men's masses vary more th	nan the women's masse	es	Outlier	A point that "lies outside" (is much smaller or larger than) most of the other values in the dataset.
Length (cm) Frequency Cumulative frequency	<u>Exam Tips!</u>			IQR	The spread of the middle 50% of
30 ≤ I < 35	Be sure to label the axis "cur		t just "trequency"	(interquartile	data. A smaller IQR shows that
35 ≤ I < 40	Note how the graphs don't h	-		range)	the data is consistent .
40 ≤ I < 45 11 25 (4 + 11 = 25)	Smooth curve going to throu	c		Other Jonie	s/Units this could appear in:
45 ≤ I < 50 12 37 (25 + 12 = 37)	When making a comparison from the graph (comparison				· · · ·
50 ≤ I < 55	from the graph (comparing	ine mealans or IQR IN C	oniexi of the question!	A Level	Statistics-Data Collection 28



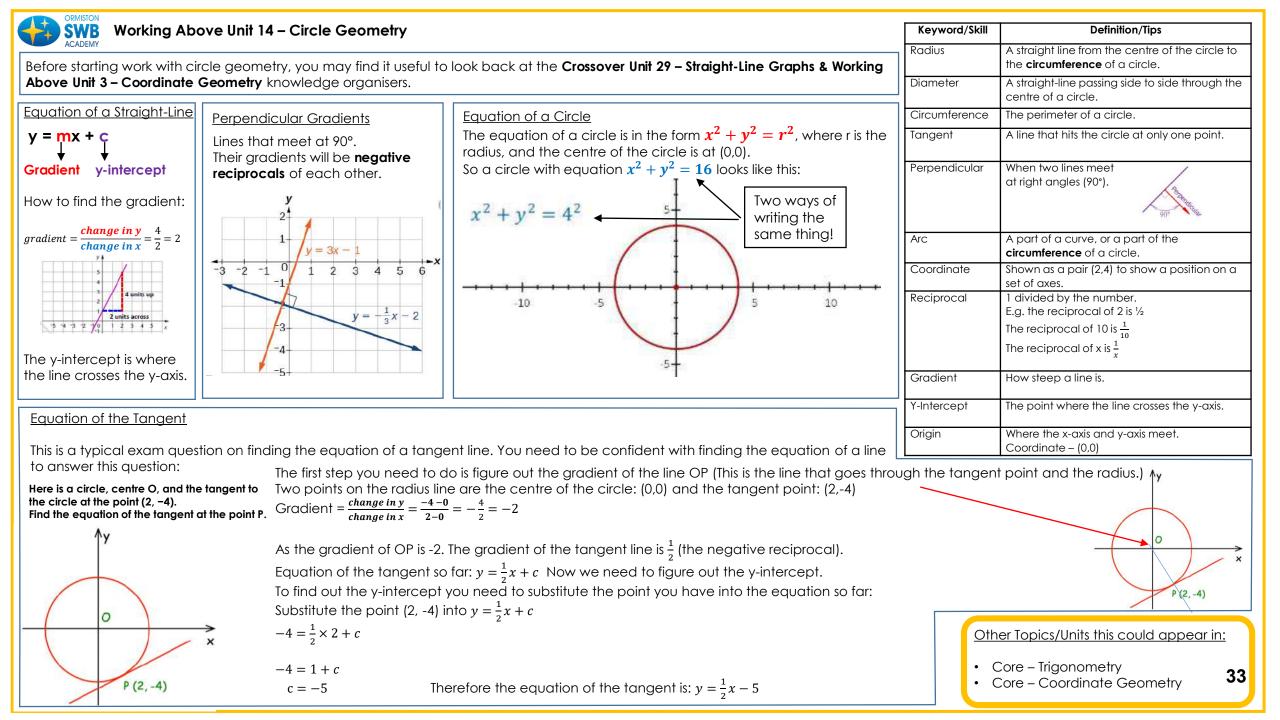
Working Above Unit 12c – Histograms	Before starting work with quartiles, you may find it useful to look back at the Crossover Unit 23 Averages from a table and U24 Averages from a group	Keyword/Skill	Definition/Tips
Interpreting Histograms – Finding the median	table knowledge organiser.	Sample	A group from the population that we are testing.
Sometimes you are given the histogram and you need to	o interpret it. The histogram below gives information on the speed (mph) of cars on a motorway. Find the median class.	Population	The whole group from where the sample is taken, i.e. a whole year group.
4 30 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	Frequency = Frequency Density x Width The area of each bar is the frequency, height (frequency density) multiplied by the width (class interval)	Discrete data	Can only take exact values (usually collected by counting), for example the number of students in a class. Bar graphs represent discrete data.
20 6 5 8.5 x 30 5	Speed (mph)FrequencyCF $0 \le x < 30$ 2424 $30 \le x < 50$ 116140 $50 \le x < 80$ 255395	Continuous data	Can take any value (collected by measuring), for example the heights of the students in a call – use continuous data. Histograms represent continuous data.
4 8.5 3 5.8 9	$80 \le x < 120$ 48 443 Position of the median = $total frequency + 1$ 2	Quantitative	Quantitative data that can be counted (discrete). Quantitative date that can be measured (continuous).
2 4 30	Position of the median = $\frac{443 + 1}{2}$	Outlier	A point that "lies outside" (is much smaller or larger than) most of the other values in the dataset.
0.8 0.8 x 30 1.2 1.2 1.2 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0	(Find this value by locking at the sumulative frequency column)	IQR (interquartile range)	The spread of the middle 50% of data. A smaller IQR shows that the data is consistent .
Speed (mph) Frequency Mid point fx	Estimating the mean from a histogram is the same as estimating the mean from a grouped frequency table.	Frequency density	Frequency Class width It is labelled as the y axis on a histogram.
	Estimated mean = $\underline{\text{Total fx}}$ Total f = $\underline{26375}$ 443 = 59.5 mph (1 d.p) Find the mid points and calculate a new fx column		ics/Units this could appear in: al Statistics- Data Collection 30

& Working Above Unit 6 - Quadratics	you may find it useful to refer to Crossover Unit 30 - Quadratic and Cubic Graphs	Keyword/ Skill	Definition/tip
Sketching Quadratic Curves Roots – where y = 0	E.g. Sketch the graph of $y = x^2 - 2x - 35$ clearly stating the roots, the y Intercept and the minimum point of the curve. Step 1: Find the roots by factorising and the solve the Quadratic Equation $x^2 - 2x - 35 = 0$ (x - 7)(x + 5) = 0 x = 7 and $x = -5$ are the two roots	Quadratic	Where the highest power of the variable (usually x) is squared . e.g, x^2 , $x^2 + 5x$, $x^2 - 2x - 8$ etc.
(-5, 0) (7, 0)	Step 2: Find the y coordinate of the y intercept by putting $x = 0$ into $x^2 - 2x - 35$ which gives a y intercept of (0, -35) Step 3: We can find the minimum coordinate of the curve by 'completing the square' $x^2 - 2x - 35$ $(x - 1)^2 - 35 - 1$	Cubic	Where the highest power of the variable (usually x) is cubed. e.g, x^3 , $x^3 + 5x$, $x^3 + 2x^2 - 5x + 2$ etc.
(0, -35)	(x - 1) ² – 36 Minimum point at (1 , -36)	Factorise	Write an expression as a product o its factors.
y-intercept –	Every Tip: These questions can be warth up to 5// marks	Root	Where a function equals zero.
where x = 0 Minimum Point	Exam Tip: These questions can be worth up to 5/6 marks Make sure you are familiar with the different techniques and hey presto, 5/6 marks in the bag.	Y Intercept	The point where a line or a curve crosses the y-axis of a graph.
ketching Cubic Curves - For GCSE, onl	y the roots (x intercepts) and the y intercept are required for cubic graphs.	Function	A mathematical relationship between two variables.
x 10,0) (0,0) (3,0)	Eg. Sketch the graph of $y = x^3 + 7x^2 - 30x$ Step 1: Factorise and solve the equation $x^3 + 7x^2 - 30x = 0$ $x(x^2 + 7x - 30) = 0$ so immediately we know one root is 0 Now solve the equation $x^2 + 7x - 30 = 0$ for the other(s) (x + 10)(x - 3) = 0 x = -10 and $x = 3There are 3 roots which are -10, 0 and 3Step 2: Find the y coordinate of the y intercept by puttingx = 0 into x^3 + 7x^2 - 30x which gives a y intercept of (0, 0)Note: In this example, the y intercept is also a root. This will not alwaysbe the case.$	 Quadrati 'A' Level Core - Dir - In 	<u>cs/Units this could appear in</u> : c and Cubic Graphs : fferentiation tegration cs - Kinematics



B Working Above Unit 13a & 13b – Quadratic/Cubic Graphs & Area under a Graph

Area under a Graph	- In a Velocity/Time (Speed/Time) gra	E.g. The velocity of a sle	edge as it slides down a hill	Keyword/ Skill	Definition/Tip
10 A B 9 -	C D E F G	travelled by the sle	oh. Estimate the distance edge. Use as many 4 s possible in your estimate.	Velocity	Speed (how fast something is moving) in a particular direction.
8		As we are estimatir	ng the distance travelled,	Acceleration	How fast velocity changes.
Velocity (m/s)		Area of Triangle Area of Trapezium	$A \frac{4 \times 5}{2} = 10$ $B \frac{4 \times (5+9)}{2} = 28$	Estimate	To find a value that is close enough to the right answer usually involving a calculation.
2			$C\frac{4 \times (9 + 8.5)}{2} = 35$	Trapezium	A flat shape with 4 straight sides that has one pair of parallel sides. Area is given by $\frac{a+b}{2} \times h$ where
		Area of Trapezium	$D \frac{4 \times (8.5 + 7)}{2} = 31$		a and b are the parallel sides and h is the distance between them.
0 2 4 6 8 Find the acceleration of t	10 12 14 16 18 20 22 24 26 28 30 Time (s) he sledge when t = 6s.	Area of Trapezium	$E\frac{4\times(7+3)}{2}=20$	Tangent	A line that just touches a curve at one point.
10		Area of Trapezium	$F\frac{4\times(3+0.5)}{2}=7$		
9• 8• 60 7•		Area of Triangle	$G \frac{(0.5 \times 2)}{2} = 0.5$	Coordinate	
Velocity (m/s)			a under the graph is 20 + 7 + 0.5 = 131.5 d is 131.5 metres.	• Real-Life Gr	raphs
	q a	work out the acceleration or decoint draw a tangent at that point. cceleration of the sledge 6 second $\frac{change in y}{change in x} = \frac{7m/s}{8s} = 0$	For instance, the ds into its journey is:	sections to split t 4 sections of Tria When calculatin	question does not specify how many he graph into, try not to use more than ngles, Trapeziums and/or Rectangles. g acceleration after drawing the right angled triangle as shown calculation.



ACADEMY	er Unit 6 & 7 – Best Buys & Exchange		Keyword/Skill	Definition/Tips
est Buys: Work out h	ow much money one item costs. The k	best value is the cheapest . Step 1 - For offer A, Convert 1.2kg to grams by multiplying by 1000 so that we have both offers in grams.	Best Buy	The cheapest price per item/unit OR the highest number of units/price .
		Step 2 - If 1200g of Apples costs 389p (or \pounds 3.89) we can work out 1g by dividing the grams and the money by 1200.	Direct Proportion	2 variables change at the same rate
		Step 3 - For offer B, If 700g costs 214p (or $\pounds2.14$) we can work out 1g by dividing the grams and the money by 700.	Unitary Method	Finding the cost of a single unit OR finding the amount of units per eg. :
		Step 4 - See which is the lowest cost per gram and that is the best		
1.2kg for £3.89	700g for £2.14	value.	Exchange Rate	The price of one currency in terms of another currency.
1.2kg = £3.89				Eg. £1 = \$1.25
1200g = 389p 1200 lg = 0.32p	\div 1200 \div 700g = 214p \div 700 1g = 0.31p \div 700			For each pound I have to spend I c buy \$1.25
fer B is the best val	ue.	offer B, $1g = \pounds 0.0031$. The answer is still the same. Offer B is cheapest.	Multiply	Repeated addition.
			·	
2. Which is better vo	alue?	Step 1 - At Tresco, the offer is 3 bottles of 500ml for the price of 2 bottles. Therefore 3 x 500ml costs $2 \times \pounds1.60$	<u></u>	
			Divide	To split into equal parts or groups.
	rmarket Sells 500ml bottles for £1.60	Step 2 – So if 1500ml of Shampoo costs \pounds 3.20, we can work out 1ml by dividing the ml and the money by 1500.		
5	3 for price of 2	Step 3 - At ASDER, the offer is buy 1 and get 1 free. Therefore we get 2 x 300ml for 1 x \pounds 1.50	Changing Format	If the rate is given in the format (for instance) $1 = £0.85$
	narket sells 300ml bottles for £1.50 buy one get one free	Step 4 - If 600ml costs \pounds 1.50 we can work out the cost of 1ml by		Divide both numbers by 0.85 to get the exchange rate in the format £1
-		dividing the ml and the money by 600.		\$1.18
SCO: x 3	$500ml = \pounds 1.60 \times 3$ $500ml = \pounds 3.20 \div 1500$			
÷ 1500	$1ml = \pounds 0.0021$	Step 5 – Now see which is the lowest cost per ml of shampoo and that is the best value. – In this example TRESCO is the	Other Topie	cs/Units this could appear in:
DER: x 2	$300ml = \pm 1.50$	cheapest/best value.	• Direct a	nd Inverse Proportion
÷ 600			Similarity	
	$1 m l = \pm 0.0025$		Mensure Fraction	ation Is and Percentages
	ue			is and Percentages

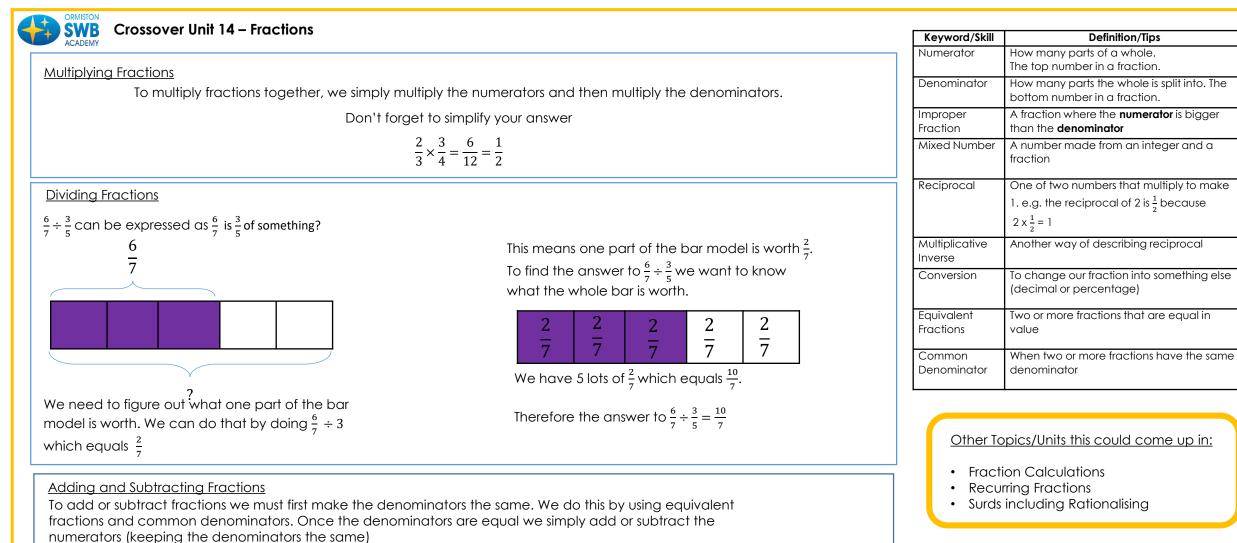
SWB Crossover Unit 6 & 7 – Best Buys & Exchange Rates		Keyword/Skill	Definition/Tips
Exchange Rates: One currency can be exchanged for another current	Best Buy	The cheapest price per item/unit OR the highest number of units/price .	
John went to Paris on holiday. He changed £1450 into Euros at Manchester Airport before his flight to Paris. The exchange rate at the airport was £1 = €1.14. On the holiday he spent €1355. On his return to England he changed his remaining euros back into pounds. Work out how much he returned to England with. Give your answer in pounds to the nearest penny. $\pounds 1 = \pounds 1.14$ $\pounds 1450 = 1450 \times 1.14 = \pounds 1653$	Therefore, for £2 he gets $2 \times \notin 1.14$, for £3 he gets $3 \times \notin 1.14$ etc so for £1450 he gets $1450 \times \notin 1.14 = \notin 1653$ Step 2 - On holiday in Benidorm he spends $\notin 1355$ so the next thing we do is subtract 1355 from 1653 which is $\notin 298$ Step 3 - When we changed from pounds to euros we used multiplication, so to convert back from euros to pounds we do the inverse of multiplication ie. division. Therefore the calculation is $298 \div 1.14 = \pounds 261.4035$ which	Direct Proportion Unitary Method Exchange Rate	another currency.
€1653 - €1355 = €298 298 ÷ 1.14 = £261.4035 = £261.40 (nearest penny)	is £261.40 to the nearest penny Note: When the exchange rate is given as £1 = "other currency" then GBP Other currency x exchange rate GBP Other currency ÷ exchange rate Otherwise: Reverse the operations. See eg2.	Multiply	Eg. £1 = \$1.25 For each pound I have to spend I can buy \$1.25 Repeated addition
Elaine is going to New York. She wants to change £500 into	tep 1 - If Elaine changes her money in the UK, for every £1 she gets \$1.26	Divide	To split into equal parts or groups
UK : 500 x 1.26 = \$630 US (New York) : 500 ÷ 0.81 = \$617.28	£2 she gets 2 x \$1.26 £3 she gets 3 x \$1.26 etc o for £500 she gets 500 x \$1.26 = \$630 ep 2 - If Elaine changes her money in New York, even though he is changing pounds to dollars the calculation is "divided by	Changing Format	If the rate is given in the format (for instance) $1 = £0.85$ Divide both numbers by 0.85 to get the exchange rate in the format £1 = \$1.18
She should change her money in the UK because she will get more dollars.	the exchange rate" because the exchange rate in New York is ven as $\$1 = \pounds0.81$ ams! As you can see, these questions are all about money, so make sure in an exam you use the correct units, e.g £ or \$ etc. You also need to make sure all of your answers are rounded to 2.d.p, as money is always in this form.	Drawin graphs	ions and substitution into simple

Intercentage of an Amount (without a calculutor) Percentage increase/decrease (without a calculutor) Percentage (without a calculutor) 1. Percentage of an Amount (without a calculutor) 2. Percentage increase/decrease (without a calculutor) Percentage (without a calculutor) 1. Percentage of an Amount (without a calculutor) 2. Percentage increase/decrease Percentage (without a calculutor) 1. Percentage increase/decrease Percentage of the given amount. Percentage (without a calculutor) 1. Percentage increase/decrease Percentage of the given amount. Percentage (without a calculutor) 1. Percentage increase Percentage of the given amount. Percentage (without a calculutor) 1. Percentage increase Percentage of the given amount. Percentage for the percentage of the given amount. 1. Percentage increase Percentage for the percen	Crossover Unit 10, 11 and 13 – Percenta	ae of an Amount, Interest, Growth and	d Decay, Reverse Percenta	ages	Keyword/Skill	Definition/Tips
$\frac{100}{9} + \frac{2}{10} + \frac{2}{10}$	ACADEMY	2. Percentage increase/decrease (with	iout a calculator)		ercentage	
$\frac{1}{10^{6}} = \frac{1}{10^{6}} = \frac{1}$	$\frac{-2}{50\%}$ $\frac{-2}{25\%}$	Then you add or subtract this amount depending on whether you are		are Fr	raction	whole. They are written as one number over another.
$\frac{1}{100}$ $\frac{1}$	$= 1$ whole $= \frac{1}{2}$ $= \frac{1}{4}$					be positive or negative. 3746.374
We can use combinations of these key percentages to find any percentage. For example: $30\% = 10\% \times 3$ 75% = 25% + 5% 16% = 10% + (10% + 2) + 1% There we can use percentage of an Amount (wh a calculator) Here we can use percentage multipliers. First of all you need to find the decimal equivalent of the percentage multiplier. For example 1: First of all you need to find the decimal equivalent of the percentage multipliers. 2% = 103 12% = 103 12% = 103 12% = 10% 12% = 1	÷10	50% of \$80 = \$40	1% of 500a = 5a		•	-
We can use combinations of find any percentages to find any percentage. Source and the amount for the starting $30\% = 10\% \times 3$ and $30\% \times 3$						
10% these key percentages to find any percentage. Sold = 10% x 3 Sold = 10% x 1 Sold = 10% x 3 Sold = 10% x 3<	We can use combinations of	men daa mis erne me staming ameerin				
$30\% = 10\% \times 3$ $75\% = 25\% + 50\%$ $16\% = 10\% \times 10\% + (10\% + 2) + 1\%$ $\frac{1}{100}$ $\frac{1}{16\%} = 10\% + (10\% + 2) + 1\%$ $\frac{1}{100}$ $\frac{1}{16\%} = 10\% + (10\% + 2) + 1\%$ $\frac{1}{100}$ $\frac{1}{16\%} = 10\% + (10\% + 2) + 1\%$ $\frac{1}{100}$ $\frac{1}{16\%} = 10\% + (10\% + 2) + 1\%$ $\frac{1}{100}$ $\frac{1}{1000}$ $\frac{1}{100}$	10% these key percentages to find any percentage.	\$80 + \$40 = \$120	amount	sidning	IOIII	than it cost to buy. It is the difference between the amount earned and the
$\frac{1}{10} 16\% = 10\% + (10\% \div 2) + 1\%$ $\frac{1}{10} 16\% = 10\% + (10\% \div 2) + 1\%$ $\frac{1}{10} \frac{1}{10} \frac{1}{1$	30% = 10% x 3	500g - 15g = 485g			OSS	
$\frac{1\%}{100}$ 1		Here we can use percentage multipliers . First of all you need to find the decimal equivalent of the percentage you			nterest	Money paid regularly at a particular rate.
$\frac{176}{100}$						The number you multiply a quantity by to
$\frac{1}{100}$ $\frac{1}$					nultiplier	
percentage multipliers.Example 1:find 25% of ± 120 find 45% of ± 250 find 48% of ± 250 To find 25% you divide by 2 then divide by 2 again (or $\div 4$) $50\% = 0.3$ $48\% = 0.48$ (this is the percentage multiplier) $\pm 120 \div 4 = \pm 30$ $30\% = 0.3$ $48\% = 0.48$ (this is the percentage multiplier) $50, 25\%$ of ± 120 is ± 30 $50, 48\%$ of ± 250 is ± 120 The increase in the value or amount of something over time. (Car prices are a commole)Example 2:Find 60% of 300kg 60% of 300kgTo find 50% you divide by 2 $50\% + 10\%$ To find 50% you divide by 2 $50\% + 10\%$ To find 10% you divide by 2 $50\% + 10\%$ To find 10% you divide by 2 $50\% + 10\%$ To find 10% you divide by 10 $50 + 10\%$ So 10\% = 30kg $16\% = 1.16$ Therefore 60% of 300kg = 180kg $50 + 10\%$ Therefore 60% of 300kg = 180kg $16\% = 1.16$ Therefore 60% of 300kg = 180kg $16\% = 1.16$ Therefore 60\% of 300kg = 180kg 16%		You pood to us			imple Interest	
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So to increase 480 by 16% So to decrease 725 by 26% So						26
480 X 1.16 = 556.8 /25 X 0./4 = 536.5		,				30
		480 X 1.16 = 556.8		$725 \times 0.74 = 536$	6.5	



Crossover Unit 10, 11 and 13 – Percentage of an Amount, Interest, Growth and Decay, Reverse Percentages

5. Interest, Growth an	d Decay	6. Reverse Percentage	
Simple Interest (Growth)	A £200 loan earns 15% simple interest over 5 years. How much will be owed at the end of the 5 years? Work out 15% of £200 = £30	value after the increas Before we do this, it is	help us to calculate the <u>original</u> price or value of something, when we only know the price or se or decrease has taken place. mportant we know that 100% represents the whole amount or the full price of something. ased by 20%, the amount we have now is worth 120%. If something is reduced by 5%, the amount 95%.
	$\pounds 30 \times 5 \text{ years} = \pounds 150$ $\pounds 200 + \pounds 150 = \pounds 350$	Non – Calculator	A shop has a 20% off sale. A shirt is now worth £24. What was the original price? So, £24 represents 80% of the value of the shirt
Compound Interest (Growth)	 A £200 loan earns 15% <u>compound</u> interest over 5 years. How much will be owed at the end of the 5 years? Here you need to use percentage multipliers. To <u>increase</u> by 15% five times (for each of the 5 years) you would multiply by 1.15 five times. A quick 	This could come up on a non- calculator paper, but they will usually give you nice numbers that will divide easily like the example here.	$100\% \longrightarrow ?$ $i = 4 \longrightarrow 20\% \longrightarrow £24 \longrightarrow 4$ $80\% + 20\% = 100\%$ $\pounds 24 + \pounds 6 = \pounds 30$
	way of writing this is by using indices. $\pounds 200 \times 1.15^5 = \pounds 402.27$	Calculator On a calculator	Some money has been put into a bank account with an interest rate of 4%. After a year, the total amount of money in the account was 291.20 . How much money was invested?
Compound	A car was brought for £12,000. It depreciates in value by 20% per year. How much will the car be worth after 3 years? Here you need to use percentage multipliers.	paper, the numbers are likely to be more difficult, so you will need to use percentage multipliers.	Starting x 1.04 End value ? $\pounds 291.20$ $\pounds 1.04 = \pounds 280$
Depreciation (Decay)	To <u>decrease</u> by 20% three times (for each of the 3 years) you would multiply by 0.8 three times. A quick way of writing this is by using indices.		phrase 'per annum' is used. This means the ear'. Annum is the Latin word for year.
	$\pounds12,000 \times 0.8^3 = \pounds6,144$	 Exams! A lot of these quest should always be reader to the second should always be reader to the second should always be reader to the second sec	 Direct and inverse proportion Ratio Ratio A-level Units



 $\frac{5}{7} - \frac{1}{3}$

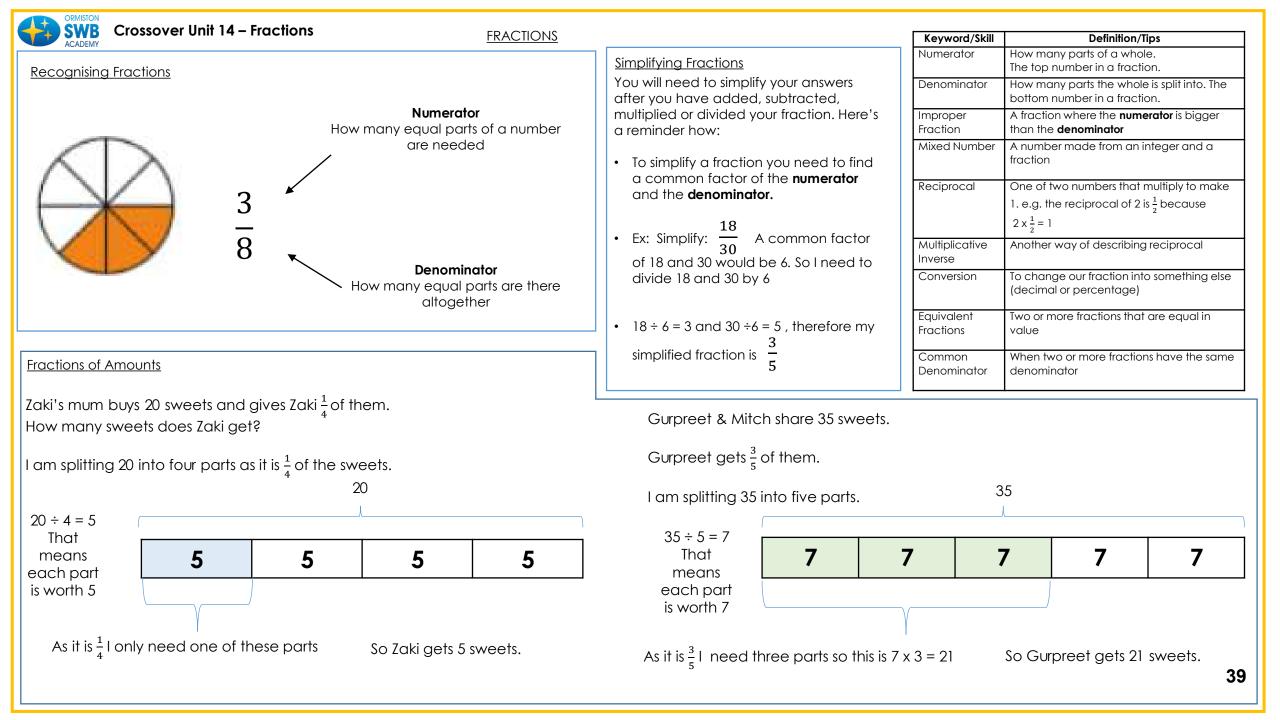
 $\frac{5}{7} = \frac{15}{21}$

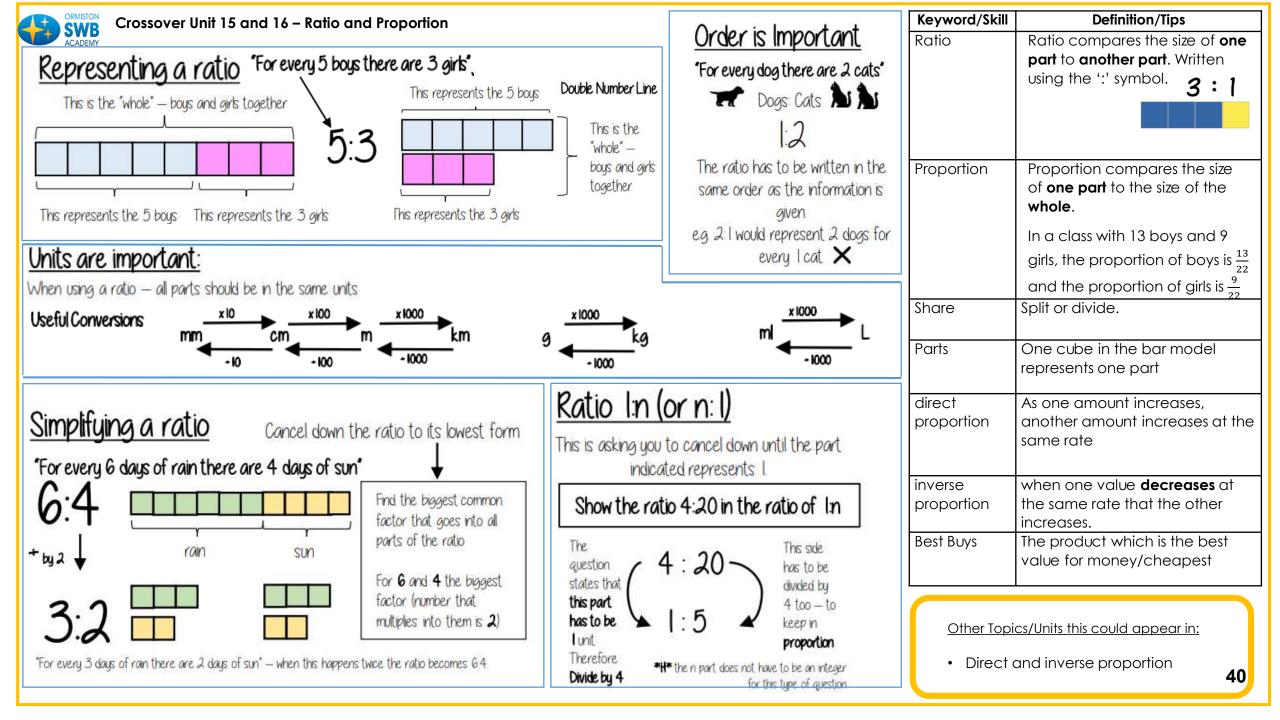
The common denominator will be 21

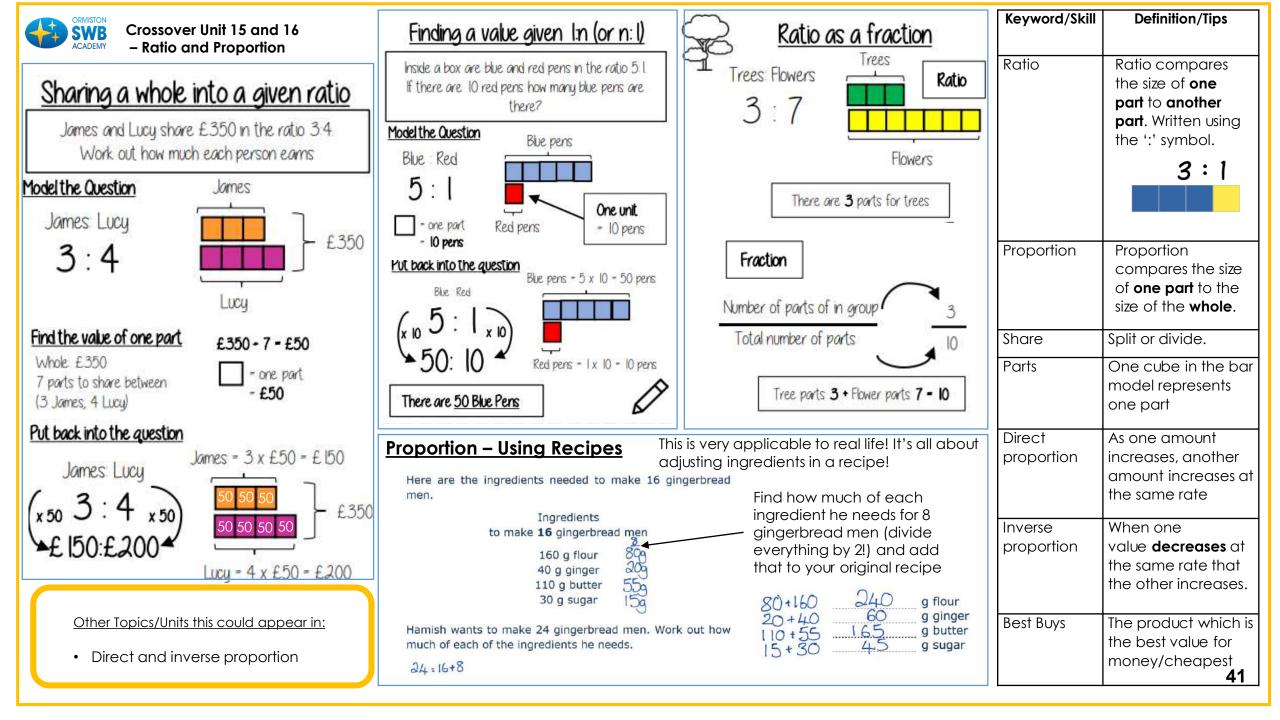
 $\frac{1}{3} = \frac{7}{21} \qquad \qquad \frac{5}{7} - \frac{1}{3} = \frac{15}{21} - \frac{7}{21} = \frac{12}{21}$

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 $\frac{1}{6} + \frac{4}{9}$ The common denominator will be 18

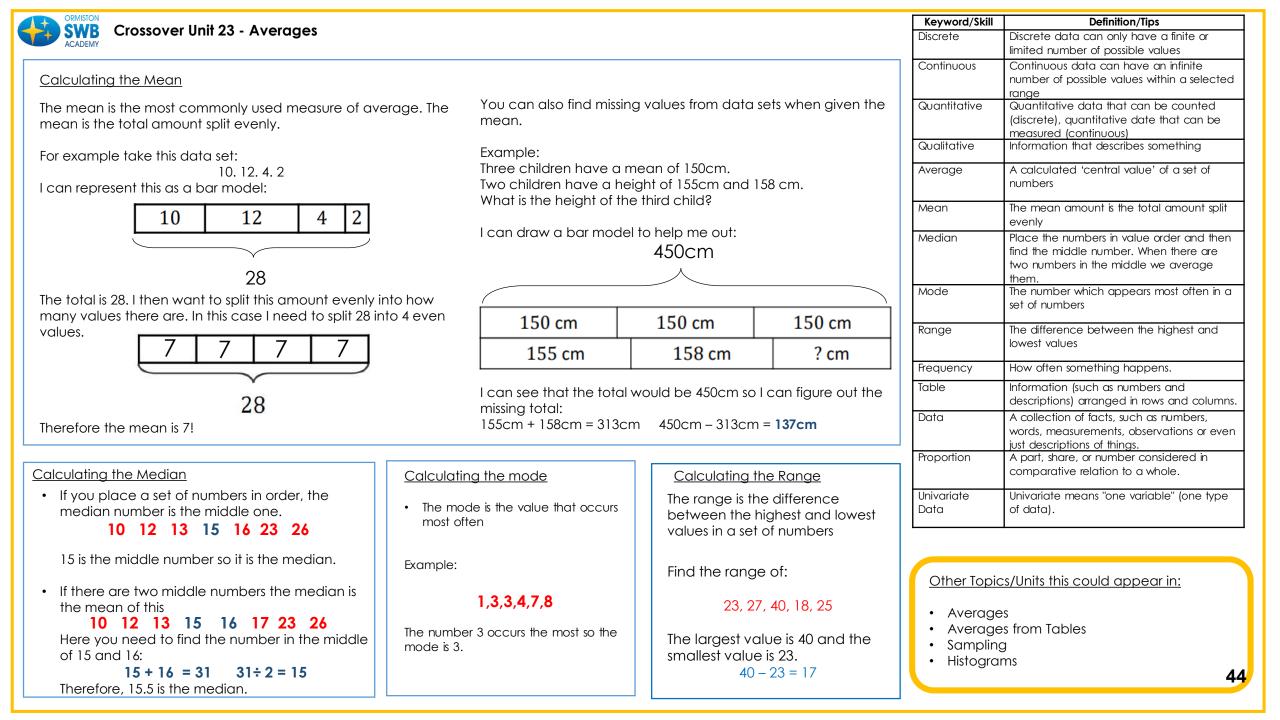






SWB Crossover Ur	nit 21 & 22– Solving Equations & 'Su	bject of'	Keyword/Skill	Definition/Tips
When we are solving equa	Expression	One or a group of symbols representing a number or a value. Can contain numbers, variables & operations		
		If the variable. $4 \times - 7 = 5$	Equation	Statement using an equals sign, to show two expressions are equal. $4x - \frac{7}{4} = 5$
		If I remove one apple from the left side, to keep it balanced I must do the same to the right side! We need to think like this when we solve equations.	Variable	A symbol for a number we do not know yet
One – Step Equations These are equations where	e you only need to do one	<u>Two – Step Equations</u> These are equations where you need to do two inverse	Operations	The four basic operations in maths: addition, subtraction, multiplication & division
inverse operation to solve		operations to solve the equations:	Inverse Operations	The operation that reverses the effect of another operation. Addition & subtraction are inverse
y + 14 = 20 -14 -14	x - 120 = 80	$ \begin{array}{cccc} \underline{Ex1} & 4x - 3 = 25 \\ & +3 & +3 \\ & 4x = 28 \end{array} $		operations Multiplication & division are inverse operations
-14 - 14 y = 6	+120 +120 x = 200	÷4 ÷4	Simplify	To remove unnecessary terms and numbers
y = 0	x = 200	x = 7	Formula	A rule or fact witten using mathematical symbols
<u>Ex3</u>	<u>Ex4</u>	$\frac{E \times 2}{5} + \frac{y}{5} + \frac{6}{6} = \frac{14}{-6}$	Solve	To find the answer/value of something
3n = 12	$\frac{k}{2} = 16$	$\frac{y}{5} = 8$ $x5 x5$	Rearranging Formulae	Use inverse operations on both sides of the formula until you find the expression/equation for the letter you need.
÷3 ÷3 n = 4	x2 x2 k = 32	y = 40	'Subject of'	A certain variable needs to be by itself on one side of the equal sign Example:
	lls you have already learnt to solve so	me equations. (If you need help expanding brackets look back at		x = 4y + 10 x is the subject of this formula
the Crossover Unit 19 – Expo Ex1	Expar	3(x + 4) = 27 and the brackets first 3x + 12 = 27 -12 - 12 3x = 15 $\div 3 \div 3$ x = 5	Formir Expan Simult Algeb Algeb	pics/Units this could appear in: ng and Solving Equations nding and Factorising aneous Equations praic Fractions praic Proof anging Equations 42

Crossover Unit 21 & 22– Solving Equations & 'Subject of'	Keyword/Skill	Definition/Tips
	Expression	One or a group of symbols representing a
Equations With an Unknown Variable on Both Sides		number or a value. Can contain numbers, variables & operations
Sometimes equations may have variables on each side of the equals sign. There is one extra step you need to do before you apply the same method for solving two-step equations. The extra step is whichever side has the smaller unknown variable, subtract that from both sides of the equation. You can then continue with the same method of using the inverse operations to solve the equation	Equation	Statement using an equals sign, to show two expressions are equal. 4x - 7 = 5
$\underline{Ex1} \qquad \qquad 8x + 12 = 5x + 30$	Function	A mathematical relationship between two
Here we will subtract $5x$ from both sides as $5x$ is the smaller variable	FUNCTION	A mathematical relationship between two values
8x + 12 = 5x + 30	Operations	The four basic operations in maths: addition, subtraction, multiplication & division
$ \begin{array}{ccc} -5x & -5x \\ 3x + 12 &= 30 \\ -12 & -12 \\ 3x &= 18 \end{array} $	Inverse Operations	The operation that reverses the effect of another operation. Addition & subtraction are inverse operations Multiplication & division are inverse operations
$\frac{1}{3}$ $\frac{1}{3}$	Simplify	To remove unnecessary terms and numbers
x = 6 Making the 'Subject of' (Rearranging Formulae)	Formula	A rule or fact written using mathematical symbols
In these types of questions, it will ask you to make a certain variable the 'subject of' the equation or formula. What this means is you need to use the solving equation method to isolate the variable it is asking for:	Solve	To find the answer/value of something
Ex1 Make y the subject of this formula: y + 81 = x + 100 -81 -81 Here y is the variable we want to isolate (have by	Rearranging Formulae	Use inverse operations on both sides of the formula until you find the expression/equation for the letter you need.
y = x + 19 itself on one side of the equals sign).	'Subject of'	A certain variable needs to be by itself on one side of the equal sign Example: x = 4y + 10 x is the subject of this formula
Ex2 Make x the subject of this formula: $C = 4x + 5y$ $-5y -5y$ $C - 5y = 4x$ $\div 4$ $\overleftarrow{+4}$	Formir Expan Simulte	<u>pics/Units this could appear in:</u> ng and Solving Equations ding and Factorising aneous Equations raic Fractions
When you are dividing an expression, just write it $\frac{C-5y}{4} = x$	Algeb	raic Proof anging Equations 43





Crossover Unit 24 – Averages From a Table

Finding the mode from a table

- The mode is the value that occurs most often.
- The mode is the only average that can have no value, one value or more than one value.
- When finding the mode, it helps to order the numbers first.

In this frequency table, the mode is the value with the highest frequency:

Shoe size	5	6	7	8	9
Frequency	2	5	11	4	1

The modal size is 7 because more people wear size 7 than any other size.

Finding the modal class from a grouped frequency table	The following table shows the weights of children in a class.
Mass (m) kg	Frequency
$30 \le m \le 40$	7
4 0 ≤ <i>m</i> < 50	6
$50 \le m \le 60$	8
$60 \le m < 70$	4

The modal class is the class that has the highest frequency. In this case the modal class is:

 $50 \le m < 60$

					Keyword/Skill	Definition/Tips
Tot	al Frequenc	<u>cy</u>			Discrete	Discrete data can only have a finite or limited number of possible values
	score	frequency	Andy thro	wa a diaa in an	Continuous	Continuous data can have an infinite number of possible values within a selected
	1	19		ws a dice in an ion. How would	Quantitative	range Quantitative data that can be counted
	2	18	,	out the total		(discrete), quantitative date that can be measured (continuous)
F	3	12	thrown the	f times he has e dice?	Qualitative	Information that describes something
┢	_				Average	A calculated 'central value' of a set of
┢	4	19			Aveidge	numbers
┝	5	9	You can	work out the	Mean	To calculate the mean, add up all of the
L	6	23	total frequ	Jency by		numbers and then divide by how many numbers there are
		100 🗡	adding u frequenc	-	Median	Place the numbers in value order and then find the middle number. When there are two numbers in the middle we average them.
Fir	nding the m	edian from a 1	able		Mode	The number which appears most often in a set of numbers
				er of hours a group of to the nearest hour.	Range	The difference between the highest and lowest values
	hours	frequency	cumulative frequency		Frequency	How often something happens.
	0	8	8	The 105.5 th value which is the	Table	Information (such as numbers and descriptions) arranged in rows and columns
	1	16	24	median would be in this category.	Midpoint	The middle point. The point halfway between.
	2	33	57			
	3	75	132 🗡	There are	e 200 data items	s, so the median must lie
	4	44	176	betweer	items 100 and	101
	5	8	184	n	+1	$n = \frac{200+1}{2} = \frac{201}{2} = 105.5$
	6	16	200	median -	2	2 2 2
		ne median val equency colu	ue, first work ou mn .	ut the Then use thi	s to help you to	e the 105.5th value in the table work out where this data 4 ive frequency column

						Keyword/Skil	Definition/Tips
Calculating the Mean from a Table To find the mean in this					• From the table, we can see that		A selection taken from a larger group 'the population' that will let you find out things about a larger group.
The mean is found adding up all the	-		•	otal number of found and then	for 2 games, no goals were scored. This makes a grand total	Population	The whole group being studied
and dividing by h numbers there ar	now many	divide game	,	number of	 of zero goals so far. The rest of the total amount of goals can be worked out in this 	Stem and Lea	'leaf' and a 'stem'. 'Stem' values are listed down and 'leaf' values are listed next to
	Number of Goals (x)	Frequen	cy <i>(</i> f)	fx	way, by multiplying goals (x) by the frequency (f). Call this column fx (f multiplied by x)	Pie Chart	them. Graph using a divided circle where each section represents part of the total.
	0	2		0 x 2 = 0		Estimate	To make an approximate or rough
	1	0		1 0 0			calculation often based on rounding.
	1	2		1 x 2 = 2	The total number of goals is 15. There were 10 football	Primary	Primary data is data that is collected by a data researchers from first hand sources.
	2	5		2 x 5 = 10	games so	Secondary	Secondary data is data gathered from
	3	1		3 x 1 = 3	15 ÷ 10 = 1.5 The mean number of goals is	Secondary	studies, surveys or experiments run by other people or for other research.
Total		10		15	1.5 goals per game.	Interval	An interval is between two points of values An interval may or may not include start and end points.
		Total number of football game		Total number of goals.	Remember to divide fx by the total of the frequencies, not by	Survey	To gather information by individual sample so we can learn about the whole thing.
	L				the amount of different items of data – the correct answer	Sort	To arrange or group in a special way (such as by size, type or alphabetically).
x	f	fx	<u>Further I</u>	<u>Example</u>	here is $\frac{15}{10}$ not $\frac{15}{4}$		
1	15	15		le shows the			
2	27	54		r of parking spaces use in a street.	Finding the Range	Othe	r Topics/Units this could come up in:
3	8	24		ut the mean r of spaces	The range is the difference between the highest and lowest values in a set of		verages
4	5	20			numbers. <- Using this table as an example:		ode, median, range and mean umulative frequency
TOTALS:	55	113	Mean =	$\frac{113}{55}$ = 2.05	The highest value is 4 and the lowest value is 1. Range = $4 - 1 = 3$	e	4

Crossover Unit 25 - Inequalities		Keyword/Skill	Definition/Tips
Inequality Symbols		Integer	Whole number including 0 and negative numbers. No fractions or decimals.
Equality and		Inequality	Compares two values showing if one is less than, greater than or not equal to each other.
Inequality		Greater than	One number is BIGGER than another number.
Thedowind	¥ A	Less than	One number is SMALLER than another number.
	greater >> greater than	Equal to	Two things have the SAME value.
equal >	than or equal	Equation	Says that two things are equal. (1 + 1 = 2).
🕇 not equal 🧹	less than so regual	Satisfy	A value that solves an equation. E.g. $2x + 1 = 9$ x = 4 so $x=4$ satisfies the equation.
	or equal	Variable	A symbol for a number we don't know yet, usually a letter.
Examples: x < 5 means x is less than 5		Coefficient	A number used to multiply a variable. E.g. 6y = 6 x y . y is the variable and 6 is the coefficient.
	Other Topics/Units this could appear in:	Inverse	Opposite of (i.e. x and ÷, + and -)
$p \ge 100$ means p is greater than or equal to 100	 Numbers, powers, roots, decimals and rounding Expressions and substituting into a formula Expand and simplify Solving equations 	Solve	Find all of the values that satisfy the inequality.
y > -2 means y is greater than -2			47

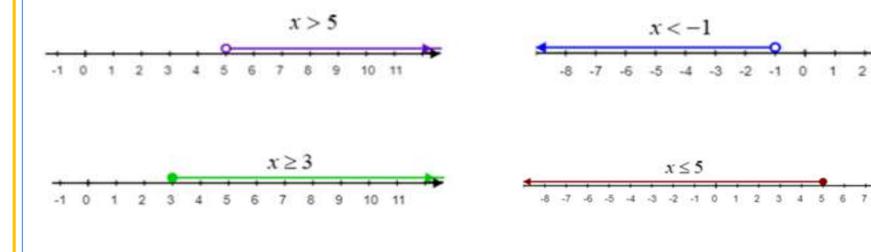


Inequalities Symbols on a Number Line

Symbol	Circle	Direction of Arrow
<	Open 🔘	Left
>	Open 🔘	Right
≤	Closed	Left
≥	Closed	Right

We use open and/or closed circles to represent inequalities on a number line. A closed circle means that the number is included in the represented group of values. An open circle means that the number is not included in the represented group of values.

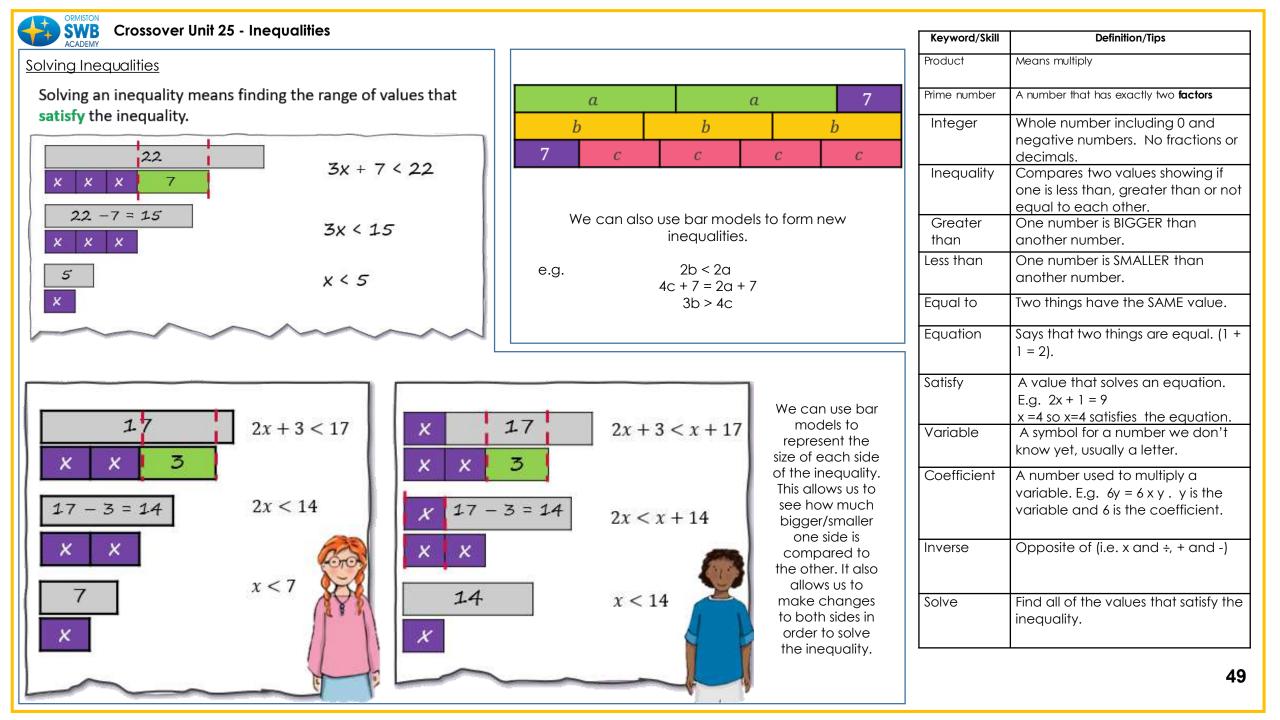
Examples:



negative numbers. No fractio or decimals.InequalityCompares two values showing one is less than, greater than not equal to each other.Greater thanOne number is BIGGER than another number.Less thanOne number is SMALLER than another number.Equal toTwo things have the SAME valueEquationSays that two things are equal (1 + 1 = 2).SatisfyA value that solves an equation E.g. 2x + 1 = 9 x =4 so x=4 satisfies the equation.VariableA symbol for a number we do know yet, usually a letter.CoefficientA number used to multiply a variable and 6 is the coefficientInverseOpposite of (i.e. x and ÷, + and	Keyword/Skill	Definition/Tips
one is less than, greater than a not equal to each other.Greater thanOne number is BIGGER than another number.Less thanOne number is SMALLER than another number.Equal toTwo things have the SAME valueEquationSays that two things are equal (1 + 1 = 2).SatisfyA value that solves an equation E.g. 2x + 1 = 9 x =4 so x=4 satisfies the equation.VariableA symbol for a number we do know yet, usually a letter.CoefficientA number used to multiply a variable. E.g. 6y = 6 x y . y is the variable and 6 is the coefficientInverseOpposite of (i.e. x and ÷, + and	Integer	Whole number including 0 and negative numbers. No fractions or decimals.
another number.Less thanOne number is SMALLER than another number.Equal toTwo things have the SAME valueEquationSays that two things are equal (1 + 1 = 2).SatisfyA value that solves an equation E.g. 2x + 1 = 9 x = 4 so x=4 satisfies the equation.VariableA symbol for a number we do know yet, usually a letter.CoefficientA number used to multiply a variable. E.g. 6y = 6 x y . y is the variable and 6 is the coefficientInverseOpposite of (i.e. x and ÷, + and	Inequality	Compares two values showing i one is less than, greater than or not equal to each other.
another number.Equal toTwo things have the SAME valueEquationSays that two things are equal (1 + 1 = 2).SatisfyA value that solves an equation E.g. 2x + 1 = 9 x =4 so x=4 satisfies the equation.VariableA symbol for a number we do know yet, usually a letter.CoefficientA number used to multiply a variable. E.g. 6y = 6 x y . y is the 	Greater than	
EquationSays that two things are equal (1 + 1 = 2).SatisfyA value that solves an equation E.g. 2x + 1 = 9 x =4 so x=4 satisfies the equation.VariableA symbol for a number we do know yet, usually a letter.CoefficientA number used to multiply a variable. E.g. 6y = 6 x y . y is the variable and 6 is the coefficientInverseOpposite of (i.e. x and ÷, + and	Less than	
(1 + 1 = 2).SatisfyA value that solves an equationE.g. 2x + 1 = 9x = 4 so x=4 satisfies the equation.VariableA symbol for a number we do know yet, usually a letter.CoefficientA number used to multiply a variable. E.g. 6y = 6 x y . y is th variable and 6 is the coefficientInverseOpposite of (i.e. x and ÷, + and	Equal to	Two things have the SAME value
E.g. $2x + 1 = 9$ $x = 4$ so $x = 4$ satisfies the equation.VariableA symbol for a number we do know yet, usually a letter.CoefficientA number used to multiply a variable. E.g. $6y = 6 \times y$. y is the variable and 6 is the coefficientInverseOpposite of (i.e. x and \div , + and	Equation	Says that two things are equal. $(1 + 1 = 2)$.
Know yet, usually a letter.CoefficientA number used to multiply a variable. E.g. 6y = 6 x y . y is th variable and 6 is the coefficientInverseOpposite of (i.e. x and ÷, + and fill)	Satisfy	x =4 so x=4 satisfies the
variable. E.g. 6y = 6 x y . y is the variable and 6 is the coefficient Inverse Opposite of (i.e. x and ÷, + and	Variable	A symbol for a number we don know yet, usually a letter.
	Coefficient	A number used to multiply a variable. E.g. 6y = 6 x y . y is the variable and 6 is the coefficient
Coluce Find all of the values that eatin	Inverse	Opposite of (i.e. x and ÷, + and
the inequality.	Solve	Find all of the values that satisfy the inequality.

0

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Crossover Unit 26 – Frequency Diagrams

Frequency Diagrams

Frequency Diagrams

A **frequency diagram**, often called a line graph or a frequency polygon, shows the frequencies for different groups of data.

Line Graphs (for discrete data)

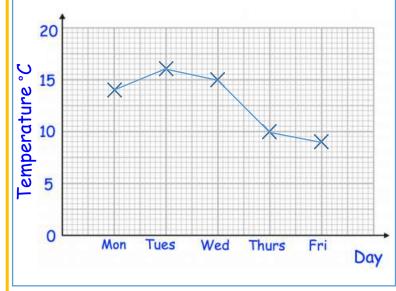
Example

The table below shows the average temperature in Belfast each day.

	Belfast
Monday	14°C
Tuesday	16°C
Wednesday	15°C
Thursday	10°C
Friday	9°C

The line graph below shows the results of the table.

A point has been plotted at the correct frequency for each day. And the points are joined with straight lines.



Frequency Polygons (for continuous grouped data)

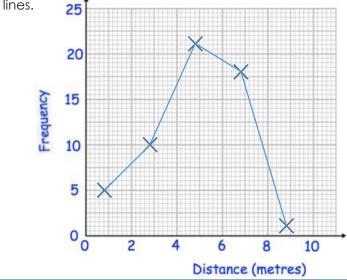
Example

The table gives information about the distances thrown, in metres, at a school sports day.

Time (seconds)	Frequency
0 < d ≤ 2	5
2 < d ≤ 4	10
4 < d ≤ 6	21
6 < d ≤ 8	18
8 < d ≤ 10	1

The line graph below shows the results of the table.

A point has been plotted at the correct frequency for each day, above the **midpant 6** each **g** oup And the points are joined with straight lines.



Keyword/Skill	Definition/Tips
Discrete	Discrete data can only have a finite or
	limited number of possible values.
Continuous	Continuous data can have an infinite
	number of possible values within a selected
	range.
Qualitative	Data categories like food, clothes and
	hobbies.
Quantitative	Data that can be counted or measured.
Derter	
Data	Collection of information.
Sample	Section of a whole group.
Population	Whole set of individuals, items or data from
Population	
	which a statistical sample is drawn.
Frequency	The number of times an item appears in a
	set of data.

<u>Exams!</u>

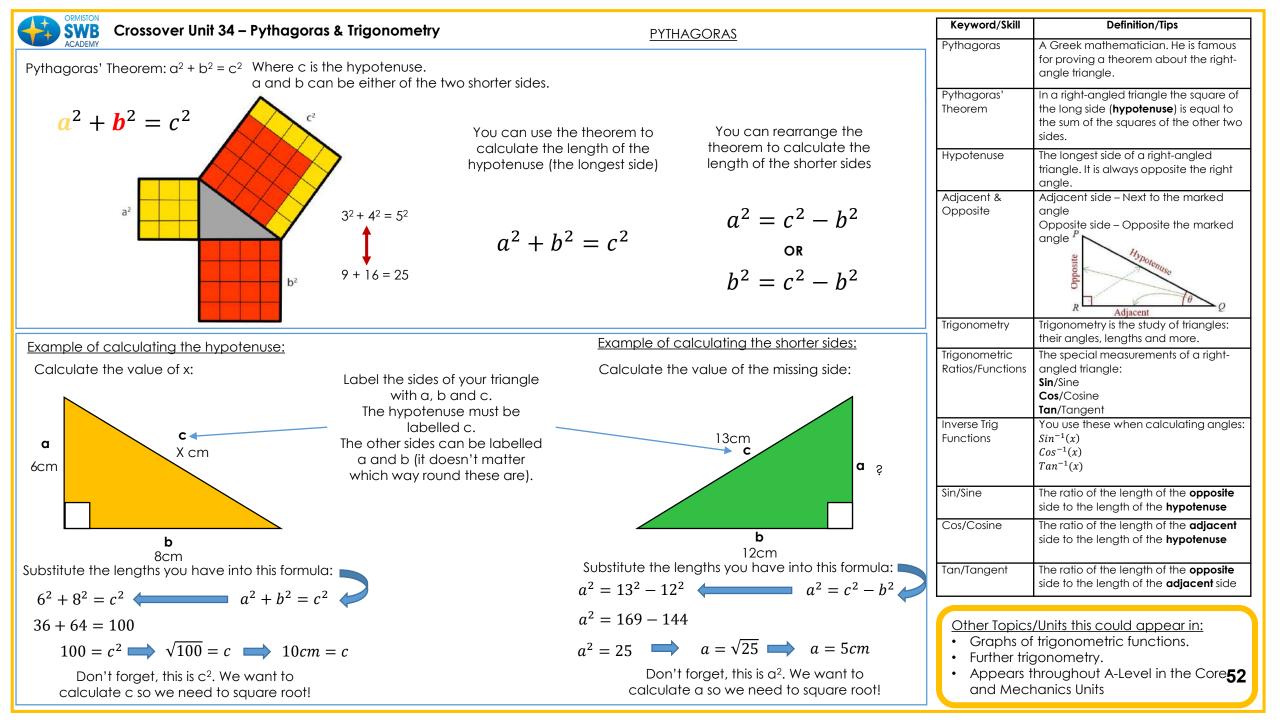
When drawing line graph or frequency polygon you must draw straight lines between the points. Not curved lines. A polygon is a shape with straight edges, so your graph needs straight edges too. You <u>do not</u> join the first and last point together

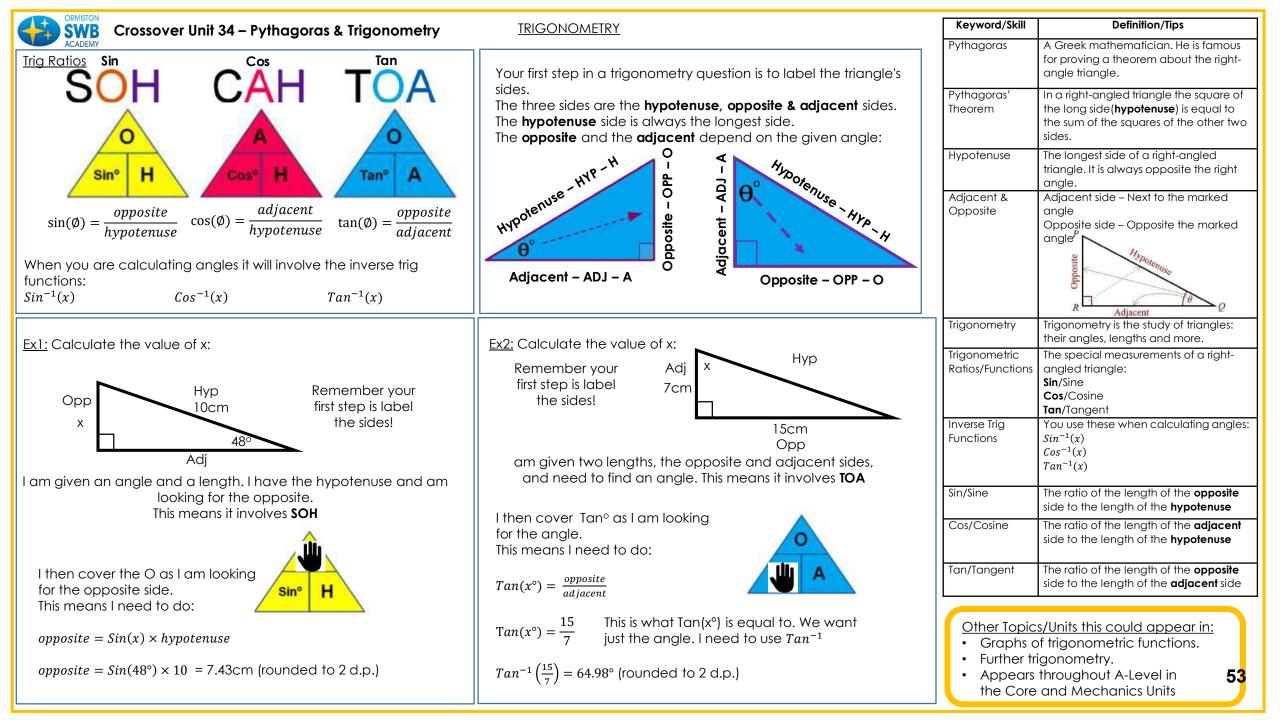
Other Topics/Units this could appear in:

- Probability
- Probability Trees
- Cumulative Frequency & Box Plots
- Histograms
- Representations of Data

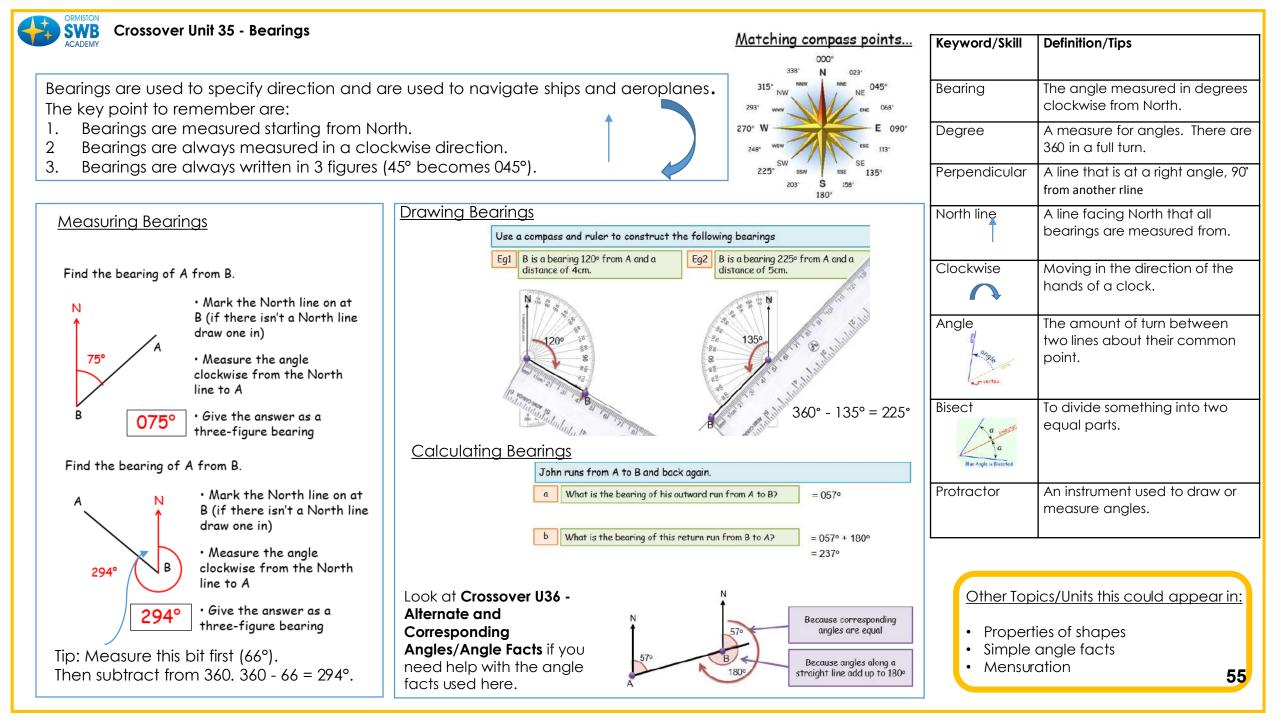
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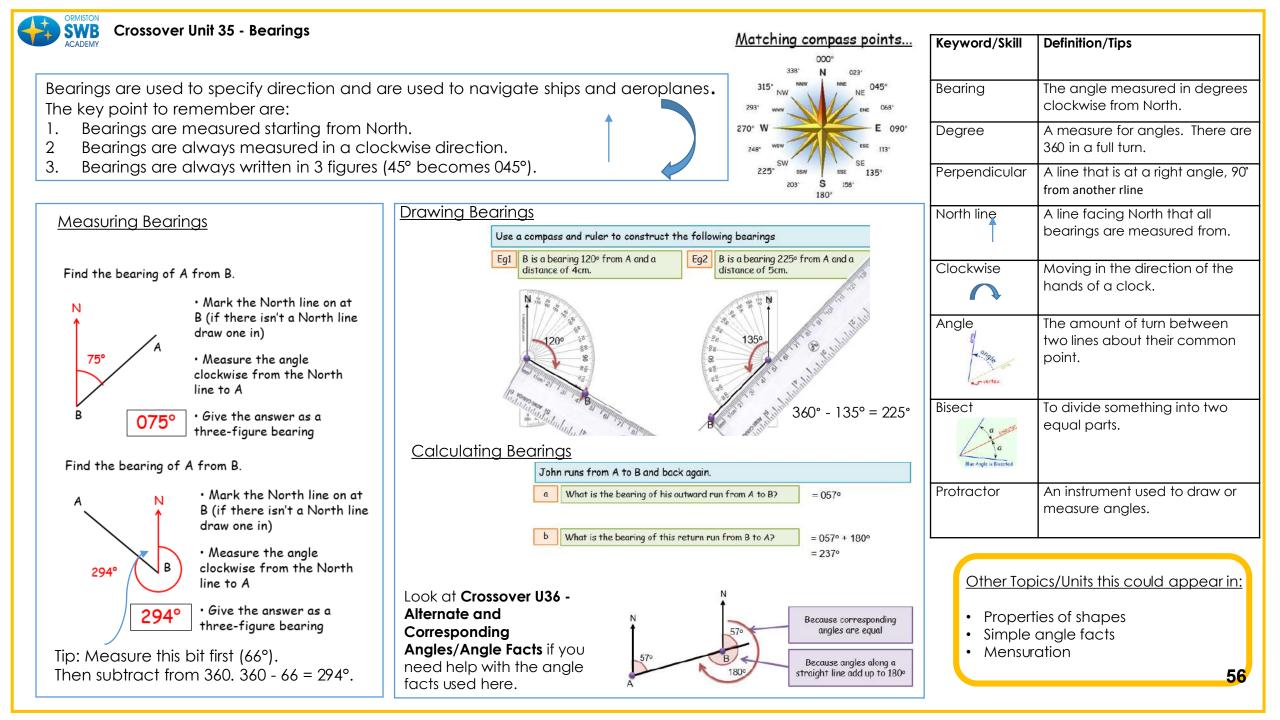
Crossover U27 & 28 -Scatter Graphs & Time-Series Graphs			Definition/tip
Scatter Graph A scatter graph is a diagram where points are plotted to show the relationship (correlation) between two variables.	Time – series graph Time series graphs show data fluctuations over time and are used to predict trends, cycles and seasonality. Example	Scatter graph Variable	A diagram with points plotted to show a relationship between two variables. A quantity that can change or
The value of one variable is shown along the x-axis and the values of the second variable is shown on the y-axis.	The time series graph below shows the amount of money invested by a company between 2005 and 2014. The general trend of the graph is an increase in the amount of money invested over time.	Line of best fit	vary, taking on different values. A straight line that best represents the data on a scatter graph.
The scatter graph to the right shows the temperature	500	Correlation	A relationship between two or more things.
compared with the number of Temperature (°C) ice-creams sold.	450 - ×**	Positive correlation	Both variables increase or both variables decrease.
Positive Negative Correlation Correlation	400-	Negative correlation	One variable increases and the other decreases or vice versa.
x x x x x x x x x	(£ billion) 350 -	No correlation	There is no relationship between the two variables.
x x x x x x x x x x x x x x x x x x x	300	Outlier	A value that lies outside most other values.
No Correlation Outlier	250 200 2005 2006 2007 2008 2009 2010 2011 2012 2013 2014 2015 2016	Time-Series	A line graph of repeated measurements taken over regular time intervals.
	Year	Trend	A direction in which something is changing.
		Other topics/L	Inits this could appear in:
 Exams! When interpreting scatter graphs always refer to what the graph is showing. For example "it has positive correlation so the hotter it is the more ice creams that are sold" 	 Exams! Once all points have been plotted, ALWAYS draw a line of best fit. (Scatter graph) Use line of best fit to estimate answers. 	CoordinateA-Level Stat	Geometry ristics - Correlation 51

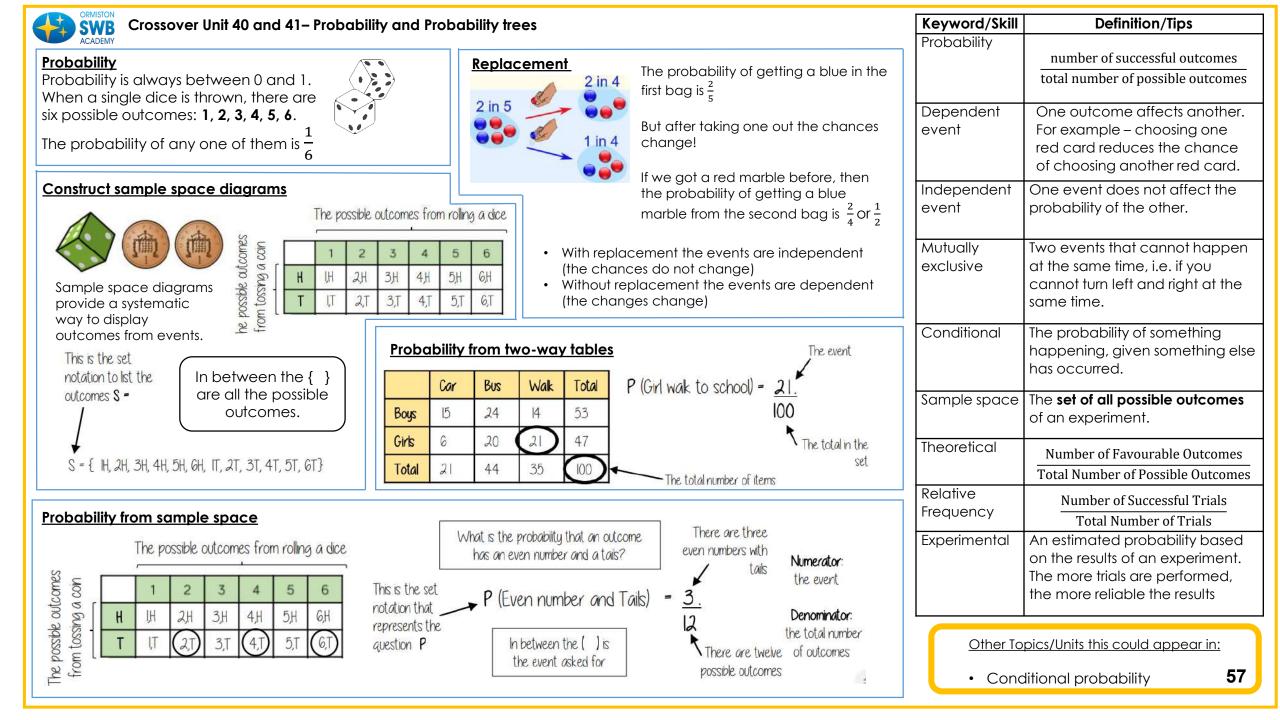




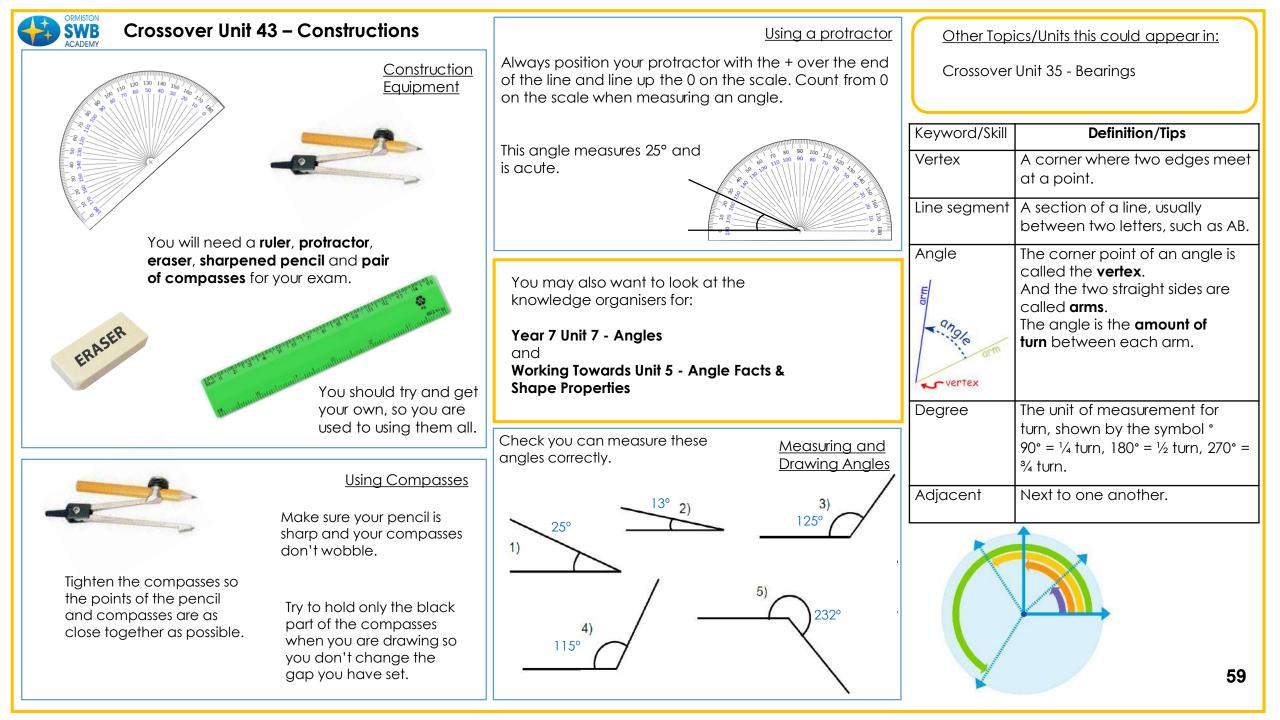
SWB Crossover Unit 34 – Pythagoras & Trigonometry	Keyword/Skill	Definition/Tips
CADEMY Calculator Help CASIO Here are the trig functions on your calculator. You use these	Pythagoras	A Greek mathematician. He is famous for proving a theorem about the right- angle triangle.
fx-300ES PLUS INTURAL-UPAID. TWO WAY 20WER	Pythagoras' Theorem	In a right-angled triangle the square of the long side (hypotenuse) is equal to the sum of the squares of the other two sides.
To get the inverse trig functions you need to press the SHIFT	Hypotenuse	The longest side of a right-angled triangle. It is always opposite the right angle. Adjacent side – Next to the marked
button first before you press the function you need. You use these ones when you are finding an angle.	Adjacent & Opposite	Adjacent side – Next to the marked angle P
Pythagoras or Trigonometry		Adjacent
Abs (x ³) (x ¹) logs) Pythagoras or Trigonometry ???	Trigonometry	Trigonometry is the study of triangles: their angles, lengths and more.
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Trigonometric Ratios/Functions	The special measurements of a right- angled triangle: Sin/Sine Cos/Cosine Tan/Tangent
STO X X X Y Y Y Are you trying to RCL ENG S S D M+ NO Does the triangle have Y Side and 1 angle?	Inverse Trig Functions	You use these when calculating angles: $Sin^{-1}(x)$ $Cos^{-1}(x)$ $Tan^{-1}(x)$
7 8 9 DEL AC find another side YES NO	Sin/Sine	The ratio of the length of the opposite side to the length of the hypotenuse
456×÷	Cos/Cosine	The ratio of the length of the adjacent side to the length of the hypotenuse
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Tan/Tangent	The ratio of the length of the opposite side to the length of the adjacent side
	Graphs ofFurther trigAppears the	<u>(Units this could appear in:</u> trigonometric functions. gonometry. hroughout A-Level in 54 and Mechanics Units





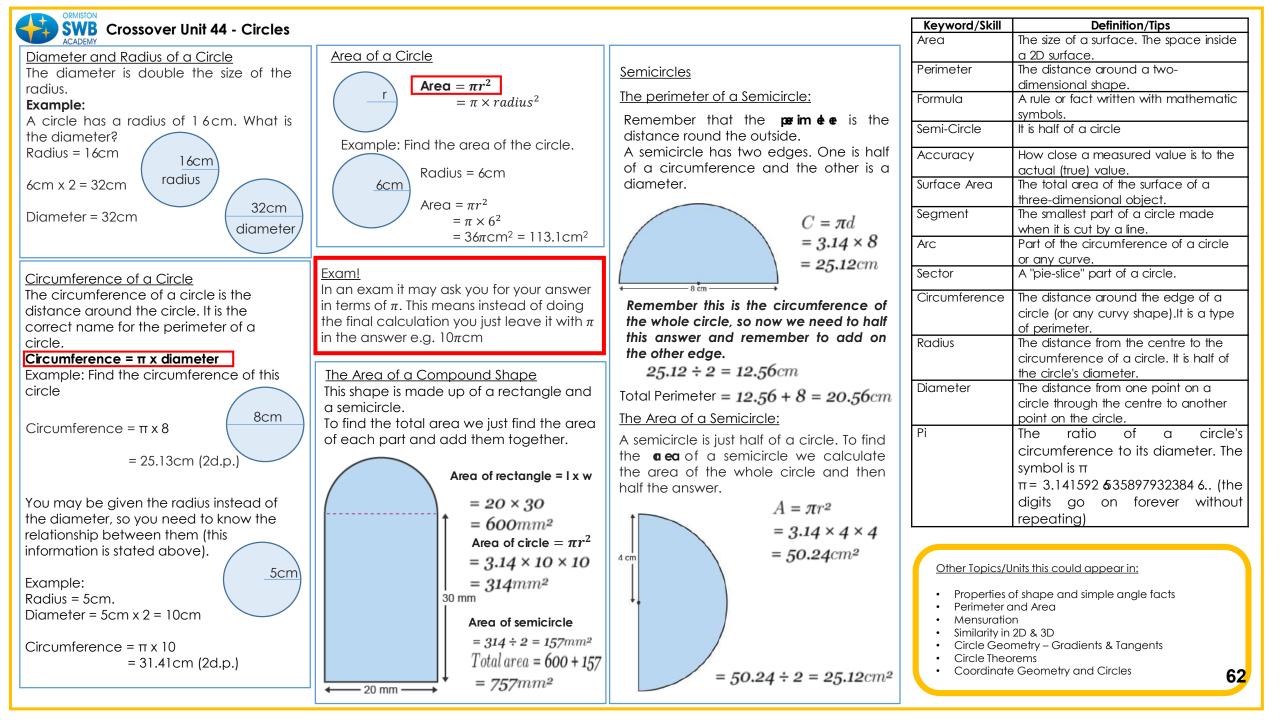


Swa Crossover Unit 40 and 41– Probability and probability tr	rees		Keyword/Skill	Definition/Tips
Mutually Exclusive When two events (call them "A" and "B") are Mutually	Aces and Kings are Mutually Exclusive (can't be both)	Hearts and Kings are not Mutually Exclusive (can be both)	Probability	number of successful outcomes total number of possible outcome
Exclusive it is impossible for them to happen together:	Aces Kings	Hearts Kings	Dependent event	One outcome affects anothe For example – choosing one red card reduces the chance of choosing another red card
Probability Trees Here is a tree diagram for the toss of a coin:	A• K•		Independent event	One event does not affect the probability of the other.
There are two "branches"		am to two tosses of a coin: 5 Head Head, Head	Mutually exclusive	Two events that cannot happe at the same time, i.e. if you cannot turn left and right at the same time.
•The outcome is written at the end of the branch	0.5	5 Tail Head, Tail 5 Head Tail, Head	Conditional	The probability of something happening, given something has occurred.
How do we calculate the overall probabilities?	0.5 Tail	5 Tail Tail, Tail		The set of all possible outcome of an experiment.
 We multiply probabilities along the branches We add probabilities down columns 		5	Theoretical	Number of Favourable Outcome Total Number of Possible Outcom
Multiply $0.5 \rightarrow \text{Head}$ Head, Head> $0.5 \times$		we can see such things as:	Relative Frequency	Number of Successful Trials Total Number of Trials
$\begin{array}{c c} 0.5 & \text{Head} \\ \hline 0.5 & \text{Tail} \\ \hline 0.5 & \text{Tail} \\ \hline 0.5 & \text{Tail} \\ \hline \end{array} \begin{array}{c} Head, Tail \\ Tail, Head \\ \hline 0.5 \\ \hline \end{array} \begin{array}{c} 0.5 \\ \hline \end{array} \begin{array}{c} 0.5 \\ \hline \end{array} \begin{array}{c} Head \\ \hline \end{array} \begin{array}{c} 0.5 \\ \hline \end{array} \end{array} \begin{array}{c} 0.5 \\ \hline \end{array} \begin{array}{c} 0.5 \\ \hline \end{array} \begin{array}{c} 0.5 \\ \hline \end{array} \end{array} \begin{array}{c} 0.5 \\ \hline \end{array} \begin{array}{c} 0.5 \\ \hline \end{array} \end{array} $	0.5 = 0.25 Add 0.5 = 0.25	e probability of "Head, ad" is 0.5×0.5 = 0.25 probabilities add 1 (which is always a good eck)	Experimental	An estimated probability base on the results of an experimen The more trials are performed, the more reliable the results
Tail $\xrightarrow{0.5}$ Tail $Tail, Tail 0.5 \times$	0.5 = 0.25 led	e probability of getting at ast one Head from two cses is 0.25+0.25+0.25 = 0.75		bics/Units this could appear in: itional probability 5

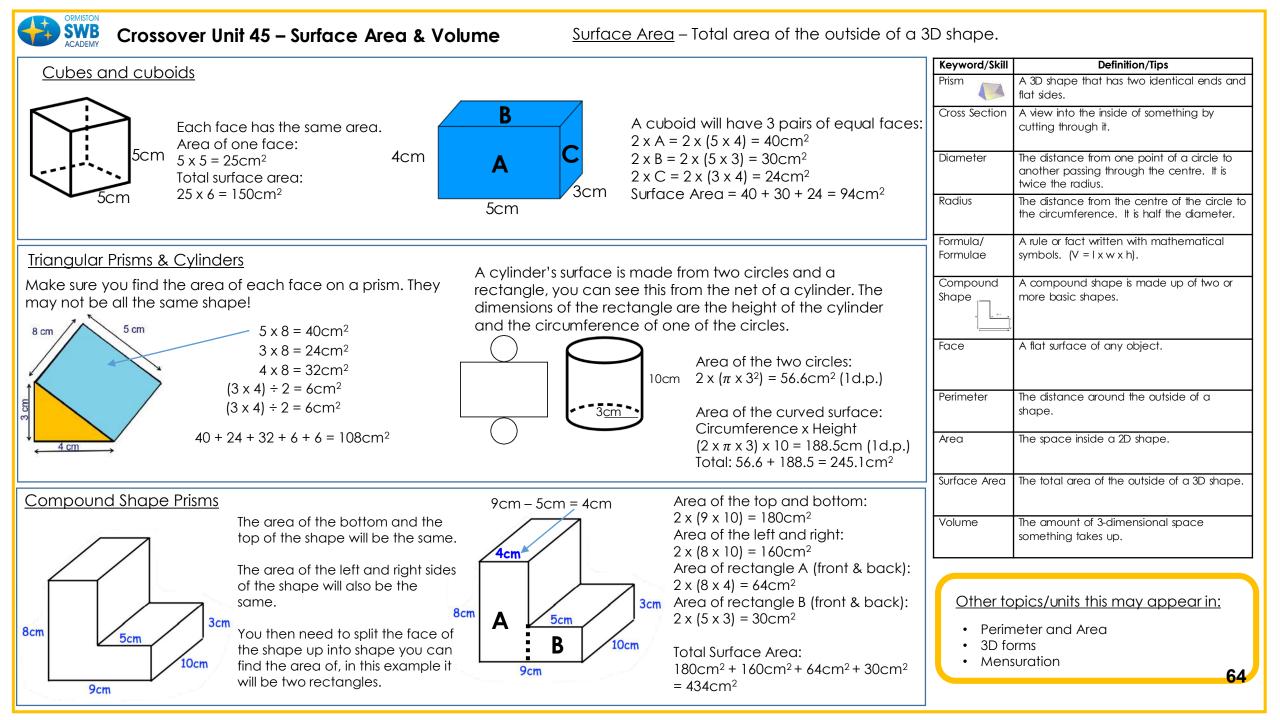


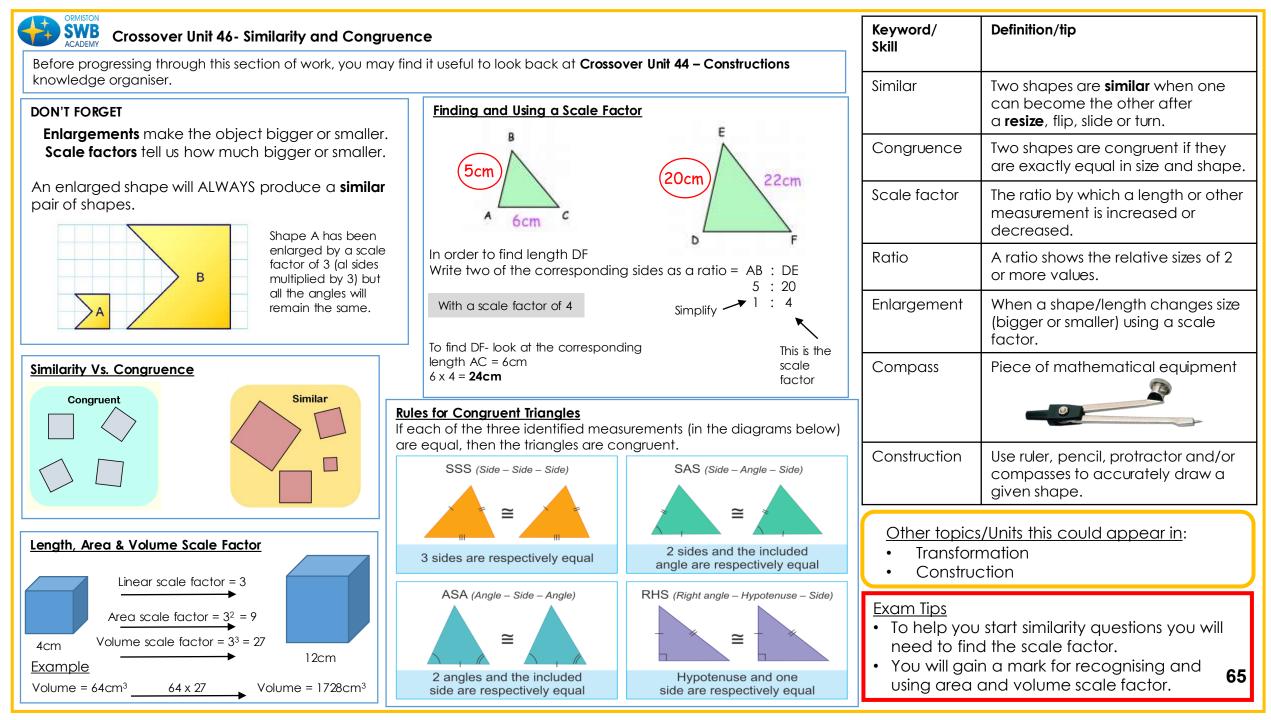
Crossover Unit 43 – Construction	5	Keyword/Skill	Definition/Tips
Constructing Triangles – given 3 side lengths	Constructing Triangles – given 2 sides and 1 angle	Arc	Part of the circumference of a circle.
Using ruler and compasses , you can construct any triangle, given its three side lengths.	Using ruler and protractor , you can construct any triangle, given two of its side lengths and the angle between them.	Vertex	A corner where two edges meet at a point.
This triangle has side lengths 6cm, 5cm and 4 cm.	This triangle has two side lengths 5cm and 4cm with a 50° angle between them.	Line segment	A section of a line, usually between two letters, such as AB.
The 6cm line was drawn with a ruler.	The 4cm line is drawn first, then an angle of 50° is measured and marked from the end of the line.	Construct	Use ruler, pencil, protractor and/or compasses to accurately draw a given shape.
5cm and 4cm were constructed at either end of the 6cm line, using compasses. B	A 5cm line is drawn at this angle.	Angle	The corner point of an angle is called the vertex . And the two straight sides are called arms .
other vertex should be.	on't rub these A 4cm B truction lines out e examiner will nt to see them!	vertex	The angle is the amount of turn between each arm.
Using ruler and protractor , you can construct any triangle, given two of its angles. These triangles both have interior angles of 48° and 25°, side lengths are different.	but the	Degree	The unit of measurement for turn, shown by the symbol ° 90° = 1/4 turn, 180° = 1/2 turn, 270° = 3/4 turn.
Draw the side in between the two angles first; sometime given to you in the question.	s this is	Adjacent	Next to one another.
Then measure the angles from each end of the line – me the protractor is lined up correctly! Extend your construct until they intersect.		Intersect (intersection)	Where two or more line segments cross or meet at a point.
The intersection points show where the other vertex (cor should be.		L	60

ORMISTON			
Crossover Unit 43 – Constructions		Keyword/Skill	Definition/Tips
	Using ruler, and compasses, you can construct <u>Constructing</u>	Bisect	Cut exactly in half.
	a perpendicular bisector, which cuts a line in <u>Perpendicular</u>	Loci/Locus of	A locus is a path formed by a
bisector, which cuts an pangle in half.	half at a right angle. Bisectors	points	point which moves according
Keep the gap on your			to a rule. The plural is loci .
compasses the same for	Remember to open the compasses	Perpendicular	Straight lines which meet or
the whole construction.	more than half the length of the		cross at right angles (90°) to
	line you are bisecting.		one another.
Draw two short arcs with the	Keep the gap on your	Scale	The scale is the ratio of a
compass point placed at the vertex of the angle.	compasses the same for		distance on the drawing or
ine venex of the dilgie.	the whole construction.		model to the corresponding
R	Draw two long arcs with the		distance in real life, eg 1:20
	compass point placed at either		means 1cm on the drawing
don't rub the	ese) \ end of the line you are bisecting.		represents 20 cm in real life.
Construction line		Region	A specific part of something,
- the examine (want to see th	Draw a straight line joining the two		usually shown by shading or
	intersection points. This is your bisector .		labelling R.
Draw two short arcs with the compass point placed on		Plan	A plan is similar to a map,
your first pair of arcs.			usually showing a small area
• P /			such as a playground or
	<u>Bisector from a</u> point to a line		house.
	When constructing a	Exams!	
	perpendicular bisector	<u>EXULLS:</u>	
Draw a straight	from a point to a line,		
line joining the	add this first step, then		n use all these construction
vertex and the	continue as above.	drawing	construct loci or scale
intersection		Ŭ Ŭ	rect part of a construction
point. This is your bisector .			n mark, so always have a go,
		even if y	/ou're not sure.
	\times		
Q R			61



Crossover Unit 45 – Surface Area & Volume <u>Volume</u>		
Volume Using Unit Cubes Example:	Keyword/Skill	Definition/Tips
Volume Using Unit Cubes Example: This is a unit cube. It has a volume of 1 cm ³ . Image: Comparison of 1 cm ³ .	Prism	A 3D shape that has two identical ends and flat sides.
You may get shapes made from unit cubes and you will need to find the volume of that The volume of this cuboid would be 24cm ³ as there are	Cross Section	A view into the inside of something by cutting through it.
shape by counting how many cubes there are.	Diameter	The distance from one point of a circle to another passing through the centre. It is twice the radius.
	Radius	The distance from the centre of the circle to the circumference. It is half the diameter.
Volume of Prisms A prism has a constant cross-section. To find the volume of it you first need	Formula/ Formulae	A rule or fact written with mathematical symbols. (V = I x w x h).
Cross-Section to find the area of the cross section and multiply it by the height/depth of the shape. Formula for Volume of Prims: Area of Cross-Section x Height	Compound Shape	A compound shape is made up of two or more basic shapes.
	Face Perimeter	A flat surface of any object. The distance around the outside of a
Example: You are given the area of the cross- Area of cross-section = $\pi \times r^2$	Area	shape. The space inside a 2D shape.
section so you need to multiply the area by the depth: Area $= \pi \times 3^2 = 28.27 \text{ cm}^2$ Volume = Area of Cross-Section x Height		
$12cm^2$ $12cm^2 \times 7 = 84cm^3$ $7cm$ $12cm^2 \times 7 = 84cm^3$	Surface Area	The total area of the outside of a 3D shape.
	Volume	The amount of 3-dimensional space something takes up.
Volume of SpheresFormula for volume of a Sphere: $\frac{4}{3} \times \pi \times r^3$ Make sure you use the correct units with your	L	
answer.		
You are given this formula in the exam, you just need to be able to use it! Area uses square units and volume uses cubic units.	• Perim	opics/units this may appear in: neter and Area
Example: Example: Example: Examples: Examples: Examples:	3D foMens	
Volume = $\frac{4}{3} \times \pi \times 3^3$ = 113.1cm ³ (1d.p.) Volume: cm ³ , m ³ , mm ³ , km ³ , etc.		63





Crossover Unit 51- Simultaneous Equations.					
In order to be able to solve simultaneous equations you will need to be able to solve linear equations (see units 2a, 21,22 and 50 for recap)	Sometimes you will h get identical terms. Example 3	nave to change both equations to	Keyword/ Skill	Definition/tip	
You also need to be able to use substitution . (Unit 2b) <u>Elimination Method</u> <u>Example 1</u> Solve the equation: $6x + y = 15$ and $4x + y = 11$ It is useful to label the equations to help with method.	Solve these equation 4x + 3y = 27 (1) 5x - 2y = 5 (2)	ns: Both equations have to be changed to get identical terms in either x or y.	Simultaneous Equation	A pair of equations with two unknown variables. Both equations need to be solved at the same time (simultaneously)	
6x + y = 15 (1) 4x + y = 11 (2) The y-term in both equations has the same coefficient. (No need to balance them)	Equation (1) x 2 Equation (2) x 3	Here it will be best to make the y- coefficients the same so that we	Eliminate	To remove a variable in order to help solve the equation.	
$6x + y = 15 (1)$ $4x + y = 11 (2)$ Equation (1) minus equation (2) $2x = 4$ $x = 2$ $\div 2$	8x + 6y = 54 (3) 15x - 6y = 10 (4) Eliminate by adding	can add the equations. (Easier than subtracting) Label new equations (3) and (4) (3) + (4)	Substitution	When a letter in an equation, expression or formula is replaced by a number, we have substituted the number for the letter.	
Substitute x = 2 into one of the original equations. (Usually the one with the smaller numbers) So substitute x = 2 into: $4x + y = 11$ 8 + y = 11 Solve	23x = 69 x = 3	÷ 23	Variable	A symbol for a number that we don't know yet. Often this is a letter such as x or y.	
y = 3 You can then test the solutions by substituting values found back into the original equations	Substitute into equation (1)= 12 + 3y = 27 Solve 3y = 15 y = 5		Coefficient	The number in front of an unknown quantity (the letter) in an algebraic term.	
Example 2 $5x + y = 22$ (1) $2x - y = 6$ (2)Both equations have the same y-coefficient but with DIFFERENT SIGNSAs the signs are different you ADD the two equations to eliminate the y-terms. [Equation (1) + equation (2)] $5x + y = 22$ (1) $2x - y = 6$ (2)	and two bubblies co and five bubblies co You need to set up t and b and then solv	ur bubblies cost 72p. Five chews ost 64p. What would three chews st? wo simultaneous equations in c	Other topics/Units this could appear in: • Working above • Unit 7 – Simultaneous equations • A-level • Core – algebra and functions • Statistics- statistical distributions		
$\frac{7x = 28}{x = 4} \div 7$		ve the simultaneous equations in example 1	 Exam Tips You will gain 1 mark by correctly starting a process to eliminate a coefficient. 		
Substitute $x = 4$ into one of the original equations, $5x + y = 22$ which gives $20 + y = 22$ $y = 2$ Solve	b = 12	o calculate 3 chews = 3 x 8 = 24p 50p		66	



Crossover Unit 51- Simultaneous Equations- Solving Graphically

To be able to solve simultaneous equations graphically, you may be asked to draw the line of given equations. In order to do this you may need to look back at Crossover Unit 29- Straight Line Graphs knowledge organiser. Example Solve the following simultaneous equations y graphically y = 2x + 1y = y = 3**Step 1-** Draw the line y = 2x + 1**Step 2-** Draw the line y = 3Step 3 - Your solution is the coordinates where the lines cross Coordinates = (1, 3)x = 1 X 9 -2 6 7 8 10 -1 v = 3Exam Tips Further questions relating to this topic may include solving simultaneous

Keyword/ Skill	Definition/tip
Simultaneous Equation	A pair of equations with two unknown variables. Both equations need to be solved at the same time (simultaneously)
Eliminate	To remove a variable in order to help solve the equation.
Substitution	When a letter in an equation, expression or formula is replaced by a number, we have substituted the number for the letter.
Variable	A symbol for a number that we don't know yet. Often this is a letter such as x or y.
Coefficient	The number in front of an unknown quantity (the letter) in an algebraic term.

Other topics/Units this could appear

in:

- Working above
- Unit 7 Simultaneous equations
- A-level
- Core algebra and functions
- Statistics- statistical distributions

equations graphically when one equation is linear and the other is guadratic. As with the example above, you would draw the graph of the equations and look for where your straight line crosses two parts of the quadratic curve.

- If you need to draw your own straight line graph you will gain marks for that.
- If the graph has been drawn for you LOOK at where the lines cross.



Crossover Unit 52 - Direct and Inverse Proportion.

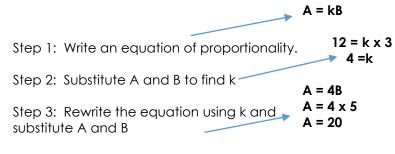
Direct Proportion means as one quantity increases or decreases, so does the other one. Example: The more hours you work, the more you will earn. A is directly proportional to B is written as:



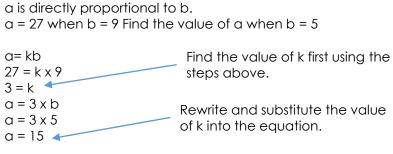
"k" is the constant of proportionality that links our two numbers. We need to put it into our formula.

Example 1

It takes 3 hours to fill 12 boxes with food. How long does it take to fill 5 boxes?



Example 2



takes 4 people 5 h people 2.5 hours.	ours to dig a hole, it			ak	e 8	3	
long would it take food?	e of food in 6 days. 3 cows to eat a tor		:				
Step 1: Write an e	qualion	y —	<u>K</u> X				
Step 2: Substitute x	and y to find k days	6 =	<u>k</u> 2 <		С	ow	'S
Step 3: Rewrite the find y	e equation using k a $y = \frac{12}{x}$ $y = \frac{12}{3}$ $y = 4$	k = 1 ind su ! days	bst	itu [.]	te	x to	D
Example 2 y is inversely propo y = 6 when x = 5. F	rtional to x ind the value of y v	vhen	x =	10			
	the same	x 1	2	4	8	16	32
steps	s as above 📝	y 32	16	8	4	2	1
$6 = \frac{k}{5}$		y					
30 = k	1 job takes 32 hours	32- X 28-	-		11		
$y = \frac{30}{x}$ $y = \frac{30}{10} = 3$	to complete. We can show the inverse relationship on a graph	24- 20- 16- X 12- 8- 4- 0-	8 1	2 16	20.24	1 28 3	$x \rightarrow x$
		4404 D.		en 5486			9995

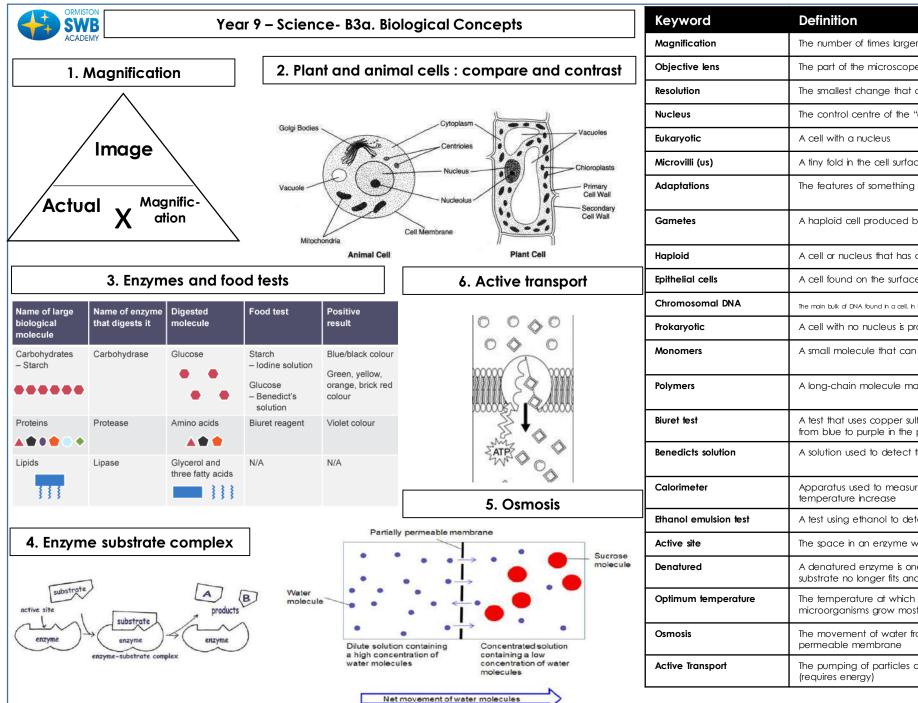
Inverse Proportion means as one quantity increases, the

Example: If one quantity doubles the other is halved. If it

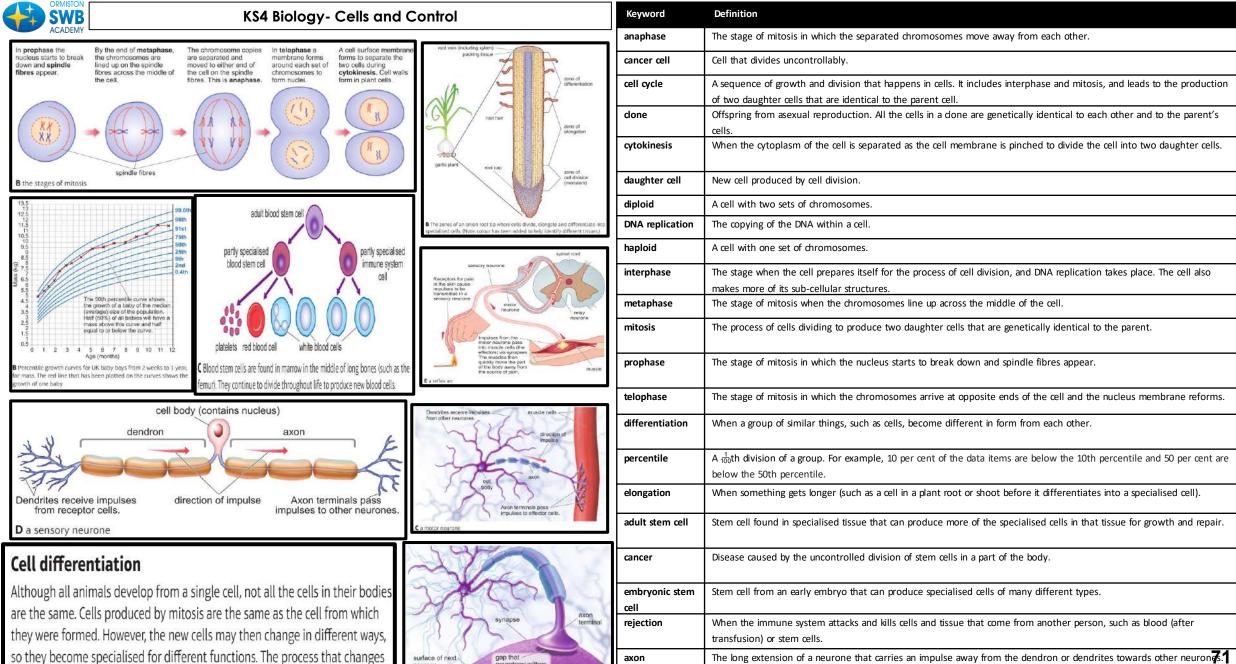
other decreases.

Keyword/Skill	Definition/Tips
Ratio	Shows the relative sizes of two or more values. E.G. 1 boy and 3 girls would be written as 1:3
Inverse	The opposite or the reverse E.g. the inverse of addition is subtraction
Proportion	Two ratios or fractions that are equal.
Direct Proportion.	Two quantities change in the same way When one increases or decrease, so does the other one.
Variable	A symbol for a number we don't know yet, often a letter x or y.
Equation Says that two things are the equal. (1+1=2).	
Linear	A graph that has a straight line.
Substitute	Putting values where the letters are.
Constant of A constant value, written as k, relating amounts that rise or fall uniformly together.	
Other topic	s/units this may appear in:
· ·	

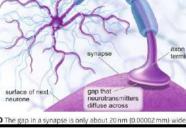
Biology Knowledge Organisers

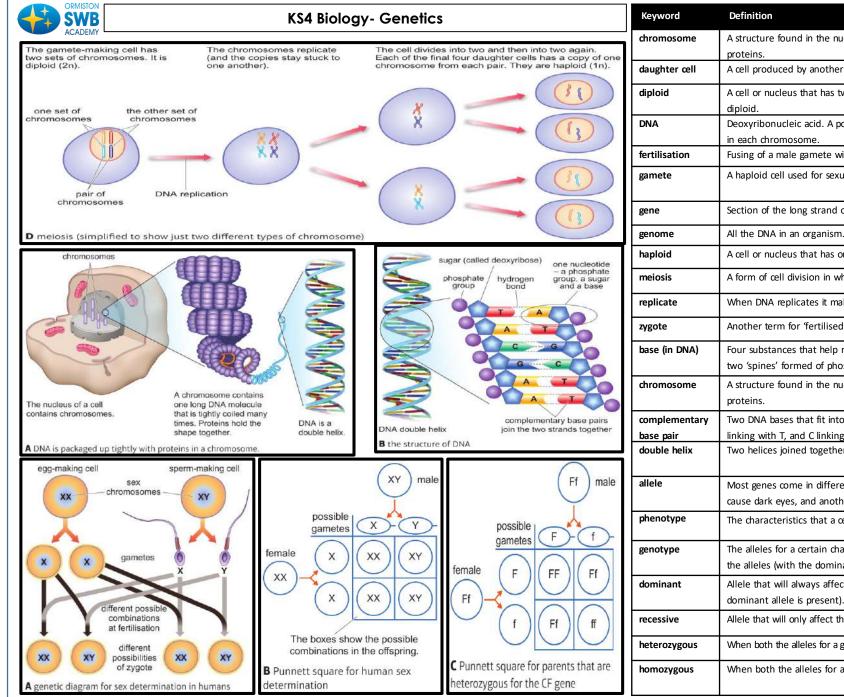


Keyword	Definition
Magnification	The number of times larger an image is than the initial object that produced it
Objective lens	The part of the microscope that is closest to the specimen
Resolution	The smallest change that can be measured by an instrument
Nucleus	The control centre of the "eukaryotic cell"
Eukaryotic	A cell with a nucleus
Microvilli (us)	A tiny fold in the cell surface membrane of a cell, increase the surface area of the cell
Adaptations	The features of something that enable it to do a certain function
Gametes	A haploid cell produced by meiosis used for sexual reproduction
Haploid	A cell or nucleus that has one set of chromosomes. Gametes are haploid
Epithelial cells	A cell found on the surface of internal organs
Chromosomal DNA	The main bulk of DNA found in a cell. In humans, this DNA is found in chromosomes
Prokaryotic	A cell with no nucleus is prokaryotic
Monomers	A small molecule that can join with other molecules ike itself to form a polymer
Polymers	A long-chain molecule made by joining many smaller molecules (monomers)
Biuret test	A test that uses copper sulfate solution and potassium hydroxide solution to test for proteins. It turns from blue to purple in the presence of proteins
Benedicts solution	A solution used to detect the presence of reducing sugars (eg. Glucose) in foods
Calorimeter	Apparatus used to measure the energy content of substances by burning them and measuring the temperature increase
Ethanol emulsion test	A test using ethanol to detect lipids (fats) in food
Active site	The space in an enzyme where the substrate fits during an enzyme-catalysed reaction
Denatured	A denatured enzyme is one where the shape of the active site has changed so much that the substrate no longer fits and the reaction can no longer happen
Optimum temperature	The temperature at which an enzymes rate of reaction is greatest, or at which a population of microorganisms grow most rapidly
Osmosis	The movement of water from a high concentration to a low concentration through a partially permeable membrane
Active Transport	The pumping of particles across a cell membrane from a low concentration to high concentration (requires energy)



less specialised cells into more specialised ones is called differentiation





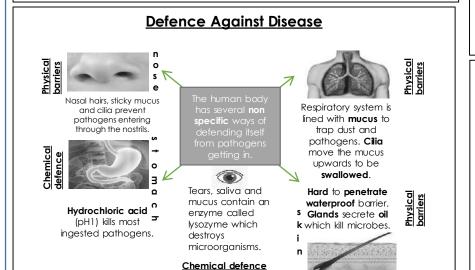
hromosome A structure found in the nuclei of cells. Each chromosome contains one enormously long DNA mole proteins. laughter cell A cell produced by another cell that has divided. liploid A cell or nucleus that has two sets of chromosomes. In humans, almost all cells except the sperm ar diploid.	cule packed up with
Iploid A cell or nucleus that has two sets of chromosomes. In humans, almost all cells except the sperm ar	
·	
	nd egg cells are
NA Deoxyribonucleic acid. A polymer made of sugar and phosphate groups joined to bases. One molec in each chromosome.	ule of DNA is found
ertilisation Fusing of a male gamete with a female gamete.	
amete A haploid cell used for sexual reproduction.	
ene Section of the long strand of DNA found in a chromosome, which often contains instructions for a p	protein.
enome All the DNA in an organism. Each body cell contains a copy of the genome.	
aploid A cell or nucleus that has one set of chromosomes. Gametes are haploid.	
neiosis A form of cell division in which one parent cell produces four haploid daughter cells.	
when DNA replicates it makes a copy of itself.	
Another term for 'fertilised egg œll'.	
ase (in DNA) Four substances that help make up DNA, often shown by the letters A, C, G and T. Pairs of bases for two 'spines' formed of phosphate groups and a type of sugar.	m 'links' between
hromosome A structure found in the nuclei of cells. Each chromosome contains one enormously long DNA mole proteins.	cule packed up with
Two DNA bases that fit into each other and link by hydrogen bonds. There are two types of complete	mentary base pair: A
ase pair linking with T, and C linking with G.	
ouble helix Two helices joined together.	
llele Most genes come in different versions called alleles. So a gene for eye colour may have one version cause dark eyes, and another allele that can cause pale eyes.	ı (allele) that can
henotype The characteristics that a certain set of alleles display.	
The alleles for a certain characteristic that are found in an organism. Written in a shorthand using le the alleles (with the dominant allele having a capital and being written first).	etters to represent
ominant Allele that will always affect the phenotype (as opposed to a recessive allele, whose effect will not dominant allele is present).	be seen if a
Allele that will only affect the phenotype if the other allele is also recessive. It has no effect if the other	r allele is dominant.
eterozygous When both the alleles for a gene are different in an organism.	70

	<u>Year 9 – Science- B3b. Na</u>	ural Selection and Genetic Modification	Keyword	Definition
1. Eviden	nce for human evolution	2. Darwin's Theory of Evolution	Binomial system	The system of naming organisms using two Latin words
Ardi (Ardipithecus ramids)	Human like female fossil	Evolution is a change in the inherited characteristics of a	Evolution	A change in one or more characteristic of a population over a long period of time
	Walked upright Long arms and short legs Small skull and brain	population over time. This occurs through a process called natural selection.		Differences between organisms caused by differences in the alleles they inherit from their parents, or differences in genes caused by mutation. Also called inherited variation
Lucy (Australopithecus afarensis)	 More human like female fossil than Ardi Walked upright better than Ardi Arm and legs were the length between ape and human 	 The differences in a population gives some individuals an advantage. This individual is more likely to survive for longer and 	Natural selection	A process in which certain organisms are more likely to survive and reproduce than other members of the same species because they possess certain genetic variations
Turkana Boy (Homo	Skull and brain slightly larger than Ardi More human like female fossil than Lucy	 be able to breed to pass on desirable genes. Nature is selecting the individual with the 	Resistance	When an organism has resistance to something, it is unaffected by it, or not affected very much
erectus) discovered by Richard Leakey	 Walked upright better than Lucy Arm and legs were human length Skull and brain larger than Lucy 	phenotypes most suited to survival ('survival of the fittest). This is called natural selection.	Pentadactyl limb	A limb that has five digits (fingers and thumbs). Amphibians, reptiles, birds and mammals share this characteristics
	tion can also be gained from looking at stone	Our understanding of evolution has also been helped by	Classification	The process of sorting organisms into groups based on their characteristics
tools, which become more		the study of antibiotic resistance in bacteria. Genetic Mutation Causes Drug Resistance	Kingdoms	There are five kingdoms into which organisms are usually divided: plants, animals, fungi, protists and prokaryotes
		Non-resistant bacteria exist Bacteria the billions Bacteria Multiply by the billions Bacteria drug resistant drug resistant bacteria drug resistant	Selective breeding	When humans choose an organism that has a certain characteristic and breed more of these organisms, making that chosen characteristic more and more obvious
3. Clo	assification	4. Genetic Engineering and Tissue Culture	Varieties	Groups of plants of the same species that have characteristics that make them different to other members of the species
The number	Kingdom	uman cell - waterium - wateria plasmid	Genetic engineering	Altering the genome of an organism, usually by adding genes from another species.
of organism s in each		ng the insulin gene Plasmid is removed and cut open using restriction enzymes.	GMOs	An organism that has had its genome genetically altered (genetic modification)
group gets	Class	s remove the insulingene. The vector DNA and the DNA being inserted have sticky ends. When mixed with ligase enzyme, the pieces of DNA combine. This is called recombinent	Yield	The amount of useful product that you can get from something
smaller, but they have	Order	Samples develop	Disease resistance	Unaffected or less affected by a certain disease
more features	Family		Stem cells	An unspecialised cell that continues to divide by mitosis to produce more stem cells and other cells that differentiate into specialised cells
in Genus			Tissue culture	Growing tiny pieces of tissue, or cells, in the lab
	Spec ies	Partiets planted Bacteria make human insulin. into compost	Pests	Animals that cause problems, such as damaging crops
		of humans and other ergenisses are sub-sub-of the DNA	Biological control	Using organisms to kill problem organisms, such as pests or weeds
		of humans and other organisms are cut out of the DNA cells of the organism to be genetically modified.	Insecticides	A substance used to kill insect pests 73



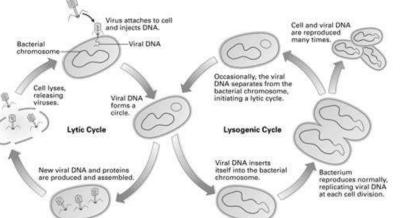
KS4 Biology - Health, Disease and the Development of Medicine

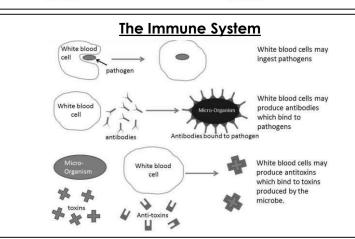
	<u>Trc</u>	ansmission of	Disease	
Disease	Symptom	Method of transmission	Control spread of disease by:	Caused by:
Malaria	Recurrent fever	Animal vector	Preventing breeding of mosquitoes or use of a net to prevent being bitten.	Protist
Chalara ash dieback	Leaf loss and bark lesions	Airborne	Remove infected leaf litter. Clean all tools, vehicles and footwear.	Fungus
Cholera	Diarrhoea	Waterborne	Clean drinking water and good hygiene and sanitation.	Bacteria
Tuberculosis	Lung damage	Airborne	Vaccination programme. Treat infection with antibiotics.	Bacteria
HIV	Flu like illness	Sexual contact or bodily fluids. Direct contact.	Use of condoms / dean needles. Treat infection with antiretroviral drugs.	Virus
Ebola	Haemorrhagic fever	Bodily fluid - direct contact	Avoid contact with people infected with Ebola.	Virus



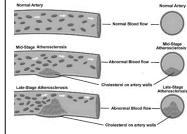
Communicable diseases:	Non-communicable diseases:					
Common cold	Diabetes					
Influenza	Cancer					
STDs	Cardiovascular disease					

Lysogenic Vs Lytic Virus Lifecycle



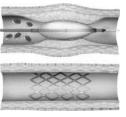


Cardiovascular Disease



Cadovascular dise ase (CVD) is a general term for dise ase which involve the heart or blood vessels. Afterosderosis is a cause of caronary heart dise ase where layers of fatty material build up inside the coronary arteries, narrowing the m This ne duces the flow of blood through the caronary arteries, ne suffing in a lack of oxygen for the heart muscle.

At he ras de rasis can be the de din two mainways by placing a stert in the coronary artery and or using **life long ne dication** addled statins. Life style changes such as a he dathy alet, exercise and no smoking are also vital in reducing the risk of CVD.



She ris are netal sylinder gids which can be inserted into an artery to maintain blood flow by leeping the attery open so that the least continues to receive enough axygen to function effectively.

Statins are alugs that lower har mild chale sterd in the blood and stop the liver producing too much chole sterd and reduce the rate of which it is at posite d P dients should change their life style and have a healthy diet. This reduces the risk of heart disease.

<u>Obesity</u>

Excess weight (obesity) can make a person at **risk of cardiovascular disease**, a stroke and Type 2 diabetes. A tool called the **Body Mass Indicator** (BMI) can be used to calculate whether a persons weight lies within a healthy range.

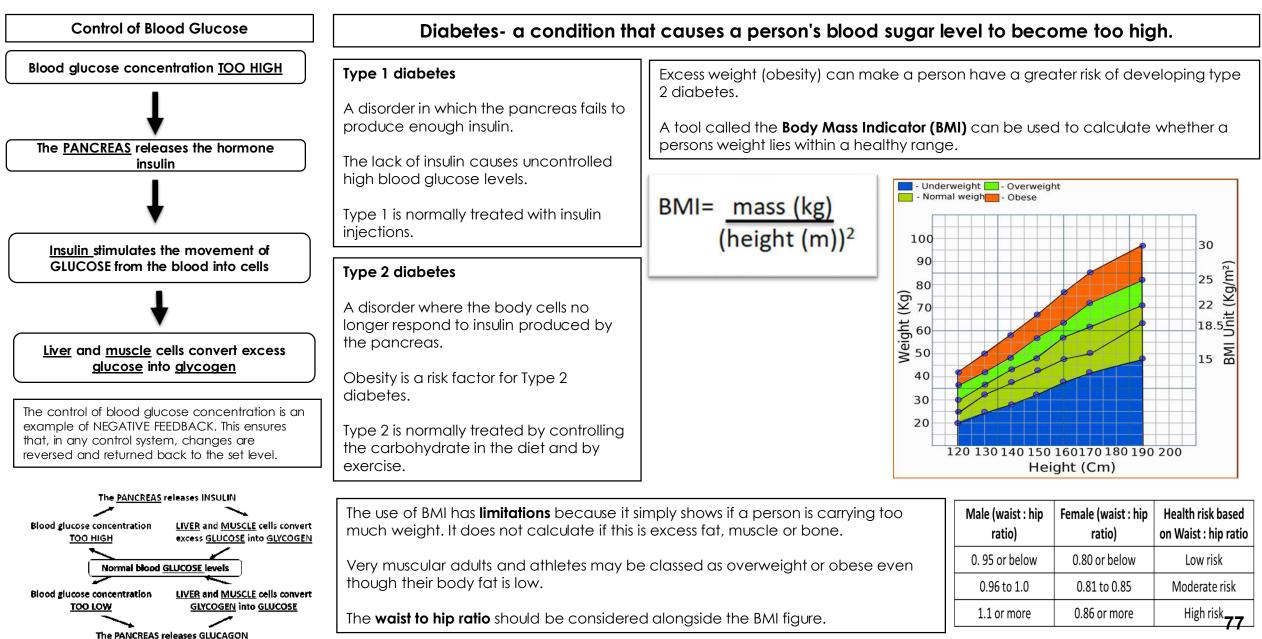
BMI= mass (kg) (height (m))²

The use of **BMI** has **limitations** because it simply shows if a person is carrying too much weight. It does not calculate if this is excess fat, muscle or bone. The **waist to hip ratio** should be considered alongside the BMI figure. **74**

	N MY	KS4 Biology - Plant S	strue	ctures and the	eir Functions		Keyword	Definition
cuticle upper epidermis palisade		chloroplast vacuole cell wall	root	w is from s to leaves minerals toughened with lignin	elis have end plates with holes two way flow two way flow	permanent vacuole cell wall cell membrane	Active Transport	Active transport moves substances from a more dilute solution to a more concentrated solution (against a concentration gradient). The energy is provided by respiration
mesophyll spongy lower epidermis			Plants make use of light energy from the environment Carbon dioxide + Water → Oxygen + Glucose			Oxygen + Glucose	Xylem	Form hollow tubes made of dead tissue. Long cells with walls toughened by waterproof lignin. Water and minerals flow from the roots
guard cell stoma		Photos	END to r	DOTHERMIC) make food glucose)	$CO_2 + H_2O \rightarrow O$	$D_2 + C_6 H_{12} O_6$		towards the leaves in one direction in a process called TRANSPIRATION. Xylem vessels also provide support to
	Waxy cuticle (top layer of the leaf)	Reduces water loss from the leaf		Factor	How the rate is affected	Limiting factors (why the rate stops going up)	Phloem	the stem of the plant. Form tubes made of living tissue. Cells have end plates with holes in
Epidermal tissues	Guard cells and stomata	Guard cells open and close the stomata to control water loss and allow for gas exchange (oxygen and carbon dioxide).			As the temperature of the environment the plant is in increases rate of	Photosynthesis is an enzyme controlled reaction. If the temperature increases		them. Sucrose in solution move from the leaves to growing tips and storage tissues in both directions. This process is called TRANSLOCATION. There are no forces causing translocation to occur and so the
Palisade mesophyll	Palisade cells	Cells near the top surface of the leaf that are packed with chloroplasts that contain chlorophyll. Both adaptations maximize photosynthesis.	of photosynthesis	Temperature	photosynthesis increases (up to a point) as there is more energy for the chemical reaction.	too much, then the enzymes become denatured and the rate of reaction will decrease and stop		
Spongy mesophyll	Air spaces in the leaf between cells	Increased surface area for gas exchange so that carbon dioxide can diffuse into photosynthesising cells.	rate of pho		Light intensity increases as the distance between the plant and the light	At point X another factor is limiting the rate	Stoma	sucrose is moved along using active uptake which requires energy.
xylem	Hollow tubes strengthened by lignified dead cells adapted for the transportation of water and mineral ions through the plant in the transpiration	Allows transport of water and mineral ions from the roots to the stem and the leaves.	Factors affecting the		sources increases. As light intensity increases so does the rate of photosynthesis (up to a point) as more energy is available for the chemical reaction.	of photosynthesis. This could be carbon dioxide concentration, temperature or the amount of chlorophyll	Guard Cell	Cells either side of the stoma that
phloem	stream Cell sap moves from one phloem cell to the next through pores in the end	Transports dissolved sugars from the leaves to the rest of the plant for immediate use or storage	Fac	Carbon	Carbon dioxide is needed for plants to make glucose. The rate of	At point X another factor is limiting the rate	Gibberellin s Ethene	End seed dormancy, promote flowering, increase fruit size. Control ripening of fruit during storage
Meristem tissue	walls New cells (roots and shoot tips) are made here including root hair cells	(translocation). Root hair cells have an increased surface area for the uptake of water by osmosis, and mineral ions by active transport.		dioxide concentration	photosynthesis will increase when a plant is given higher concentrations of carbon dioxide (up to a point).	of photosynthesis. This could be light intensity, temperature or the amount of chlorophyll	Auxins	and transport. Weed killers, rooting powders, promoting growth in tissue culture. 75

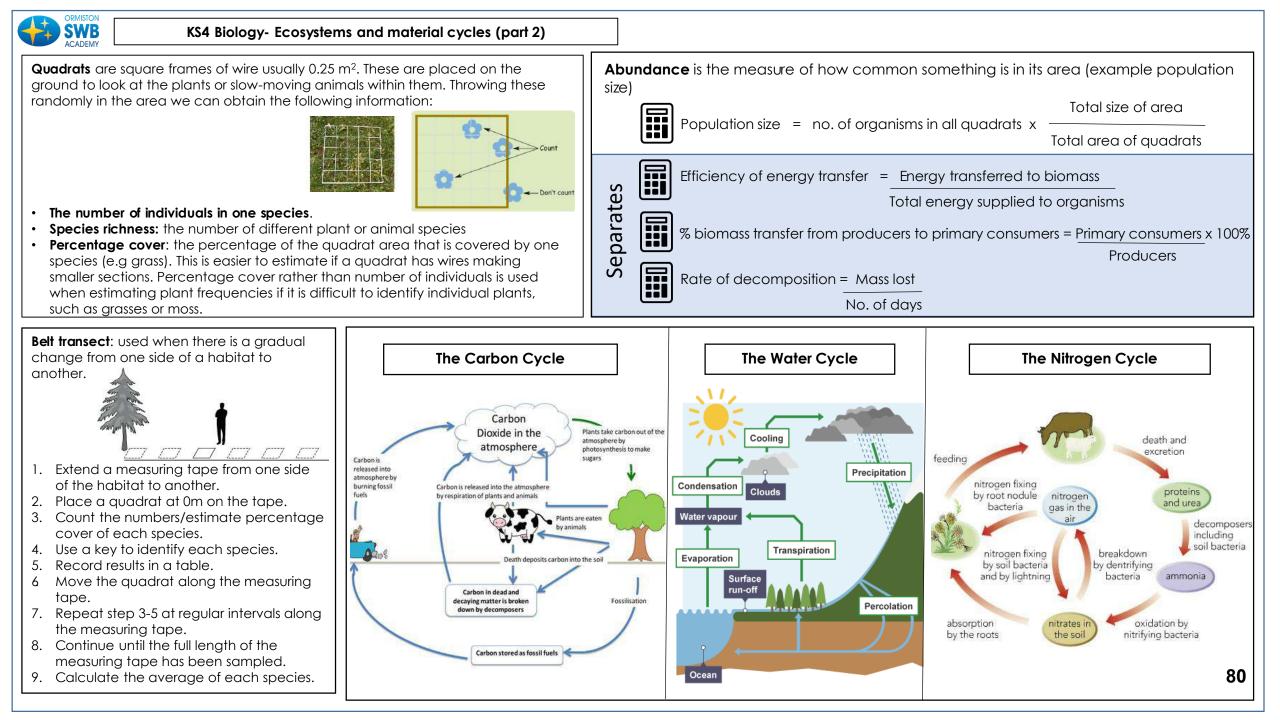
KS4 Biology- Animal Co	ordination, Control and Hom	neostasis (p	part 1)	Keyword	Definition		
The Human Endocrine System	The <u>PITUITARY GLAN</u>		s of 1	Endocrine system	Collection of organs/glands in the human body that release hormones		
Thalamus Pituitary gland	OESTROGEN stimulate the production of LH (positive feedback) <u>RELEASES LH</u> <u>RELEASES LH</u> <u>FSH STIMULATES</u> egg ripening and <u>OESTROGE</u> production (in ovaries)	EN Pe f FSH tback) Menstruation Lining of the uterus builds up maintained down	Hormones	Chemical messengers released from endocrine glands that cause a change in bodily responses			
Thymus Thymus Thymus An		Target organ Metabolic rate	The organ where a particular hormone is released and effects The rate at which the energy stored is				
Adrenal glands Pancreas Uterus	Pancreas Hormone Produced in Causes						
Ovaries (female)	Calliala atimuslating	Pituitary Gland	Stimulates egg ripening and oestrogen production (in ovaries)	Glycogen	How glucose is stored as a polymer		
Anti-diuretic hormone (ADH) Kidney Controls water levels in the blood	LH Pi Luteinising hormone	Ovaries Pituitary Gland Ovaries	Lining of the womb to develop. Stimulates pituitary gland to make LH Stimulates egg release and progesterone production in the ovaries Maintains the lining of the womb	Fight or flight response	When the body prepares to fight or run away from a perceived danger. This response is caused by the hormone adrenaline		
Thyroid-stimulating hormone (TSH) Thyroid Stimulates the thyroid gland to secrete thyroxine Luteinising hormone (LH) Ovaries Stimulates egg release and progesterone production in the ovaries Follicle-stimulating Stimulates egg release and progesterone production in the ovaries				Menstrual cycle	A roughly 28 day cycle of changes that occurs in the female reproductive system		
hormone (FSH) Ovaries production (in ovaries) Prolactin (PRL) Breasts Stimulates the breasts to produce milk Growth hormone All cells in the body Stimulates growth and repair	The pill - oral contraceptives the oestrogen to inhibit FSH product no eggs develop and mature. A for a while egg development a	ction so that . After taking	Barrier methods such as condoms and diaphragms prevent the sperm reaching	Menstruation	When the lining of the uterus breaks down and passes out through the vagina, Also known as a period		
Low levels of thyroxine detected in the blood The hypothalamus in the brain releases TRH into the blood TRH acts upon the pituitary gland	will stop completely. The mini pill and injections comprogesterone. High levels stimul cervix to produce a thick much	ulate the	an egg The 'coil' or other	Ovulation	When an egg is releases from its follicle in the ovary, happens roughly on day 14 of the menstrual cycle		
Pituitary gland releases TSH into the blood TSH acts upon the thyroid gland	stops sperm entering the uterus Implant or skin patches of slow	vs. v release	intrauterine devices which prevent the implantation of an embryo	Contraception	A method that prevents fertilisation and therefore pregnancy. Contraception can be hormonal or physical/barrier methods		
Thyroxine is released	progesterone inhibit the mature release of eggs for a number of years.	ration and of months or	Sterilisation or vasectomy - surgical methods of male	ART	Assisted reproductive technology that uses hormones and other techniques to increase the chance of pregnancy in infertile women.		
and production of TSH. This is called negative feedback.	Spermicidal agents which kill or sperm.	or disable	and female sterilisation.		Clomifene therapy is an example of ART 76		





ORMISTON SWB ACADEMY			KS4 - Biology	Exchange and T	ransp	ort			Keyword		Definition	
		alveolus		aorta	ខ អ្ន ម្ន			Pumps blood to the lungs where gas exchange takes place.	Diffusion		of the particles of a gas n solution, resulting in a	
trachea	trachea		vena cava	A have different	Left ver	ntricle	Pumps blood around the rest of the body. Thicker cardiac muscle in the wall.		net movement of particles from a region where they are of a higher concentration to an area of lower			
lung	bronchus	apillary CO ₂ out		veins left atrium	the heart h	Pacemake	•	Controls the natural resting heart rate. Artificial electrical pacemakers can be fitted to correct irregularities.	Surface	concentration to an area of lower concentration The surface area to volume ratio can		
diaphragm	dlaphragm		right atrium	left ventr	structure in aloi	Coronary	arteries	Carry oxygenated blood to the cardiac muscle.	areas to volume ratio		by dividing an object's A) by its volume	
Trachea		ings of cartilage protect the rway.	right ventricle		Different	Heart v	valves	Prevent blood in the heart from flowing in the wrong direction.	Cardiac Output	Cardiac outpu rate	t = stroke volume x heart	
Bronchioles		plits into multiple pathways preach all the air sacs.	Aero	Aerobic respiration		Anaerobic respiration			Stroke		t is the volume of blood	
Alveoli	0 0	laximises surface area for		h oxygen. Occurs in: 10ndria continuously		Respiration when oxygen is in short supply. Occurs during intensive exercise		Volume	pumped by a ventricle per minute. The units are cm ³ min ⁻¹ Heart rate is the number of beats per			
		fficient gas exchange.		be is oxidised by oxygen to During hard exercise, muscle cells are respiring so fast that blood cannot			Heart Rate	minute (bpm)	e number of beats per			
Capillaries	between into/out of bl	ood and carbon dioxide ffuses out.	transfer the energy the organism needs to perform it's functions.			transport enough oxygen to meet their needs. Glucose is partially oxidised to produce lactic acid which builds up in		Stroke Volume	Stroke volume is the volume of blood pumped by one ventricle per			
muscle layer	cannective	e layer	Glucose + oxyg	gen → carbon dioxi water	de +	muscle ti painful a		uusing them to become gued.	Volonie	contraction (cr	n ³)	
	endothelium						Gluco	se \rightarrow lactic acid	Plasma (55%)	Pale yellow	Transports CO ₂ , hormones and	
	Jumen -	endothelium one cell thick	Factors affec	ting rate of diffusion			5			fluid	waste.	
Carry blood away from the	Carry blood to the heart	Capillary Connects arteries and veins	Surface area	Concentration gradient Diffusion is from		fusion tance			Red blood cells (erythrocytes)(45	Carries oxygen	Large surface area, no nucleus, full of haemoglobin.	
heart Thick muscular walls, small lumen, carry blood under high	Thin walls, large lumen, carry blood	d One cell thick to	Increased surface area on exchange surface	area of high concentration to low concentration. A	The smaller th diffusion distance to			 plasma 55% buffy coat <1% (white blood cells and platelets) 	White blood œl (phagocytes an lymphocytes)(<1	d immune	Some produce antibodies, others surround and engulf pathogens.	
pressure, carry oxygenated blood (except for the pulmonary artery).	flow in the wrong direction, carry deoxygenated bloc (except for the pulmonary vein).	Carry blood under	increases diffusion.	large difference in concentration will increase rate f diffusion.		r the rate liffusion,		 red blood cells 45% (erythrocytes) 	Platelets (<1%)) Fragments of cells	Clump together to form blood clots. 78	

KS4 Biology- Ecosys	rems and material cycles (part 1)	Keyword	Definition
A food web shows the feeding	Species availability that indicate water	Ecosystems	a biological community of interacting organisms and their physical environment.
relationship between organisms in	pollution	Biodiversity	the variety of plant and animal life in the world or in a particular habitat
a community.	Clean → Stonefly Some → Shrimp Moderate → Bloodworm Very polluted → Sludge worm	Resources	a substance or object in the environment required by an organism for normal growth, maintenance, and reproduction.
Tertiary consumer/		Community	two or more populations of organisms.
Tertiary consumer/ carnivore/ predator	Species availability that indicate air pollution	Populations	the number of organisms of the same species that live in a particular geographic area at the same time.
Primary consumer/ herbivore	Clean → Bushy lichens and blackspot fungus Some → Leafy lichens Very polluted →Crusty lichens	Interdependent	this means that all the organisms in an ecosystem are dependent upon each other. If the population of one organism rises or falls, then this can affect the rest of the ecosystem.
		Habitats	the place where an organism lives.
Producer		Biomass	the total dry mass of one animal or plant species in a food chain or food web
Examples of abiotic factors	Examples of biotic factors	Abiotic	non-living chemical and physical factors in the environment which affect ecosystems.
Extremes of an abiotic factor can reduce the biodiversity of the ecosystem. For example, ecosystems with a very low	Biotic factors are the interactions between the living things in an ecosystem. This can include grazing and predation. Animals	Adaptations	the biological mechanism by which organisms adjust to new environments or to changes in their current environment.
temperature tend to have low biodiversity Example: Temperature, light, water and pollutants.	which graze on the plant species in an ecosystem can influence which species of plants can survive	Eutrophication	the gradual increase in the concentration of phosphorus, nitrogen, and other plant nutrients in an aging aquatic ecosystem such as a lake.
	Example: competition and predation.	Indigenous	a species can be indigenous to a given region or ecosystem if its presence in that region is the result of only natural processes, with no human intervention.
Eutrophication process	No oxygen	Preservation	The act or process of keeping the environment safe from harmful effects of human activity.
Fertiliser is added to crops & gets b Nitrates and phosphates dissolve in soil water b b b soil water b b sophosphates b b sophosphates b b sophosphates b b sophosphates b b sophosphates b b sophosphates b b sophosphates b b sophosphates b b sophosphates b b sophosphates b b sophosphates b b sophosphates b b sophosphates b sophosphates b solve in b sophosphates b solve in b sophosphates b solve in b solve in solve in b solve in b solve in b solve in b solve in b solve in i solve in i solve in in i solve in i i i solve in i solve in i i i in in i i i i i i i i i i i i	ac. of Excess surface produced → Oxygen	Parasitism	relationship between two species of plants or animals in which one benefits at the expense of the other, sometimes without killing the host organism. Examples include tapeworms and fleas.
washed off and are promo with heavy rain stream/river grow	bte photosynthesis → death of lgae leads to plant aquatic	Mutualism	The interaction between two or more different species in an environment where each species has a net benefit. Example includes an oxpecker bird landing on the back of zebra eating ticks. The oxpecker gets fed and the zebra gets pest control.

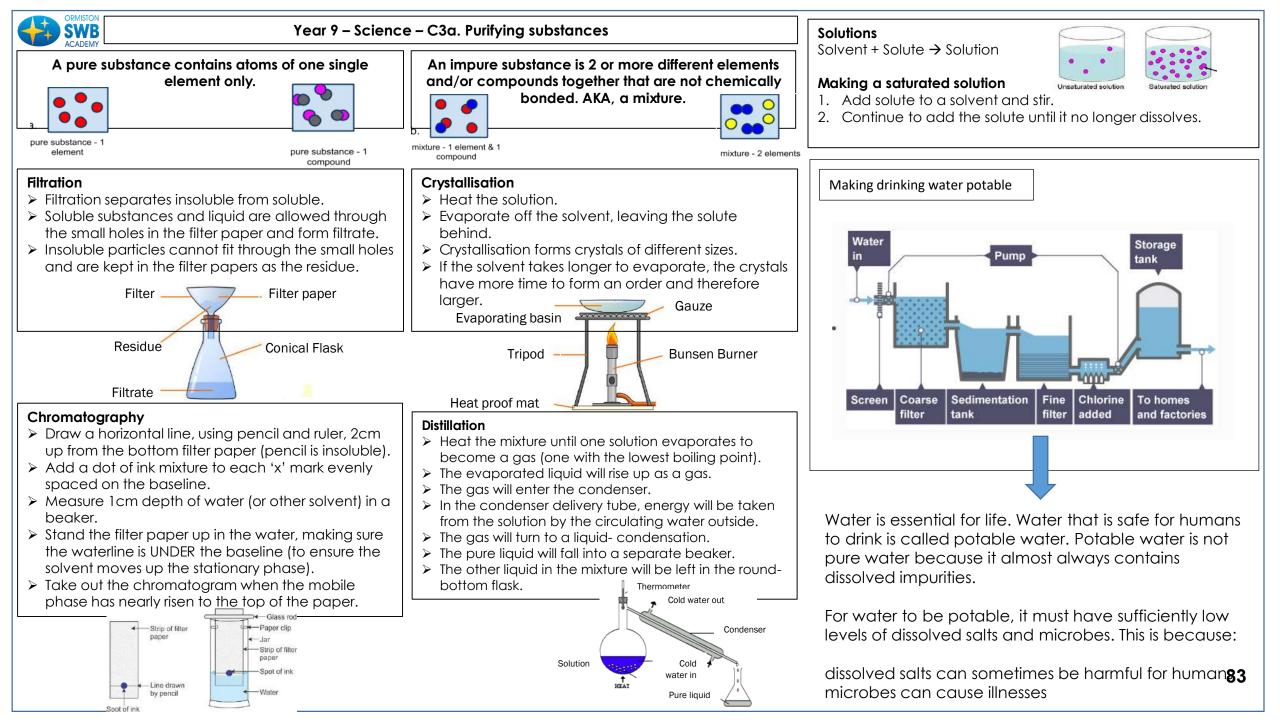


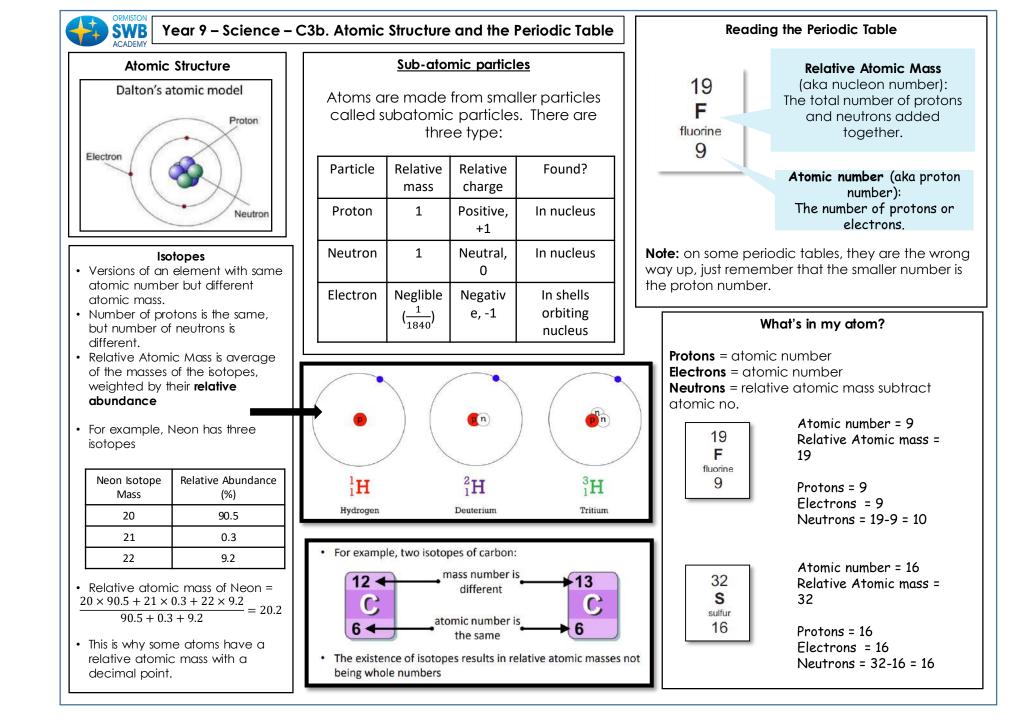
Chemistry Knowledge Organisers



Year 9 – Science – C3a. Purifying substances

	Solids	Liquids	Gases	Chromatogram	The end product in chromatography (paper with separated components).
		20222		Solute	The solid that dissolves.
		SSEE S		Solvent	The liquid that dissolves the solute.
				Solution	Formed when a solvent dissolves a solute.
Arrangement	Particles are close together, next to	Particles touch each other and are next to	Particles are not in a regular arrangement.	Dissolve	The act in which a solution is made (forming a solution).
	each other. Particleseach other. ParticlesThe particles areare in rows . Regularare not in a regularspaced out.arrangement.pattern.			Saturated	When no more solute can be dissolved in a solvent.
Novement	Very little movement,	Particles have some	Particles in gases	Unsaturated	When more solute can be dissolved in a solvent.
	particles vibrate in their fixed positions.	movement. The particles are able to	have lots of movement and	Atom	Smallest component of an element.
	They do not move from one place to another.	rollover each other.	move in all directions.	Molecule	A group of atoms chemically bonded together.
Challenge – energy	Particles have very	Particles have some	Particles have lots of	Compound	Two or more different atoms chemically bonded together.
and attraction of particles	little energy. The particles are	,	energy and there is no or very little	Evaporation	Change of state where a liquid turns to a gas.
	attracted to each other.	other.	attraction between the particles.	Condensation	Change of state where a gas turns to a liquid.
		Physical change	Chemical change	Filtration	Separation technique where insoluble particles are separated from soluble particles and liquid.
		(Reversible)	(Irreversible)	Crystallisation	Separation technique where the solvent in a solution is left to evaporate, leaving the solute behind.
5	olid	For example – melting chocolate	For example – frying an egg	Distillation	Separation technique where liquid mixtures or soluble solutions can be separated based on their boiling points.
Subliming	Freezing Melting	Freezing water into ice	- rusting	Soluble	Can dissolve in water.
Depositing	······································	No new substances or products formed.	One or more new substances has been	Insoluble	Cannot dissolve in water.
Gas	oiling Liquid	There has just been a change of state	formed.	Baseline	The pencil line drawn at the base of the chromatography papeduring chromatography.
-Con	densing→	(solid, liquid, gas)		Mixture	Two or more different atoms not chemically bonded together
			!	Boiling point	The temperature that a liquid turns into a gas.







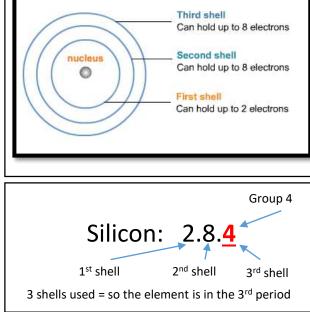
Year 9 – Science – C3b. Atomic Structure and the Periodic Table

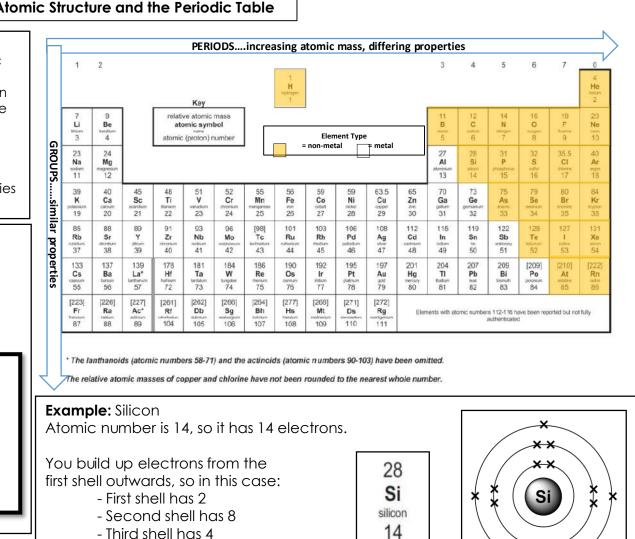
Mendeleev

- Arranged elements by increasing atomic mass but....
- He broke this rule and left some gaps if an element's properties weren't similar to the one above it.
- He thought the gaps were for elements that hadn't been discovered yet and predicted their properties.
- When they were discovered, the properties matched the predictions

Electron Configuration

- Electrons orbit the nucleus in shells.
- First shell holds up to two electrons
- Second shell can hold up to 8 electrons
- Third shell can also hold up to 8 electrons





- Third shell has 4

This can be written as: 2.8.4: or drawn as:

Electron configuration and how it links to the Periodic Table:

Group number: shows the number of electrons in the outer electron shell Period number: shows the number of electron shells In the above example, we can see Silicon belongs to group 4, and is in period 3.

хх

	KS4 Ch	emistry - Chemical Bo	nding and Types of Substances (part 1)	Keyword	Definition			
lonic	Particles are opp	oositely charged ions	Occurs in compounds formed from metals combined with non metals.	lon	An atom with an electric charge, caused by the loss or gain of electrons.			
Particles are atoms that share pairs Oc of electrons		oms that share pairs	Occurs in most non metallic elements and	Cation	A positively charged ion.			
000	of e	electrons	in compounds of non metals.	Anion	A negatively charged ion.			
_	U Particles are atoms which share I D <td></td> <td>Electrostatic force</td> <td>The attractive or repulsive force between two electrically charged objects.</td>			Electrostatic force	The attractive or repulsive force between two electrically charged objects.			
Meto			Occurs in metallic elements and alloys.	Attraction	The electric force that acts between oppositely charged bodies, tending to draw them together.			
		1		Intermolecular force	Forces of attraction which act between molecules.			
lonic b	Keyword ond	-	Definition c force of attraction between oppositely	Atom	The smallest unit into which matter can be divided without the release of electrically charged particles.			
Covale	charged ions. Covalent bond The bond formed when a pair of electrons is shared between two atoms.			Element	An element is a substance whose atoms all have the same number of protons.			
Metalli		The type of bonding found in metals. Positively charged ions in a 'sea' of negatively charged electrons.		Compound	A substance formed when two or more chemical elements are chemically bonded together.			
Lattice	Structure	-	many particles that are bonded together in a	Transfer	Movement of a particle from one place to another.			
Melting) point	-	ke pattern which a substance changed fro the solid ate when heated, or from the liquid state to	Share	Two bodies having equal portions distributed between the two.			
Boiling	point	solid state when cod	•	Delocalised electron	An electron that is not associated with a particular atom within a shell, or held in a covalent bond.			
Charge	9		ric charge, is a characteristic of a unit of	Proton	A particle found in the nucleus of an atom, having a positive charge and the same mass as a neutron.			
Electric	electrons that			Neutron	A particle found in the nucleus of an atom having zero charge and a mass of 1.			
Electrical conductivityAllowing electricity to pass through.Aqueous solutionA mixture that is formed when a substance is dissolved in water.		Electron	A finy particle with a negative charge and very little					
Molten			s been liquefied by heat.		mass.			
Electro		Two electrons occup	oying the same orbital in an atom or	Shell	Area around a nucleus that can be occupied by			
			y forming a nonpolar covalent bond		electrons and usually drawn as circles. 86			
		between atoms.		Nucleus	The central part of an atom or ion.			



KS4 Chemistry - Chemical Bonding and Types of Substances (part 2)

Me	etallic bonding		Ionic bonding								
Giant structure of	Electrons in the outer shell of metal atoms are delocalised and free to move through the	High melting and boiling points	Large amounts of energy needed to break the bonds.	Electrons a	-	Metal atoms lose electrons and become positively charged ions	Group 1 metals form +1 ions Group 2 metals form +2				
whole structure. This sharing of electrons leads to strong metallic bonds.		Do not conduct electricity when solid	lons are held in a fixed position in the lattice and cannot move.	transferred so that all atoms have a noble gas configuration		Non metals atoms gain electrons to	Group 6 non metals form 2 ions				
• • • • <td>Do conduct electricity when molten</td> <td>Lattice breaks apart and the ions are free to move.</td> <td>(full outer she</td> <td>elis).</td> <td>become negativel charged ions</td> <td colspan="2">Group 7 non metals form 1 ions</td>		Do conduct electricity when molten	Lattice breaks apart and the ions are free to move.	(full outer she	elis).	become negativel charged ions	Group 7 non metals form 1 ions				
		Dot and cross	$(I) \rightarrow \left[\left(N_{a} \right)^{+} \left[\left(C_{I} \right)^{-} \right]^{-} \right]$	Structure		Held together attraction be	ist of a regular arrangement of atoms by strong electrostatic forces o tween oppositely charged ions				
High melting and boiling points	This is due to the strong metallic bonds.	diagram (2, 8, 1) (2, 8	✓ L J L ∞ J 3, 7) (2, 8) (2, 8) (2, 8)			Forces act in all directions in the lattic					
Pure metals can be bent and	Atoms are arranged in layers that can slide over			-ide	en Usua	ompound name ds in –ide, it Ily contains only wo elements.	For example: calcium + oxygen → calcium oxide				
shaped Good conductors of electricity and heat	each other. Delocalised electrons transfer energy.	Giant structure	Cl.	-ate	ena usual or ma	ompound name ds in -ate, it ly contains three ore elements one which is always oxygen.	For example: Calcium + carbon + oxygen - calcium carbonate				



	ACADEMY												
	Covalent bonding												
	Sir	nple mole		-			Diamond	Giant covalent s			mhana	and fulleren	
	Low melting boiling po		he needed to overcome the						∬ 1 ┌───			and mileren	
				nolecular forces.	Each	Ţ	Very hard.	Rigid structure.				Excellent conductor.	Contains delocalised
	Poor conduc electrici		No free e	electrons to transfer energy.	carbon atom is bonded		Very high melting point.	Strong covalent bonds.	hene				electrons.
	ze of atoms d molecules	joined by	Simple molecular structures consist of atoms oined by strong covalent bonds. This means hat atoms are smaller than simple molecules.		to four others		Does not conduct electricity.	No delocalised electrons.	Graphen	Single lay graphite atom th	one	Very strong.	Contains strong covalent bonds.
	Dot and cross :			and cross :	Used for cutting tools due to being very hard.						<u> </u>		
		H	X /	now which atom the ectrons in the		Graphite							Hexagonal
suo	Can be sma) bor - All ider	nds come from electrons are ntical	Each carbon atom is		Slippery.	Layers can slide over each other.		<i>(</i>			rings of carbon atoms with
rs of electrons	molecules e.g. ammon	monia H—N—H 2D with bonds: H H bonded together		w which atoms are	bonded to three others forming)	Very high melting point.	Strong covalent bonds.	Fullerenes	Fir:		uckminsterfull erene, C ₆₀ st fullerene to e discovered.	hollow shapes. Can also have rings of five
Atoms share pairs		2	inco 3D ball o + Attem	prrectly at 90° and stick model: pts to show the H-C- angle is 109.5°	layers of hexagonc rings with no		Does conduct	Delocalised electrons	Delocalised			liscovered.	(pentagonal) or seven (heptagonal) carbon atoms.
Ato	Can be gian covalent structures			Simple polymers consist of large chains of hydrocarbons.	covalent bonds between the layers		electricity.	between layers.	g	rapnire,	Very hig melting	n need	of energy led to break
	e.g. polyme	rs		nychocchoons.	Used for e	lectrodes as is	inert.			silicon dioxide	points		

	KS4 Cher	nistry – Acids	and Alkalis (part 1)	Keyword	Definition	
	4 5 6 7 8 9	2 10 11 1	2 13 14		H ⁺ ion	A positively charged hydrogen ion	
				The pH scale and	OH ⁻ ion	A negatively charged, diatomic hydroxide ion.	
			indicators		Aqueous solution	A mixture that is formed when a substance is dissolved in water.	
acidi		alkaline			Acid	A solution that reacts with alkalis, turns litmus red and has a pH of less that 7.	
acial	c neutral			Red in gold groop in	Alkali	A solution which contains an excess of OH ⁻ ions, turns litmus blue and	
	Acids produce hyd	rogen Univ	ersal indicator	Red in acid, green in neutral and blue in alkali		has a pH greater than 7.	
Acids	ions (H+) in aque	ous			Base	A substance that will react with an acid to form only a salt and water.	
	solutions.		Litmus	Red in acid, purple in neutral and blue in alkali	pH scale	A scale going up to 14 showing acidity or alkalinity.	
	Aqueous solution	s of			Indicator	A substance which can change colour depending on the pH of a solution.	
Alkalis	alkalis contain hyd	oxide 🛛 Me	thyl orange	Red in acid, yellow in	Concentration	The amount of a solute dissolved in a certain volume of solvent.	
	ions (OH ⁻).		_	neutral and yellow in alkali	Concentrated	Containing a large amount of solute dissolved in a small volume of	
				Colourless in acid and in		solvent.	
Base	that reacts with ar	acid	neutral and pink in alkali		Dilute	A low concentration of solute in a solution.	
	to form a salt and water		to form a salf and water		In neutralisation reactions,	Strong acid Weak acid	An acidic solute that dissolves completely into ions when it dissolves.
	only		reaction is	hydrogen ions react with		An acidic solute that does not dissociate completely into ions when it dissolves.	
Examples of solub	Alkalis e.g. sodiu	^{jm} II hot	between an acid and a base hydroxide ions to produce water: $H^+ + OH^- \rightarrow H_2O$		Salt	A compound formed by neutralisation of an acid by a base.	
bases	hydroxide, potas hydroxide	sum H			Filtration	Using a filter to separate insoluble substances from a liquid.	
		/ L		2	Crystallisation	Separating the solute from a solution by evaporating the solvent.	
	Read	ctions with ac			Soluble	A substance that can be dissolved in a certain liquid.	
Matala	Metal + acid → me	etal salt +	Magnesiu	m + hydrochloric acid \rightarrow	Insoluble	A substance that cannot be dissolved in a certain liquid.	
Metals	hydrogen		magnesi	um chloride + hydrogen	Solute	Describes a substance that dissolves in a liquid to make a solution.	
					Solvent	Describes the liquid in which a substance dissolves to make a	
Metal oxides	Metal oxide + acid \rightarrow	metal salt +	1	de + sulfuric acid \rightarrow copper	Solution	solution. Formed when a substance has dissolved in a liquid.	
	water			sulfate + water			
	Metal hydroxide + acid	\rightarrow metal salt	Sodium hydr	oxide + nitric acid \rightarrow sodium	Burette	A piece of apparatus used to accurately measure the volume of solution that has been added during a titration.	
Metal hydroxides	+ water		,	nitrate + water	Pipette	A piece of apparatus used in a titration to accurately measure a set	
						volume of a solution.	
Metal carbonates	Metal carbonates + a			arbonate + sulfuric acid \rightarrow	End-point	When just enough solution has been added from the burette to react	
	salt + carbon dioxid	e + water	calcium sulta	te + carbon dioxide + water	lonic equation	with all the solution in the flask in a titration experiment.	
					A balanced equation that only shows the ions that react together. The spectator ions are not included.		
Gas	Test		Posil	ive result	Half equation	A chemical equation written to describe an oxidation or reduction	
Hydrogen	Burning splint	'squeak	y pop' sound.			half-reaction.	
Carbon dioxide	e Limewater	Goesch	oudy (as a solid	calcium carbonate forms).	Spectator ion	These are ions that do not change within a reaction. 89	

	KS4 Chem	istry – Acids and Alkalis (part 2)			Producing salts from soluble reactants			
	Making pur	e, dry insoluble salts	Soluble salts	Soluk	ble salts can be made from reacting acids with solid insoluble substances			
Step 1	Add insoluble reactant (e.g. metal oxide) to	Add until there is an excess of insoluble			(e.g. metals, metal oxides, hydroxides and carbonates).			
	acid	reactant.	Production of	Add t	the solid to the acid until no more dissolves. Filter off excess solid and then			
Step 2	Filter the solution	Collect the filtrate in a conical flask and dispose of the residue.	soluble salts	alts crystallise to produce solid salts.				
					Solubility			
Step 3	Crystallisation	Heat the filtrate using a Bunsen burner to evaporate the water from the solution.	Sodium, potas and ammoni		All common sodium, potassium and ammonium salts are soluble e.g. sodium chloride and potassium fluoride.			
Step 4	Evaporation	Leave the evaporating basin with the heated filtrate to evaporate any remaining water	Nitrates		All nitrates are soluble e.g. potassium nitrate.			
	used to work out the pred	and make pure, dry insoluble salts.	Sulfates		Common chlorides (e.g. sodium chloride) are soluble, expect those of silver and lead.			
	ach other to form salt and		Carbonates of	and	Common carbonates and hydroxides are insoluble except those			
-	Use the pipette to a	dd 25 cm ³ of alkali to a conical flask and add a	hydroxide	S	of sodium, potassium and ammonium.			
Step		few drops of indicator.	Strong and weak acids (HT ONLY)					
Step 2	Fill the burette with the acid from the bu	acid and note the starting volume. Slowly add irette to the alkali in the conical flask, swirling to mix.	Concentra	ited	High mass of substance in a given volume of solution			
Step 3	appropriate colour c	e acid when the end-point is reached (the hange in the indicator happens). Note the final Repeat steps 1 to 3 until you get consistent readings.	Dilute		Low mass of substance in a given volume of solution			
Sta	ite Symbol	Meaning	Strong ac	ids	Completely ionised in aqueous solutions e.g. hydrochloric, nitric and sulfuric acids.			
	S	Solid		• -1 -	Only partially ionised in aqueous solutions e.g. ethanoic acid, citric			
	I	Liquid	Weak aci		acid.			
	g	Gas	Hydrogen		ion As the pH decreases by one unit (becoming a stronger acid), the			
	aq	Aqueous solution	concentra	tion	hydrogen ion concentration increases by a factor of 10.			



KS4 Chemistry – Calculations involving masses

formula mass:

2. How to deduce the molecular formula from the empirical formula and relative

1. How to find an empirical formula:

Example: 10.0g of calcium reacts with 17.8g of chlorine. Find the empirical formula of the product that is formed.

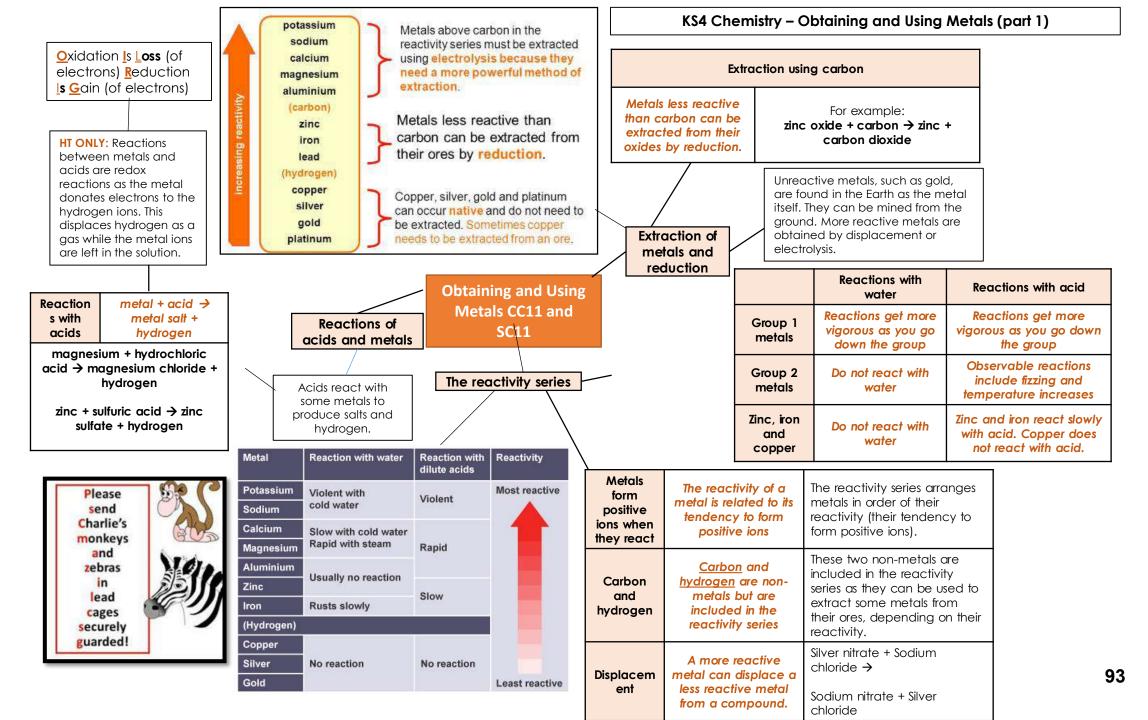
1. Symbol	Ca	Cl			
2. Mass (g)	10.0	17.8			
3. A _r	40.0	35.5			
4. Divide mass by A _r	$\frac{10.0}{0.25} = 40$	$\frac{17.8}{35.5} = 0.50$			
5. Divide answers by smallest number	<u>0.25</u> = 1 0.25	$\frac{0.50}{0.25} = 2$			
6. Empirical formula	CaCl ₂				

	'	formula of the					
product that is fo	rmed.			ole: The empirical formu			
1. Symbol	Са	Cl		and its relative formula ce the molecular formu			
2. Mass (g)	10.0	17.8	1. Find	1. Find the empirical formula C + H + H			
3. A _r	40.0	35.5		by adding up the e atomic masses of all	12 + 1 + 1 + 16 = 30		
4. Divide mass by A _r	$\frac{10.0}{0.25} = 40$	$\frac{17.8}{35.5} = 0.50$	of the	atoms	100 - (
5. Divide answers by smallest numbe	$\frac{0.25}{0.25} = 1$	<u>0.50</u> = 2 0.25		de the relative formula by the empirical formula	<u>180</u> = 6 30		
6. Empirical formul	a CaCl ₂		empir	tiply the numbers in the ical formula to get the cular formula	 CH₂O x 6 So C₆H₁₂O₆ 		
	ate the mass	ses of reactan	ts or				
3. How to calcul products: Example: Calcula 53.4g of aluminiur 1. Write the balanced equation	te the mass of	f chlorine need		A, or M,			
products: Example: Calcula 53.4g of aluminiur 1. Write the balanced	te the mass of m chloride. 2AI + 3Cl ₂ -> 2 • M, Cl ₂ =	f chlorine need	ed to make	A, or M,	mass		
products: Example: Calcula 53.4g of aluminiur 1. Write the balanced equation 2. Calculate M, of substances in the	te the mass of m chloride. 2AI + 3Cl ₂ -> 2 • M, Cl ₂ = • M, AICl ₃ = • (3)	f chlorine need 2AICI ₃ 2 x 35.5 = 71	ed to make 33.5 AICI ₃ 2 x 133.5) AICI ₃	A _r or M _r (g/mol)	nass (g) number o		
products: Example: Calcula 53.4g of aluminium 1. Write the balanced equation 2. Calculate M, of substances in the question 3. Calculate the	the the mass of m chloride. 2AI + $3CI_2 \rightarrow 2$ • $M_{\rm r} CI_2 =$ • $M_{\rm r} AICI_3 =$ • (3)	f chlorine need 2AICl ₃ 2 x 35.5 = 71 27 + (3 x 35.5) = 1 3Cl ₂ makes 2 x 71) Cl ₂ makes (2 213g Cl ₂ makes 2 213g Cl ₂ makes 2	ed to make 33.5 AICI ₃ 2 x 133.5) AICI ₃ 57 g AICI ₃ 67 g AICI ₃	A _r or M _r (g/mol)	nass (g) number o moles (mo 5. ass of ute (g) ion_volume		

	6. Keyword	7. Definition							
	atom	The smallest neutral part of an element that can take place in chemical reactions.							
	Avogadro constant*	The number of particles in one mole of a substance (6.02 x 10^{23} atoms, molecules, formulae or ions).							
	closed system	Substances cannot enter or leave such as a precipitation reaction in a stoppered flask.							
	concentration	The amount of solute dissolved in a stated volume of a solution. Units include g/dm3.							
	conservation of mass	During a chemical reaction, the overall mass of substances does not change so the total mass of reactants is equal to the total mass of products.							
	empirical formula	The simplest whole number ratio of atoms or ions of each element in a substance.							
	excess reactant	There is more of this reactant present than is needed so it is not completely used up in a reaction.							
	limiting reactant	There is less of this reactant present than is needed so it is completely used up in a reaction. The mass of product formed is controlled by this reactant.							
	mole*	One mole of particles of a substance is defined as: a) the Avogadro constant number of particles (6.02 x 10 ²³ atoms, molecules, formulae or ions) of that substance b) a mass of 'relative particle mass' g. The SI unit symbol is mol.							
	molecular formula	This represents the actual number of atoms of each element in one molecule.							
	molecule	A particle consisting of two or more atoms joined together by bonds.							
	open system	Substances can enter or leave such as a reaction in an open flask that takes in or gives out a gas.							
7	precipitate	An insoluble substance that is formed when two soluble substances react together in solution.							
	precipitation	A reaction in which a precipitate is formed.							
	product	A substance formed in a reaction.							
	reactant	A substance used up in a reaction.							
	reaction	A process in which reactants are converted to different substances called products.							
	relative atomic mass	$(A_{\rm r})$ The mean mass of an atom relative to the mass of an atom of C-12 which is assigned a mass of 12. Unit is g/mol.							
	relative formula mass	$\left(M_{r}\right)$ The sum of the relative atomic masses of all the atoms or ions in its formula. Unit is g/mol.							
	stoichiometry*	The ratio of moles of each substance in a reaction. 91							
	volume	The amount of space hat a liquid takes up. Units include cm^3 and dm^3 .							

	KS4 Chemistry – Ele	mistry – Electrolytic Processes		Splitting up	water, the ions	compound is melted or dis are free to move. These	are then	<u>O</u> xidation	l <u>I</u> s <u>L</u> oss, <u>R</u> educ	tion <u>I</u> s <u>G</u> ain
ACADEMIT	Key Word	<u>Definition</u>	Process of electrolysis	using electricity	able to conduct electricity and are called electrolytes. Passing an electric current the electrolytes causes the ions to move to the electrodes.		t though to the	buisr		tracted from molten sing electrolysis.
	Electrolysis	Decomposition/break down of a compound using electrical energy.	Electrode	Anode Cathode		electrode is called the a electrode is called the a	node. athode.	Extracting metals u electrolysis	too reactive to	sed when the metal is o be extracted by with carbon.
	lons	Charged particle.	Where do the ions go?	Where do the ions Cations		ons are positive ions and they move to the negative cathode. ns are negative ions and they move to the		xtracting elec	The process is expensive due to large amounts of energy needed to produce the electrical current. Example: aluminium is extracted in this	
	Cations	CationsPositively charged ions, formed by losing electrons. Usually metal form cations.AnionsNegatively charged ions, formed by gaining electrons. Usually non-metal form anions.		Electrolytic p	processes	positive anode.				way.
	Anions			KS4 EC	DEXCEL		Bromide ions Br ·	happer equation At the	<u>Higher tier:</u> You can display what is happening at each electrode using half- equations: At the cathode: $Pb^{2+} + 2e^{-} \rightarrow Pb$ At the anode: $2Br^{-} \rightarrow Br_2 + 2e^{-}$	
	Electrodes	A rod made of metal or carbon which carries the	solutions			°••.	Molten lead (II) — bromide		potassium sodium	most reactive K
	Cathode	current in the electrolyte. An electrode that is negatively charged.	At the negative electrode Metal will be produced on the electrode if it is less reactive than hydrogen. At the positive electrode Oxygen is formed at positive electrode. If you have a halide ion (Cl ⁻ , l ⁻ , B ⁻) you will get chlorine, bromine or iodine formed at that electrode. / / The ions discharged when an aqueous solution is electrolysed using inert electrodes depend the relative reactivity of the elements involved.						calcium magnesium	
	Anode	An electrode that is positively charged.							aluminium carbon	Al C Zn
	Electrolyte	lonic liquid where moving ions carry the current during							zinc iron tin	En Fe Sn
	Reduction	electrolysis When a positive ion gains		Copper is a ver			icn't pure opoug		lead hydrogen	Pb H
	NeutritionWhen a positive ion gains electrons.OxidationWhen a negative ion loses electrons.DischargedWhen ions convert to elements due to transfer of electrons during electrolysis.		Using copper	electrical cond		Much of the copper available isn't pure enough for this use so it is purified using electrolysis.			copper silver gold	Ag Au
				The anode is m impure copper a cathode is made	and the solution	the Both electrodes are placed in copper sulfate solution. Copper ions (Cu^{2+}) leave the anode and			platinum	least reactive Pt
				copper		are attracted to the cathode.			the second s	nent given off at positive electrode rine, Cl2
	(H) Half equations	lonic equation showing transfer of electrons in oxidation and reduction.	Electrodes The cathode of pure copper builds up The anode decreases in size. The impu- behind form a sludge.					bron iodic	nide, Br ⁻ bror de, 1 ⁻ lodi	nine, Br2 ne, 12 gen, O2
								sulfa	ate, SO ₄ oxy	2 2





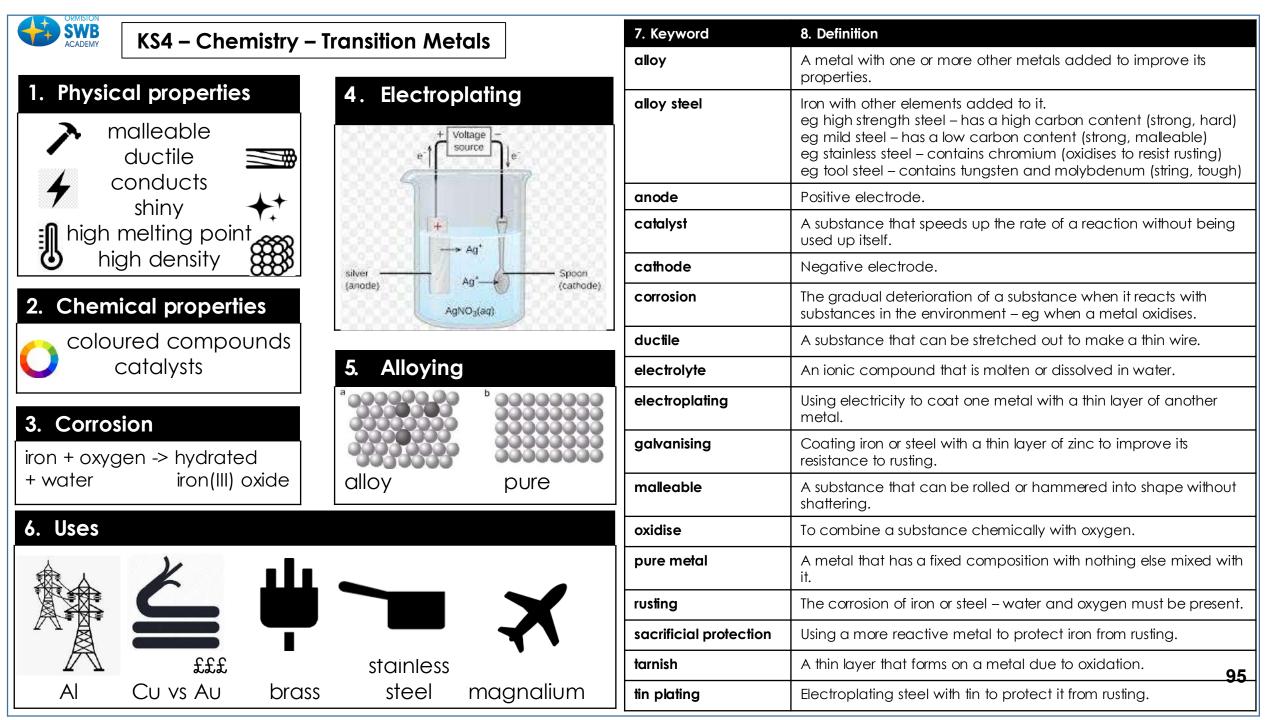


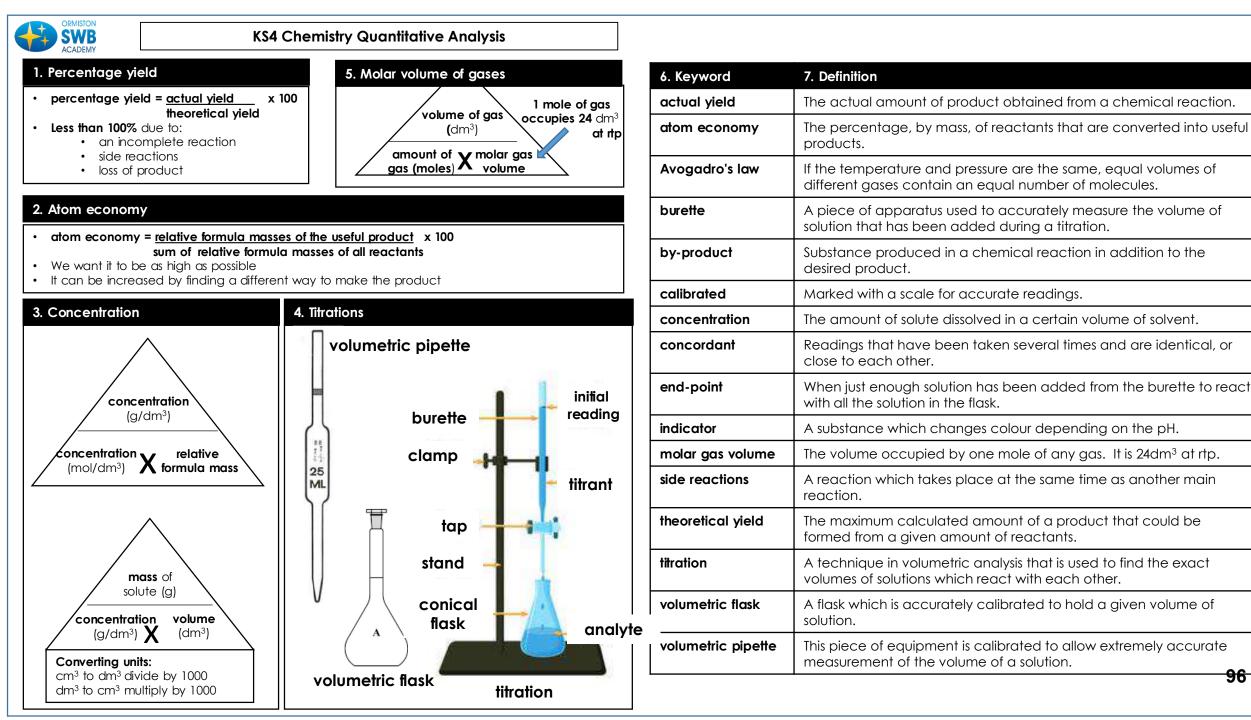
KS4 Chemistry – Obtaining and Using Metals (part 2)

Biolo	gical methods o	f metal extractior	I	Oxidatio	on and reduction OI	n in terms of e NLY)	electrons (HT			Ionic half eq	uations (HT only)						
Metals ores	These resources are limited	Copper ores especially are becoming sparse. New ways of extracting copper from low-grade ores are being developed. These plants are then harvested and burned; their ash contains the metal compounds.		Metals and oxygen	Metals react with oxygen to form metal oxides	magne 2Mg	$\begin{array}{c} \mathbf{m} + \mathbf{oxygen} \rightarrow \\ \mathbf{sium oxide} \\ + O_2 \rightarrow \\ MgO \end{array}$		For	lonic half equations show what	For examp The ionic equati reaction betwee copper (II) i Fe + Cu²⁺ → Fe	on for the in iron and ons is:					
Phytomining	Plants absorb metal compounds			developed. These plants are then harvested and burned their ash contains the	These plants are then harvested and burned their ash contains the metal compounds.	These plants are then harvested and burned; their ash contains the		These plants are then harvested and burned; their ash contains the		Reduction	This is when oxygen is removed from a compound during a	with hydroge	oxides reacting n, extracting lov vity metals	~	displace - ment reactions	happens to each of the reactants during reactions	The half-equation is: Fe → Fe²⁺ - The half-equation
Bioleaching	Bacteria is used to produce leachate solutions that	The metal comport can be processed obtain the metal e.g. copper can	d to from it be	This is when oxygen is e.g. metals reacting with				(II) ions i Cu²+ + 2e ⁻ -	is:								
	contain metal compounds	obtained from its compounds by displacement or electrolysis.		Oxidation Compound during a reaction		oxygen, r	oxygen, rusting of iron				Word Reactivity series	Definition List of metals in the order reactivity					
		6 1 1 1					Life cycle assessment				Cations	Positive ions					
Reduce, reuse		ys of reducing the u	This, there	fore, reduces e	energy sources		Life cycle assessments	stag	/ are assessed jes: Extraction and		Displacement reactions	When a reactive metal a less reactive metal	replaces				
recycle		ed resources		d, reduces wa: environmental i	ste (landfill) and mpacts.		are carried out to	p	processing raw		Extraction	Taking a metal out of a compound	, r				
	line of fea	matela alma	Most of th	e energy requi	red for these	LCAS	assess the environment	p	Manufacturing backaging		Native state	Unreactive metals found naturally					
Limited raw materials	building m	Used for metals, glass, building materials, plastics and clay ceramics building materials, plastics building materials, plastics and clay ceramics building materials, plastics and clay ceramics building materials, plastics building materials, plastics and clay ceramics building materials (building materials) building materials (building materials)		products		Jse and operc during lifetime Disposal	ition	Electrolysis	Passing electricity through molte ionic compound to decompose into it's elements								
				les can be reu:	and They are		Allocating numerical		ue judgments		Oxidation	Gain of oxygen by a substance					
Reusing and recycling	me	melting and glas		and melted to r lucts. Products		Values	values to pollutant effects is		cated to the e utants so LCA ely objective p	is not a	Reduction	Loss of oxygen by a sub	ostance				
			reused dre	e recycled.			difficult				Redox	Reactions in which oxic reduction occurs	lation and				

When a metal reacts with oggen and sometimes with water

Corrosion



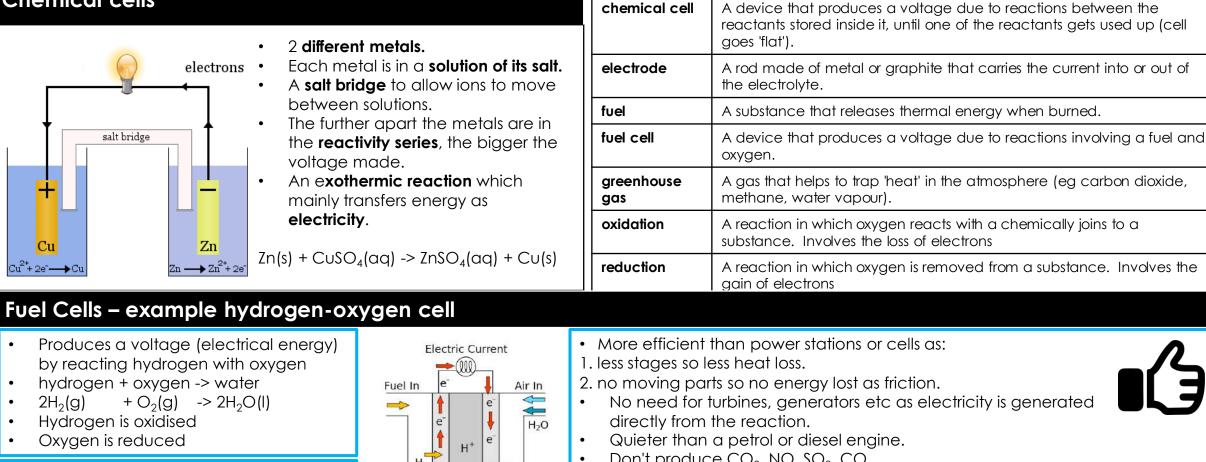


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KS4 – Chemistry – Chemical and Fuel Cells

Chemical cells



Keyword

by-product

Definition

A substance that is made in addition to the desired product.

- Need non renewable natural gas to make hydrogen
- Making hydrogen produces carbon dioxide as a by-product
- Hydrogen has to be stored safely (explosive, takes up space)

H₂ 07 H^+ Excess Fuel H₂O Anodé Cathode Electrolyte

- Don't produce CO_2 , NO, SO_2 , CO.
- Only by-products are water and heat.

 $2H_2(g) \rightarrow 4H+(ag) + 4e$ - $O_2(g) + 4H^+(aq) + 4e^- > 2H_2O(I)$

Unused

Gases

Out

	KS4 Chemistry Dyr	KS4 Chemistry Dynamic Equilibrium and Fertilisers								
	Reversible Read	tions and Equilibria	3	Le Chatelier's Principles	States that when a system experiences a disturbance (change in condition), it will					
Reversible reactions	In some chemical reactions, the products can react again to re-form the reactants.	Changing	Changing If the concentration of a reactant is increased, more products will be formed .		respond to restore a new equilibrium state.					
Representing		concentration	If the concentration of a product is decreased, more reactants will react.	Changing	If the concentration of a reactant is increased, more products will be formed .					
reversible reactions		Changing	If the temperature of a system at equilibrium is increased:	concentration	If the concentration of a product is decreased, more reactants will react.					
The direction	The direction of reversible reactions can be changed by changing conditions:	temperature	 Exothermic reaction = products decrease Endothermic reaction = products increase 	Changing	If the temperature of a system at equilibrium is increased:					
	heat A + B C + D cool	Changing pressure	For a gaseous system at equilibrium: - Pressure increase = equilibrium position shifts to side of equation with smaller number of	temperature	 Exothermic reaction = products decrease Endothermic reaction = products increase 					
ouce	Equilibrium Equilibrium Equilibrium Equilibrium Equilibrium		 Pressure decrease = equilibrium position shifts to side of equation with larger number of molecules. methane steam 	Changing pressure (gaseous reactions)	 For a gaseous system at equilibrium: Pressure increase = equilibrium position shifts to side of equation with smaller number of molecules. Pressure decrease = equilibrium position shifts to side of equation with larger 					
Time	This process uses nitrogen from the air an	d hydrogen from	methane + steam → hydrogen + carbon monoxide	Equilibrium in reversible reactions	number of molecules. When a reversible reaction occurs in apparatus which prevents the escape of reactants and products, equilibrium is reached when the forward and reverse reactions occur exactly at the same rate.					
The Haber pro	natural cas to form ammonia. The reacti	on is reversible	hydrogen air	Equilibrium in reversible	When a reversible reaction occurs in apparatus which prevents the escape of reactants and products, equilibrium is					
Optimum tempe		•	hydrogen + oxygen → water This reaction removes oxygen from the air to leave nitrogen	reactions	reached when the forward and reverse reactions occur exactly at the same rate.					
Optimum pres	Optimum pressure Interception of the optimum pressure for the Haber process is 200 Optimum conditions The optimum temperature for the Haber process is 450°C and optimum pressure is 200 atmospheres. These are economically viable conditions as they produce the best yield to cost ratio. Interception of the equilibrium but it does not alter the position of the equilibrium but it does increase the rate The use of a catalyst The Haber process uses an iron catalyst. This does not alter the position of the equilibrium but it does increase the rate 450 °C		from the all to leave hittogen		This process uses nitrogen from the air and					
Optimum cond			450 °C pitragen + budtagen - ammeria	The Haber Process	hydrogen from natural gas to form ammonia. The reaction is reversible and uses optimum conditions and a catalyst in order to reach dynamic equilibrium.					
The use of a cc			200 atmospheres	NPK fertilisers	Formulations of various salts containing appropriate percentages of the elements.98					

4		B	KS4 (Cher	nistry	y Gro	oups in the Pe	riodic	: Table			Keyword	Definition		
		Alkali metals	Halogens	N	loble ga	ases			Group 0				The atoms get larger as you go down, so		
<u>1</u> H	2 Be	Transiti	n metals		0 He Ne		das	active, mole	do not form cules		is is due to having full ter shells of electrons.		the single electron in the outermost shell (highest energy level) is attracted less strongly to the positive nucleus. The		
	n Mg	Ti V Cr M		S C	Ar		Boiling P OZ	points ir the g	ncrease down group		Increasing atomic number.	Reactivity of group 1	electrostatic attraction with the nucleus gets weaker because the distance between the outer electron and the		
Cs		Hf Ta W Re	RuRhPdAgCdInSnSbeOsIrPtAuHgTIPbBi		Xe Rn		Used in bo	alloons		Due to being less dense than air, which means balloons will float.			nucleus increases. Also the outer electron experiences a shielding effect from the inner electrons, reducing the		
Fr	Ra Ac	Rf Db Sg Bl		Troup b	we the		Used in	signs	Glows when	elect	ricity flows through it.		attraction between the oppositely charged outer electron and the nucleus.		
	arranged in ler of atom number	properties a	Elements with similar properties are in columns called groups Elements in the same period (row) have the same number of electron shells.		with unreactive argon		When Group 7 elements react, the atoms gain an electron in their outermost shell. Going down the group,								
		Group 1			Group 7					Reactivity of	the outermost shell's electrons get further away from the attractive force of the nucleus, so it is harder to attract and				
	Metal	Reaction with water	Word equation				f molecules made of a pair of atoms Have seven electrons in their outer shell. Form -1 ions.				heir outer shell. Form -1 ions.	group 7	gain an extra electron. The outer shell will also be shielded by more inner shells of		
	Lithium	Fizzing	Lithium + water → Ithium hydroxide + hydrogen	Hologens			and boiling points inc he group (gas → liqui solid)		Increasing atomic mass number.			electrons, again reducing the electrostatic attraction of the nucleus for			
	Sodium	Fizzing more vigorously than	Sodium + water → sodium hydroxide + hydrogen			Reacti	vity decreases down group	decreases down the group Increasing proton number means an electron is less easily gained as outer shell is further away from nucleu therefore the attraction force is weaker.			is further away from nucleus,		an incoming electron. Elements in Group 0 of the periodic table		
Po	otassium	lithium Fizzes and burns with a lilac flame	Potassium + water → potassium hydroxide + hydrogen	w	ith met	tals	Forms a metal halide		Aetal + halogen → meto halide e.g. Sodium + chlorine - sodium chloride		halide . Sodium + chlorine →		e.g. NaCl metal atom loses outer shell electrons and halogen gains an outer shell electron		are called the noble gases. They are unreactive because their atoms have stable arrangements of electrons. The atoms have eight electrons in their outermost shell, apart from helium which
4		t and easily cut	Low melting and bailing points.	h	With ydrogo		Forms a hydrogen halide	e.g. I			hydrogen halide Hydrogen + bromine →		drogen halide rogen + bromine → Dissolve in water to form acidic solutions.		has just two but still has a complete outer shell. The stable electronic structure explains
Alkali meta	Rec	ry reactive with gen, water and chlorine activity increases own the group	Only have one electron in their outer shell. Form +1 ions. Negative outer electron is further away from the positive nucleus so is more easily lost.	so so	With aqueou lution alide s	us ofa	A more reactive halogen will displace the less reactive halogen from the salt	C	Chlorine + potassium bromide → potassium chloride + bromine		(HT) These are redox reactions. The halogen gains electrons and the halide ion from the compound loses electrons.		why they exist as single atoms; they have no tendency to react to form molecules. The boiling points of the noble gases get higher going down the group. For example, helium boils at -269 °C and 99 radon boils at -62°C.		



KSA Chemistry Pates of Peaction and Energy Changes in Peactions

Kowword

	KS4 (Chemistry Rates of Reac	Keyword	Definition				
F	Rates of React	ion		Energy Cho	anges			
Rate of chemical reaction Rate of used or produced in a give	ring the reactant uct formed Rate	= <u>quantity of reactant used</u> time taken = <u>quantity of product formed</u> time taken	Endothermic	Activation energy Products	Products are at a higher energy level than the reactants. As the reactants form products, energy is transferred from the surroundings to the reaction	Collision theory	Chemical reactions can only occur when reacting particles collide with each other with sufficient energy.	
Factors o	affecting the rate		E	Reactants	mixture. The temperature of the surroundings decreases because energy is taken in			
Temperature	-	temperature, the quicker rate of reaction.			during the reaction.		This is the minimum amount of	
Concentration		concentration, the quicker rate of reaction.		Activation energy	Products are at a lower energy level than the	Activation energy	energy colliding particles in a reaction need in order to react.	
Surface area		surface area of a reactant oker the rate of reaction.	Exothermic					
Pressure (of gases)		act, the higher the pressure the quicker the rate of reaction.	Exot	Products	surroundings. The temperature of the surroundings increases because energy is released		Occur in the following:	
Volume.cm ³ Slope of tangent $= \frac{25 \text{ cm}^3}{60 \text{ s}}$ 100 $= 0.42 \text{ cm}^3 \text{ s}^{-1}$	Quantity	Unit			during the reaction.		- Salts dissolving in water	
80 80 70 25 cm ²	Mass	Grams (g)		Calculate the overall energy change for the forward		Heat energy changes	Neutralisation reactionsDisplacement reactions	
50 50 40 50 50 50 50 50 50 50 50 50 50 50 50 50	Volume	cm ³		$N_2 + 3H_2$		C C	- Precipitation reactions	
20 40 60 80 100 120 140 Trees	Rate of reaction	Grams per cm ³ (g/cm ³) HT: moles per second (mol/s)	ation	Bond energies (in kJ/mol)	: H-H 436, H-N 391, N≡N			
Catalyst Catalyst chor rate of a cher reaction but is in the reaction	nical s not used		Bond energy calculation	Bond breaking: 945 + (3 2253 k	3 x 436) = 945 + 1308 =	Exothermic reactions	Heat energy is given out as bonds are being formed.	
Enzymes These are biolicatalysts.	°	REACTARTS	lu pu	Bond making: 6 x 3	391 = 2346 kJ/mol			
Catalysts prov different react How do pathway whe	How do Catalysts provide a different reaction pathway where		Bor	Overall energy change = 2253 - 2346 = -93kJ/mol		Endothermic reactions	Heat energy is taken in as bonds are being broken.	
as much energy	actants do not require s much energy to react hen they collide. Therefore reaction is exothermic overall.						100	

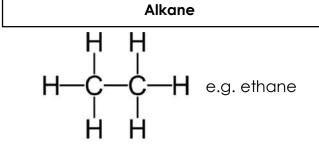
	20 °C	Butane & Propane	KS4 C	Chemistry Fue	ls	Keyword	Definition
ACADEMY	200 °C <u> </u>	Petrol	Boiling poin (temperature at which		What happens as the hydrocarbon chain length increases?	Crude oil	A finite resource. Consisting mainly of plankton that was buried in the mud, crude oil is the remains of ancient biomass.
Crude Oil	370 °C	Fuel Oil	Viscosity (how easily it fl Flammabilit (how easily it b	y		Hydrocarbons	 make up the majority of the compounds in crude oil made up of hydrogen and carbon only.
The oil i heated in furnace	a	Lubricating oil, Parrafin Wax, Asphalt	Methane (CH ₄)	-H H H H-C-C-	Display formula for first four alkanes H H H H H H H H H - C - C - C - H H - C - C - C - H	General formula for alkanes	C_nH_{2n+2} for example: C_2H_6 or C_6H_{14}
1. Hydrogen fuel	Hydrogen reacts with oxygen in the engine to power the vehicle Crude oil,		duct gases released , i diesel oil are non-renewable.	Η Η Ethane (C ₂ H ₆) Con Methane	$\frac{H}{H}\frac{H}{H}\frac{H}{H}\frac{H}{H}\frac{H}{H}\frac{H}{H}\frac{H}{H}$ Propane (C ₃ H ₈) Butane (C ₄ H ₁₀) The plete combustion of methane: + oxygen → carbon dioxide + water + energy	Cracking	 The breaking down of long chain hydrocarbons into smaller, more useful chains. Helps supply meet demand. Can be done by various methods including catalytic cracking and steam cracking.
2. Fossil fuels 3. Sulfur dioxide	natural gas and coal Released from burning hydrocarbo ns with sulfur impurities in	sulfur dioxide dissolve This damages plant lif	natural gas and is also non- es in rain water to form acid rain. ie and can make water habitats n also erode limestone and	nplete bustion	$(1) + 2O_2(g) \rightarrow CO_2(g) + 2 H_2O(I)$ The carbon and hydrogen in the fuels are oxidised . Carbon dioxide, water and energy are released.	Fractions	 The hydrocarbons in crude oil can be split into fractions. Each fraction contains molecules with a similar number of carbon atoms in them. The process used to do this is called fractional distillation.
4. Oxides of nitrogen	Oxygen and nitrogen react under high temperatur es to form these.		of nitrogen can damage the dso dassified as greenhouse piratory problems.	ω⊆	There is not enough oxygen available for complete combustion.	Using fractions	 Fractions can be processed to produce fuels and feedstock for petrochemical industry. We depend on many of these fuels; petrol, diesel and kerosene. Many useful materials are made by the petrochemical industry; solvents,
5. Incomplete combustion issues	1. Carbon monoxide is an odourless, toxic gas that can kill	atmosphere and can	o produced that builds up in the cause global dimming. This of sunlight that reaches the Earth patterns.	Incomplet combustio	The products of the reaction are carbon monoxide, carbon and water.		lubricants and polymers.



KS4 Chemistry Earth and Atmospheric Science

				Volcano activity	Billions of years ago the was intense volcanic		This released gases (formed to early atmos	osphere and water	Effects of climate change		
argon		<u>C</u> m	Deve entran	1 st Billion years	activity		vapour that conden oceans.	sed to form the	Rising sea levels		
oxyge		Gas Nitrogen	Percentage 78%		Released from volcani	ic	Nitrogen was also re building up in the ati		Extreme weather events such as severe storms		
04985		Oxygen	21%	Other gases	eruptions		proportions of ammo also produced.	onia and methane	Change in amount and		
		Argon	0.9%	Reducing carbon	When the water vapor		This formed carbonc		distribution of rainfall		
nitrogen		Carbon dioxide	0.04%	dioxide in the atmosphere	formed and the carbo	condensed, the oceans formed and the carbon dioxide dissolved into it		his reduced the kide in the	Changes to distribution of wildlife species with some becoming extinct		
		u]			Hu	man activities and	greenhouse gases		
		Но	w oxygen ir	ncreased		_	Carbon dioxide		s that increase carbon dioxide levels ing fossil fuels and deforestation.		
Algae and plants		produced the ox n the atmospher photosynth	e, through	carbon dioxide + water	carbon dioxide + water \rightarrow glucose + oxygen 6CO ₂ + 6H ₂ O \rightarrow C ₆ H ₁₂ O ₆ + 6O ₂			include raising li	ities that increase methane levels ivestock (for food) and using landfills		
Oxygen in the atmosphere	First pr	oduced by algae ago.	2.7 billion yea	rs gradually produce more	Over the next billion years plants evolved to gradually produce more oxygen. This gradually increased to a level that enabled animals to evolve.			(the decay of organic matter released meth There is evidence to suggest that human act			
		How co	arbon dioxic	le decreased	ecreased			will cause the Earth's atmospheric temperature to increase and cause climate change.			
dioxide in the	dioxide in the Algae and plants leve			e ,	e gradually reduced the carbon dioxide s in the atmosphere by absorbing it for osynthesis.			Carbon xide, water spour and methane Examples of greenhouse gases the temperatures on Earth in order to			
Formation of sedimentary rocks and fossil fuelsThese are made out of the remains of biological matter, formed over millions of yearsof oc sedim and of the remains of sedim of the remains of sedim sedim sedim 		f oceans. Over millions ediment settled on top pressures turned them in and sedimentary rocks.	ains of biological matter falls to the bottom eans. Over millions of years layers of nent settled on top of them and the huge ures turned them into coal, oil, natural gas edimentary rocks. The sedimentary rocks ain carbon dioxide from the biological er.			atmosphere ar this radiation is re (including car	Radiation from the Sun enters the Earth's atmosphere and reflects off of the Earth. Some of this radiation is re-radiated back by the atmosphere (including carbon dioxide, methane and water vapour) to the Earth, warming up the global temperature.				

ORMISTON
SWB
ACADEMY

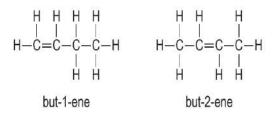


- General formula: C_nH_{2n+2}
 Saturated molecule

Number of carbons in the chain	Prefix	Alkane	Molecular formula	Structural formula
1	meth-	methane	СН	H H-C-H H
2	eth-	ethane	C ₂ H ₆	H H H-C-C-H H H
3	prop-	propane	C ₃ H ₈	H H H H-C-C-C-H H H H
4	but-	butane	C ₄ H ₁₀	H H H H

B the first four members of the alkane homologous series

lsomers



KS4 – Science – Hydrocarbons				
	Alkene			
ane	H C=C H	e.g. ethene		

- General formula: C_nH_{2n}
- Unsaturated molecule
- C=C functional group

Name	Molecular formula	Structural formula
ethene	C ₂ H ₄	H H C=C H H
propene	C ₃ H ₆	$\begin{array}{c} H & H & H \\ I & I & I \\ H - C - C = C \\ I & I \\ H & H \end{array}$
butene	C ₄ H ₈	$\begin{array}{c cccc} H & H & H & H & H \\ I & I & I & I \\ H - C - C - C = C \\ I & I & I \\ H & H & H \end{array}$

C the first three members of the alkene homologous series

D The numbers in the butene isomer names show the position of the double bond. The 1 shows that the C=C bond starts at the end of the molecule on the first carbon, the 2 shows that it is in the middle. Carbon atoms are numbered starting from the end closest to the double bond.

Keyword	Definition		
Alkane	A hydrocarbon in which all the bonds between the carbon atoms are double bonds.		
Alkene	A hydrocarbon in which there are one or more double bonds between carbon atoms.		
Homologous series	A family of compounds that have the same general formula and similar properties, but have different numbers of carbon atoms.		
Carbon monoxide	A poisonous gas produced from carbon burning without enough oxygen.		
Carbon dioxide	A colourless, odourless gas with the molecular formula CO ₂ .		
Saturated	A molecule that contains only single bonds between the carbon atoms in a chain.		
Unsaturated	A molecule that contains one or more double bonds between carbon atoms in a chain.		
Complete Combustion	Combustion of hydrocarbons with enough oxygen present to convert all the fuel into carbon dioxide and water.		
Incomplete Combustion	When a substance reacts only partially with oxygen, such as when carbon burins in air producing carbon monoxide and soot (unburnt carbon).		
lsomer	Molecules with the same molecular formula but different arrangements of atoms.		
Addition reaction	A reaction in which reactants combine to form one larger product and no other products.		
Functional group	An atom or group of atoms in a molecule that is mainly responsible for the molecule's chemical reactions and properties.		



KS4 – Science – Hydrocarbons

KS4 – Science – Hydrocarbons				
Combustion		Addition Reactions		
Complete Incomplete		Two reactant molecules add to one another to form just one product molecule.		
 Plentiful supply of oxygen Products: Carbon dioxide Water Poor supply of oxygen Products: Carbon monoxide Carbon (soot) 		A reaction in which reactants combine to form one larger product molecule and no other products.		
Blue flame on Bunsen burner	Yellow flame on Bunsen burner	Example:		
• Example:		ethene + bromine — 1,2-dibromoethane		
$CH_4(g) + 2O_2(g) \rightarrow CO_2(g) + 2H_2O(g)$		$\begin{array}{ccccc} H & H & H & H \\ $		
Bromine	Water Test			
alkane + bromine water The product retains the colour from the bromine solution.	Bromine water is a dilute solution of bromine in water, Br ₂ (aq) _. It has an orange-brown colour.	D Ethene reacts with bromine to form 1,2-dibromoethane.		
	 Alkenes – react with Br₂ Alkanes – do not react with Br₂ 	•		
BROMINE WATER	The C=C double bond reacts with the bromine to form a colourless product. The bromine is therefore removed from the solution, which loses its colour.			

alkene + bromine water The product is colourless.

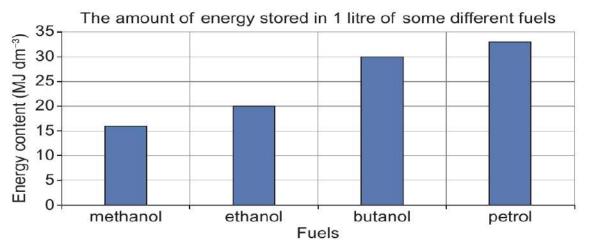
		34 – Science – Alcoho	ols and Carboxylic Acids	Keyword	Definition
ACADEMY				Alcohol	An homologous series of compounds that contain the –OH functional group.
General formula: C _n H _{2n+1} OH • -OH functional group			÷ .	Fractional distillation	In fractional distillation a mixture of several substances, such as crude oil, is distilled and the evaporated components are collected as they condense at different temperatures.
Name methanol	СН,ОН	Structural formula H H-C-O-H	 Combustion of alcohols: alcohol + oxygen → carbon dioxide + water Oxidised to form carboxylic acids React with reactive metals to form 	Fraction Organic compound	In fractional distillation, such as that of crude oil, the different parts of the original mixture are called fractions. The substances in each fraction have similar boiling points to each other. Chemical compounds that contain carbon. Atoms such as
ethanol	C ₂ H ₅ OH	H H H H-C-C-O-H H H	hydrogen gas	Distillate	hydrogen, oxygen, nitrogen or chlorine are also common in organic compounds. A distillate will contain the compound that boils at the lowest
propanol	C ₃ H ₇ OH	Н Н Н Н-С-С-С-О-Н Н Н Н		Renewable	temperature Energy sources that are replenished and not exhausted, eg solar power.
butanol	C [*] H [*] OH	H H H H HC-C-C-O-H H H H H		Homologous series	A family of compounds that have the same general formula and similar properties, but have different numbers of carbon atoms.
	B The naming of compounds uses a set of rules produced by the International Union of Pure and Applied Chemistry (IUPAC).			Carboxylic acid	An homologous series of compounds that contain the –COOH functional group.
				Oxidation	Oxidation occurs when an atom, molecule, or ion loses one or more electrons in a chemical reaction
	Carboxylic Acids			Oxidising agent	A substance that can oxidise other substances in chemical reactions.
Name	Molecular formula	Structural formula	The carboxylic acids have similar chemical properties. They all:	Fermentation	Anaerobic respiration occurring in microorganisms.
methanoic ac	tid HCOOH	О Н-С	 form solutions with a pH less than 7 (if soluble) react with metals to form a salt and hydrogen 	Functional group	An atom or group of atoms in a molecule that is mainly responsible for the molecule's chemical reactions and properties.
		0-н	 react with bases to form a salt and water react with carbonates to form a salt, water and carbon dioxide. 	Sugar	Basic unit of carbohydrates.
ethanoic acid	сн,соон	H O H C C	oxygen An oxygen molecule collides with the functional group of an ethanol molecule and two	Carbohydrates	Food belonging to the food group consisting of sugars, starch and cellulose.
propanoic ac	id C ₂ H ₅ COOH	н о-н	hydrogen atoms are removed.	Respiration	Process in living organisms involving the transfer of energy, typically with the intake of oxygen and the release of carbon dioxide from the oxidation of glucose.
propundie de	2,1,500,11	H-C-C-C	has lost the hydrogens forms a double bond	Anaerobic	Without oxygen.
butanoic acid	с,н,соон	нно-н	ethanol with an oxygen atom.		A protein which catalyses or speeds up a chemical reaction.
		н−¢−¢−¢−с′́ н н н о−н	The hydrogen atoms from the ethanol combine with an oxygen atom to form a		OOH \leftrightarrow CH ₃ COO ⁻ + H ⁺ + metal → salt + hydrogen 105
C the first four o	arboxvlic acids		water molecule. water ethanoic acid A oxidising ethanol		+ here \rightarrow salt + water



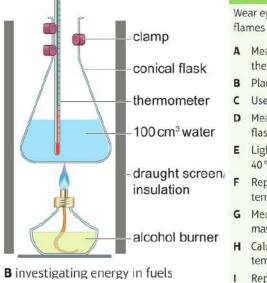
KS4 - Science - Alcohols and Carboxylic Acids

Ethanol Production (C_2H_5OH)

Combustion of alcohols



D Butanol has some advantages over some other alcohols used as fuels.



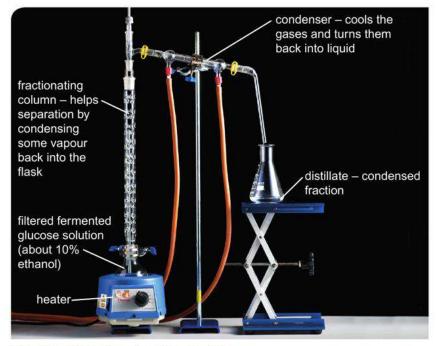
Method Wear eye protection. Do not refill the alcohol burner if there are any naked flames nearby.

- A Measure the mass of an alcohol burner and cap. Record the mass and the name of the alcohol.
- B Place the alcohol burner in the centre of a heat-resistant mat.
- C Use a measuring cylinder to add 100 cm³ of cold water to a conical flask.
- **D** Measure and record the initial temperature of the water and clamp the flask above the alcohol burner.
- E Light the wick of the burner and allow the water to heat up by about 40 °C.
- **F** Replace the cap on the burner and measure and record the final temperature of the water.
- **G** Measure the mass of the alcohol burner and cap again and record the mass.
- **H** Calculate the mass of the alcohol burned to produce a 1 °C rise in temperature.
- Repeat steps A to H using fresh, cold water and a different alcohol

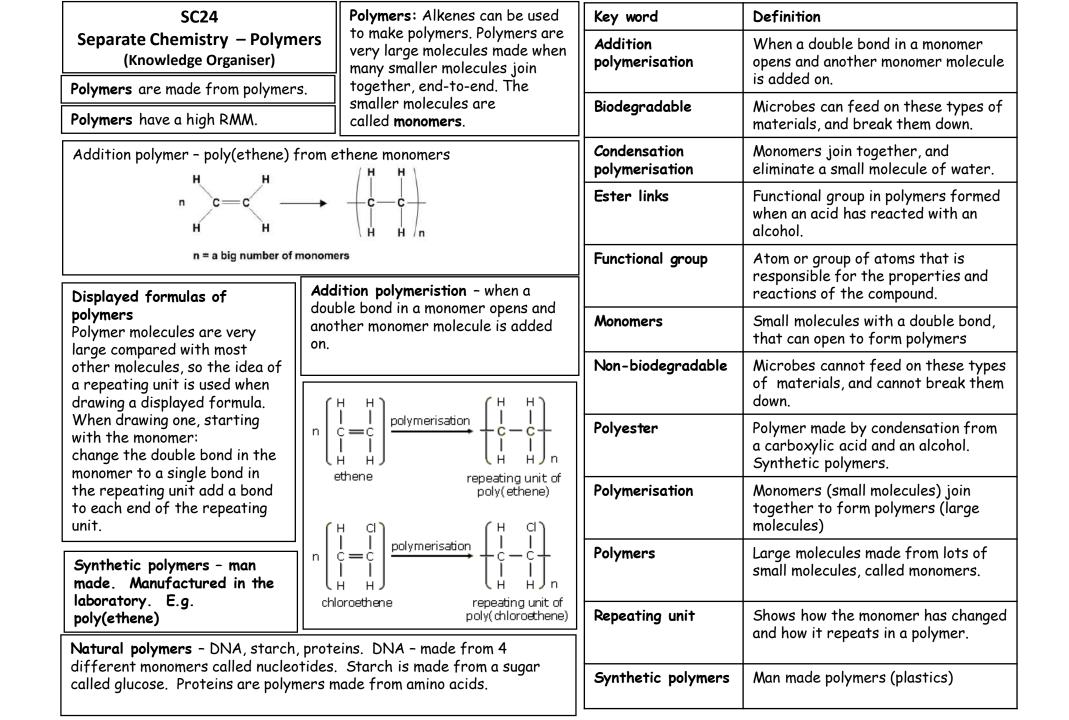
- Sugars mixed with water and yeast
- Fermentation enzymes in the yeast turn the sugars into ethanol and carbon dioxide
- Temperature and pH must be carefully controlled
- Yeast undergo anaerobic respiration (occurs in absence of oxygen)
 Example:

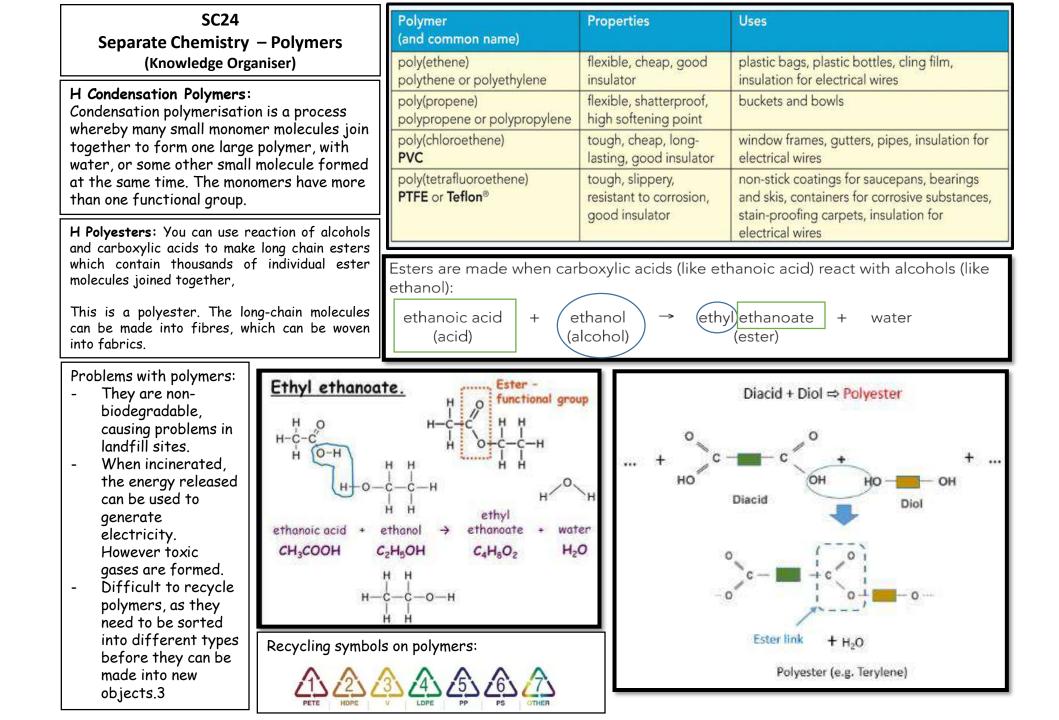
glucose \rightarrow ethanol + carbon dioxide

- Fermentation produces alcohol concentrations <15%
- Fractional distillation used to form more concentrated alcohol solutions
- Yeast undergo anaerobic respiration (occurs in absence of oxygen)



D fractional distillation of an ethanol solution





	S	C25	Key wor	rd	Definition	
Separate Cho	emistry – Qualitat	ive Analysis (Knowledge Organiser)	Anions		A negatively charged ion formed from an atom that has gained electrons.	
Qualitative analy		emical analysis type of substance present in a sample.	Cations		A positively charged ion formed from an atom that has lost electrons.	
		e amount of substance present in a	Confirm	atory test	A chemical test carried out to check the conclusion from the results of another test.	
lonic Compo	ounds	Flame tests	Flame p	hotometer	A machine used to identify metal ions in solution and to determine their concentration.	
are made up c and anions. Cat positively charg	tions are	3. Dip the wire salt to be tested	Halides		A compound formed between a halogen and another element such as a metal or hydrogen.	
formed by the le electrons. Anio	oss of	in concentrated HCl 4. Hold sample in the edge of a hot flame	Halide	ions	A negatively-charged ion formed from one of the Group 7 elements.	
negatively charg formed by the g			Standar	rd solutions	A solution containing a known substance.	
electrons.	8		Precipit	ates	An insoluble substance that is formed when two soluble substances react together in solution.	
Element	Colour flames	To carry out a flame test on an ionic substance:		An	Analyses the concentration of ions	
Lithium Red		Clean a metal loon in dilute		instrumental method used to	I compares the results to known	
Sodium	Yellow	• Dip it into the sample solution	metry	analyse metal	data in order to identify the metal	

• Hold the loop at the edge of a

• Observe the colour of the flame

and use this to determine which

Bunsen burner flame.

metal ion is present.

or solid.

Potassium

Calcium

Copper

Lilac

Orange-red

Blue-green

used to identify metal ion and to determine ntration. formed between a another element such as ydrogen. -charged ion formed the Group 7 elements. ontaining a known substance that is n two soluble substances her in solution. the concentration of ions e solution using a on curve. The user s the results to known data in order to identify the metal ions and their concentration.

Using scientific instruments may improve: sensitivity (detect much smaller amounts), accuracy (give values closer to the true value and speed.

ions

SC25 Separate Chemistry – Qualitative Analysis (Knowledge Organiser)				Sodium hydroxideIs added to solutions to identify metal is These are precipitation reactions.				
		Vill turn blue in the presence f ammonia.		White precipitates	Aluminium, calcium and magnesium ions for this with sodium hydroxide solution.			
ammonia	ammonia paper Can als			Can also be identified by its characteristic smell.		Copper (II) = blue-green Iron (II) = green Iron (III) = brown		
red litmus paper turns blue			Carbonates		React with dilute acids to form carbon dioxide. Halide ions can be identifi silver nitrate solution acid			
ammonium chloride sodium hydroxide heat			Halide ions	precipit	When in a solution, they produce precipitates with silver nitrate solution in the presence of nitric acid.			
F Testing for ammonium ions			Sulfate ions	produce barium (When in a solutions they produce a white precipitate with barium chloride solutions in the presence of hydrochloric acid.			
Test carried out usingIons identifiedflame testNa ⁺ , K ⁺ , Ca ²⁺ , Cu ²⁺ dilute acid and lime water CO_3^{2-} dilute hydrochloric acid and barium chloride solution SO_4^{2-} silver nitrate solution and dilute nitric acid CI^- , Br, I ⁻ sodium hydroxide solution AI^{3+} , Ca ²⁺ , Cu ²⁺ ,			to form a sodiun 2Na	For example, sodium hydroxide solution reacts with copper sulfate solution to form a blue precipitate: sodium hydroxide + copper sulfate \rightarrow sodium sulfate + copper hydroxide 2NaOH(aq) + CuSO ₄ (aq) \rightarrow Na ₂ SO ₄ (aq) + Cu(OH) ₂ (s)			n sulfate + copper hydroxide O ₄ (aq) + Cu(OH) ₂ (s)	
sodium hydroxide solution and heat Al ³⁺ , Ca ²⁺ , Cu ²⁺ , Fe ²⁺ , Fe ³⁺ , NH ₄ ⁺ E Some of the tests that a forensic chemist might perform on unknown substances.			These precipitation reactions can also be modelled using balanced ionic equations. For example, aqueous iron(III) ions react with aqueous hydroxide ions to form solid iron(III) hydroxide:) ions react with aqueous	

 $Fe^{3+}(aq) + 3OH^{-}(aq) \rightarrow Fe(OH)_{3}(s)$

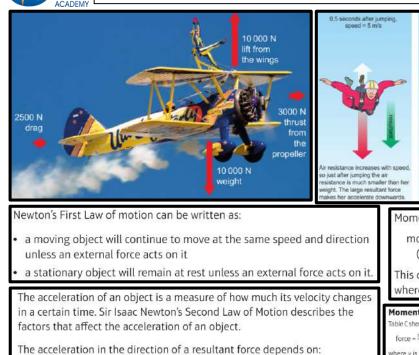
KS4 Bulk and Surface Propertie	s of Matter, including Nanoparticles	Keyword	Definition
Glass and clay ceramics	Brick, porcelain and china are clay ceramics. They	Alloy	An alloy is a mixture of two or more elements, at least one of which is a metal
Ceramics are a range of durable compounds that	are made from clay moulded into the desired shape. When the clay is heated to a very high temperature,	Brittle	If something is brittle it is easily broken
change very little when heated. They are chemically unreactive, hard and stiff but brittle. They are also	tiny crystals form and join together. Bricks are usually decorated by adding a coloured substance to the clay	Composite material	Material made from two or more different materials with contrasting properties
poor electrical and thermal conductors, and have high melting points. Ceramic materials consist of giant	before heating. A pattern may also be moulded into surfaces that will be visible in a finished wall. Porcelain and china are dipped in a 'glaze' and heated strongly	Compressive strength	A measure of how well a material resists being crushed when a force is applied
structures with many strong bonds (covalent or ionic), giving them their typical properties.	again. The glaze forms the hard, waterproof, smooth surface you see on tiles, washbasins and toilet bowls. mers	Density	A measure of compactness and the ratio of mass to volume. It is usually measured in kilograms per metre cubed (kg/m ³) or grams per centimetre cubed (g/cm ³)
cool and solidify. Glass and clay ceramics have similar	ners are substances with high average relative formula masses. They are from monomers – smaller molecules that join together to form repeating For example, poly(ethene) is made from ethene, and poly(chloroethene)	Ductile	A ductile material is capable of being drawn into thin sheets or wires without breaking
However, the atoms in glass are not arranged in a	PVC, is made from chloroethene (see <i>SC24 Polymers</i>). alymers can be moulded into complex shapes. The properties of a polymer spend on its structure and chemical composition, but polymers are usually rong and chemically unreactive. They are also poor electrical and thermal	Malleable	Capable of being hammered or pressed into a new shape without being likely to break or return to the original shape
regular way to form crystals, so glass is transparent	ictors. Rigid PVC is useful for underground pipes and window frames. an be made softer by including substances called plasticisers in its facture. Flexible PVC is useful for indoor water pipes and waterproof	Matrix	The substance that binds the reinforcement together in a composite material
A composite material is a mixture of two		Nanoparticles	Tiny particles which are between 1 and 100 nanometres (nm) in size
produce a material with improved proper		Nanoparticulate materials	Useful substances containing nanoparticles
have contrasting properties. The individuation the composite material, and can often be	separated out by physical separation	Polymer	A large molecule formed from many identical smaller molecules known as monomers
wood can be seen in pykrete, and they see		Reinforcement	Fibres or other material that make up the bulk of a composite material
Reinforcement and matrix	Laminates	Resin	Raw plastic, especially when in semi-liquid form
Concrete is made by mixing cement, sand, aggregate (small stones and gravel) and water together. As the concrete sets hard, chemical reactions happen that bond the solid components together. The sand and aggregat	Wood is a natural composite material consisting of cellulose fibres in a matrix of a polymer called lignin. It is stronger along its grain than it is across its grain. Plywood typically consists of	Tensile Strength	The tension a material can withstand without breaking
form the reinforcement of the concrete. The reinforcement is bonded together by cement, which forms the matrix .	odd numbers of thin sheets of wood, each glued at right angles to the sheet below (see diagram E).	Tension	Pulling force exerted by each end of a n 1 object such as a string or rope

Physics Knowledge Organisers

		Motion		Speed	Scalar measurement that shows how fast an object is moving. Measure in m/s
Scalar	Vector	Calculating speed/velocity			(meters per second).
Distance	Displacement	$\mathbf{S}_{\mathbf{r}} = \mathbf{s}_{\mathbf{r}} $	\wedge	Velocity	Vector measurement that shows how
Speed	Velocity	Speed (m/s) = distance (m) ÷ time (s) How to remember the equation?	$D = S \times T$ $S = D \div T$		fast an object is moving in a specific direction. Measured in m/s (meters per
Power	Momentum	"Don't Step on Turtles"		Distance	second). Measurement of how far an object is
Mass	Acceleration		$\mathbf{S} \mathbf{T} \mathbf{T} = \mathbf{D} \div \mathbf{S}$	Distance	moving/has moved. Measured in m (meters).
Volume Temperature	Weight	Distance-time graph		Time	Measurement of time. Measured in s (seconds).
Force		Key features:		Acceleration	When an objects speed increases over time.
Pressure Calculating a gradie	ent	fast, aetting	You can calculate speed from this distance-time	Conversion	Changing a measurement to another form.
Gradient - ^{Change} in y	YA Change in Y	speed. faster	graph. Steeper gradient=	Deceleration	When an objects speed decreases over time.
$\frac{Gradient}{Change in x}$	Change in X	Steady stationary speed	faster speed.	Scalar	A measurement that shows magnitude only.
Calculating acceler	x	returning to start		Vector	A measurement that shows magnitude and direction.
<u></u>			-	Plateau	A straight horizontal line on a graph.
Acceleration is the <u>rat</u>	<u>te of change of</u>	TIME Velocity-time graph		Gradient	Difference between two values, shown by a incline or decline on a line graph.
Acceleration (m/s/s) =	Change in velocity (m/s)	Key features:	You can calculate acceleration	Constant	When something does not change.
		Constant	from this velocity-time	Magnitude	Shown by a straight line on a line graph. Another term used for size.
	$a = \frac{v - u}{t}$	speed/velocity	, graph.	Direction	The course which an object is moving.
a= acceleration v= final velocity u= initial velocity		Acceleration Deceleration	Calculating the area beneath the lines , is the same as the		We show North, West, East, South or a combination of two.
t= time	v-u		overall distance travelled	Initial	The beginning.
				Final	The end.
	a t	0 10 20 30 40 50	Steeper gradient= faster acceleration.	Displacement	A vector measurement to show the shortest distance to the final place 4/13
		Time/s			object ends up.

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KS4 Physics- Forces and Motion



- the size of the force (for the same mass, the bigger the force the bigger the acceleration)
- the mass of the object (for the same force, the more massive the object the smaller the acceleration).

Newton's Third Law is about the forces on two different objects when they interact with each other. This interaction can happen:

- when objects touch, such as when you sit on a chair
- at a distance, such as the gravitational attraction between the Earth and the Moon.

On Earth the **gravitational field strength** has a value of about 10 newtons per kilogram (N/kg). This means that each kilogram is pulled down with a force of 10 N. The gravitational field strength is different on other planets and moons.

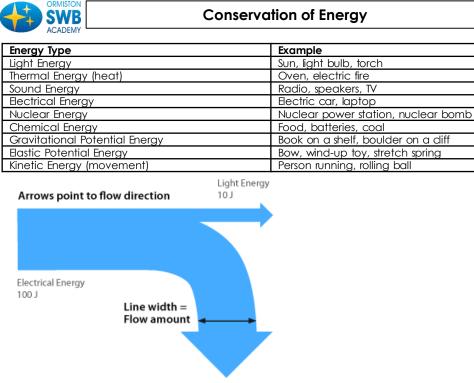
The weight of an object can be calculated using the following equation:

weight = mass × gravitational field strength (N) (kg) (N/kg)

This is often written as: $W = m \times g$

orces	and	Motion		Keyword
econds after jur speed = 5 m/s	mping,	3 seconds after jumping, speed = 25 m/s	12 seconds after jumping, speed = 55 m/s	acceleration
			- 1 -	balanced force
Ĵ.		-	221000	resultant force
				scalar quantity
T	rosultan		resultant	speed
+	•	• • •	•	unbalanced
ince increases er jumping the		Her air resistance is larger but her weight stays the same. The	She is moving so fast that the air resistance balances her weight.	forces
e is much small ne iarge resulta r accelerate do	ler than her ant force	resultant force is smaller, so she is still accelerating, but not as much.	She continues to fail at the same speed.	vector quantity
	Mom	entum is calculated u	using this equation:	velocity
ction	1000	omentum = mass × ve		centripetal for
s on it.	5487925		m/s)	mass
anges	1112 227 232-4	can also be written as e p stands for mome	And a second sec	weight
he		tum and acceleration	ing These can be combined to give	gravitational
		mass × change in velocity or m	1823	field strength
		time the final velocity and u is the star	1	inertial mass
gger	As mass × written as	velocity is the momentum of an o	bject, this equation can also be	action-reaction
as 10	force =	change in momentum or mv - time or t	<u>mu</u>	forces
oject	H Mo	mentum and collisions		balanced force
n they	When mo same bel	oving objects collide the total mor fore the collision as it is after the o nal forces acting. This is known as	collision, as long as there are	
	Rememb you add direction	er, momentum is a vector so you n the quantities together. If two obje s, we give the momentum of one egative sign.	need to consider direction when ects are moving in opposite	equilibrium
th and				conservation o
		uu 🔵 🛶 🥊		momentum
ns per		before collision	×	momentum
orce of			-	kinetic energy
noons.				
n:		Jule -		work done
		after collision		acceleration
	co	The total momentum of the tw loured balls will be the same a omentum of the white ball that	s the	

Keyword	Definition
acceleration	A measure of how quickly the velocity of something is changing. It can be positive if the object is speeding up or
	negative if it is slowing down.
balanced forces	When the forces in opposite directions on an object are the same size so that there is a zero resultant force.
resultant force	The total force that results from two or more forces acting upon a single object. It is found by adding together the
	forces, taking into account their directions.
scalar quantity	A quantity that has a magnitude (size) but not a direction. Examples include mass, distance, energy and speed.
speed	How fast something is moving. Often measured in metres per second (m/s), miles per hour (mph) or kilometres per
	hour (km/h).
unbalanced	When the forces in opposite directions on an object do not cancel out, to there is a non-zero resultant force.
forces	
vector quantity	A quantity that has both a size and a direction. Examples include force, velocity, displacement, momentum and
	acceleration.
velocity	The speed of an object in a particular direction. Usually measured in metres per second (m/s).
centripetal force	A force that causes objects to follow a circular path. The force acts towards the centre of the circle.
mass	A measure of the amount of material there is in an object. The units are kilograms (kg).
weight	The force pulling an object downwards. It depends upon the mass of the object and the gravitational field strength. The
	units are newtons (N).
gravitational	A measure of how strong the force of gravity is somewhere. It is the force on a 1 kilogram mass, so the units are
field strength	newtons per kilogram (N/kg).
inertial mass	The mass of an object found from the ratio of force divided by acceleration. The value is the same as the mass
	calculated from the weight of an object and gravitational field strength.
action-reaction	Pairs of forces on interacting objects. Action-reaction forces are always the same size, in opposite directions, and acting
forces	on different objects. They are not the same as balanced forces.
balanced forces	Forces acting on the same object. Balanced forces are always equal, in opposite directions, and always act on the same
	object. They do not have to be the same type of force An object acted on by balanced forces will not change the way it
	is moving
equilibrium	When a situation is not changing because all the things affecting it balance out.
conservation of	The total momentum of moving objects before a collision is the same as the total momentum afterwards, as long as no
momentum	external forces are acting.
momentum	The mass of an object multiplied by its velocity. Momentum is a vector quantity, with units kilogram metres per second
momentum	(kg m/s).
kinetic energy	A name used to describe energy when it is stored in moving things. The amount of energy stored depends on the mass
	of the object and on its speed (or velocity) squared.
work done	The energy transferred when a force acts through a distance to move an object or change its speed. It is calculated
	using the size of the force and the distance moved in the direction of the force. The unit for work done is the joule (J).
acceleration	A measure of how quickly the velocity of something is changing. It can be positive if the object is speeding up or 114
	negative if it is slowing down.



Heat Energy 90 J

Energy Efficiency = Useful energy/total energy input

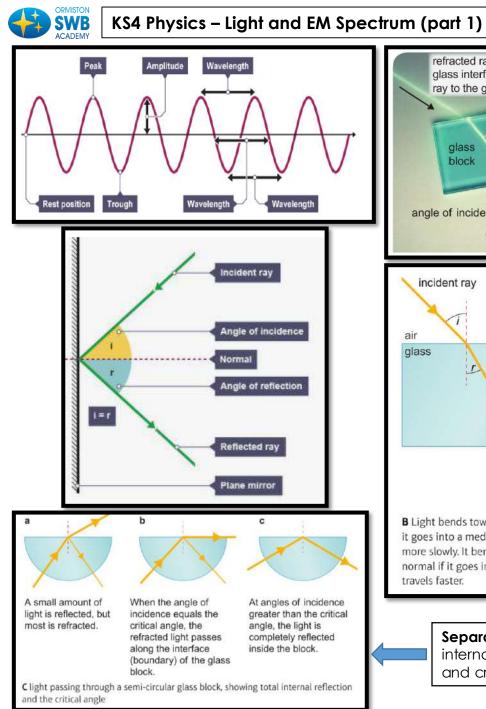
heating	Put more jumpers on and turn off central heating
Hot water	Take showers, only boil the amount of water you need
Electrical appliances	Turn off devices that are on standby
Washing clothes	Air dry clothes, wash on a lower temperature
Heat lost from home	Install insulation – double glazing, loft/floor insulation
	·

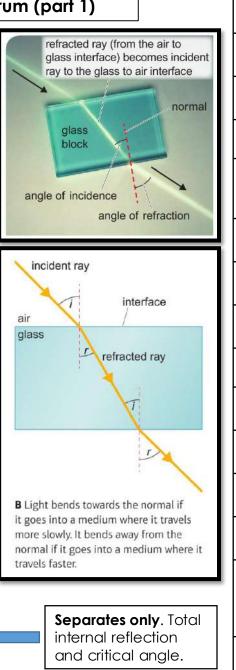
Renewable Energy	Quickly replenishes its energy used. Infinite	Wind power, solar power, hydroelectric power, tidal power, geothermal power, biomass
Non-renewable Energy	ls finite (will run out). Does not quickly replace energy used	Fossil fuels – coal, oil and natural gas Nuclear power

Energy Source	Advantages	Disadvantages
Fossil Fuels	Cheap to set up, power stations already present	Limited (will run out), causes pollution – greenhouse gases and gases that make acid rain, running costs
Nuclear power	Does not produce carbon dioxide or sulphur dioxide	Finite (will run out) danger from radioactive material
Wind power	Infinite, cheap to run, no pollution, cheap to run	Costly to build, only works when windy, noisy and ugly
Tidal power	Good for islands, potential to generate lots of energy, reliable – tide will always go in and out, doesn't release pollution	Costs a lot to build, hard to find suitable locations, could damage environment
Solar power	Infinite, building can have their own power supply, doesn't release pollution, cheap to run	Expensive to set up, only works when sunny
Geothermal power	Doesn't create any pollution, potentially infinite	Expensive to set up, only works in volcanic areas, volcanic activity may stop making station useless
Hydroelectric power	Doesn't create pollution, creates water reserves	Costly to build, can cause flooding, can have major ecological impacts
Biomass	Cheap, if replaced can be sustainable	Burning releases atmospheric pollution, replanting
	30310110010	required

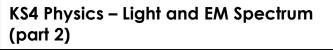
Keyword	Definition		
Chemical	Energy store that is emptied during chemical reactions when energy is		
	transferred to the surroundings.		
Conduction The transfer of heat by passing on energy (or electrical charge) to near			
	particles.		
Convection	The process by which heat travels through fluids (gases and liquids).		
Eastic potential	An energy store that is filled when a material is stretched or compressed.		
Bectrical	Energy store resulting from the movement of electrical charge (electrons).		
Energy	This is the ability to make something happen when it is transferred.		
Gravitational potential	Energy store that is filled when an object is raised.		
Joule	Unit of energy, represented by the symbol J.		
Kinetic	An energy store filled when a moving object speeds up.		
Light	A form of radiation that can transfer energy in a wave.		
Non-renewable	An energy resource that will be used up, and not replenished in our lifetime.		
Nuclear	An energy store associated with nuclear interactions.		
Radiation	Radiation is the transfer of internal energy in the form of electromagnetic		
	waves. This radiation lies in the infrared region of the electromagnetic		
	spectrum. It does not require particles to move, it can travel through a vacuum.		
Renewable	An energy resource that can be readily replenished in our lifetime.		
Sound	A form of energy transferred by sound waves. 115		
Thermal	An energy store that is filled when an object is heated.		
Transformation	Energy transformation is the process of changing one form of energy to		
	another.		

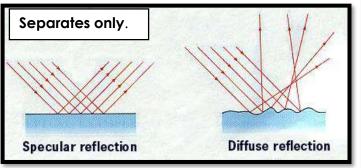
	KS4 Physics – Wave	es	Frequency	The number of waves	Higher frequency =	Increasing frequency	Longitudinal Wave	
Wave	Vibrations that transfer energy from	place to place.		pass a point in a second.	more energy transferred	= higher pitch	Wardength Amplitude	oreal wavelength
Transverse	A wave where the vibrations are at direction in which the wave is trave		Amplitude	Maximum disturbance	Increasing amplitude	Increasing amplitude		S / In coulibrum
Longitudinal	A wave where the vibrations are po direction in which the wave is trave			from its undisturbed position.	increases energy transferred	= increase in volume	Capterment of ar molecules	Trough
Frequency	The number of vibrations (or the nur second, measured in hertz.	mber of waves) per	WavelengthThe distance between a point on one wave and the same point on theIncreasing wavelength = decrease energy transferred				vs. Transverse Wave	
Period	The time taken for one complete w It is measured in seconds.	ave to pass a point.			energy		Longitudinal Particles oscillate (vibrate) in the direction	Transverse Particles oscillate (vibrate) at right angles
Wavelength	The distance between a point on a same point on the next wave.	one wave and the			the		of the wave's movement	to the direction of the wave movement
Amplitude	The size of vibrations or the maximu particle moves away from it resting waves passes.		Transverse Wave			`]	Sound waves, ultrasound Electromagn (light), wat	
Refraction	The change in direction when a wo medium to another.	ave goes from one	undisturbed position (equilibrium) Wavelength			Reflection light	from surfaces. When off waves reflect, they	Diagram
Normal	An imaginary line drawn at right an of a mirror or lens where a ray of lig					bouncing a surface		
Wave Formula Wave speed = waveleng			gth x freque	ncy		Refraction	incidence equals the angle of reflection. n = Waves change	place minur
Example	Wave speed is measured in meters per second (m/s)	Wavelength is mean meters (m)	asured in Frequency is measured in Hertz (Hz)				across the boundary between two	Incident Ray Normal
Dy Ian is standing 1.3m. He counts Wave speed = fi	ing past him. T	he wavelength c	of each wave is		substances with different densities, such as air and glass. This causes them to change	Angle of incident Angle of refraction		
Wave speed = 2 x 1.3							direction and this effect is	116
= 2.6 m/s							called refraction.	





Angle of incidence (i)	The angle between incidence ray and the normal.
Angle of refraction (r)	The angle between normal and the refracted ray.
Separates only - Critical angle	The angle at which total internal reflection happens. In a glass block this is 42°.
Separates only - Diffuse reflection	Reflected light scattered in all directions
Electromagnetic radiation	Form of energy transfer including radio waves, microwaves, infrared, visible light, ultraviolet, x- rays and gamma rays.
Electromagnetic spectrum	The entire frequency range of electromagnetic waves.
Electromagnetic waves	A group of waves that all travel at the same speed in a vacuum, and all are transverse.
Incident ray	The light ray approaching the interface (mirror edge, or edge of a perspex block)
(H) Oscillations	Movement backward and forward
Radiotherapy	Medical technique to kill cancer cells using gamma rays.
Ray diagram	A diagram that models what happens when light is reflected or refracted.
Refracted ray	The light ray that leaves a material like a glass Perspex block. This ray has changed direction.
Separates only - Specular reflection	Light is evenly reflected from smooth surfaces such as a mirror surface.
Separates only – Total internal refection	When a ray of light is shone into a curved glass block and when the angle (critical angle) has been achieved to allow all the light to be completely reflected inside the glass.
Transverse waves	Vibrations are at right angles to the direction 117 which the wave is travelling

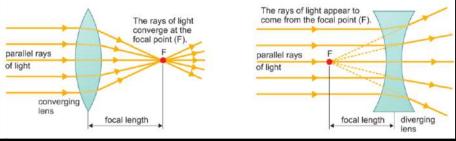




10 ⁻⁶ m	10 ⁻³ m		
	10-11	1 m	10 ³ m
visible light	waves	radio waves	within the
	•	visible light spec spectrum (not to scale)	visible light spectrum are put in

Separates only.

A **converging lens** is fatter in the middle than at the edges. It makes parallel rays of light converge (come together) at the **focal point**. The **focal length** is the distance between the focal point and the centre of the lens. A **diverging lens** is thinner in the middle than at the edges. The focal point is the point from which the rays seem to be coming after passing through the lens.



Separates only.

Heat transfer by radiation - Heat can be transferred by infrared radiation. Unlike conduction and **convection** - which need the vibration or movement of **particles** - **infrared radiation** is a type of electromagnetic radiation. When infrared radiation is absorbed by an object it is heated and its temperature rises.

Separates only.

Dark matt surfaces are better at absorbing heat energy than ight shiny surfaces.

 •Dark matt surfaces are better at radiating heat energy than light shiny surfaces.
 Separates only.

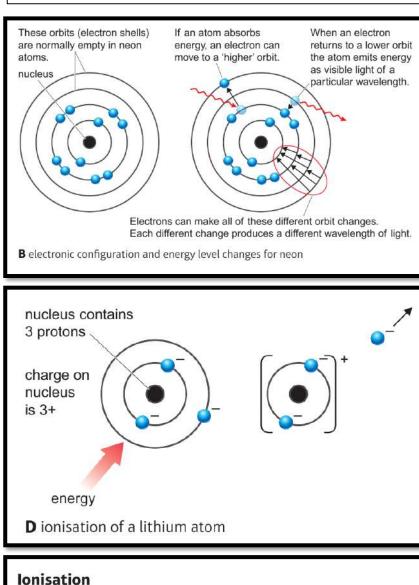
Surface	Absorption	Emission
Dull, matt or rough	radiation	Good emitter of heat radiation
Shiny	Poor absorber of heat radiation	Poor emitter of heat radiation

Long wavelength, low frequency	Uses	Dangers
Visible light	Light bulbs, our eyes detect it	From a laser can damage the retina in the eye
Infrared	Communication – TV remote, grills, toasters	Felt as heat, and can cause skin to burn
Microwaves	Communications, mobile phones, microwave for food	Can cause internal heating of body tissue
Radio waves	Radio broadcast, communications	Very large doses can cause cancer
Short Uses wavelength, high frequency		Dangers
Ultraviolet	Used to kill microorganisms in	Too much exposure can lead to skin cancer

	water, detecting forge bank notes	
X-rays	Hospitals – to check for broken bones	High frequency, transfer a lot of energy and can penetrate the body. Excessive exposure may cause DNA mutation, possibly leading to cancer.
Gamma rays	Cancer treatment, sterilising hospital equipment	High frequency, transfer a lot of energy and can penetrate the body. Excessive exposure may cause DNA mutation, possibly leading to cancer.

CRMISTON SWB ACADEMY KS4 Phy	ysics – Radioactivity	(part 1)		Atomic number (also called proton number)	Number of protons in an atom		
Dalton's model	Plum Pudding - Thomson	Rutherford Bohr/Chadswick		Background radiation	Naturally radioactive substances in the environment that produce radiation.		
Charles -	.979					Becquerels (Bq)	Unit to measure radioactivity. One Bq is one nuclear decay each second.
000				Count rate	Number of clicks on a GM tube when radiation is detected. It is the amount of radiation per second or minute.		
John Dalton thought that all matter was	Thomson carried out experiments and	Rutherford suggested a new model for the atom,	Bohr did calculations that led him to suggest	Elements	Substances that contain the same type of atoms		
made of tiny particles called atoms, which he imagined as tiny spheres	discovered the electron. This led him to suggest the plum pudding model	called the nuclear model. In the nuclear model: the mass of an atom is	that electrons orbit the nucleus in shells. The shells are at certain	Geiger-Muller (GM) tube	An instrument to measure radioactivity.		
that could not be divided.	of the atom. In this model, the atom is a ball	concentrated at its centre, the nucleus	concentrated at its centre, the nucleus	concentrated at its centre,	distances from the nucleus. Chadwick found	Half-life	Time taken for half the unstable nuclei in a sample of a radioactive isotope to decay.
	of positive charge with negative electrons embedded in it.	the nucleus is positively charged	evidence that the nucleus contains no charged particles called the neutron.	Isotopes	Atoms of a single element that have different numbers of neutrons, but same number of protons.		
Atomic Mass	Atomic Mass - The number of particles in the nucleus in the nucleus. - The number of particles in the nucleus - The number of particles in the nucleus. - Electrons orbit			Kinetic theory	Model that helps explains the properties of solids, liquids and gases.		
protons &			protons and neutrons found	Mass number (also called nucleon number)	Total number of protons and neutrons.		
			Electrons orbit	Nucleons	Smaller particles that make up the nucleus.		
just protons in the Number		Proton Pr		Neutrons	Sub-atomic particle found in the nucleus, with no charge.		
of elect	= number rons		nree Isotopes of Hydrogen	Particle theory	Model that helps explains the properties of solids, liquids and gases.		
relative c proton +1	harge relative mass 1	element that have different numbers of		Protons	Positively charged sub-atomic particle found in the nucleus.		
neutron 0 neut electron -1	tral 1 1 / 1840	neutrons, but same number of protons.	· · · · · · · · · · · · · · · · · · ·	Subatomic particles	Particles smaller than atom, and make uppersatom. Protons, neutrons and electrons.		

KS4 Physics – Radioactivity (part 2)



Sometimes an atom gains so much energy that one or more of the electrons can escape from the atom altogether. An atom that has lost or gained electrons is called an **ion**. Radiation that causes electrons to escape is called **ionising radiation**.

Type of nuclear radiation

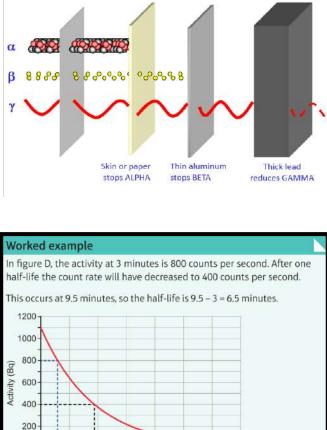
- Treating cancer

Checking thickness of paper

Smoke alarms – contains a source of alpha particles

Property	Alpha			Beta	Gamma
What is it?	Nucleus of a helium atom			electron	EM waves
Charge	+2			-1	None
Mass	Relative	e 4	Rela	tive 0 (1/1840)	None
Range in air	3-5cn	n		15cm	Long range
Penetration ability	Low, stopp pape	-	s	Increased, topped by ninium or lead	Great slowed by concrete, lead
lonising ability	Highly ion	isinig		Fairly	Least ionising
ffects of a magnetic field	Deflected			Deflected	Unaffected
Effects of an electric field	Attracted to negative electrode			Attracted to tive electrode	Unaffected
Particle	Sym	nbol		Dangers of	f radioactivity –
Alpha	α	⁴ ₂H€	õ	can dama	ige the DNA (II. This damage
Beta	B⁻	⁰ ₋₁ e	2	is called m mutation t	utation. Gene hat occur in
Positron	β+	⁰ ₊₁ €	è	on to the r	can be passed next generation.
neutron		n		Some mute cause can	

The penetration power of the three types of radiation.



25

30

[Separates] Radioactivity in medicine

Radioactive materials are used to diagnose medical conditions without having to cut into a patient's body.

15

Time (minutes)

raph of activity against time for a radioactive substance

20

- Tracers that emit positrons used to detect medical problems.
- Treating cancer external radioactivity which uses a beams of gamma rays, x-rays or protons directed the tumour.

KS4 Physics Astronomy	Keyword	Definition	The Solar System and Orbit Dwarf Planet: too small to b			
Life Cycle of Stars:	Solar System	Contains Sun, 8 planets, dwarf planets, comets, asteroids and meteors.	planet Orbit: A path of one object around another.			
Main sequence	Protostar	A very young star that is still accumulating mass from its main molecular cloud.	Satellite: Any object which orbits another. Planetary orbits:	Versue Mars Cature Nesture		
Red giant Red super giant star	Main Sequence Star	A star that merges hydrogen into its core/nucleus and has a stable balance of the external pressure of central nuclear fusion and the gravitational forces that push inward.	 occur because of gravi (acts on an object towe the centre of the more massive object). Weight and gravity: 	not to scale		
White Supernova	Red Giant	A dying star in the later stages of stellar evolution.	 Your weight is a force of gravity acting on you. It depends on your 	 Suggests that the Universe is expanding. If a wave source is moving relative to an observer, there galaxy 		
Black dwarf	White Dwarf	They mark the evolutionary end point of mass stars from low to intermediate like our Sun.	mass and the gravitational field strength (g) of the	will be a change in the observed frequency and wavelength		
Neutron star Black hole Nebula: Cloud of dust and gas from which stars are made. Image: Cloud of dust and gas from which stars are made. Gravity and Thermal Pressure: The two forces that	Black Dwarf	All that remains after a white dwarf star burns all its heat but retains its mass.	earth. (g = 9.81 N/Kg) • Weight (N) = mass (kg) x gfs (N/kg)	 There is an increase in the 400 500 600 700 wavelength of light coming from me galaxies. The further away the galaxy, the more redshift, the faster it is moving away. Creation of Universe Theories Steady State The Universe has always existed, and is expanding As the universe expands, new matter is being constantly 		
	Red Super Giants	A huge giant star that has consumed its core hydrogen reserve. Helium has accumulated in the nucleus and hydrogen is undergoing nuclear fusion in the outer layers.	 Models of Solar System: 1. Geocentric model: The earth is at the center of everything. 2. Heliocentric model: 			
 determine whether a star is stable (balanced), shrinking or growing. Nuclear Fusion: process of nuclei combining that releases energy 	Super Nova	The explosion of a star, the largest explosion that takes place in space.	 Copernicus. The sun at the center of universe. 	created. Evidence: • Red shift because galaxies are moving away from us.		
 in a star in the main sequence Hydrogen fuses to make Helium requires huge pressure from gravity Which path does a star evolves along? 	Neutron Star	Created when giant stars die in supernovae, their nucleus collapses, and protons and electrons fuse together to form neutrons.	Evidence from Galileo using telescopes to observe Jupiter's moons.	 2. The Big Bang Theory – accepted theory as there is more evidence the Universe began as a very tiny point of concentrated energy. The evengetion is still going on 		
 depends on mass stars like our sun become Red Giants more massive stars become 	comet	Balls of ice and dust in elliptical orbits around the Sun.	 3. Elliptical orbit: Most bodies in the solar system are in alliptical orbits 	 The expansion is still going on. Evidence: Cosmic Microwave Background Radiation: Left over radiation from the beginning of the Universe. 494 		
Red Super Giants.	asteroid	Made of rock and metal in orbit around the Sun between Mars and Jupiter.	elliptical orbits.This is the current model.	 Redshift because galaxies are moving away from us due to the continual expansion. 		



KS4 Physics- Energy – Forces Doing Work

			_			
Keyword	Definition	Units of measurement		Keyword	Definition	Units of measurement
Work done	The energy transferred by a force	Joules (J)		Power	The rate at which energy is transferred	Watts (W) or Joules per second (J/s)
600 - CC	s work done force	rection of the force $ \begin{array}{c} F \times d\\ B \end{array} $		This can be write $P = \frac{E}{t}$ where E represent P represent		
	ng a box weighing 200 g floor using a force o by Danny. The force must direction of mo	f 150N. Calculate			Crane A lifts a weight of <u>10</u> in <u>10second</u> Crane B lifts the exact <u>same</u> <u>same distance</u> , but in <u>4 sec</u> We can say that <u>crane B</u> is has done the <u>same amour</u> force a certain distance), b <u>time.</u>	<u>e weight</u> the exact conds . more <u>powerful</u> as it



KS4 Physics - Forces and their Effects

	n be placed into two ntact and there are	Free body force diagrams	A diagram show all the forces ac		
Contact Forces		Non-Contact Forces	Free foi diag	on an isolate object or a sys	
Ai	r Resistance	Gravity			
	Friction	Magnetism	tant ce	Forces acting a	
	Tension	Electrical Force	Resultant force	the same line	
N	ormal Force	Nuclear Force	<u>۳</u>		
Force	A vector quantity	. A push or a pull on an object.	rams	A diagram wh forces do not a	
Contact forces	Two objects have to touch for the force to act. Interact at zero distance.	Caused by objects interacting. E.G. Friction, man pushing a wall, a book on a table, Upthrust of water on a boat.	Vector diagrams	the same line. scale diagram find the result force	
Non- contact forces	Two objects do not have to touch for the force to act. Can interact at a distance.	E.G. Magnetic forces, electrostatic forces, gravitational forces. Everyday ex		s: A force or a s t to rotate. y examples of f nclude door ha	

When two children are on a see-saw the see-saw may be balanced and the children will not move. In this case the clockwise moment is balanced by the anti-clockwise moment – so the two moments are equal.

As both the clockwise moment and anti-clockwise moment are balanced:

 $F_{c} x_{c} = F_{a} x_{a}$

Where the subscript denotes the direction (clockwise or anticlockwise).

torce diagrams	A diagram showing all the forces acting on an isolated object or a system	The size and direction of the pairs of forces acting upon an object or system.
force	Forces acting along the same line	Add together the forces acting in the same direction. Subtract the forces acting in opposite directions.
Vector diagrams	A diagram where forces do not act in the same line. Use scale diagrams to find the resultant force	Draw all the forces acting upon an object. Make sure they are to scale and in the right directions. Draw a joining line rom the start of the first force and the end of the last force.

Moments: A force or a system of forces may cause an object to rotate.

Everyday examples of force causing a rotation motion include door handles, steering wheels and see-saws.

The turning effect of a force is called the moment of the force. The size of the moment is determined by the equation:

moment of a force (Nm) = force $(N) \times$ distance (m)

 $M = F \times x$

The distance, x, is normal to the direction of the force

Levers are used to increase the force applied to an object, usually to lift it up from a surface. Levers must have a pivot to rotate around and will work on the principle of moments.



A gear is a wheel that has teeth on it (also known as a cog), as shown in the diagram opposite. For gears to do work you need at least two gears. Gears are used to transmit rotational forces from one place to another



When two cogs are in contact with their teeth interlocking, the driven cog will rotate in the opposite direction to the drive cog. If the drive cog in a gear spins clockwise then the driven cog will spin anti-clockwise.

When a large cog is driving a small cog, then the small cog will rotate faster than the large cog. Halving the number of teeth on the small cog will double the speed of the small cog. Going from a large cog to a smaller cog will increase the speed of rotation.



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KS4	Physics-	Electricity
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	Mass	Charge	Location	$ \sim$ $-$	switch (o
Proton	1	+ (positive)	nucleus	-0-0-	switch (c
Neutron	1	no charge	nucleus	-+	cell battery
Electron	1/1835 negligible	- (negative)	shells		diode

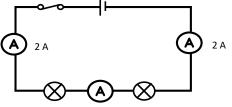
Useful formulas

- energy transferred (J) = charge moved (C) x potential difference (V) $E = Q \times V$
- charge (C) = current (A) \times time (s) Q = 1 t
- potential difference = current × resistance
- energy transferred = current × potential difference × time $E = I \times V \times t$
- electrical power (W)= potential difference (V) x current (A) P = VI
- power = current squared x resistance $P = I^2 R$



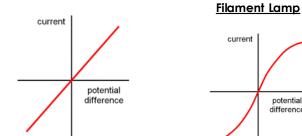
Conductors:

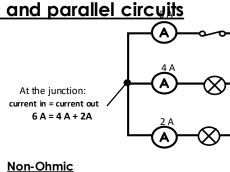
potential difference

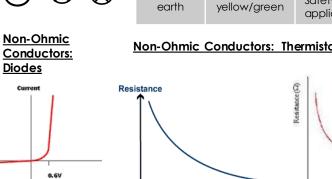




2 A





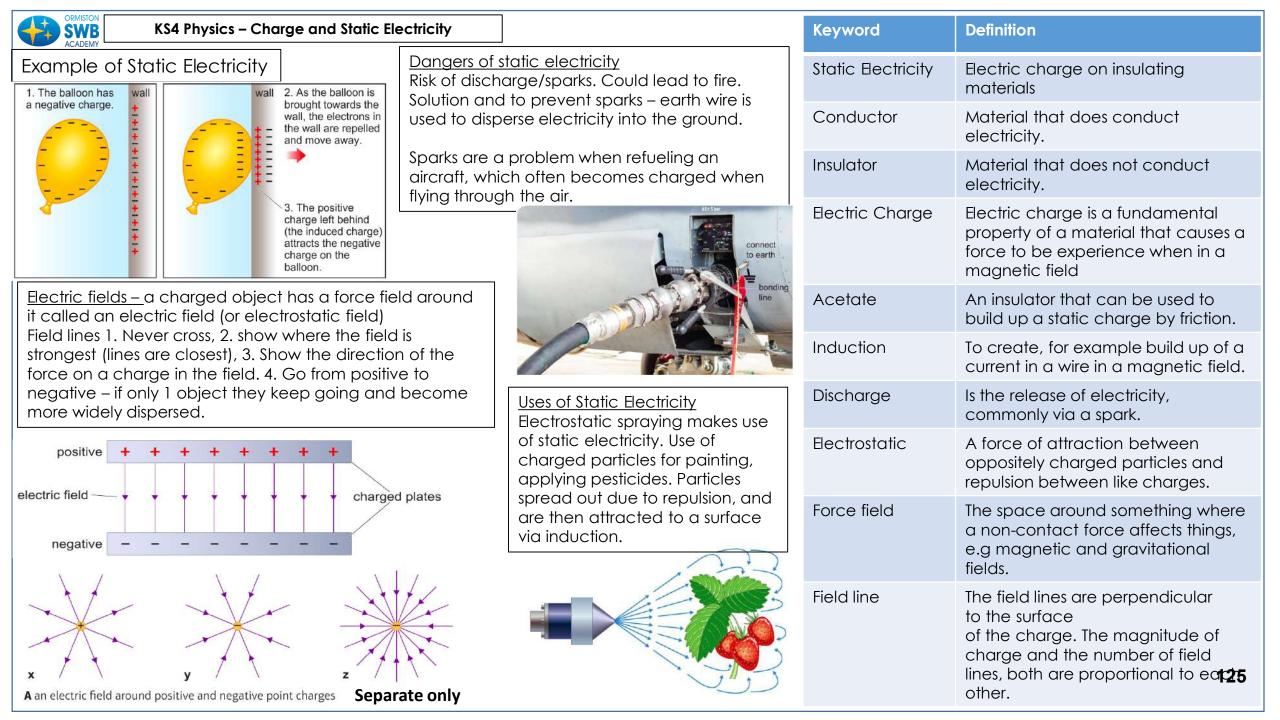


Name

live

neutral

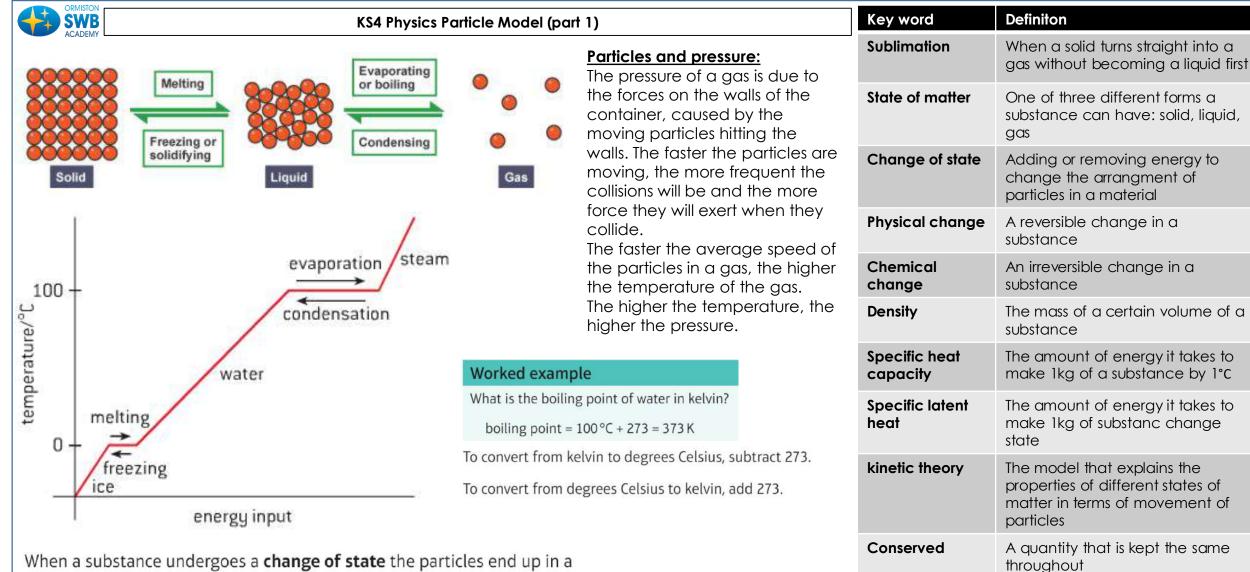
		Keyword	Definition
• •	switch (open)	Current	The rate of flow of charge in a circuit.
1. 	cellv voltmeter pattery(A) ammeter	Potential Difference	Also called voltage. The difference in potential between two points of a circuit. Causes a current to flow.
U	diode	Charge	Charge is the amount of electricity travelling through a circuit.
,	variable resistor LDR	Resistance	Anything that slows the flow of charge around a circuit. Resistance is usually caused by electrons colliding with ions in a material.
		Series Circuit	A circuit with a single loop of wire.
		Parallel Circuit	A circuit with two or more loops (branches) of wire.
Colour brown blue	FunctionCarries alternating potential difference from the supply.Completes the circuit.Safety wire to stop	Variable resisitor	A variable resistor is a component where the resisitance changes as you slide the connector along it. It can be used to investigate how the resisitance of the lamp changes as you change the resisitance.
ductors: Thermistors and LDRs		Ohm's Law	the current through an ohmic conductor (at a constant temperature) is directly proportional to the potential difference across the resistor
		Ohmic Conductors	Ohmic conductors will produce a straight line I – V graph that goes through the origin.
		Non-Ohmic Conductors	The resistance of components such as lamps, diodes, thermistors and LDRs is not constant. It changes with the current through the component.



KS4 Physics – Magnetism and	I the Motor Effect	Keyword	Definition
Maxwell's Right Hand Grip rule A circular magnetic field forms around a current carrying wire. If you point your right thumb in the direction of the current (+ to -), the magnetic field goes in the direction your fingers are pointing.		Magnet	An object that has its own magnetic field around it.
		Magnetic field	The area around a magnet where it can affect magnetic materials or induce a current.
direction of magnetic field A current flowing through a wire causes a magnetic		Permanent magnet	A magnet that is always a magnet such as a bar magnet.
	field. Electric motors and other devices	Solenoid	A coil of wire with electricity flowing in it. Also called an electromagnet.
	depend on the magnetic effect of electric currents.	Motor Effect	The force experienced by a wire carrying a current that is placed in a magnetic field.
Fleming's Left Hand Rule (HT) The motor effect describes the force that acts on a current carrying wire in a magnetic field. upwards force on wire Earth's magnetic field		Magnetic flux density	A way of describing the strength of a magnetic field. Measured in Teslas (T).
		Split ring commutator (HT)	This reverses the direction of the current in the coil each half turn. This allows the motor coil to rotate continuously in one direction.
direction of current	A plotting compass can also show the Earth's magnetic Tield.	core North Pole	Magnetic field diagrams (HT) Magnetic field flows from north to south. Also around a current carrying wire.
force F magnetic field	The behaviour of compasses is evidence that the Earth has a magnetic field, which is similar in shape to the magnetic field of a par magnet.	Pole	S N Image: S N Image: S Image: S

KS4 Physics Electro	omagnetic Induction	Keyword
Inducing a potential difference A potential difference can be induced (created) in a co	onductor when there is movement between the	A.C
conductor and a magnetic field. This can occur in two d a coil of wire is moved in a magnetic field a magnet is moved into a coil of wire	lifferent ways:	Alternator
This is called electromagnetic induction and is often refer	rred to as the generator effect .	Conductor
The direction of the induced potential difference or indu movement. The current is reversed when: The magnet is moved out of the coil The other pole of the magnet is moved int		Dynamo
An induced potential difference or induced current will in The speed of movement is increased The magnetic field strength is increased The number of turns on the coil is increase	ncrease if:	Electromagnetic Induction
Transformers and how they work	Transformer Calculations	Generator
ac input ac output	The primary coil of a transformer has a current of 0.5A with a potential difference of 100V. The current in the secondary coil is 25A. What is the potential difference across the secondary coil? Use $V_p \times I_p = V_s \times I_s$	Generator Effect
	$100V \times 0.5A = V_s \times 25A$ $50 = V_s \times 25$ $V_s = \frac{50}{25} = 2V$	Magnetic Field
Primary Magnetic field in core 1 A primary voltage drives an alternating	A radio runs off the 230V mains supply but only needs 23V. Its transformer has 100 turns of wire in the primary coil. How many turns are needed in the secondary coil?	National Grid
 current through the primary coil. 2.The primary coil current produces a magnetic field, which changes as the current changes. 3.The iron core increases the strength of the magnetic field. 	$\frac{V_p}{V_s} = \frac{N_p}{N_s}$ $\frac{230 V}{23 V} = \frac{100}{N_p}$ $10 = \frac{100}{N_p}$	Potential Differenc
4. The magnetic field passes through (or cuts) the secondary coil. 5. The changing magnetic field induces a changing	$10 \times N_p = 100$ $N_p = \frac{100}{10} = 10$	Power
potential difference in the secondary coil. 6.The induced potential difference produces an alternating current in the external circuit.	So the secondary coil must have 10 turns.	Transformer

Keyword	Definition
A.C	Alternating current, current that flow in both directions
Alternator	An electrical generator which produces alternating current
Conductor	Electrical conductors are materials that have low resistance and allow current to pass through them easily
Dynamo	An electrical generator which produces direct current
Electromagnetic Induction	The production of potential difference (voltage) when a conductor e.g. wire moves through a magnetic field. If the conductor is part an electric current, an induced current will flow
Generator	Device that converts kinetic energy into electrical energy
Generator Effect	When motion between a conductor and a magnetic field creates electricity, i.e. a magnet is moved into a coil of wire.
Magnetic Field	Area surrounding a magnet that can exert a force on magnetic materials
National Grid	The network that connects all of the power stations in the country to make sure that everywhere has access to electricity.
Potential Difference	The potential difference (or voltage) of a supply is a measure of the energy given to the charge carriers in a circuit. Units = volts (V). This is the voltage between two points that makes an electric current flow between them.
Power	The energy transferred each second, measured in watts (W). Power = work done ÷ time taken.
Transformer	An electrical device that increases, or decreases, 127 potential difference (voltage) of an alternating current.



Joule (J)

Kelvin (K)

Pascals (Pa)

A unit for measuring energy

scale. One kelvin is the same

The units for pressure

The unit in the Kelvin temperature

temperature interval as 1° C 128

When a substance undergoes a **change of state** the particles end up in a different arrangement. There are the same number of particles so the mass stays the same (mass is **conserved**). This is a **physical change**, because no new substances are formed and the substance recovers its original properties if the change is reversed. Mass is also conserved in **chemical changes**, but the change in the substances often cannot be reversed.



KS4 Physics Particle Model (part 2)

$\rho = m / v$

Density $(g/cm^3) = \frac{mass(g)}{volume(cm^3)}$

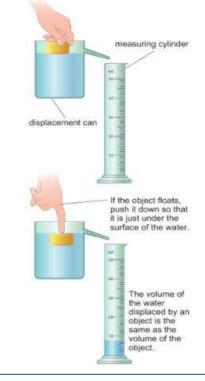
 $\Delta Q = m \times c \times \Delta \theta$

 $Q = m \times L$

thermal energy for a change of state $(J) = mass (kg) \times specific latent heat <math>(J/kg)$

change in thermal energy (J) = mass (kg) × specific heat capacity (J/kg ° C) × change in temperature (° C)

Core Practical: Investigate the densities of solids and liquids



Method Liquids

- A Put an empty beaker on a balance, and set the balance to zero.
- **B** Use a measuring cylinder to measure 50 cm³ of a liquid and then pour it into the beaker. Write down the reading on the balance. This is the mass of 50 cm³ of the liquid.

Solids

C Find the mass of the solid and write it down.

Diagram B shows how to find the volume of an irregular shape:

- **D** Stand a displacement can on the bench with its spout over a bowl. Fill it with water until the water just starts to come out of the spout.
- **E** Hold a measuring cylinder under the spout and carefully drop your object into the can. If your object floats, carefully push it down until all of it is under the water. Your finger should not be in the water.
- F Stand the measuring cylinder on the bench and read the volume of water you have collected. This is the same as the volume of your object. Write it down.

Core Practical: Investigate the properties of water by determining the specific heat capacity of water

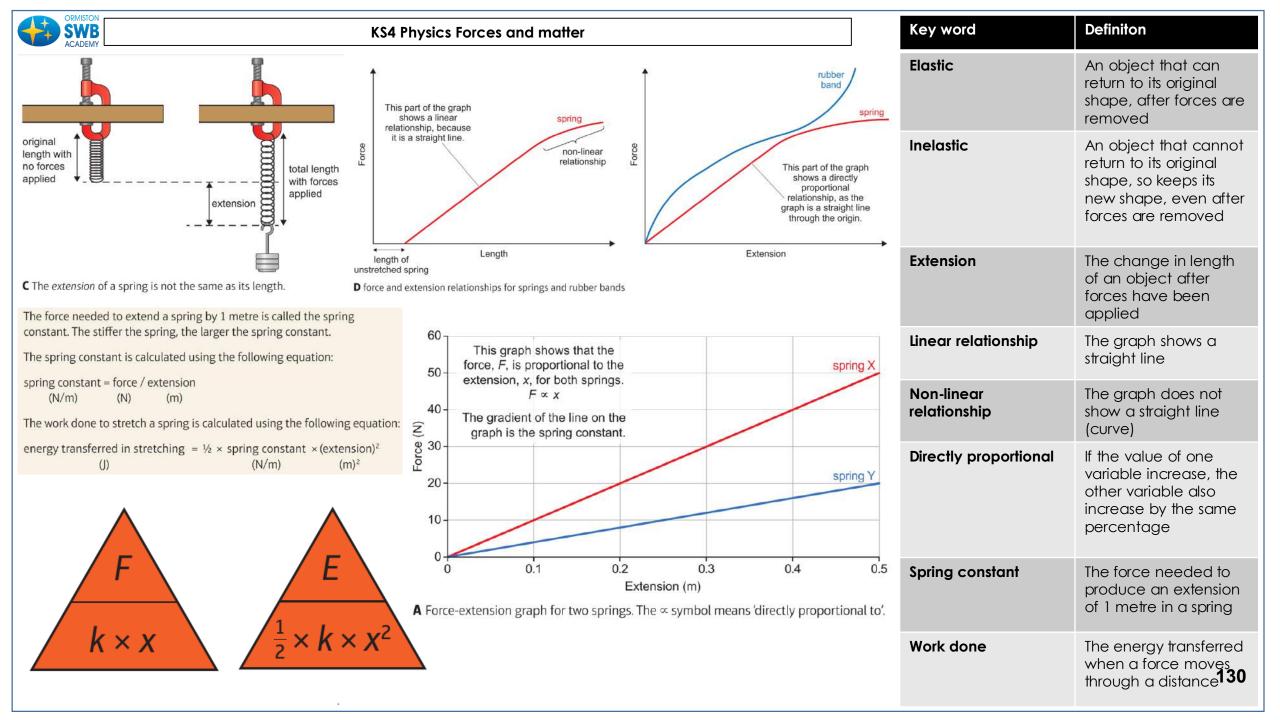
Method Melting ice

Wear eye protection.

- A Put a boiling tube full of crushed ice into a Pyrex [or heatproof] beaker. Put a thermometer in the ice and note the temperature.
- B Put the beaker onto a tripod and gauze. Pour hot water from a kettle into the beaker, and keep it warm using a Bunsen burner.
- C Measure the temperature of the ice every minute and record your results in a table. Stop taking readings three minutes after all the ice has melted.
- D Note the times at which the ice starts to melt and when it appears to be completely melted.

Specific heat capacity

- E Put a polystyrene cup in a beaker onto a battery-powered balance and zero the balance. Then fill the cup almost to the top with water and write down the mass of the water. Carefully remove the cup from the balance.
- F Put a thermometer in the water and support it as shown in photo B. Put a 12V electric immersion heater into the water, making sure the heating element is completely below the water level. Connect the immersion heater to a joulemeter.
- **G** Record the temperature of the water, and then switch the immersion heater on. Stir the water in the cup gently using the thermometer.
- H After five minutes record the temperature of the water again and 29 write down the reading on the joulemeter.





DEVELOP

EXPERIMENT

AO3

RECORD

AO

PRESENT

Anna Allanti Della

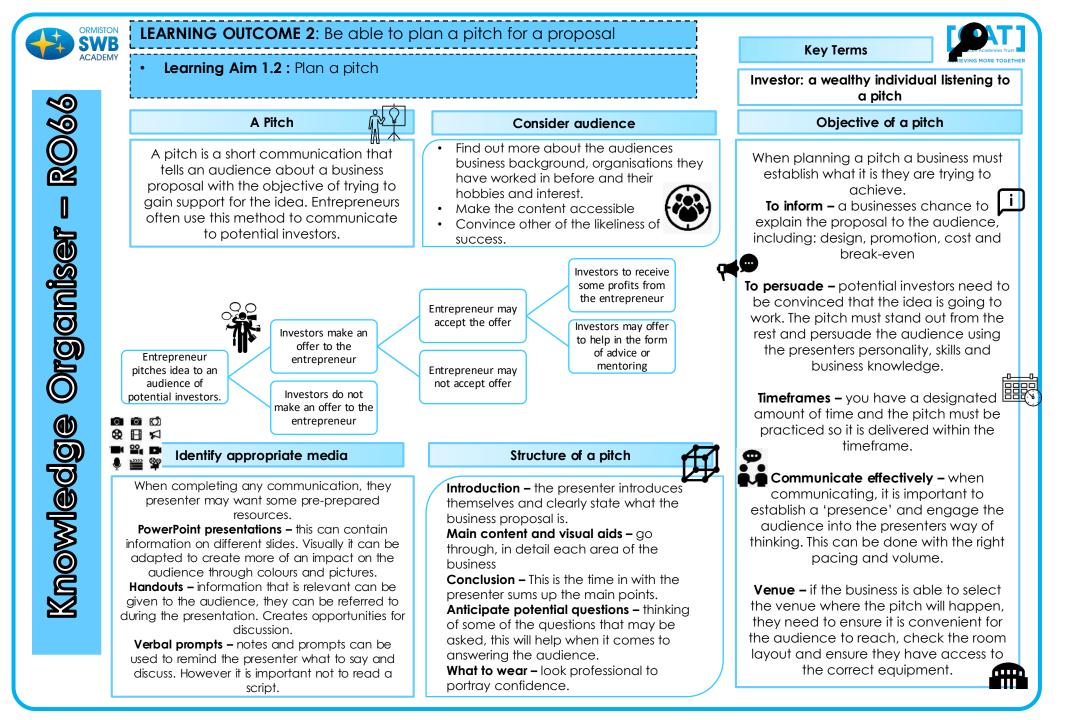
Y11 TEXTILES KNOWLEDGE ORGANISER SWEETS & CAKES

5/LE



Y11 TEXTILES	How do Lidentify the formal elements of my major project; Cakes, Biscuits & Sweets?	Keyword	<u>Definition</u>
SWEETS & CAKES	 Artist's information/Inspiration What specific theme/genre are you going to study. Name a well known artist to take influence from 	Observation	The action or process of closely observing or monitoring something or someone.
DEVELOP ideas through investigations informed by ARTISTS and other sources, showing analytical and cultural UNDERSTANDING	 within your chosen genre. Define Form/shape/pattern/experiment. Apply numerous techniques during development. 	Silk Printing	A design is cut out of paper or another thin, strong material and then printed by rubbing,
REFINE ideas through EXPERIMENTING and	 How do I create a response to chosen Artists work? Use the ideas behind an artists work to inspire your 		rolling, or spraying paint or ink through the cut out areas.
SELECTING appropriate resources, media, techniques and processes	 Watch a demonstration by your teacher. Use decorative/dyeing/printing/experimental 	Fabric Manipulation	Experimenting with the fabric to change its appearance, drape or shape.
RECORD ideas, observation and insights RELEVANT to your INTENTIONS	 Textiles techniques with skill and control. Create a response to your chosen artists work using influence from their work. 	Influence	Something or someone that influences a person or thing, then, has
Present a PERSONAL response, showing analytical understanding and realising INTENTIONS for your project, making connections in your work	What needs to be included to ensure a successful final piece?	millence	an influence on that person or thing.
Manuary Contiles 2020	 Commit to design throughout project. Use shape, scale and proportion accurately. Make your work as detailed as possible using the Textiles techniques explored. 	Moodboard	An arrangement of images, materials, pieces of text, etc. intended to evoke or project a particular style or concept.
Digital Ad	 Take inspiration from your chosen artist and show clear development in response to their work. Create a mock-up of a final product. Create final product signifying the conclusion to the journey you have created throughout your sketchbook. 	Batik	A method (originally used in Java) of producing coloured designs on textiles by dyeing them, having first applied wax to the parts to be left undyed.
Artitle Responding to the work of		Applique	Layering pieces of fabric that are sewn or stuck on to a larger piece to form a picture or pattern.
other artists is a great way to generate ideas Starting with a direct response showing their understanding of the ideas and aesthetic of an artist.		Toile	An early version of a finished garment made up in cheap material so that the design can be tested and perfected.

SWB YEAR 11 OCR GCSE FIN	IE ART. Watch and learn.	Keyword	Definition
ACADEMY Asking Artistic questions? Practice the use of these words by asking the follow	ving Bitesize	Brushwork	Refers to the way paint is applied in a painting, describing texture of the paint surface applied with a brush.
questions about your artwork or the work of others: <u>Composition</u> : How is the space used? Does	Ditesize	Colour wash	A term used to describe the transparent layers of colour in a watercolour.
the image fill the canvas/paper? <u>Light:</u> Highlighted areas? Shadows? Time of day? Natural light or artificial? Harsh or soft?	https://www.bbc.com/bitesize/subjects/z6hs34j https://www.studentartguide.com Pinterest is a great way to collect the work of	Complementary colour	Red and green, yellow and purple, blue and orange. These colours lie opposite each other on the colour wheel.
Reflected or direct? Abstract? Line: Are there any lines that are prominent? Are	individual artists, themes or cultures. These images can motivate you and help formulate the kinds of imagery and art that	Composition	The arrangement of elements within an artwork.
they straight, curvy? Thin or thick? Do the lines create direction? Do they outline? Do the lines show movement or energy?	you enjoy and are inspired by. Task: Create your own Pinterest boards on your favourite artists, cultures and inspirations.	Contrast	The difference in colour found between the light and dark parts of an image.
<u>Repetition</u> : Are there any objects, shapes or lines which repeat and create pattern?		Contextual	Connections made to the work of other artists from different and similar times, places and cultures.
Shape: Do you see geometric or organic shapes? What are they?		Style	The visual appearance of a work of art. Could also link to art movements – for example 'expressive style.'
Space : Positive or negative? Is there a depth to the work?	OINTEREST	Develop	Bring out potential.
Texture: Smooth or rough? Does the work give the impression of texture?		Explore	Try out the qualities of materials, techniques or processes through practical investigation.
<u>Tone</u>: Is there a range of tones from dark to light? Where is the darkest? Where is the lightest?		Imaginative	Develop ideas and concepts in new, engaging and inventive ways.
<u>Colour:</u> Are the coloursprimary? Secondary? Tertiary? Complementary? Harmonious?		Refine	Improve initial work taking into account feedback and aims. 13





R066

Organiser

Knowledge

LEARNING OUTCOME 3 + 4: Be able to pitch and review a proposal to an audience

Learning Aim 3.1+3.2+3.3+3.4+4.1+4.2: feedback, delivering and reviewing a pitch.

Key Terms



Professionalism: consistently displaying appearance and conduct of the highest quality and is associated with the impression that is given to others when working in a business

Future developments

When any kind of project or business has been completed or launched, it is a good idea to think about how the idea can be developed further. Always look to improve or develop your new products (product development)

Reviewing a business proposal OF

Businesses will need to ensure that their staff have the right knowledge and skills when reviewing any new business idea. A business will put together a plan of what information is required.

Product proposal – a description of the product and its unique selling point, etc.

Pricing strategy – the pricing methods that they intent to adopt to try and obtain sales.

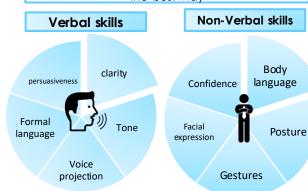
Brand – the brand personality, identity and image which is individual to the pitch.

Promotional plan – the different promotional methods that were created to sell the product.

Relevant and appeal to the customer – research that shows why customers will want to purchase the product.

Use and develop personal presentation skills Personal presentation – it is important to

create the right impression. You want to show your best attributes and demonstrate that you have taken the time to present yourself in the best way



Support peers – feedback

Constructive feedback –giving your opinion and in a polite manner make suggestions. Feedback sandwich – two slices of 'bread' represent two positive comments and then the 'filling' is one constructive articism. Phasing feedback – be sensitive to the persons feelings. Sharing opinion – everyone has their own opinions, listen to everyone and be respectful. Encouragement – encourage others by focusing on the positives of their work.

Review a practice pitch

Practicing a pitch allows you the opportunity to gain feedback and make improvements prior to the professional pitch.

nal presentation

Deliver the pitch.

To deliver a pitch businesses should consider the following: the business idea, customer profile, market research, product design, costing and risks

Reviewing a pitch

Ø

Self assessment

•When you individually assess different aspects of your work, it is a useful method of reviewing but you need to be honest with yourself.

Feedback from others

•Using different types of feedback can help you gain several different views on your pitch

Lessons learned from practice pitch

•Focus on the changes that need to be made and documents them to provide you with a source of review.

Lessons learned from professional pitch

• Verbal feedback from the audience can be recorded so that it can be reviewed later on.

Compare the outcomes with the objectives

A business will often complete a review and will focus on the positive aspects and on things that could be improved. This could be in the form of a meeting and participants will

- be asked to consider:What went well
- What could have been improved
- The format and content of the pitch
 - Visual aids



Anticipation and preparation of responses
 to potential questions

Target Audience:

You need to know your target audience. Who are they? What kind of things do they do? What are their likes and dislikes? What are they interested in? Getting an understanding of these individuals helps you create with ease and make something you know will relate to them.

Planning:

Create a work plan which lists all of the tasks involved in the whole project. Estimate how long each task will take and create a chart or diary to record how long they REALLY take to complete. Build in some contingency time in case things go wrong!

Create a site map to show the pages of the website and how they will be linked together with navigation features.

Create a visualisation diagram to plan the content and layout of the individual web pages.

Test Plans:

There are a range of elements that all need work to produce a successful product. Create a test plan to check these functions:

- Navigation
- Links to take the user to the correct page
- Display of images and content
- Playback of video and audio

Asset Table:

Create an asset table to show the range of audio, video and images you will be using. This will include listing where you got the assets from and describing any legal issues with using them.

Methods of internet connection:

- Wired broadband
- Wi-Fi
- 3G, 4G and 5G wireless broadband

Devices used to access webpages:

- Laptops and personal computers
- Tablets
- Mobile devices and smartphones
- Game consoles and digital television
- Smart Speaker
- Smart Watch

Purposes of websites:

- Education
- Online retail
- Information
- Services
- Advertising
- Promotion
- Entertainment

How does the appearance of websites differ on different devices?

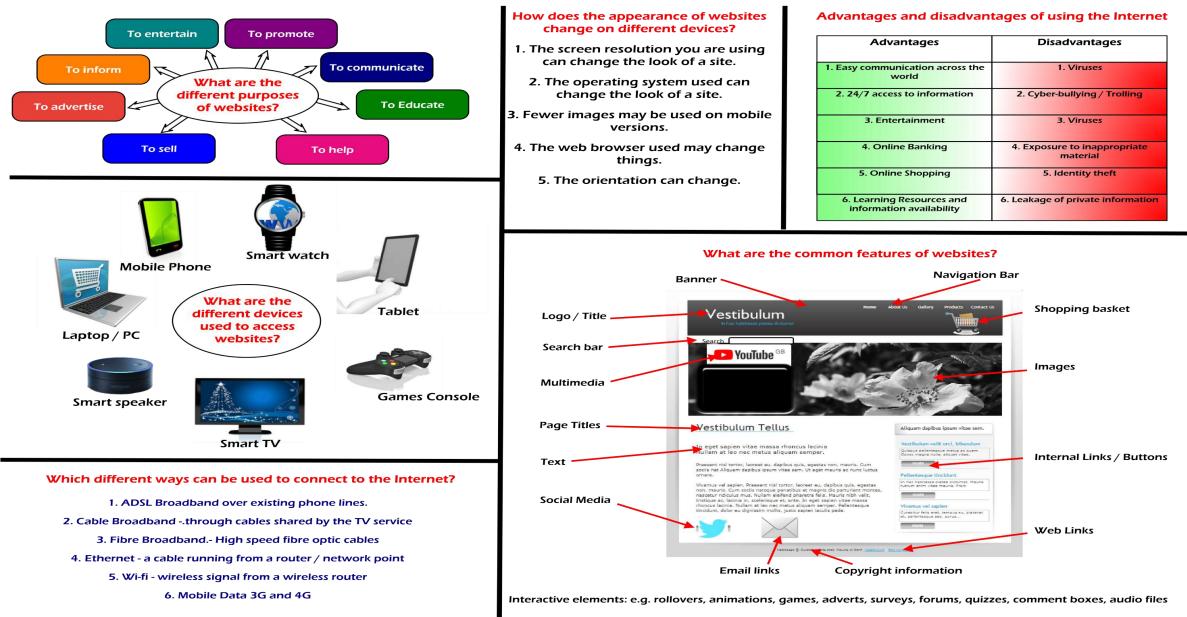
The screen resolution used can change the look of a site Operating system used can change the look of a site Fewer images may be used on mobile versions The web browser may change things The orientation can change

Client Requirements:

Your client is the person you will be working for. They will tell you what to plan, design or create for them. The client will set out requirements that they want you to follow when you plan the project. Features of websites:

House style Navigation features Hyperlinks Search facility Website footer Images/image gallery Ordering forms Downloadable content Logo/Title Page Titles Email links Links to social media Internal links Shopping basket Interactive features

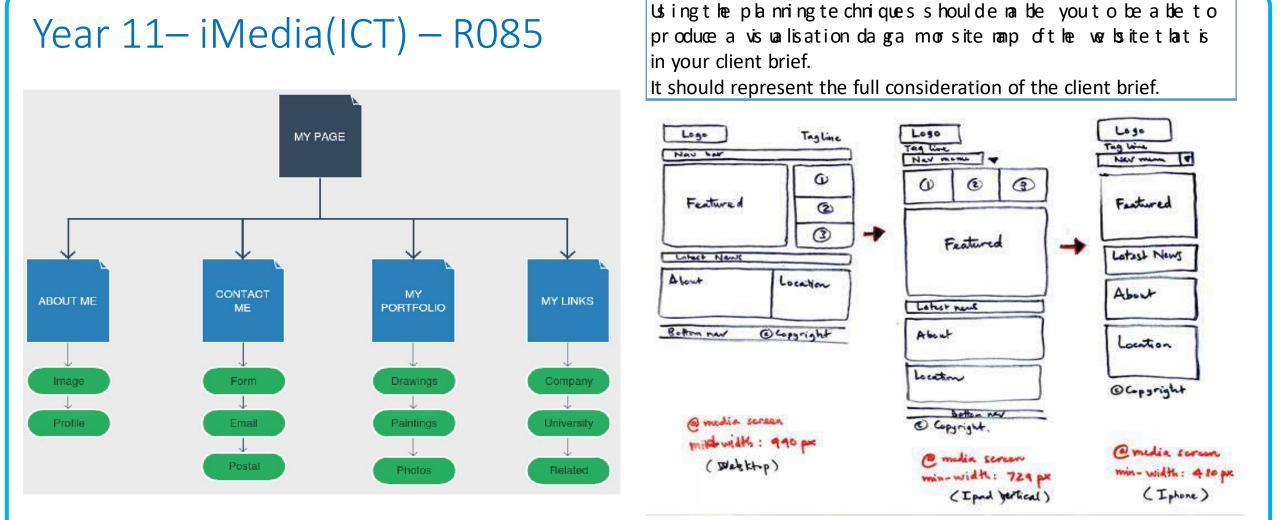
- Rollovers
- Animations
- Adverts
- Surveys
- Forums
- Quizzes
- Comment boxes
- Audio/video files



 Layout Colours Fonts Media Content 	Þ Logos	Quality Þ Testing Þ Fix errors
FontsMedia	Þ Images Þ Logos	Þ Testing
o Media	Þ Logos	Þ Testing
	Þ Logos	J
P Content	b Tevt	b Fix orrors
Duser needs	Þ Media	Þ Obtain feedback
P House Style	P Hyperlinks	Þ Check fit for
P Charts		purpose Þ Improvement
P Equipment	Þ Testing plan	Þ Meets client
		requirements
		Þ Use target
		audience and
		client feedback
		 Charts Equipment Testing plan

	Keywords				
	Word	Meaning/Description			
1	Website	A collection of web pages linked together.			
	Webpage	A document which can be displayed in a web browser such as Firefox, Google Chrome, Microsoft Internet Explorer or Edge, or Apple's Safari.			
	Navigation Bar	A navigation bar is a user interface element within a webpage that contains links to other sections of the website.			
	Rollover Image	Allows you to have two images and when the pointer hovers over one, it changes to another image.			
	Hyperlink	A button, text or image that allows you to move around a website.			
	Hotspots	An area on a computer screen which can be clicked to activate a function, especially an image or piece of text acting as a hyperlink.			

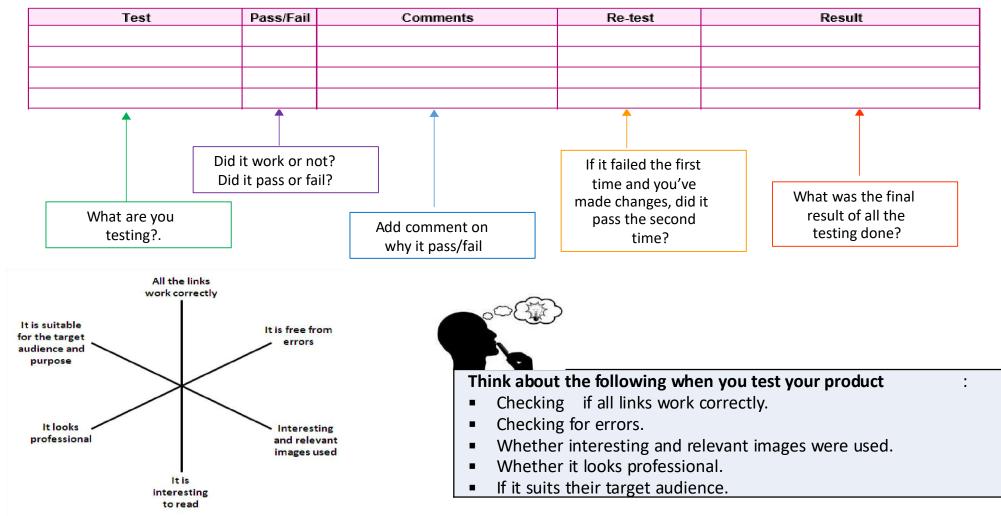
Required Evidence	Examples of evidence	
Written and	Electronic files/evidence	
presentation files	Written report/presentation	
Client requirements	 Written report, presentation, audio 	
	commentary	
Planning Documents	• Work plan, asset table, visualisation diagram	
	and test plan	
Finished product	• An website product.	
Review	• Written report, presentation or recording. 138	



Visualisation diagrams and storyboards are always produced prior to creating the multimedia project. They assist in the development of the project ensuring that all the client's requirements have been fulfilled before the expressive task of creating the media begins. Sitemaps and visualisation diagrams are never edited once the multimedia product has been completed, so some differences are expected to be seen.

What is a test plan and where will I use it?

Example of a test plan



Copyright Protection

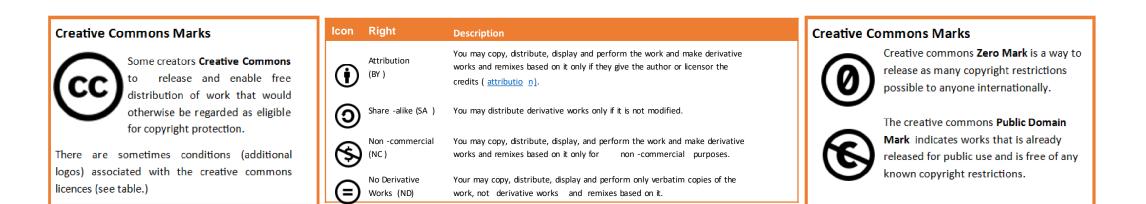
- Copyright protection starts as soon as a work is created .
- Once your copyright has expired, anyone can use or copy your work.
- The length of copyright depends on the type of work .

Type of Work	How long Copyright Lasts
Written, dramatic, musical and artistic work	70 years from when it's first published
Films	70 years after the death of the director, screenplay author and composer
Broadcasts	50 years from when it's first broadcast
Layout of published editions of written, dramatic or musical works	25 years from when it's first published

Copyright Infringement

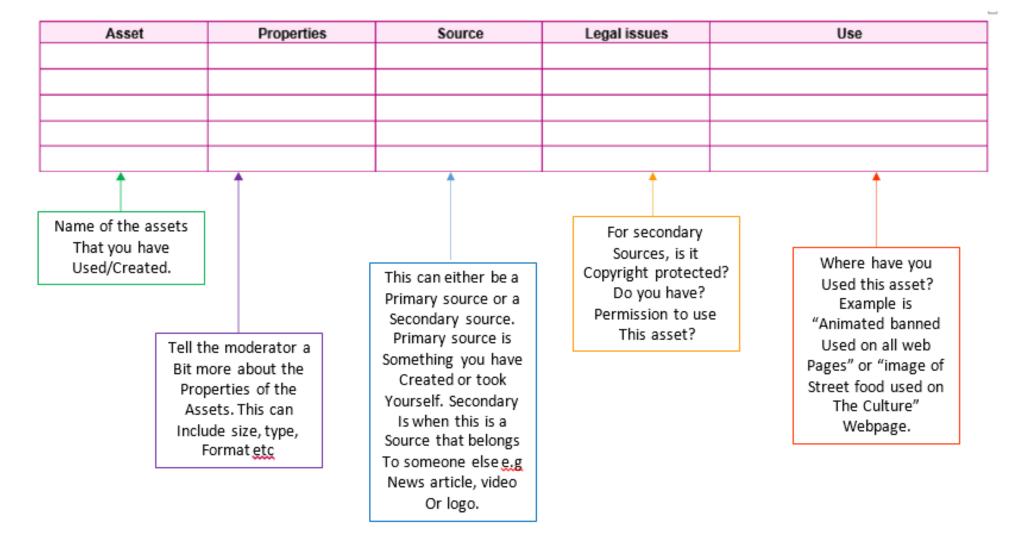
Copyright is infringed when any of the following acts are done **without permission**, whether directly or indirectly and whether the whole or a substantial part of a work is used:

- · copying the work in any way
- issuing copies of the work to the public
- renting or lending copies of the work to the public
- · performing, showing or playing the work in public
- broadcasting the work or other communication to the public by electronic transmission
- · making an adaptation of the work.
- Conviction in the magistrates' court the maximum term of incarceration in the UK for copyright infringement is 6 months and/or a fine of up to £50,000.
- Conviction in the Crown Court the maximum term of incarceration in the UK for copyright infringement is 10 years and/or an "unlimited" fine



What types of sources/assets can be collected to use in website design?...

Example of an assets table – this is how you can keep a record of all the assets you have collected or created to use while creating your website.



Year 11– iMedia(ICT) – RC	Possible Careers:	Keywords	Definition
The Internet $V_{Your} \rightarrow V_{Yourksp} \rightarrow V_{Internet}$	• Web designer • Data Analyst • Programmer	Tag/s	are the hidden keywords within a web page that define how your web browser must be formatted and displayed e.g. <title></td></tr><tr><td>The Internet also known as WWW which stands
for World Wide Web is a network of online
content formatted in a code called HTML. These
are interlinked HTML pages that can be
accessed over the Internet.</td><td rowspan=4> (1) When connecting a computer to a website, the user needs to have an internet service provider which is also known as an ISP. (2) The ISPs are responsible for making sure you can access the Internet, routing Internet traffic, resolving domain names, and maintaining the network infrastructure. (3) The website host server stores the webpages for individuals and organisations. Websites are hosted, or stored, on special computers called servers </td><td>Html</td><td>Stands for Hypertext Markup
Language is the standard
markup language for
documents designed to be
displayed in a web browser</td></tr><tr><td>It provides space for a wide range of information like documents, content and videos</td><td>Http</td><td>transfers web pages from
web servers to the client. All
web page addresses start
with http</td></tr><tr><td></td><td>Code</td><td>Is the set of instructions
forming a computer
program which is executed
by a computer</td></tr><tr><td>Hyperlinks e</td><td>CSS</td><td>Cascading style sheets are
used to format the layout of
Web pages</td></tr><tr><td>A hyperlink, or simply a link, is a link from a document to another document or part of the document that the user can follow by clicking or tapping on.</td><td></td><td>Webpage</td><td>are HTML documents that
present images, sound and
text accessed through a
web browser 143</td></tr></tbody></table></title>





CSS Script	Definition – What does it do?
Colour	Font colour
Text-align	Horizontal alignment
Background – Colour	Changes background colour
Background – Image	Change background image
Background - Repeat	Changes the background to stay in place or move when scrolled

HTML TAG	Definition – What does it do?	
<html></html>	Root of a HTML document	
<boy></boy>	Content of the page	
<head></head>	Information about a page	
<title></td><td>Tab title/ defines title</td></tr><tr><td><h1>, <h2>, <h3></td><td>Headings</td></tr><tr><td></td><td>Paragraphs</td></tr><tr><td></td><td>Image</td></tr><tr><td><a></td><td>Anchor (used in hyperlinks with href)</td></tr><tr><td>/</td><td>Ordered/unordered list</td></tr><tr><td></td><td>List item</td></tr><tr><td></td><td colspan=2>Creates and defines tables</td></tr><tr><td></td><td colspan=2>Table row</td></tr><tr><td></td><td>Table data</td></tr><tr><td><div></td><td>Divider 144</td></tr></tbody></table></title>		

		Improv	ement Plan	
	What should you do?	Short Term Effects of negative choices	Long Term effects of negative choices	Suggestions
Smoking	Not smoke	 Increase pulse rate Decrease: blood flow, smell, taste Possible mood swings 	 Addicted to Nicotine Wheezing Wrinkles Weaker Immune System Increased risk of: Heart disease and strokes Bronchitis Lung Cancer/disease Cataracts Decrease in Bone Density 	 Make a plan Nicotine Replaceme nt Therapy Increase Exercise Support groups (face to face and virtual)
Alcohol	Recommended Weekly Alcohol Intake = 14 units 1 unit = 1 glass of wine/ ½ pint	 Skin flushing Increased Heart Rate & blood pressure Dehydration Passing our and vomiting Hangover 	 Heart Disease Alcoholism Beer Belly Diabetes Cancer Liver disease Anxiety/Depression 	 Make a plan Alcohol free days/week s Don't keep alcohol in house Support groups (face to face and virtual)
Diet	Balanced Diet including all 7 nutrients – Carbohydrates, Proteins, Fats, Minerals, Vitamins, Water, Fibre	 Tiredness Stress and mood swings Weight gain/loss Headaches Bloated 	 Obesity & Tooth decay Lower self-esteem/anxiety High Blood Pressure Osteoporosis High Cholesterol Coronary Heart disease and strokes Cancers 	 Weekly Food Plan Diet Groups High Protein breakfast Eat Slowly
Exercise	Child – 5 days a week 1 hour Adults –30 mins a day, 5 times a week Heart Rate above 60% Max	 Increased tiredness/lower energy levels Weight gain Reduced memory & concentration Not as happy 	 Obesity Heart disease & strokes High Blood Pressure High Cholesterol Type 2 diabetes Cancers Osteoporosis Increased feelings of depression & anxiety 	 Exercise programs ie Conch to 5km Join a group/gym Exercise with friends Include sort 10 mins of exercise into day

		Illness & Disease	
Illness & Disease	Memory Picture	Definition	Lifestyle Causes
Asthma	٩Ž	Your airways are sensitive and become inflamed and tighten when they breathe if anything irritates them. This ca cause tightness and wheezing and make it hard to breathe	
Diabetes Type 2		A condition that causes high levels of glucose in your blood because you have a problem with producing insulin. This means the glucose stays in the blood stream and can't be used to give you energy.	
Dementia	·	Memory loss can be a problem. Risk increases with age	
High Blood Pressure	3	When your blood pressure , the force of your blood pushing against the walls of your blood vessels, is consistently too high .	
Bronchitis	AB	Bronchitis is an infection of the main airways of the lungs (bronchi), causing them to become irritated and inflamed. The main symptom is a cough, which may bring up yellow-grey mucus.	
Stroke	E R	A stroke is a serious life-threatening medical condition that happens when the blood supply to part of the brain is cut off.	
Coronary heart disease	V	Coronary heart disease (CHD) is when your coronary arteries become narrowed by a build-up of fatty material within their walls. These arteries supply your heart muscle with oxygen-rich blood.	
Mobility	۵.	Mobility refers to whether you can move an injured body part, like a joint or a limb.	
Liver disease		The liver has become so damaged that it can no longer work as effectively. The liver produces bile to help break down and absorb fats. Waste products and toxins are removed through bile. If the liver does work as well these toxins will harm the body	
Osteoporosis		Osteoporosis is a bone disease that occurs when the body loses too much bone, makes too little bone, or both. A lifelong lack of calcium plays a role in the development of osteoporosis. Low calcium intake contributes to diminished bone density, early bone loss and an increased risk of fractures. Eating disorders . Severely restricting food intake and being underweight weakens bone in both men and women	146

1	HEALTH & WELLBEING IMPROVEMENT PLANS				
	Health Improvement Plans	When creating improvement plans follow these steps:			
ŀ	Health Improvement plans should be based on the individuals physiological and lifestyle indicators.	1 Identify the health issue & goal			
ŀ	Should be person centered and include goals, actions, targets and identify support	2. Recommended actions to take			
ŀ	Person Centered Approach -Puts the individual at the heart of their health improvement plan. Takes into account the individual's needs ,	3.Set Targets for health improvement			
	circumstance and wishes • Needs = PIES	4. Identify support needed			
	 Wishes = Likes, dislikes, choices, desires Circumstances =Ilnesses, accesses to facilities, time, commitments, experience 	5. Identify possible obstacles and ways to overcome			

Sections of an Improvement Plan	Definitions	Improvement Pla	n Examples	
Goals	From the case study you will be about to identify goals	Reduced BMI	Stop smoking	Improve mobility
Recommended Actions		Eat a healthier diet and exercise more	Use nicotine substitutes	Start an exercise routine
Targets	Challenges to help a person complete the action. SMART Targets are more likely to be successful. These are: <u>Specific</u> = an exact goal, clearly explained <u>Measurable</u> =so that progress can be	Short Term = Within 6 weeks exercise twice a week and stop snacking in- between meals	Short Term = Cuts down to 5 cigarettes a day within 1 month	Short Term = Within 6 weeks walk for 20 mins every other day.
	assessed <u>Achievable</u> = possible for the person <u>Realistic</u> = suitable for the needs and circumstances <u>Time-related</u> = has a deadline	Long Term = Within 6 months exercise 3 times a week and eat less	Shorts completely within 10 months.	Long Term = Join a gym and attend regularly within 6 months
	Short Term Targets = less than 6 months Long Term Targets = 6 months to a year	than 1800 calories a day.		

	Formal Support	Types of f	ormal support & o		
Provided by heal	th professional who are paid and trained. Either primary, secondary or tertiary		ormansupport a c		
Primary care	Healthcare provided in the community for all individuals. The individual makes the initial approach to a	Physiologi	cal measuring aids -		
	medical professional.	pressure. V	veighing scales etc		
Health centre	 Measures & interprets BMI, pulse, blood pressure & peak flow 	Medicatio	• •		
	 Advises on lifestyle risks – ie smoking, alcohol and drugs Works with people to produce health improvement plans & monitor health 		upport – DVDs, heal		
	 Prescribes treatment 		••		
	 Refers to other health professions 		outine advice		
Dental Surgery	Run daily dinics to diagnose and treat dental issues. Tasks include: advice, clean teeth, perform minor	 Advice and 			
	surgeries.	 Emotional 	Support		
Opticians	Examine eyes for vision problems, diagnose and treat eye disease, prescribing glass & lens if needed.	 Support G 	oups:		
Pharmacy	Give advice on minor conditions, recommend medication and dispense prescriptions. Provide aids	o Alcoho	: Alcoholics Anonym		
-	such as nicotine replacement therapies and blood pressure monitors	o Diet: W	eight Watchers		
Secondary	Primary care professional refers you to a specialist, you are then in secondary care.		Exercise: Change4Lif		
Cardiologist	Specialises in diagnosing and treating diseases of the heart. they may carry out tests, and they may some do procedures and surgeries.	o Smokin	g: QUIT		
Psychologist	sess, diagnose and treat individuals suffering from mental distress and mental illness o Drugs: Talk to Frank, Action				
Physiotherapist	Treats people who have mobility, breathing and neurological problems	 Social Is 	solation of elderly :		
Orthopaedics	Specialise in disorders of eye movements and diagnostic procedures related to disorders of the eye and visual system.				
Tertiary Care	Patient needs higher level of care within the hospital. Tertiary care requires highly specialised equipment and expertise.		otential Obstacles		
Dermatology	Dermatologist specialises in treating skin, nail and hair disorders.	Emotional			
Psychiatry,	They make a diagnosis and work with you to develop a management plan for your treatment and recovery for mental illnesses.		motivation, a current health		
Allied	Professionals who may not be medically trained but use their knowledge to support peoples health.	Time	Lack of time b		
Professiona Is		constraints			
Podiatrist	Provides essential foot care for individuals with diabetes, circulatory and nerve damage.	Resources			
Art therapist,	Helps people who have behavioural and emotional problems by using drawing, painting and other art.		resources, e		
Dietician,	Uses their expert knowledge about the science of food to advise and support individuals in their	Lack of	Lack of inform		
	dietary needs.	Support	support. Not a		
Social worker	Provide advice, support and resources to individuals and families to help them solve their problem		services		
Youth Worker	Personal and social development. Support for young people between 11-25.	Access to	Geography,		
Informal Sup		services	or language		
Provided by peo	ple who are not paid to provide help		you cann		
Partners			to/access ser		
Family	They can:	Individual	Factors specif		
Friend Neighbour	\circ Follow same health plan \circ Reassure	specific	individual, su		
s Work	 Praise progress Praise progress Help overcome barriers 		gender, disab		
Colleagues			or addiction		
J		J L			

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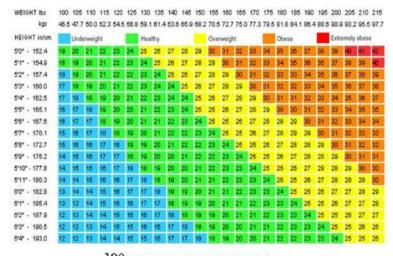
Pote	Potential Obstacles			
Emotional Time constraints	Low self-esteem, lack of motivation, acceptance of current health situation Lack of time because of work or family			
Resources	Lack of financial resources, equipment or opportunities			
Lack of Support	Lack of informal or formal support. Not able to access services			
Access to services	Geography, culture or language means you cannot get to/access services			
Individual specific	Factors specific to the individual, such as age, gender, disability, illness or addiction			

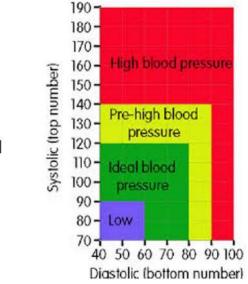
Hea	alth Indicators t	o discuss	What?	Abnormal Reading?	Risks	Causes
		BLOOD PRESSURE	Pressure exerted by the blood against the artery walls.	High Blood Pressure is 140/90 mm Hg Low Blood Pressure is 90/60 mm Hg or lower A Blood pressure chart is used to interpret measurements	Hypertensio n Heart Disease Stroke Kidney Disease Dementia	Lifesty le Diet Gene tic Lack of Exercise Stress Overweight
Physiological		BOD Y MAS S INDE X	 A way of measuring the amount of body fat. Based on height and weight and can be found on a published chart. Normal Reading 18.5- 24.9 	Underweight = <18.5 Overweight = 25-29.9 Obese = 30-34.9 Severely Obese= 35>	Underweight – Anaemia, Weak immune system, osteoporosis Overweight – Heart Disease, stroke, diabetes, arthritis	Poor diet (too many calories, too much fat) Lack of exercise Alcohol intake
Рһу		PEAK FLO W	 Measures the speed a person can expel air from their lungs Assesses health of lungs 	depend on sex and height and you need to use a peak flow chart to assess	 Asthma Emphysema Bronchitis Cystic Fibrosis Lung Cancers 	Any lung condition that decreases air flow will result in abnormal readings.
	-11-2-1111	PULS E RATE	Measures how fast the heart beats per minute (bpm) Indicates the level of a person's health & physical fitness.	Average is between 60- 100bpm for an adult High than 100bpm at rest is abnormal	Dizziness Heart Attack Stroke High Blood Pressure	Lack of exercise Overweight High stress levels Smoking
Lifestyle	SN.	KING	ALCORO IL III INTAKE	T è	LEVE	

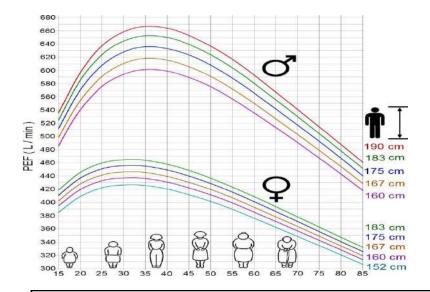
BTEC Health & Social Care Learning Aim B – Health Indicators

Need to be able to use these charts to understand people's current physical health









Resting Heart Rate Chart

Men (beats per minute)

Age	18 - 25	26 - 35	36 - 45	46 - 55	56 - 65	65 +
Athlete	49 - 55	49 - 54	50 - 56	50 - 57	51 - 56	50 - 55
Excellent	56 - 61	55 - 61	57 - 62	58 - 63	57 - 61	56 - 61
Great	62 - 65	62 - 65	63 - 66	64 - 67	62 - 67	62 - 65
Good	66 - 69	66 - 70	67 - 70	68 - 71	68 - 71	66 - 69
Average	70 - 73	71 - 74	71 - 75	72 - 76	72 - 75	70 - 73
Below Average	74 - 81	75 - 81	76 - 82	77 - 83	76 - 81	74 - 79
Poor	82 +	82 +	83 +	84 +	82 +	80 +

Women (beats per minute)

Age	18 - 25	26 - 35	36 - 45	46 - 55	56 - 65	65 +	
Athlete	54 - 60	54 - 59	54 - 59	54 - 60	54 - 59	54 - 59	
Excellent	61 - 65	60 - 64	60 - 64	61 - 65	60 - 64	60 - 64	
Great	66 - 69	65 - 68	65 - 69	66 - 69	65 - 68	65 - 68	
Good	70 - 73	69 - 72	70 - 73	70 - 73	69 - 73	69 - 72	
Average	74 - 78	73 - 76	74 - 78	74 - 77	74 - 77	73 - 76	
Below Average	79 - 84	77 - 82	79 - 84	78 - 83	78 - 83	77 - 84	
Poor	85 +	83 +	85 +	84 +	84 +	85 +	
	agelessinvesting.com						

	Short term target	Long term target	Support	Ways to overcome obstacles
Stop Smokin J	 Make a plan to reduce the number of cigarettes you have each day from 40 to 5 a day over 2 months (using nicotine patches to support). 	 Completely stop smoking in 6 months and within 9 months stop the use of nicotine patches 	 Use nicotine patches Support group – and online advice - QUIT 	 Friend could stop smoking near them Find social hobbies that don't include smoking
Reduce A coho Intake	 When having a drink, drink water alongside, starting immediately. Have at least 4 non drinking days per week 	 Reduce alcohol consumption to less than 14 units per week and not binge drinking 	 Join a support group i.e. AA 	 Replace alcohol with non- alcohol drinks Find social hobbies that don't include drink
Reduce calorie intake	 Follow a slimming world diet plan to lose 0.5kg each week for 4 weeks. Have 2 "health days" per week for 3 months where he eats healthy food. 	 To reduce weight by 5kg within 6 months and reach a healthy BMI within 1 cheat day per week and 6 healthy days per week which will improve the intake of nutrients. 	 Slimming World consultant and slimming world app - These will give lots of advice and inspiration. They will also help review diet plan and ensure it will help her lose weight. See a dietician to improve his diet and include more 	 Stop buying take ways and un healthy snack Friends and family could do the diet with you
Increas e activity levels	 Exercise 2 times a week for at least 30 mins with the heart rate at 65% max. Complete a minimum of 7000 steps per day. Take 2 short walks per week to build up his stamina. 	 Exercise 5 times a week for at least 30 mins with the heart rate at 65% max. Complete a minimum of 10000 steps per day. Build up his exercise to be 30-45 minutes per day. 	 John Brown With Wether Times fiftheless instructor Friends and family exercise with them Fitness Apps 	 Family could suggest opportunities to take a walk instead of going in the car Use free apps and research free ways to exercise Log amount of exercise to motivate
Social isolatio n	 Within 6 weeks, be able to leave the house every day for a short walk and have a short conversation with one person 	Join a social dub and be able to attend regularly within 6 months	 Friends and Family could visit regularly and accompany to social events 	 Friend could encourage the person and discuss risks of isolation Friends could suggest ways to increase interactions

		Key Terminology	
Restricted gross motor skills	A child is unable to control the large muscles in their bodies compared to other children their age	Care/educational providers	Settings that provide formal care or education for children
Restricted fine motor skills	A child is unable to control the small muscles in their hands and fingers compared to other children their age	Family structure	The way in which a family is organised
Delayed gross motor skills	The large movements of a child's body are not progressing as quickly as other children their age	Expected milestones	Development that is expected at a particular age
Delayed fine motor skills	The small movements of a child's hands are not progressing as quickly as other children their age	Initiate play	To start play
Poor concentration levels	When children find it difficult to focus on what they are doing	Navigate	Move with planned direction
Delayed literacy skills	When a child's reading and writing skills are not progressing as quickly as other children the same age	Preferences	Things that children prefer to do
EAL	English as an additional language	Lack of responsivene ss	Not responding to people
Negative role model	Someone who does not set a good example	Emotional resilience	A person's ability to adapt to stressful situations
Social norms and values	Attitudes and behaviours that are considered normal in society	Positive relationships	A relationship between two people that makes them happy
Disruptive behaviour	Unwanted behaviour that disturbs and interrupts activities	Expression	The action of making known ones thoughts and feelings
Transition	A change in a child's life	Routine	A sequence of actions that is regularly followed

	Circumstan	ces that may impact on a child's learning	How they may affect learning
Physical circumstances	F	These may include sensory impairments, restricted fine and gross motor skills, and delayed gross and fine motor skills.	A child may not be able to access learning at varying levels, grasp and manipulate small objects and to navigate play areas. Children may also tire easily and not be able to sustain involvement in activities,
Cognitive circumstances		These may include poor concentration levels and delayed literacy skills	A child may not be able to understand the rules of play,
Communication and language circumstances	P	These may include English as an additional language and a child who has a language and communication delay	A child may have difficulty communicating preferences and choices, and play with others may be limited due to lack of responsiveness
Social and emotional circumstances		These may include negative role models, difficulty forming friendships with other children, disruptive behavior and transitions such as death of a loved one, birth of a new sibling and moving house.	A child may have poor emotional resilience, may isolate themselves, refuse to join in play with others and may have low self-esteem. A child may also have limited expression of thoughts and feelings and find it difficult to build positive relationships with others.

	Key Terminology			
Risk	Likelihood of an environment, activity or resource causing harm	Health and safety conside	erations of inside environ	ments
Hazard	Potential for an environment, activity or resource to cause harm			
Risk assessment	A process of evaluating what might cause harm to people and making sure things are in place to manage the risk		ſ	17
Positive risk taking	Balancing the potential risk of harm against the benefit of children participating in activities	Lavout of furniture	Width of doorw	j_ avs and corridors
Adult to child ratio	The number of adults to the number of children			
Stimulation	Giving something interest, enthusiasm or excitement			
Role model	A person looked to by others as an example to be imitated			363
Intrusive	Causing disruption or annoyance through being unwelcome			
Smart device	Allows us to connect different devices or networks	Types of flooring and floor coverings		
Parental controls	Software and tools that can be installed on internet enabled devices		Layout o	f furniture
Personal informati	Private details about someone e.g. date of birth, full name, address			
on				Ъ
Inappropriat	Information online that could upset a child, including violence and			
e content	bad language			
Trip hazard	Objects on the floor that cause someone to trip and fall	How resources are organised	Use of specific areas for play	
Toileting needs	The need to use the toilet		Use of specific	
Accessibility	How easy it is for an area to be reached]		
	Ensuring Children are safe	Health and safety consid	erations of outside en	ivironments
Manage	An adult must consider the hazards and risks when planning			
ri sks and hazards	an activity. Children should also be taught how to explore and take risks in a positive way, learning how to judge risks for themselves. Adults should use safety features such as the BSI kite mark, age advice symbols and the CE mark to ensure the suitability	Î		
	of resources.	Appropriate dothing	Planning ahead for hur	nger, thirst, toilet break etc.
The role for the adult	Adults have a responsibility to plan play activities, ensure the correct adult to child ratio and model appropriate behaviours. Adults must be available but not intrusive and ensure the play is age appropriate.	ΟĠ	ô,	Ð.(((
Internet enabled technology	Adults need to teach children how to be safe online, including not sharing personal information or befriending strangers. Controls must be put in place by adults, and also talk to children about internet safety.	Accessibility – how children may enter and exit buildings	Choice of outdoor play resources	Noisy or quiet play spaces – use of signs and maps

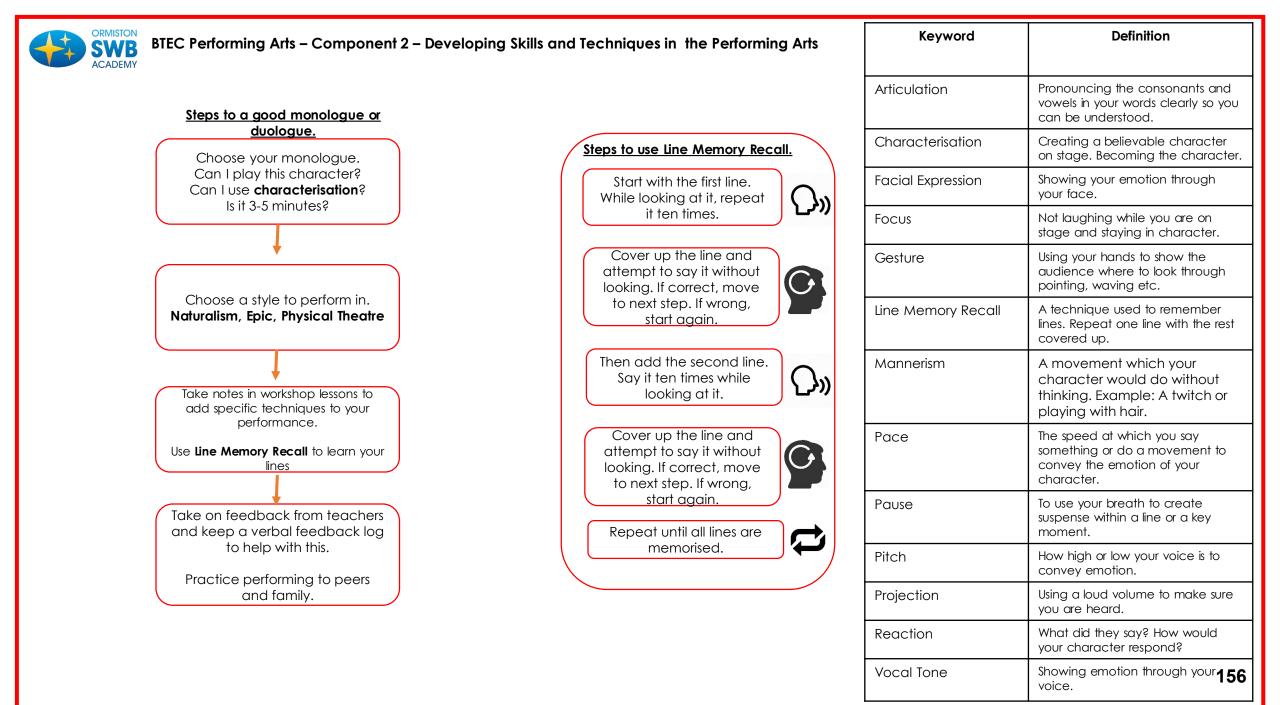


Practitioner	Style	Key Stylistic Features		Book Musical	A musical where the music, lyrics and script follow a well thought out narrative.
Bertolt Brecht	Epic Theatre	Alienation: Using sing, placards, pitch and pace to make the audience distance from the action on		Creative Intentions	The theme, issues within a play or the style. Why did the director choose each of these?
		stage. Gestus: To give a character a clear and over exaggerated gesture they must use when they come on stage		Epic Theatre	A form of didactic theatre where the scenes are episodic and follow no narrative. Often political.
		Political themes: Brecht wanted the audience to think about the corruption of the world they live in.		Focus	Not laughing while you are on stage and staying in character.
Frantic	Physical	Chair Duets:	\bigcap	Genre	The style of a theatre.
Assembly	theatre	Using two chairs create a continuous string of movements. Add emotion. Add pace to speed up or slow down sections. Hymn Hands: Use hands to mirror what your partner is doing- or grab the hands or shoulders for effect.	Ã	Physical Theatre	a form of theatre which emphasizes the use of physical movement, as in dance and mime, for expression.
		Round-by-through:Moving around the body.Go through another actors' arms to create a spiral effect.	6	Practitioner	A person who pioneered a style of theatre.
Jerome Robbins	Book Musical	Allegory for Romeo and Juliet: based on this story- however focusses on the love of a Jew and a Catholic		Purpose	The reason a piece of theatre exists. Example: to put across a political message.
		Theme of Love:The love Tony felt for Maria and Romeo for Juliet made them defy their families, theirfriends and their social world. Their love is strong and forceful, so much so that it madethem revolt against the very world they revolved in and, sometimes, even againstthemselvesTheme of Society:Racial inequality Jets Vs Sharks= Jews vs CatholicsGender Roles in the song "America" we see how men and women view America verydifferently.	¢ i Q		

Keyword

Definition

BTEC Performing Arts – Component 1 – E	Exploring the Performing Arts- Learning Aim B	Keyword	Definition
		Responsibilities	What someone in the theatre is required to do. Director- have a vision and tell the actors what to do on stage.
Director	To have a vision for the production. To be in charge of telling actors where to go and what to do To tell the other role holder on the production what	Role	A job role within theatre: director, actor, stage manager, lighting designer etc.
	they need to do to bring your vision to life.		
Choreographer	To design the movement for the show. To teach the movement to the actors/ dancers To the liaise with the director about their vision		
Costume designer	To design the costumes for the actors. To take accurate measurements or the actors. To liaise with the director and ensure that costumes are in line with the setting and time of the production.		
Set designer	To design the set for the show. To build and paint any set required. To liaise with the director about the context and vision of the show.		



BTEC Performing Arts – Component 3 – R	esponding to brief	Keyword	Definition
ACADEMY Milestone 1- Ideas Log:	Milestone 2- Skills Log:		
 2. What is your target audience and why did you decide on 	 Milestone 2- skills tog. What was your role in the group? (director, performer etc) 	Articulation	Pronouncing the consonants and vowels in your words clearly so you can be understood.
this?	2. Which style did you choose? Why?	Characterisation	Creating a believable character on stage. Becoming the character.
3. What resources do you think you will need to develop and perform your performance?	3. Which techniques and skills did you choose? Why did you choose them?	Facial Expression	Showing your emotion through your face.
 4. How do your ideas for the performance meet the brief ? 5. How has the work of Pantomime, Stanislavski, Frantic 	4. What work have you done individually to help the group? (research, rehearsal leading, choreographing	Focus	Not laughing while you are on stage and staying in character.
Assembly or any other practitioners influenced your performance? 6. What ideas have you contributed to the performance plan? Were these successful? Why?	movement etc)5. How did the resources you chose aid your	Gesture	Using your hands to show the audience where to look through pointing, waving etc.
	performance?6 What would you change or improve?	Line Memory Recall	A technique used to remember lines. Repeat one line with the rest covered up.
7. How did you come up with the ideas you have <u>Milestone 3- Workshop Performance:</u>	Milestone 4- Evaluation Report: 1. How did the result of your performance meet the brief?	Mannerism	A movement which your character would do without thinking. Example: A twitch or playing with hair.
 You must present your group workshop performance or pitch/presentation to an invited audience. The group workshop performance must be between 10 	 2. How did you process and ideas develop through this project? 3. Was the outcome of the performance what you wanted? What did the audience learn? 4. What were the key strengths of your group's 	Pace	The speed at which you say something or do a movement to convey the emotion of your character.
 And 15 minutes long. You will need to perform as part of a group and work well together. 		Pause	To use your breath to create suspense within a line or a key moment.
You will be assessed on your individual skills and	performance?	Pitch	How high or low your voice is to convey emotion.
techniques, collaboration with others and communication of creative ideas to the audience	5. What were the key strengths in your individual performance?	Projection	Using a loud volume to make sure you are heard.
through your role.	6 What would you improve upon given the chance again? Why would you change this? How would that	Reaction	What did they say? How would your character respond?
	help your performance meet the brief?	Vocal Tone	Showing emotion through your 157 voice.

Knowledge Organiser: Year 11 Autumn Term



Section A	Section A Research			
Design Brief:	Primary Research: involves the collection of data that does not yet exist.			
 A Design Brief is a short paragraph explaining the situation you have been given and the problem you need to solve. What is your problem? Can you find one that needs solving? How could you solve it? Client: A Client is a group or a single person you are developing a page design on some of solve.	Site Visit: Ask a shop or manufacturer of What are the dimensions of product How is it made? Whats popular? Evaluate a product at first hand: See how How many components? Fixings? Materials? Dimensions?	ts?		
developing a new design or concept specifically for to meet their needs.	Secondary Research: involves the collection of data that does not yet exist.			
 Gender specific Age specific Disabilities Sports person Adults Teenagers 	 Anthropometrics and Ergonomics: Data if the human body What are the 50th, 95th and 5th percentile of your client Specific body sizes to suit the products function Assess how a product interacts with the user. 	Ethical Issues in design: issues around design Social Issues: Communicate, Language, Understanding, Meanings. Moral Issues: Health and Safety, Religious Views or Beliefs, Environment/Sustainability. Environment Issues: Possible Issues: Recycling, Renewable Energy, Materials, Disposal.		
Analyse the NEA situation through a spider diagram and list the following: • Existing products • Materials	Existing Products: • Find out information about	Research materials/finishes relevant to your product; does it meet manufacture needs, user needs and how can it be processed?		
 Materials Finishes Function Function Function Function Client 	parts Prices Features/Functions Reviews What can be improved? 	Research Tools/processes and machinery: Accurately manufacture your product, most efficient manufacturing process, safety and what parts could they be applied for?		



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Knowledge Organiser: Year 11 Autumn Term

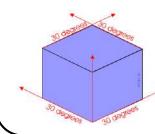


Design Specification

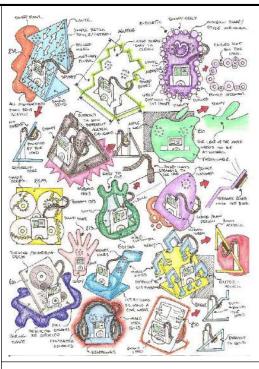
A list of points that have been developed from research carried out that your design concept should meet t function with your client effectively.

- <u>Aesthetics:</u> What colour is the product? What shape is the product? What texture does the surface have? Does the product look attractive?
- <u>Cost:</u> What is the retail cost (the price you would pay for it in a shop)? How much do you think it would cost to make the product?
- <u>Customer/Client:</u> Who might use the product? Who else, might be affected by the product? What market is the product aimed at?
- <u>Environment:</u> Are recycled materials used in the product? How do you know? Could the materials used be recycled? Are there alternative options which would be kinder to the environment?)
- <u>Safety:</u> What safety issues have been considered when the product was designed? Other safety issues which need to be considered?
- <u>Size:</u> What size is the product? How long, wide and tall is it? (Hint: use dimensions) Is the product comfortable to use? Are its proportions appropriate for its use? How has the designer considered human factors (ergonomics) in the product's design?
- <u>Function:</u> What will the product be used for? How does the product have to perform? How is it tested? How well does it work? What market or client quality standards does the product meet?
- <u>Material/Manufacturing</u>: What materials or components is the product made from? Which manufacturing processes were used to make the product?

Isometric Projection



Isometric drawing is way of presenting designs/drawings in three dimensions. In order for a design to appear three dimensional, a 30 degree angle is applied to its sides. The cube opposite, has been drawn in isometric projection.



<u>Rendering</u>

This is applied to a drawing to show the client how the product is finished or what material it may be manufactured from.

The example opposite shows the lighting effect applied to all three visible sides of the cube.

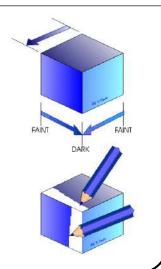
When shading a side of a cube using this technique, a coloured pencil is used in a linear fashion parallel to an edge of the cube.

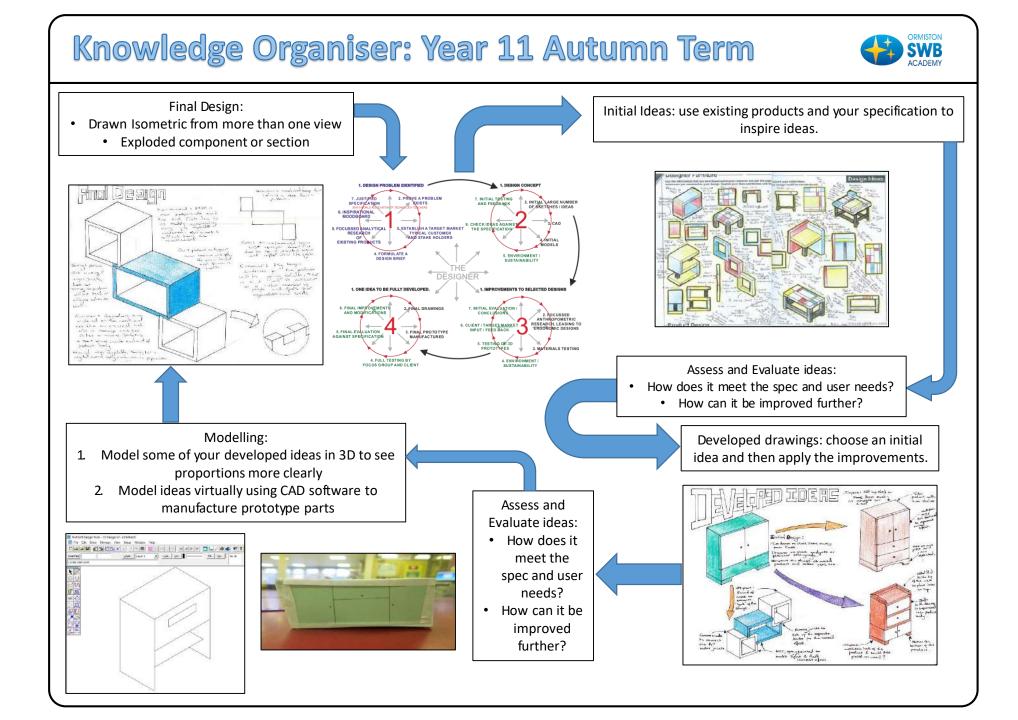
What makes an effective design sheet?

Clear Drawings: Drawn in 3D or 2D but very clear and neat so the idea can be understood.

Annotation: used to explain parts of the design that are not clear from the drawing alone.

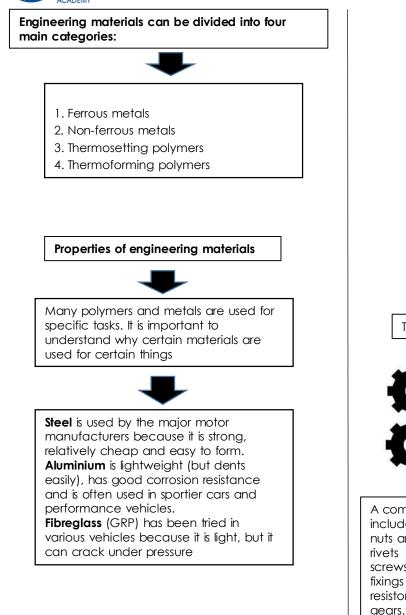
Rendering: Colour or tone added to the idea to show the anticipated finish of the product or the material to be used.



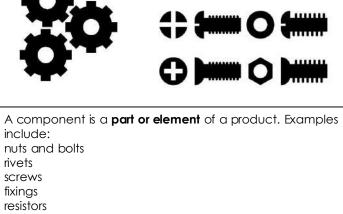




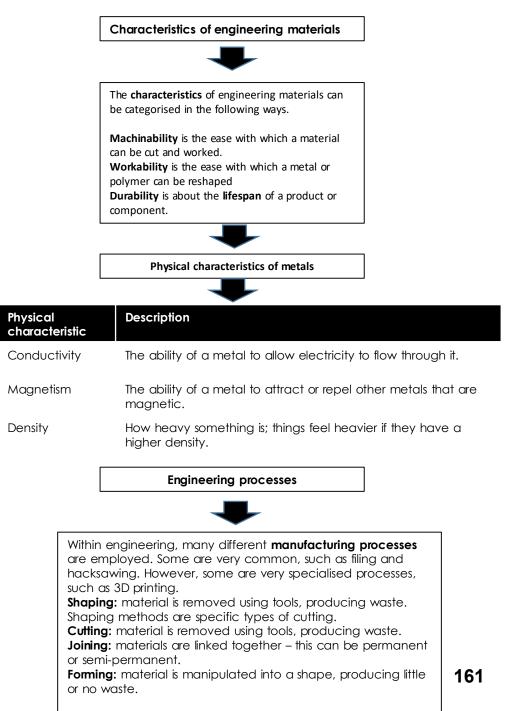
BTEC Engineering Component 2 Learning Aim A



The most important properties of engineered products to remember are their strength, hardness and toughness. **Strenath** - the strenath of a material is its ability to resist stress and strain. Hardness - the hardness of a material is its resistance to wear, abrasion and indentation. **Toughness** - a material that is tough is resistant to sudden impact or shock loadina. Types of components



A manufacturing company may make the components they need, themselves, or it may be cheaper to buy **proprietary** items.

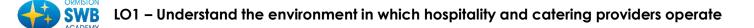


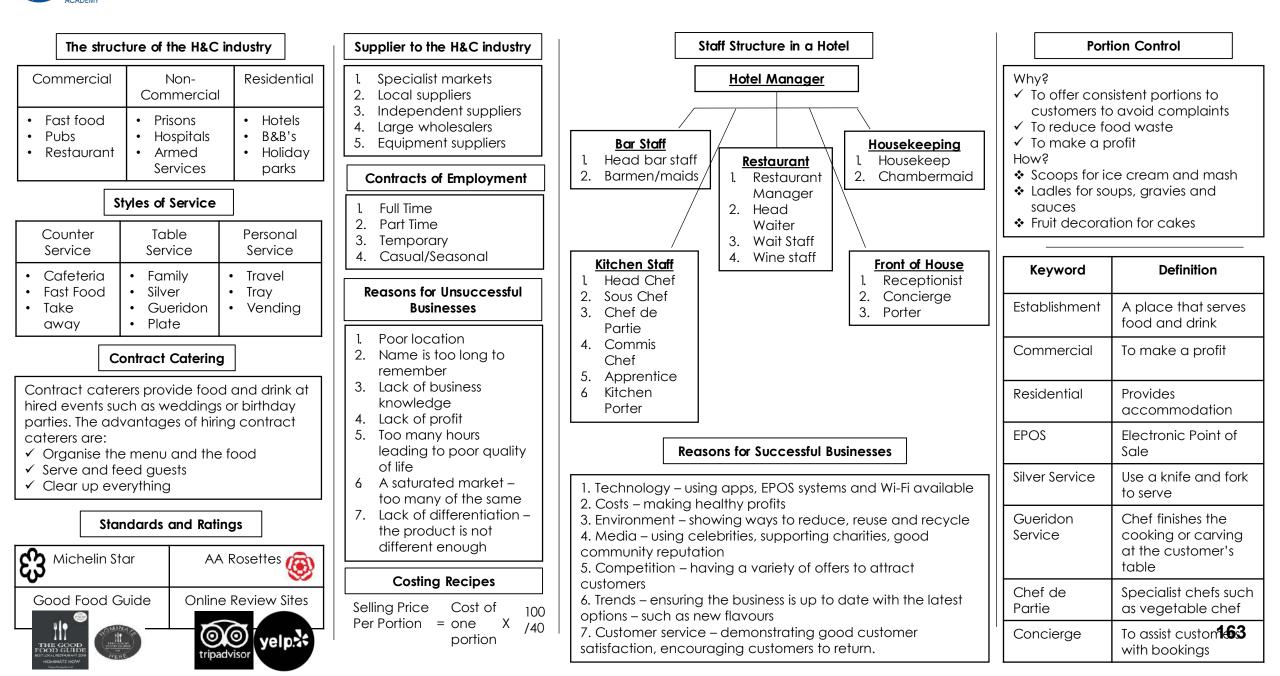


BTEC Engineering Component 2 Learning Aim A

Materials used in Engineering including key properties and characteristics

ACADEMY		Туре	Properties	Composition (what it's made of)	Examples of use
	Ferrous metals are materials that contain iron. This means that they have magnetic properties greater than other materials. Iron in its pure form is a soft, grey material that is difficult	Mild steel	ToughDuctileMagneticMalleable	 Iron 0.1–0.3% carbon 	ScrewsNailsBolts
Ferrous metals	to machine as it gives a poor surface finish and does not cast or forge well. The addition of carbon changes and improves its properties, allowing the production of steel and cast iron . Ferrous metals – except for stainless steel – are also very vulnerable to rust when exposed to water or moisture	Wrought iron Stainless steel	 Hard Brittle Magnetic Hard 	 Iron 2-6% carbon Alloy 	 Machine parts Vices Manhole covers Cutlery
			 Tough Sometimes magnetic Difficult to cut 	 10.5% + carbon 10.5–18% chromium 8% nickel 8% magnesium 	SinksMedical equipment
	Non-Ferrous metals				Thermoforming polymers
	Non-ferrous metals are metals that do not contain iron		Thermosetting polymo	Ther by h It is	moforming polymers can be softened neating, and often remoulded . becoming increasingly important for
Туре	Properties Composition Examples of use		thermosetting polym do not melt when he applied and cannot	at is impo	rmers to be recycled and to have less act on the environment.
Aluminium	 Corrosion-resistant Malleable Ductile Easily machined Pure metal Pure metal Foil Drinks cans 		re-formed by applying heat.	the	rethylene , also known as polythene, is most common plastic in use in the world. a highly ductile material, which means
Copper	 Corrosion-resistant Malleable Ductile Tough Easily machined 	Se	Once formed, thermosetting polyme cannot be altered. The means that they can be recycled and use	ers re-fo	it can be stretched and prmed.
Brass	 Good technical conductor Casts well Harder than copper 65% Door fittings 	ngs	again for another engineered product.	Acry poly or tu	ylic is a strong, rigid and transparent /mer that can be supplied as sheet, rods ubes. It is often used to replace glass in enhouses, lenses and visors 162
	 Easily machined copper Locks 35% zinc Musical instru 	ments			

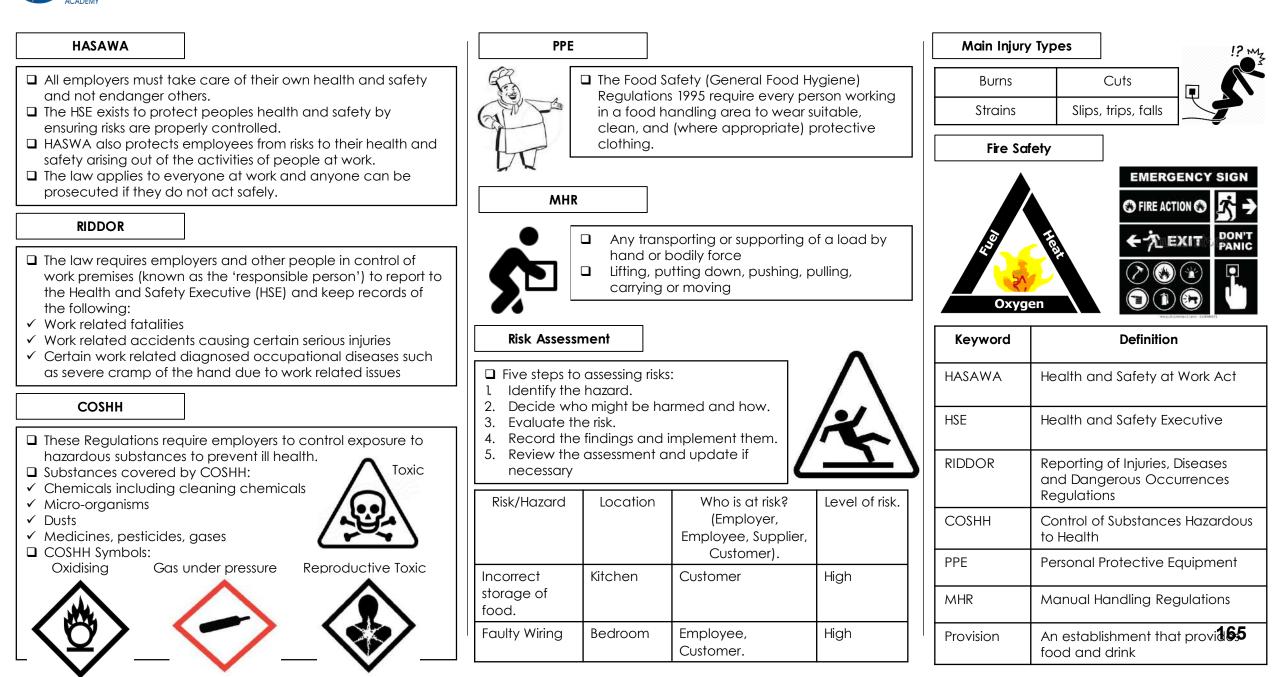




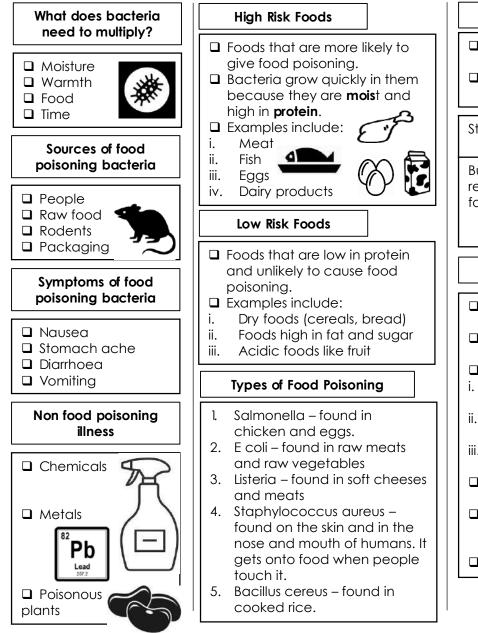


Kitchen Workflow	Small Equipment	[Types of Custome	er		Good Customer Service
1. Delivery 2. Storage	2. Storage Chef's Knife			Business/ Corporate	 ✓ Respectful and polite ✓ Helpful and attentive ✓ Smart and restance 	
 3. Food preparation 4. Cooking 5. Holding 6 Food service 7. Wash up 8. Waste disposal 	Bread Knife Boning Knife	Customers who visit the establishments in their leisure time e.g. a meal with friends, a family day out, tourists.	Customers who live in the local area who visit the establishment often e.g. regular Sunday	E.g. business lunches. Use business facilities in establishment for meetings or presentations, courses and	 Smart and professional Friendly Patient Deals with customer problems effection Knowledgeable Makes the customer feel welcome, to want to return. 	
Types of Documentation Image: Temperature charts	Filleting Knife		lunch	conferences	Keyword	Definition
 (fridge, freezer) Time sheets (staff working hours) 	Paring Knife Cleaver Knife		Safety and Securi	r health and safety	Perishable Foods	Foods with a short shelf life such as meat, fish and dairy
 Accident report form (to log first aid/accidents) 	form form form form form form form form			Staple Foods	Foods with a longer shelf life such as canned products	
Equipment fault reports	Large Equipment	you whether you a risks. Ways to achie	are doing enough to eve this is through:	mitigate these	Stock Control	Using the First In First Out (FIFO) rule to rotate food stock
Kitchen Dress Code (PPE) Chef Hat Neckerchief	All large scale equipment such as a floor standing mixer, walk in fridge/ freezer and a large oven must be: o Turned off before cleaning o Use correct cleaning materials	CCTV Staff	f Lockers Safes for		Consumer Rights Act 2015	A legal right to reject goods that are of unsatisfactory quality, unfit for purpose or not as described
Jacket	 Any attachments should be stored correctly If equipment is not working correctly it must be reported Ensure there is no food left on the 		Customer Rights	Š Š	Food Safety Act 1990	Businesses do not include, remove or treat food in any way that would be damaging to the health of people.
Apron	equipment as it will contaminate		ght to be protected (against hazardous goods) ght to be informed (quality, quantity, allergies)		Equality Act 2010	All must be treated equally regard to age, gender, race
Trousers Chequered Trousers			e their complaints be k compensation eive satisfactory goo	e heard	Disability Discriminat ion Act 2010	Unlawful to discriminate against mental and physical disabilities 164

LO3 – Understand how hospitality and catering provision meets health and safety requirements







HACCP

- By law, catering establishments have to do a risk assessment for food hygiene.
- This is to stop food being contaminated before it reaches the customer.

Step	Hazard	Hazard Prevention
Buying and receiving food	High risk food could be contaminate d with bacteria	Buy from suppliers with a good reputation.

EHO

- EHOs check that catering establishments are obeying laws such as the Food Safety Act.
- They can inspect a catering establishment at any time.
- □ The EHOs check the hygiene of:
- . The **food premises** (the place where the food is)
- ii. The **employees** (the people working with the food)
- iii. The **working practices** (the ay that things are done around food)
- EHOs advise catering establishments to improve their hygiene.
- They have the power to close establishments, impose fines, or take legal action against catering establishments that break the law.
- □ They issue a food hygiene rating from 1 to 5

FOOD HYGIENE RATING (0) (1) (2) (3) (4) (5)

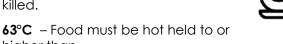
Key Temperatures

Monitoring and controlling temperatures during cooking, serving and storage of food limits bacterial growth and helps prevent food poisoning.

100°C – Bacteria is dead

higher than.

75°C – Food must be cooked to or higher in the middle. Most bacteria is killed.



5°C to 63°C – Bacteria multiplies rapidly, this is the danger zone. Food should spend no more than 90 minutes in the danger zone.

0°C to 5°C – Food is stored in a fridge, bacteria has slow growth

-18°C – Food is stored in a freezer, bacteria is dormant(too cold to multiply).

Keyword	Definition			
EHO	Environmental Health Officer			
Due Diligence	Establishments have done everything in their power to safeguard consumer health			
НАССР	Hazard Analysis Critical Control Point			
Hot Held/holding	When food is kept hot until it is served 166			



Unit 2 – Hospitality and Catering in Action

AC 1.1 – Functions of Nutrients	AC 1.3 – Nutritional Deficiencies (-)	AC 1.3 Nutritional Excesses (+)
Carbohydrates provides the body with energy. There are two main types, complex and simple. Complex carbohydrates give long lasting energy. These are found in foods such as bread, pasta and cereals. Simple carbohydrates make blood sugar levels go up very quickly. This provides a short burst of energy. These are found in sugary; foods such as cakes, jams and sweets.	 Lack of energy/ tiredness Weight loss Severe weakness 	 Tooth decay (simple sugars) Raised blood sugar levels Weight gain
Protein is needed for growth and to repair cells. Protein is made up of amino acids. Proteins that are high in essential amino acids are called high biological value (HBV) proteins. These are found in milk, cheese, fish, eggs, meat and soya beans. Proteins that are low in amino acids are called low biological value (LBV) proteins. These are found in nuts, cereals and pulses.	 Weak nails Hair loss Weakened immune system Poor growth (children) Food is nor digested properly 	 Weight gain Strain on kidneys and liver
Fats are used by the body for energy. Fat also forms an insulating layer under your skin to keep us warm and protect our organs, such as our kidneys. There are two main types of fat, saturated and unsaturated. Foods such as meat, cheese and butter are high in saturated fats. Foods such as seeds, fish and vegetable oils are high in unsaturated fats. We should eat less saturated fats.	 Weight loss Bruising of the bones Lack of vitamin A, D, E and K 	 Weight gain Type 2 diabetes High blood pressure (adults) High cholesterol (adults) Heart disease (adults) Organ failure
Fibre helps food to move through our bowels and prevent constipation. Foods such as vegetables, wholemeal bread and beans are high in fibre.	 Constipation Bowel cancer 	 Children feel full and so miss out on other nutrients
Water is needed for lots of reasons, keeping our body at the right temperature, digesting food, lubricating our bones and keeping us hydrated. Water is found in drinks, fruits and vegetables.	 Dehydration Headaches Kidney stones 	✤ Water intoxication
Vitamin A good vision, especially when it is dark. B group vitamins releasing energy from carbohydrates. Vitamin C Fighting diseases and helping the body to absorb iron. Vitamin D along with calcium, it helps our body make strong bones and teeth.	 Vit A infected mucus membranes. Vit B1 beriberi. Vit B2 sores. Vit B3 pellagra, dementia, dermatitis. Vit B9 megaloblastic anaemia. Vit B12 pernicious anaemia. Vit C scurvy. Vit D rickets (children), osteomalacia (adults) 	Vit A night blindness. Vit D kidney damage
Iron to make red blood cells to carry oxygen around the body. Calcium Along with vitamin D, calcium helps make strong bones and teeth.	 Iron deficiency anaemia Calcium same as Vit D 	Iron nausea, stomach pain. 167



AC 1.4 - Cooking Methods

Boiling – A liquid is heated to 100°C to cook foods like pasta, rice and potatoes. A quick and simple method of cooking that does not require any fat. Boiling makes the texture of food soft, over boiling really soft. If boiling vegetables, B vitamins and vitamin C is lost during boiling.

Steaming – Food is cooked from the steam of boiling water. Steaming food such as vegetables, fish and rice helps to preserve colour, texture and water soluble vitamins.

Baking - Food is cooked using the dry hot air of the oven. Foods that are baked such as cakes, pastries and biscuits cause the outer layers to turn brown and crisp.

Grilling – Dry heat is used to cook food at a very high temperature, it is fairly healthy because no fat is added, and fat drips out of the food as it cooks. Because the heat is so high the food cooks quickly and browns the outside.

Stir-fry – Food is cooked in a wok with very little oil. This is a healthy method as food cooks quickly due to very small pieces of meat and vegetables, leading to crunchy and colourful vegetables that have limited vitamin loss.

Roasting – Food is also cooked using dry heat in a hot oven. Fat is added to the outside of roasted food, causing the outside to go brown and crisp, while the inside stays moist.

Poaching – Food is cooked very gently in liquid that is below boiling point. Delicate foods such as meat, fish and eggs.

 ✓ Use seasonal ingredients ✓ Use ingredients from local farmers and markets 	 ✓ Cover pans ✓ Use the correct size hob and pan ✓ Cook different foods together ✓ Don't boil more water than you need ✓ Use water carefully – bowls to wash 	AC 2.2 – Environmental Policy An establishment requires an environmental policy because : 1. It is the law 2. Saves the establishment money	 Recycle mater Use biodegrad taking away fo Give foods to c Weigh ingredie 	 ✓ Reuse glass bottles and plastic containers ✓ Reuse leftover food for stock, compost or animal feed ✓ Use cardboard for wet floors 		
 ✓ Use ingredients from rooftop garden ✓ Use less plastic 	 Order in bulk to reduce packaging use FIFO to rotate stock Prepare the correct amount of ingredients Use accurate portion control 	 3. Builds a good reputation 4. It saves energy, water and reduces waste 5. Reduced the harm to the environment ✓ Fully load dishwashers 	 Recycle materials using recycling bins Use biodegradable packaging for taking away food Give foods to charities Weigh ingredients accurately 			
Keyword	Definition	 Fully load washing machines Carry out maintenance checks 	Keyword	Defin	ition	
Biodegradable Decomposes naturally in the ground		 ✓ Turn equipment off ✓ Use energy efficient equipment 	Recycle Product is broken down			
Food miles	The distance food has travelled from field to plate		FIFO	something new First In First Out	168	

Geography – Topic 1 Chang	ging Physical La	ndscapes		Keyword	Definition
• The study of rocks and	<u><u> </u></u>	oastal Landforms	Formation of a Wave-Cut Platform 1. Weath or weakens the top of	Erosion	The wearing away and removal of material by a moving force e.g. rivers or waves.
 The study of rocks and different rock types Sedimentary = formed in 	Deposition		the cliff. 2. The sea dit oc ks the bose of	Abrasion	Erosion caused by sediments rubbing against the river bed, bank or cliff.
layers (e.g. chalk) Igneous = formed from cooled lava (e.g. granite) 	cooled notch.		Attrition	Erosion where particles in rivers or the sea are worn down as they collide with each other.	
Metamorphic = formed under intense heat and pressure	Spit	Coves Caves, Arches,	causing the cliff to collapse. 4. The boc kwash carries the rubble to wards the sea	Hydraulic Action	Erosion caused by the sheer force of water wearing away cliffs, river beds or banks.
(e.g. marble)	forming a wave-cut platform.	Solution	Erosion where some rock minerals slowly dissolve in water, which is slightly acidic.		
Erosion • There cre four types of erosiond processes thetoccurinboth coast alarrad	Tombolo	Wave cut platforms	cliff continues to retreat.	Transportation	The movement of sediment by rivers or waves.
river landscapes. • Abrasion			<u>ce Coastal Erosion</u>	Traction	Transport of sediment through a rolling action along the river bed or sea floor.
 Attrition Hydraulic Action Solution 	Strategy Type Hard	Strategy Sea Wall – concrete walls	Advantages and Disadvantages + Effective at stopping the sea	Saltation	Transport of sediment being bounced along the river bed or sea floor.
Weathering	Engineering built at the top of a beach		 Very expensive to build and maintain + Force waves to break, protecting cliffs 	Suspension	Transport of sediment carried within the water flow.
 Weathering There are three main types of weathering processes that can affect rocks. 	Hard Engineering		 + Relatively cheap and easy to maintain - Restrict access to beach - Do not fit in with local geology 	Solution	Transport of sediment particles that have been dissolved in the water.
 Biological Weathering (e.g. plant roots) Chemical Weathering (e.g. 	Hard Engineering Groynes – wooden or rock structures built along the beach at right angles	+ Quick to construct + Trap sediment and widens the beach	Longshore Drift	The movement of material along a beach transported by wave action.	
acid rain) • Mechanical/Physical Weathering (e.g. freeze-thaw)		reducing wave energy - Stopping movement of sediment can affect elsewhere on coast	Weathering	The breakdown and decay of rock by natural processes acting on rocks, on cliffs and valley sides.	
O	Soft	Beach Nourishment –	- Can be ugly + Can absorb more wave energy	Mass Movement	The movement of material down a slope due to gravity.
• There crefour main ways that sediment is transported in river and coastal	Engineering	adding sediment to a beach	+ Easy and cheap to maintain - Needs constant maintenance	Deposition	A process where sediments are dropped by the river or waves that carried them.
environments. • Traction • Saltation • Suspension	Soft Engineering	Sand Dune Regeneration – grasses and bushes are planted to stabilise dunes	+ Maintains a natural coastal environment - Areas of beach have to be fenced off	Hard Engineering	Strategies using artificial structures (e.g. concrete) to prevent river or coastal flooding
 Solution long shore diff is diso a method of transportation in coastal environments. 				Soft Engineering	Flood defences that work with natural processes to reduce the risk of river or coc \$69 flooding.



Geography – Topic 1 Changing Physical Landscapes

Formation of a Spit

- 1. Prevoiling wind blows de an angle to the coastline.
- 2. Wores come in di an angle, resulting in lonashore drift.
- 3. Longshore drift noves sediment dong the coastline.
- 4. A spit is formed when the material is deposited.
- 5. Over time the spit grows and c and evelop a hook if wind direction changes further out.
- 6. Where the soit meets a river, or where there is foster woter, the spit stops forming os deposition no longer occurs.
- 7. There is a shelf ered area behind the spit. Sit (fine sediment) is deposited and mud flots or salt marshes form.



Formation of Caves, Arches, Stacks and Stumps

- 1. Coresoccur when woves forcetheir woyinto crocksinth ecliff oc e Theworter contoins sand and other moterials that gind away at the rock until the crocks become a corre Hydraulic action is the predominant process.
- 2. If the correis for med in a head land, it may eventudly break through to the other side forming an arch.
- 3. The crch will grod udlybecome bigger until it c on no long er support the top of the orch. When the orch colloo ses, it leaves the headland on one side and a stack (a tall column of rock) on the other.
- 4. The slock will be dit ocked ditheb ose in the same wayth of a way ecol notchisformed. This weakens the structure and it ŵll eventually collapse to form a stump.

How does a river channel change as it travels downstream?

- The Bradshaw Model is used by geographers to describe how the characteristics vary between the upper course and lower course of a river.
- Key characteristics are that width, depth, velocity and discharge all increase further downstream whereas particle size and aradient of the channel decrease further downstream.

Causes of River Flooding

- Intense rainfall
- Impermeable rocks

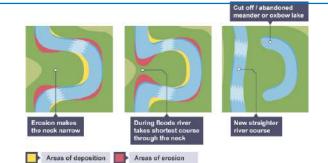
Strategies to Reduce River Flooding

Strategy Type	Strategy	Advantages and Disadvantages
Hard Engineering	Levees – high banks on/near riverbanks	+ Stop water spreading into areas where it could be problematic - Can burst under pressure
Hard Engineering	Channelisation – deepening or straightening the river	+ Allows water to run through channel more quickly - Water taken downstream may put other places at risk
Soft Engineering	Washlands – areas on floodplain allowed to flood	+ Give a safe place for floodwater to go - May limit the use of the land (e.g. for recreation)
Soft Engineering	Flood-plain Zoning – allocate areas to different uses	+ Prevents using high risk zones for businesses and housing - May cause accessibility issues for the public

Formation of a Meander

A stheiver makes its way to the middlecourse itg oins no rewoler and therefore no re energy. Lot erot erosionstartsto widenthe river. When the river flows over flatter land they develop large bends called meanders.

- 1. A sativer goes around abend, nots of the water ispushed to wards the outside This causes increased speed and therefore increased erosion (through hydraulic action and abrasion).
- 2. The lot erclierotion on the ootside bend causes under cotting of the bank to form a river cliff.
- 3. Weter on the inner bend is slower, causing the water to slow down and deposit the eroded material, creating a gentle slope.
- 4. The build-up of deposited sediment is known as a slip-off slope (or sometimes river beach).



Formation of an Oxbow Lake

- 1. Due to erosion on the ostist deofaberd and deposition on the inside, the shape of a meander will change over a period of time.
- 2. Exp sign norrows the neck of the land within the meander and as the process continues, the meanders move closer together.
- 3. When there is avery high discharge (usually duing aflood) theriver cuts across the neck, taking a new, straighter and shorter route.
- 4. Deposition will occur to cut off the original meander, leaving a horseshoe-shaped oxbow lake.

Formation of a Waterfall

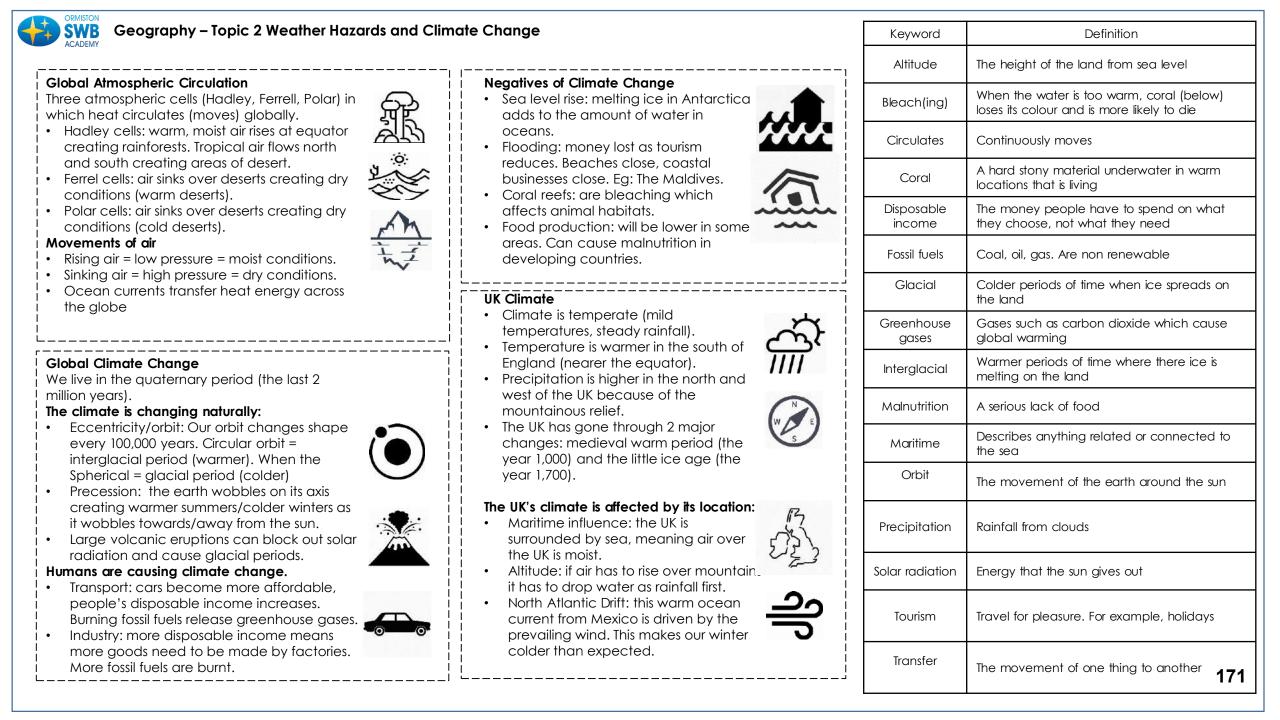
W drefdlsform when there creho izonlob card so fress and nock (hord rock) positioned over exposed, less resistant rock (soft rock).

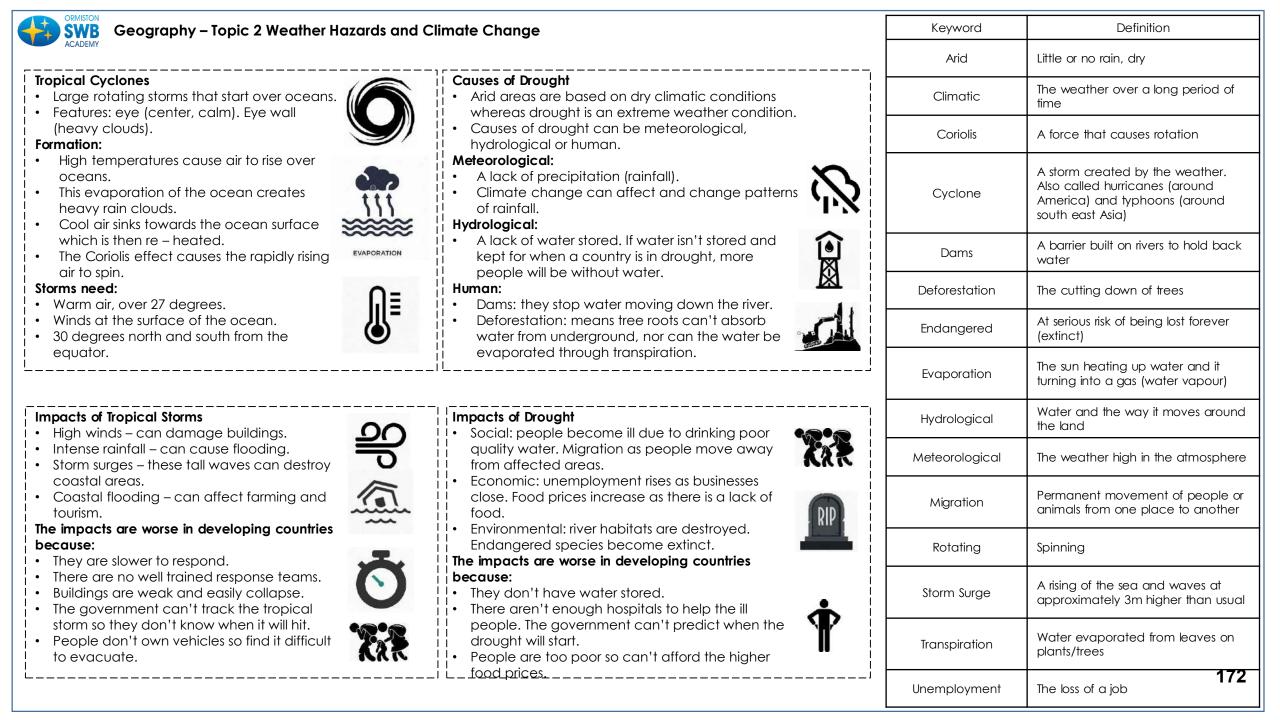
- 1. The soft lock is evoded quicker than the hard lock and this creates a step.
- 2. As erosion continues, the hard rock is undercut forming an overhang.
- 3. Abrasion and hydraulic action erode to create a plunge pool.
- 4. Over time thisg desbigg ex, increasing the size of the oveh ang until the hard rock is no longer supported and it collapses.
- 5. This process continues and the waterfall retreats upstream.
- 6. A steep sid set volley is left where the wonterfollonce was. This 750 called a gorge.

Downstream Upstream Discharge oad quant

- Long duration of rainfall
- Steeper relief of land Deforestation
 - Urbanisation









Geography – Topic 3 Ecosystems. Biodiversity & Management

The World's Ecosystems

- Biomes are large ecosystems spread across the world.
- Each biome has a different climate and type of vegetation.

Examples of Biomes

- Tropical Rainforest: hot & wet all year, tall trees.
- Deserts: very hot all year, very dry all year, plants have deep routes to find water.
- Tundra: freezing temperatures for most of the vear, low precipitation. Very few plants arow.

The Biosphere (The Earth's Surface)

- Humans use the biosphere for water, for fossil fuels and minerals like gold, silver and metal.
- Humans have exploited the biosphere.

The Nutrient Cycle

- Nutrients move around the biosphere.
- Nutrients move between soil, biomass and litter.



- As animals die the nutrients fall into the litter store. As the animal body decomposes the nutrients move into the soil. As plants take nutrients from the soil the nutrients move back into the biomass (it's a cycle).
- The climate affects how quickly this cycle

UK Ecosystems

- UK terrestrial ecosystems are: woodland (deciduous trees), moorland (upland areas), wetlands (areas around rivers and lakes), heathland (sandy soil, can't be farmed).
- UK marine ecosystems are used for tourism and leisure activities.

The Benefits of UK Ecosystems

Bring in £3bn into the economy and provide 200,000+ jobs.

Marine Ecosystems – Human Exploitation

- Chemicals used by farmers are being washed into rivers/lakes/seas.
- Large windfarms change bird migration routes.



- High temperatures & rainfall = high biodiversity (lots of plant and animal species).
- Trees grow tall but have shallow routes because only the top layer of the soil is fertile.
- Nutrient cycle: biomass is the biggest store of nutrients. Small litter store because of decomposition.

Tropical Rainforests – Importance

- They provide goods: timber, oxygen, medicine.
- They provide services: store carbon dioxide, provide animal habitats, bring in tourists.

Adaptations To The Rainforest

- Buttress roots think roots that spread across the soil to give tallest trees stability.
- Poison dart frog its poison skin helps protect it from predators.

Tropical Rainforests Are Chanaina

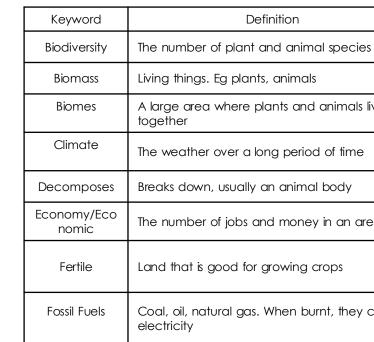
- Climate change: creating drier conditions which is increasing the risk of forest fires.
- Climate change: drier conditions are reducing biodiversity and causing animals to migrate away.

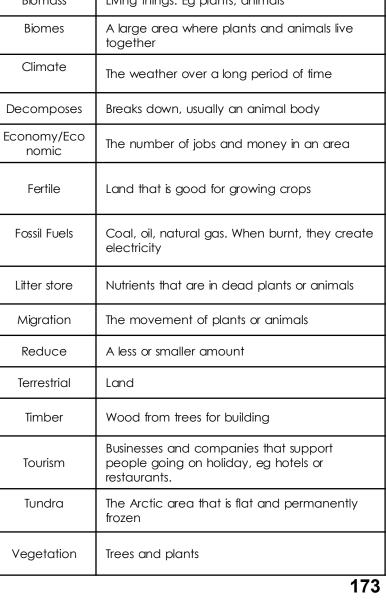
Deforestation

- Trees are cut down for social and economic reasons.
- Social: population is increasing which means we need more space for housing and farming (growing food).
- Economic: more money can be made from farming animals, trees need to be removed so we humans can extract minerals.

Deforestation Affects Animals

- Habitats are destroyed which might cause extinction of plant and animal species.
- Food chains are affected which means the number of animals is affected.





Definition









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Geography – Topic 3 Ecosystems. Biodiversity & Management

D . C . . . Managing & Protecting The Rainforest **Temperate Deciduous Woodland - Importance** • Selective logging: only older trees are cut They provide goods. • Timber used to build with. down to allow younger trees to grow taller. Fuel: some UK power stations now burn wood Aaribusiness: growing crops in between the create electricity. trees of the rainforest. They also provide services. Reforestation/afforestation: re – planting Recreation: woodland is popular for dog walking, trees. cycling, horse riding. Governmental policies: where the government Carbon capture: trees absorb and store carbon. encourage the protection of the rainforest. This helps stop alobal warmina. Conservation: forest areas are protected by law and protect animal/plant species. **Temperate Deciduous Woodland Are Changing** Case Study: Amazon Rainforest Climate change: if winters become milder • The Amazon rainforest, Brazil, is being (slightly warmer) then diseases amongst trees cut down for: logging, minerals, building can spread. roads, houses and farming. Climate change: climate becomes drier = The rainforest is being protected by: more droughts and forest fires. • Agribusiness: farmers are encouraged to Deforestation of the Woodland plant crops amongst the trees. • Space is needed for people moving into Education: local people are being cities (urbanisation). educated about why the rainforest • The timber from trees is used for building needs protecting. houses etc. • Afforestation: trees are being replanted. Space is needed for land for farming. Temperate Deciduous Woodland Case Study: The New Forest in the south of Climate: 4 seasons. Cool summers. Mild Enaland winters. Rainfall all year. • Tourism in the area brings in £500m. • High biodiversity – lots of plants and animal • The 15m visitors damage the area by: species. Walking, cycling, driving, horse riding. Mostly deciduous trees. Nutrient Cycle In The The Area is Protected and Managed **Deciduous Woodland** • Conifer trees cut down are replaced by • Summer: warm temperatures mean rapid deciduous trees. decomposition = soil full of nutrients. Chemicals are rarely used in The New Forest

• Winter: decomposition is slow = soil lacks nutrients.

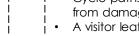
Deciduous Trees

- Drop leaves in the winter because of the lower temperature.
- Deep roots to find water and nutrients.

- The New Forest is used Sustainably
- Cycle paths and routes keep people away from damaged areas.
- A visitor leaflet called '5 ways to love the Forest' explains how people can protect the forest.

Keyword	Definition
Carbon	A gas produced when burning fossil fuels
Conifer	A tree that has needle like leaves
Conservation	Protect or look after, make last longer
Deciduous	A tree that drops its leaves every year
Decompositio n	Breaks down, usually an animal body
Drought	A severe lack of water and rainfall in an area
Logging	Trees being cut down
Policies	Something that is recommended by a government
Rapid	Very quick
Recreation	Activities that people do to have fun or relax
Seasons	Autumn, Winter, Spring, Summer. Changes in temperature and weather
Selective	Chosen, in this example chosen trees
Timber	Wood from trees for building
Tourism	Businesses and companies that support people going on holiday, eg hotels or restaurants.







Geography – Topic 4 Changing Cities – Mumbai, India (Major City in an Emerging Country)

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Mumbai's Location

- Coastal city in the west of India.
- Has an international airport with major connections.
- Good road + rail connections to all of India.

Mumbai's History

- Was a British colony called Bombay.
- City developed because of its textiles industry, coastal location and its port.
- The CBD is the historic center of the city.

Mumbai's Rapid Growth

- Population: 20m in 2014. Predicted to be 30m in 2021.
- Industry: now a global financial hub and home to Bollywood.
- Birth rate is declining as women work.
- Life expectancy now higher as healthcare improves.

Dharavi Slum (Informal Housing)

- Rapid growth has led to land being very expensive.
- Cost of housing is high in Mumbai led to 9m people living in slums. Im people live in Dharavi.
- Dharavi slum is Asia's 2nd largest slum.

Negatives of Dharavi

- Health issues. Diphtheria + typhoid spread.
- 1 toilet per 1, 4440 people.
- Poverty. Many live on less than \$1 a day. ٠

Positives of Dharavi

- Strong community, people help each other.
- Slum businesses create \$40m for local economy.
- 85% of people have a job in the slum.

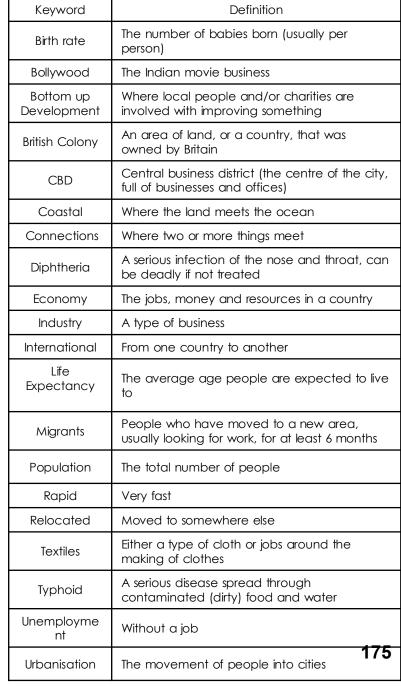


- Urbanisation brings in migrants, they move to the slums, eg Dharavi.
- Rapid increase in population has led to huge unemployment.
- Richer people live in gated communities because of the drug + crime problems in the city.













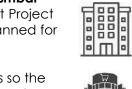
Improving the Quality of Life in Mumbai

- The Dharavi Slum Redevelopment Project
- \$2bn development project planned for the land Dharavi is built on.
- The people of Dharavi will be relocated to apartment blocks so the land can be used for gardens and shopping centers.
- The people do not want this as it will break up their communities. The people would prefer small

improvements.

- Bottom up development.
- Build areen outdoor spaces for local people.
- Build clean water pumps and toilet blocks.











Geography – Topic 4 Changing Cities – Birmingham, United Kingdom (Major City in a Developed Country)

Birmingham's Location History

- 1700s went from a small market town to creating industries in jewellery and guns.
- 1830s canals and railways were built. This connected Birmingham to the country.
- Was an industrial city secondary industry.



Housing – tower blocks and terraced housing in the inner city. Larger semi detached housing in the suburbs.

Birmingham – Changes in the 1900s

*Urbanisation: happened in the 1800s because of the jobs in the city.

1920s/30s

- Suburbanisation happened.
- Semi detached homes built with greenery.

1970s - 1990

- Counter urbanisation happened.
- People left the city for a quieter lifestyle in rural areas.
- People could afford cars so they could commute back into the city for work.

After 1990

- Re urbanisation happened.
- New apartments were built in the city center.
- Old factories were changed into apartments.

Population

- 1.1m people live in Birmingham.
- Birmingham is ethnically diverse. 42% from non – white backgrounds.



- Young people move into the city for work or for university (there are 78,000 students in Birmingham).
- 1950s migrants came from old colonies such as India + Pakistan to work in transport and steel.
- 2000 migrants came from eastern Europe to _work._____

Industry in Birmingham

- Secondary industries are now in decline.
- Factories are moving abroad because of cheaper wages, land and better global transport.
 - This has created unemployment and brownfield sites.

Deprivation in Birmingham

• Areas are deprived because of deindustrialisation. Eq: South West of the city.

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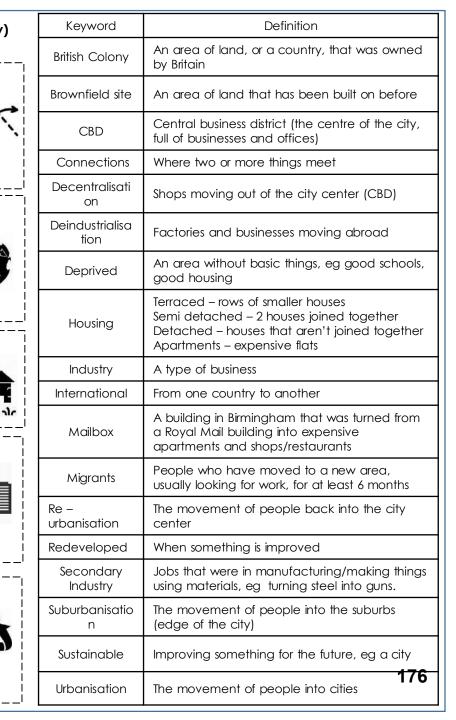
- · Sparkbrook: deprived area, high unemployment, poor housing, health issues.
- Sutton Coldfield: wealthy area.

Retail (Shopping) in Birmingham

- 1970s: decentralisation. Shops moved to the suburbs for space and cheaper land.
- 1980s onwards: CBD is redeveloped. The Mailbox has expensive shops and restaurants in.
- The Bullring shopping center is built in 2003 to attract people back into the city.

Birmingham – trying to become sustainable

- The city is recycling 30% of waste. This is increasing.
- Bus + cycle lanes are being built to reduce areenhouse emissions.
- More schools are being built to improve education.
- Homes are built more energy efficient with insulation

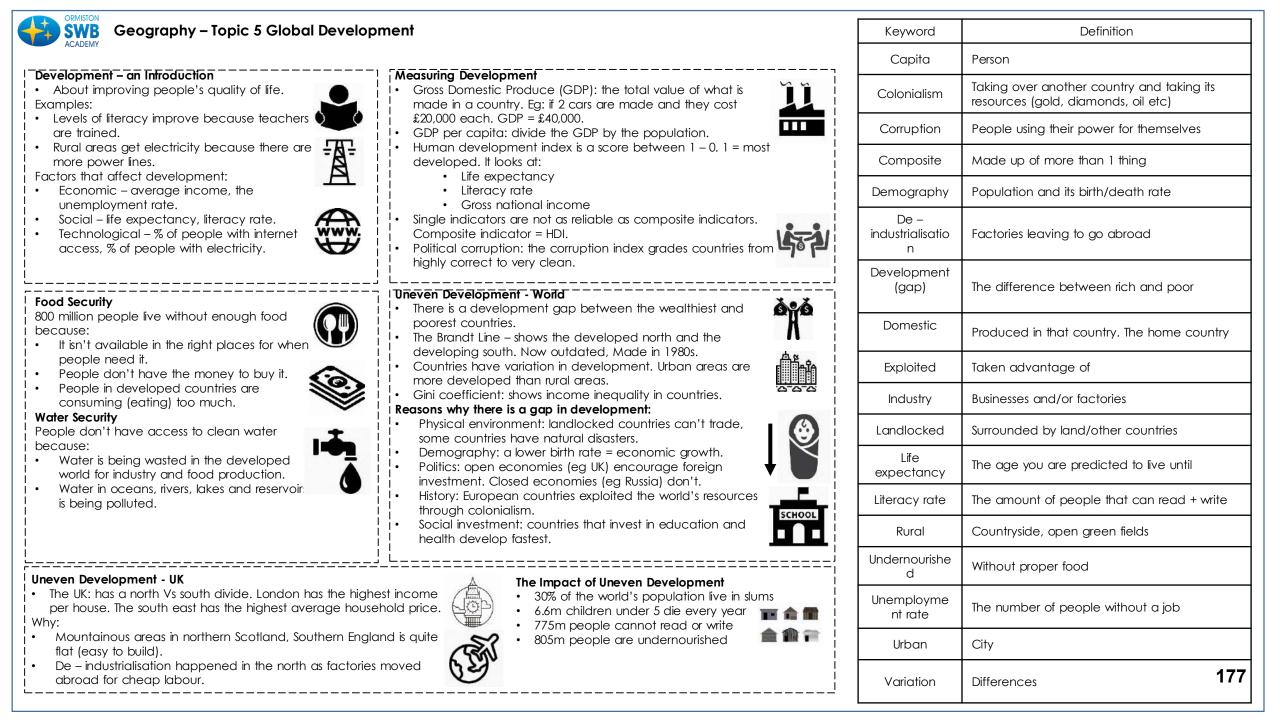


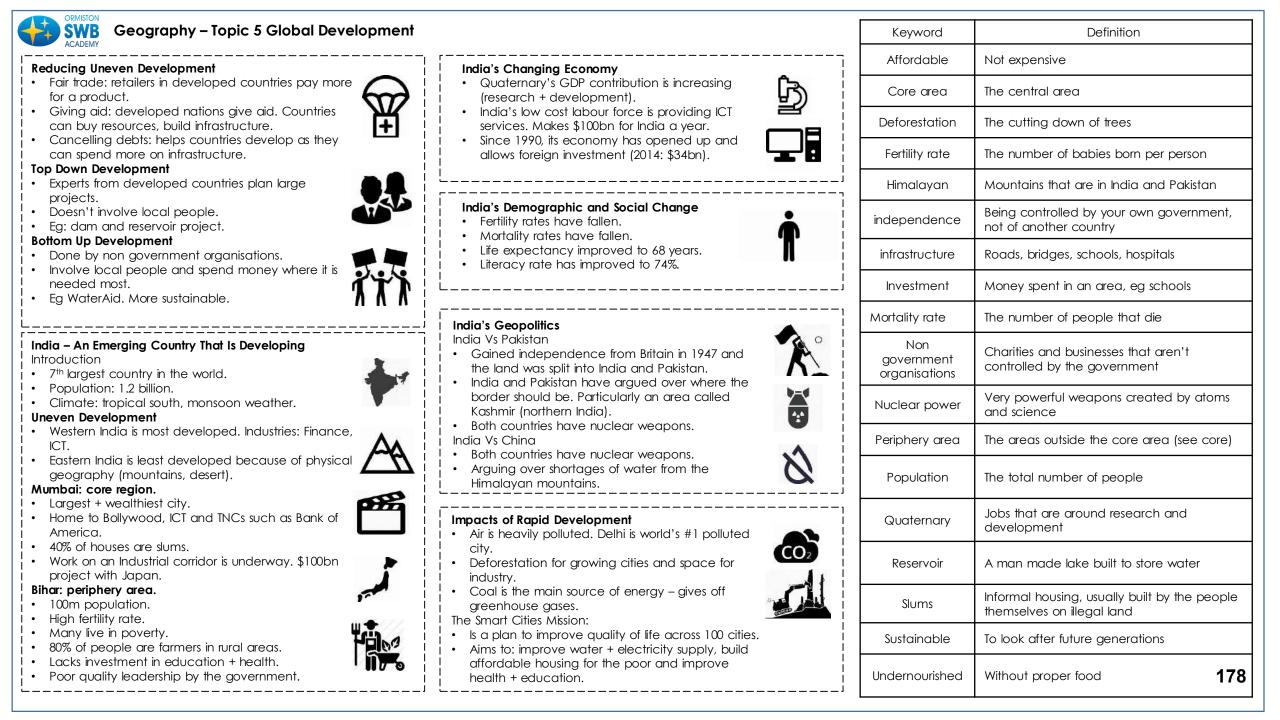


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Geography – Topic 6 Resource Management

Types of Resources

- Abiotic: found from things that can't reproduce. Eg: soil, water.
- Biotic: found from things that can reproduce. Eg animals and plants.
- Non renewable: resources that either cannot be remade or would take millions of years to make again. Eg: fossil fuels.
- Renewable: resources than can be used again and again or re-created in a short amount of time. Eg: wind, solar, hydro electric power.

The Location of Natural Resources Around the world

- Gold and diamonds are found near volcanoes.
- Fossil fuels are found where there is sedimentary rock.
- Oil is found in countries such as America, Iraq, United Arab Emirates.

In the UK

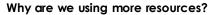
- Iron and coal helped Britain in the industrial revolution but there is little left in Britain.
- Oil and gas are found in the north sea but these supplies are running out.

Problems in the UK

- Rainfall is higher in the north and west of the UK but this isn't where people live.
- The water is in the wrong place. The supply of water cannot meet the demand.

Consumption of Resources Around the world

- People are using more resources everywhere in the world, but the biggest increase is in Asia.
- <u>_</u>
- America is eating up the most calories. This leads to undernourishment in other countries.



- People are richer and can afford more cars.
- The population is increasing.
- People are buying more technology.

All of this means we need to build/make more which takes up resources.

Where do we get our electricity from?

- In the past, we have always burnt fossil fuels (non renewable energy sources).
- Recently, we have started to use renewable energy sources.
- All energy sources have positives & negatives

The Energy Mix (where countries get there electricity from)

- 80% of the world's energy comes from non renewable sources.
- 60% of the UK's energy comes from non renewable sources.
- Factors that affect a country's energy mix: size of population, wealth of the country, what energy resources are nearby.

Coal – a fossil fuel

Advantages

- $\checkmark\,$ Should last for another 200 years.
- \checkmark Cheap and easy to mine.
- ✓ Creates large amounts of electricity.

Disadvantages

- Releases greenhouse gases, polluting the air.
- Destroys animal habitats.

Wind energy – renewable energy Advantages

- ✓ Does not pollute greenhouse gases.
- ✓ Creates cheap electricity for customers.

Disadvantages

- Can ruin the look of the landscape.
- Doesn't work when it isn't windy.
- Can be expensive to build.

	-
Keyword	Definition
Calories	Energy that comes from eating food
Expensive	Costs a lot of money
Fossil fuels	Coal, oil, natural gas. All non renewable
Greenhouse gases	Harmful gases that are released into the air. For example when burning fossil fuels
Habitats	Places where something lives
Hydro electric power	Electricity that is made from water powering machines
Industrial revolution	Happened in the 1700s where people had jobs in factories
Landscape	The land, the environment. Eg fields
Population	The total number of people in an area
Reproduce	Make more of
Sedimentary rock	Layers of rock that have fossils (dead plants/animals) in them
Undernrouish ment	People that do not have enough food to eat



















Geography – Topic 6 Resource Management

Nuclear Power Plants

Advantages

- ✓ Produce electricity all year round.
- ✓ Produces huge amounts of electricity.
- ✓ Produce less carbon dioxide than fossil fuels.

Disadvantages

- Expensive to build.
- Can be dangerous if they explode.

Frackina Natural Gas

- This is a new way of finding natural gas.
- Involves drilling down to shale rock.
- Blasting water, sand and chemicals into the rock.



This breaks the rock and allows the gas to come out.

Advantages

- \checkmark Produces large amounts of gas.
- \checkmark Has made natural gas cheaper to use/buy. $I \checkmark$ Produces less areenhouse gases than coal.

Disadvantages

- Can be dangerous as gas enters the water supply and can come through kitchen taps.
- The use of chemicals can damage animal habitats.

Managing & Protecting Our Energy Individuals can:

• Measure their carbon footprint.



- Use public transport like buses, not private transport like cars.
- Insulate homes and use solar panels.

Companies can:

- Restaurants can re use cooking oil as fuel for lorries.
- Use LED lights which use less electricity.

Governments can: • Use smart meters in homes.



 Build renewable energy sources. I • Improve renewable energy sources by improving technology.

Case Study: China, an emerging country

- 70% of its energy comes from burning coal.
- 7 cities in China are classed as 'heavily polluted'.

Why does China have pollution problems?

- More people are owning cars.
- China's roads are heavily congested.
- Poor quality petrol causes air pollution.

What has China done to solve these problems?

- 2006: China introduced the 'China Renewable Energy Law'. Which aims to minimise the use of coal and use more nuclear and wind power.
- China built the 'Three Gorges Dam' to create hydro electric power.
- Solar farms (rows of solar panels) have been built in the Gobi desert.

Case Study: Germany, a developed country

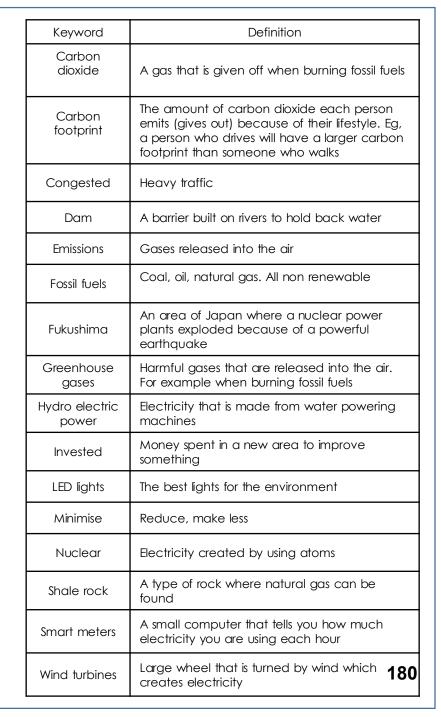
- Germany wants to use less nuclear power because of the Fukushima disaster in Japan, 2012.
- By the end of 2020 Germany plans to reduce greenhouse emissions by 40%, and by 80% in 2050.

How will it meet these targets?

- Germany has built the Bavaria solar park with 60,000 panels. This will reduce carbon emissions by 100,000 tonnes.
- Germany has invested in new wind turbines in the north sea, called Norsee Ost farm. This will increase how much energy is created.



















Geography – Topic 8 UK Challenges

The UK's Population

• Is increasing. 2020: 67m. 2050: 77m people.

This increase will put pressure on resources:

- More housing means more greenfield sites are needed to be built on.
- More food will mean natural habitats are destroyed to make space for farms.
- Î

• More energy might mean more greenhouse gases if fossil fuels are burnt.

To solve these problems:

- Use brownfield sites for building.
- Encourage people to limit their food intake and have a healthy lifestyle.



Use sustainable, renewable energy sources for electricity.

Transport in the UK

their cars at home.

• Global transport releases emits huge amounts of greenhouse gases.

The UK is trying to reduce greenhouse gases:

Improve public transport so people leave



- London has created cycle routes and Boris bikes for people to rent for the day or longer.
- Congestion charges: charging people for driving through city centers has reduced traffic and pollution.

Greenfield Sites

Advantages:

- Cheaper and quicker to build on.
- Environment is usually cleaner and more pleasant to look at.

Disadvantages:

- Valuable farms or open spaces are lost.
- Animal habitats are destroyed.



- Less countryside is lost.
- Old dis used urban areas are cleaned up.

Disadvantages:

- Often more expensive because old buildings need to be destroyed/rebuilt.
- Higher levels of pollution in these areas.

The UK's Economy

The 2 speed economy: London and the south east's economy grows faster than the north of England.

Why is this?

- More government money is invested in the south to attract more businesses.
- The big businesses in London attract the brightest people from the north of England. These people move south to work.

How is the government trying to stop the north Vs south divide?

- A high speed railway (HS2) is being built from London to northern cities. This will create jobs in the north.
- Airports in the north are improving to encourage travel and businesses to set up in the north.

Migration

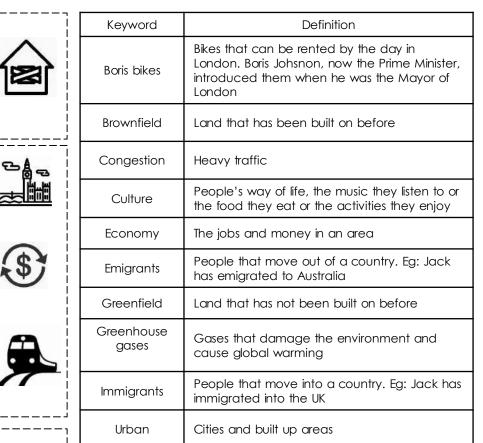
- Immigrants: mainly from India + China.
- Emigrants: mainly leave to Australia.
- Reasons for migration: better jobs, better education, family reasons or to retire.

Advantages of migration nationally:

- Brings more workers so more taxes are paid.
- Different cultures bring different music and
- food.

Disadvantages of migration locally:

- More people means we need more hospitals, schools and doctors. This costs money.
- Some people think migrants take jobs from British people.





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Geography – Topic 8 UK Challenges

UK's Landscape Challenges

- How can we protect national parks? Employ young people to work in the park so
- they grow up seeing its importance.
- Encourage public transport into the national park.

The UK is at risk from river flooding because:

- The population is increasing so more people have to live nearer rivers.
- More urban areas means more impermeable surfaces.
- More extreme rainfall because of climate change.

The UK government is planning to:

- Reduce the number of buildings near rivers.
- Build flood defences hard/soft engineering.
- Help people prepare through warnings.

Specific example: Somerset floods, 2013/14.

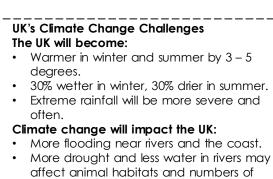
- Heavy rainfall flooded 600 homes.
- Flood defences failed to protect people.
- Rivers were blocked with silt which meant rivers couldn't hold much water.

UK's Coastal Challenges

- Storm surges cause severe flooding because of the wave height and wind power.
- Storm surges can destroy some coastal defences.

Reducing coastal flooding in the UK:

- Build sea walls to reduce flooding and erosion.
- Use managed retreat in certain areas.
- Specific example: 2013, across east England.
- The Thames flood barrier protected London from floodina.
- Flood barriers were quickly built in Norfolk.
- Advice was passed out through social media which meant that 800,000 homes were protected.



More heatwaves could mean more illnesses like heatstroke. This puts pressure on the NHS.

Responding to climate change Individual people can:

- Walk or cycle to reduce greenhouse
- emissions.

species.

- Recycle waste to reduce resource consumption.
- Build solar panels and insulate homes to conserve heat.

Governments can:

- Place limits on carbon emissions from businesses.
- Create adverts which encourage recycling.
- Sign agreements like the Paris agreement to work with other governments.

Problems aovernments face:

- Encouraging economic growth and development usually means using more resources.
- Reducing climate change can damage economic growth.

Keyword	Definition	
Drought	Little or no rainfall for a long time, usually months or years	
Economic growth An increase in the number of jobs in an area and an increase in the number of money spent in an area		
Emissions	When gases are released into the atmosphere (air)	
Extreme rainfall	Very heavy rain	
Flood defences	Things that aim to try and stop rivers from flooding	
Greenhouse gases	Gases that damage the environment and cause global warming	
Heatwaves	Very hot weather for a few days or possibly weeks	
Heatstroke	An illness that makes the body weak. Is caused by being in hot temperatures for too long without drinking enough water	
Impermeable surfaces	Building materials that don't allow water to pass through them. Eg concrete, tarmac	
Insulate	Keep warm	
Managed retreat	Allowing certain parts of the land to become flooded	
National parks	Parks that cannot be built on. They are protected by laws	
Severe	Very dangerous	
Storm surge	This is when the wind 'picks up' the waves and makes them higher. About 6 feet higher than usual	







Thames flood barrier

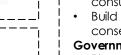




A barrier on the Thames river that protects

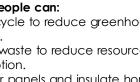
London from flooding





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Year 10 – History – Early Elizabethan England – Elizabethan society in the Age of Exploration 1558-1588

Education and Leisure

 Education expanded during Elizabeth I's reign – however it was mostly boys who received an education



- Most people in the Elizabethan times were illiterate
- There was not much difference in the classroom education of airls and boys but boys were more active outside of the classroom
- Every town had a grammar school by 1577
- Elizabethan past times were similar to modern ones but sport was more violent
- The theatre was very popular and appealed to all classes of people many new theatres were built
- Protestantism led to many new plays being written

Exploration and vovaaes of discovery

- Trade was the driving force behind exploration
- Another reason was to challenge Spain's position in the New World
- New technology made it possible to undertake longer journeys and increase accuracy of maps and navigation
- The printing press enabled the reproduction of maps, navigation manuals and accounts of exploration
- Drake's circumnavigation of the globe started as a mission to attack Spain and eventually led people to invest in exploration
- Nova Albion encouraged the English to attempt further colonisation of North America













The problem of the poor

- Poverty and vagabondage were seen as a growing problem in Elizabethan Enaland
- The poor were divided into 'idle', 'deserving' and 'impotent'
- Population increase, laws about land (enclosure), disruption to trade and inflation made the issue of poverty worse
- Attitudes changed when unemployment was seen as a genuine issue
- Elizabeth I passed laws to help the poor
- One involved giving people raw materials so they could make goods and sell them
- Vagabonds faced harsh punishment but these were rarely enforced
- There were local initiatives to help the poor too -e.g. lpswich









Raleiah and Virainia

- Walter Raleigh attempted to set up a colony in Virginia twice
- These attempts were a failure due to inexperience, the suitability of the colonists and the relationship with the Native Indians
- Conditions in Virginia were harsher than colonists expected
- Many of the 1585 colonists did not co-operate with each other
- The English were very dependent on the Native Indians in Virginia in order to survive
- The local chief Winging did not trust the English and became hostile









Year 10 – History – Early Elizabethan England – Elizabethan society in the Age of Exploration 1558-1588

Key Word	Definition	
Colony	Area of land owned by another country where people live	
Settlement	Living and establishing a community where people haven't typically lived before	
Exploration	Finding and discovering new areas	
Vagabondage	Vagrancy, homelessness and wandering without purpose	
Bear Baiting	Fight between a chained Bear and dogs	
Illiterate	Unable to read or write	
Protestantism	The religion of the Church of England that Elizabeth was in charge of	
Poverty	The state of being extremely poor	
Impotent Poor	Helpless, unable to work	
Idle Poor	Those seen as able to work but chose not to	
Deserving Poor	Those whose poverty was not their fault and wanted to work	
Circumnavigation	The process of sailing all around something – usually the world	
Colonist	Someone who lives in a colony	18



Year 11 – History – Weimar and Nazi Germany 1918-1939 – The Weimar Republic 1918-1929

The origins of the republic 1918-1919

- The Kaiser abdicated on 9th November 1918, the war ended 2 days later
- War had not been kind to Germany the SPD (social democratic party) had to work hard to establish order
- Despite revolts and riots, Ebert and the SPD established a new government
- The National Assembly met in Weimar and created a constitution for the Weimar Republic
- The structure meant no single party could hold all the power
- But it also made making decisions hard and divided people

The recovery of the republic 1923-1924

- Gustav Stresemann was appointed Foreign Secretary in 1923
- He introduced a new currency (the Rentenmark) in 1923 which ended hyperinflation
- The Dawes Plan (1924) and Young Plan (1929) reduced the strain of reparations
- The Locarno Treaties (1925) secured Germany's Western borders
- The Kellogg Briand Pact (1928) and Germany's joining of the League of Nations (1926) helped Germany's reputation to recover
- Not all economic problems were solved by 1929 when Stresemann died



Early challenges to the republic 1919-1923

- The Treaty of Versailles (1919) and the 'stab in the back' myth made the Weimar Republic unpopular
- The T of V included terms such as: a reduced army, reparations, war guilt and a loss of land
- From 1919 to 1923 the Weimar Republic suffered opposition from the extreme Left (Spartacist Uprising and the Communists) and extreme Right (Kapp Putsch and the Freikorps)
- 1923 was a crisis year for 3 reasons: hyperinflation, the occupation of the Ruhr and the Munich Putsch





Changes in society 1924-1929

- Some improvements in the standard of living improved in areas such as unemployment, wages, housing and women
- There were some improvements in the position of women in politics, work and leisure – not everyone was happy about this!
- Dramatic changes occurred in culture particularly art, cinema and architecture – this also did not please some Germans









Key Word	Definition
Coalition government	Two or more parties because neither party has a majority in Parliament
Social Democratic Party (SDP)	The main left-wing party, supported mainly by the working class
Constitution	Ideas for how a country is governed
Proportional Representation	Parties gain seats in proportion to the number of votes cast for them
Spartacist Uprising 1919	Left wing revolt by radical SDLP members, believed in communist ideas and tried to overthrow the Weimar government in 1919.
Treaty of Versailles 1919	Document signed by the countries involved in WWI. It imposed certain conditions on Germany such as the Reparations.
Kapp Putsch 1920	Right wing uprising. to seize Berlin and try to set up a new right-wing government. The plan failed.
Reparations	Money to be paid by Germany agreed by the Treaty of Versailles for war damage
War Guilt	Germany having to accept full responsibility for the war
Occupation of the Ruhr	France sent troops in the Ruhr (border between the two countries), which was the main industrial area of Germany.
Hyperinflation	Very extreme high inflation which makes money and currency worth less
Kellogg Briand Pact 1928	Germany and 64 other nations signed this deal to agree that their armies would be used for 'self-defence' and that international disputes would be sorted peacefully.
Locarno Pact 1925	Signed between Germany, Britain, France, Belgium and Italy. Agreed to keep the existing borders between Germany and France/Belgium to help both countries feel secure.
Dawes Plan 1924	New financial deal to help Germany. It promised American loans and lowered the amount of Reparations being paid
The Young Plan 1929	A new agreement that reduced the amount of reparations Germany had to pay and agreed that they could pay them 18 a longer period of time.

Year 11 – History – Weimar and Nazi Germany 1918-1939 – Nazi Control and Dictatorship 1933-1939

The creation of a dictatorship 1933-1934

- The Reichstag Fire (February 1933) gave Hitler the opportunity to form a dictatorship
- After the fire, there were attacks on Communists
- The Nazi party gained more seats in the Reichstag
- The Enabling Act (March 1933) gave more power to Hitler he could pass laws without the support of the Reichstag
- The Night of the Long Knives (July 1934) saw many of the SA killed, enabled Hitler to eliminate threats and strengthen his power
- After Hindenburg's death in August 1934, Hitler became the official leader of Germany

Controlling and influencing attitudes

- In Nazi Germany, propaganda and censorship were used to influence peoples opinions
- Joseph Goebbels was the head of the Ministry of People's Enlightenment and Propaganda and organised propaganda
- Nazi control and influence was exerted using the media, rallies and sport including the Olympic Games of 1935
- The Nazis also exerted control over the Arts, including literature, art and film









The police state

- Nazi Germany was a police state controlled by the SS, SD and Gestapo
- From 1933 concentration camps were used to deal with 'undesirables' such as political opponents to Nazism
- The legal system was Nazified law courts and judges were placed under Nazi control
- Religion was closely controlled although both Catholics and Protestants resisted this

Opposition, resistance and conformity

- Most Germans supported or at least conformed to Nazi practices and beliefs
- Resistance was limited due to propaganda and the police state
- Also, the Nazis had improved foreign policy and employment
- Opposition came from youth groups, the church, trade unions, the army and political opposition groups
- Some young people set up alternative youth groups to those approved by the Nazis
- Very few people were brave enough to openly oppose the Nazis and many just grumbled privately about their views











Key Word	Definition
SD	Security Service, set up in 1931 to gather intelligence of potential enemies
Police State	A dictatorship that has absolute control by secretly monitoring the belief and activities of its people and taking act
Concentration Camp	Prison for political prisoners and Nazi enemies, placed there without trial. The first one was opened in 1933.
The SA	The Sturmabteilung, often called Brownshirts. Were protection squads/private Nazi party army
Gestapo	Nazi official secret police
Enabling Act	1933 Reichstag law gave Hitler and his government full power. He would have more powers than the president under the law.
Night of the Long Knives	This was where Hitler's political and military rivals in the SA were removed on 30th June 1934.
SS	(Schutzstaffel). Hitler's personal bodyguard
Decree for Protection of People and State	Law passed after the Reichstag Fire, due to state of emergency. It stopped basic civil rights
Concordat	Nazi agreement with the Pope. The Pope agreed to stay out of politics if Hitler agreed to stay out of the Catholic Church.
Propaganda	Spreading of false or exaggerated information normally by a government to increase support for them
The Edelweiss Pirates	Often young working class opposition. Listened to banned swing music, did anti-Nazi graffiti, created no go Hitler Youth areas
Swing Youth	Middle class opposition. They listened to swing music, men grew their hair long, women wore make up in protest at Nazi
Totalitarian State	Country where government controls all areas.
indoctrinate	To teach a person to accept a set of beliefs without questioning them
Aryan	'Pure' Germans, with no Jewish family.
Rallies	Large meetings of people, usually to show support for Hitler and the Nazis.
Censorship	Not including information that may make someone/something seem unpopular and only allowing a certain viewpoint to be share 1

Year 11 – History – Weimar and Nazi Germany 1918-1939 – Life in Nazi Germany 1933-1939

Nazi policies towards women

- The Nazis believed women should adopt a traditional appearance, leave professional jobs to men and get married and have children
- The Nazis used propaganda to convince women of this
- Nazis laws included the Law for the Encouragement of Marriage, changes to divorce laws, Lebensborn and the Mothers Cross
- Women were banned from certain jobs and discouraged from going to University
- Nazi policies towards women had some impact but not as much as they had hoped

Employment and living standards

- Reducing unemployment was a priority for Hitler and the Nazis
- The Nazis used a range of strategies to do this: The National Labour Service, building autobahns and rearmament
- As well as official unemployment levels, Nazi Germany also had 'invisible unemployment'
- People experienced many changes under the Nazis, these included: unemployment, wages, prices and the use of luxuries
- The Labour Front, including Strength through Joy and the Beauty of Labour also affected living standards



Nazi policies towards the young

- The Nazis believed the youth should be brought up to be useful to Germany and support Nazi ideas
- They believed boys and girls should be brought up differently
- The Hitler Youth (boys) and the League of German Maidens (girls) were set up to reinforce Nazi ideas in young people's 'free time'
- Schools in Nazi Germany were organised to create useful German adults and Nazi supporters
- The Nazis shaped the development of young Germans by controlling teachers and the curriculum



The persecution of minorities

- Treatment of minority groups was shaped by eugenics, racial hygiene and anti-Semitism
- The Slavs, gypsies, homosexuals and people with disabilities were all mistreated
- Nazi persecution began in 1933, became worse in 1935 with the Nuremburg Laws and became worse still in 1939 after the events of Kristallnacht
- The Final solution was not enforced until 1942, when Germany was involved in WW2

















Key Word	Definition	
Lebensborn	'Fountain of life' programme. unmarried pure women could donate a baby to the Fuhrer by becoming pregnant by a SS man.	
Law for the Encouragement of Marriage 1933	Law gave couples a loan when they married if the wife quit work. Could keep more of loan if had more kids	
The Mothers Cross	An award given to mom's for the amount of children they had	
Anti-Semitism	Hatred & persecution of the Jews	
Reich Labour Service (RAD)	Scheme to give jobs/manual labour to young unemployed men Rearmament: Providing the German armed forces with weapons and military equipment	
Volksgemeinschaft	ksgemeinschaft Creating of a people's community where the strongest races e.g. Aryan would dominate the weakest races in Germa	
Strength through Joy	Aim was to improve Germans leisure time by organising leisure/cultural activities for cheap	
Labour Front (DAF)	DAF was set up to replace Trade Unions and control German workers	
Nuremburg Laws 1935	Racial laws. Consisted of Reich Citizenship Law & Law for the Protection of German Blood and Honour	
Kristallnacht 1938	The smashing of Jewish windows: businesses, homes and synagogues	
Concentration Camps	Prison for political prisoners, Nazi enemies and 'undesirables' like Jews placed there without trial.	
Ghetto	A slum part of a city where a particular group of people live.	
Master Race/Ubermenschen	'Pure' Germans. Pure race (often blue eyed, blonde, tall and athletic)	
Subhuman/Untermenschen	All non-pure German groups. They were treated as second class citizens who had 'contaminated' the master race. Jews were an example.	

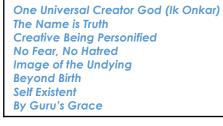


Sikh Beliefs: Part 1 – Key Beliefs



The Nature of God: The Mool Mantra

- Opening words of the Guru Granth Sahib (GGS 1a)
- 'Mool Mantra' means 'Main Chant'.
- Sikhs believe the words were the first teachings of Guru Nanak after he became enlightened
- Most important part of the Guru Granth Sahib (GGS) and most important statement for Sikhs
- Said daily in Sikh prayers and recited in worship.



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God as Creator

- God (Waheguru) created everything.
- There are no creation stories in Sikhism, and Sikhs accept scientific views about how the universe came to be here, but nothing would have happened without it being God's will (hukam).
- Sikhs believe God is both separate from and part of His creation: 'He possesses all qualities; He transcends all qualities'

God as separate from the Universe

- God is transcendent beyond human understanding
- He does not have a physical form, is timeless and spaceless, and has no limits
- God is without gender (Sikhs use 'He' to have a simple way to talk about God) and has no beginning or end.
- 'Nirgun' without qualities or form.
- 'He is the Perfect Transcendent Lord, from the very beginning and through the ages'

He transcends all qualities'					
	God shown in and through the				
	<u>Universe</u>				
-	God is present within creation				
	and within human beings, as				
	a soul or Divine Spirit				
-	There are ways that God can				
	be understood by humans,				
	e.g. through the Mool Mantra,				
	the teachings of the Gurus,				
	other parts of the GGS or				
	through his creation.				
-	'Sargun' – with qualities or				
	form				

The Lord is seen to be manifest and present'
He Himself is the water... He Himself abides in each and every heart'

The Virtues				
 God has given people an opportunity to reunite with Him (Mukti) To achieve this, Sikhs must build good karma in the hope of being released from the cycle of birth, death and rebirth One way to build good karma is to live a good life, developing certain positive characteristics known as virtues. 				
Truth and Truthful Living	Telling the truth, living an honest life. Includes promoting justice and not discriminating. 'Truth is higher than everything; but higher still is truthful living'			
Compassion and Patience	Being kind and aware of the needs of others. Being able to accept/put up with delays/problems with a calm mind and attitude. 'Show kindness and mercy to all life'.			
	Not being greedy, being satisfied with what you have, maintaining detachment from material things.			
Humility St	To be humble, not proud. Not full of your own importance.			
Love	To show a loving attitude to everyone, to show kindness, respect and forgiveness (just as God would do for them)			
Wisdom	Having experience, knowledge and good judgement – understanding all of the virtues and being able to put them into practice.			
Courage	Being brave. Many Sikhs throughout history have shown bravery in remaining true to their faith, even if they have suffered for it.			
Temperance/ Self-Control →	Showing self-control and moderation, can include not partaking of alcohol or drugs. Being able to control one's temper and behaviour.			
	Working to make all things fair, or to bring equality.			
	Gurmukh and Manmukh			
 Gurmukh: God centered. Someone who prays, worships, follows the virtues etc and keeps God in mind at all times. 'The Gurmukh acts in harmony with God's will; the Gurmukh finds perfection' 				

Manmukh: Man centered. Someone who is selfish, thinks they are above God and others, succumbs to the evils. 'The foolish, self-willed manmukh is blind in the world'.

	Beliefs about Life After Death		
Rebirth	 Reincarnation: when a human dies, their soul is reborn into another body This rebirth is part of a cycle of being born, dying and reborn, known as samsara. The cycle will repeat until the soul is freed/ liberated and becomes united with God All animals, including humans, have souls, so a human may be reborn as an animal. 'They die and die, over and over again, only to be reborn, over and over again' 		
Karma	 Sum total of a person's actions and words which determines their afterlife Reincarnation is based on the good or bad karma they built in a previous life. A human is the best being you can be reborn into as it gives you the best chance to build good karma and be liberated from samsara. Good actions = good karma = a good reincarnation/ liberation from samsara Bad action = bad karma = a lower reincarnation e.g. animal. 'The body is the field of karma in this age; whatever you plant, you shall harvest' 		
Mukti	 Liberation, freedom and release from the cycle of samsara The final goal for Sikhs - individual soul reunites with God Negative aspects: To achieve mukti, a person must rid themselves of all that stands in the way of getting close to God. This can be challenging. Positive aspects: the soul is free to unite with God. This is indescribable and can only be experienced. 'Through selfless service, eternal peace is obtained' 		
Exam Terminology Influence: The capacity to have an effect on people's character, behaviour or actions Contrasting: To show a difference Contemporary: Occurring in the present time Sacred Writings: Writing that is believed to contain words of God e.g. The Guru Granth Sahib Evaluate: Consideration of different viewpoints before arriving at a final judgement			

Justified Conclusion: A final decision which is based upon a range of evidence.



Sikh Beliefs: Part 2 – Key Beliefs/ Beliefs about the Nature of Life

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	Key Beliefs Continued	The Oneness	s of Humanity	Sewa: Selfl	ess Service
- Stages a hum	The Stages of Liberation: The 5 Khands nan being must pass thorough on the way to mukti. of all happen in one lifetime	God.Following this experience, he taught God; there is no need to convert ot	During that time, he said he met with that there was not only one way to	anythir - It will build good karma a achiev - Through selfless service,	b help others without expecting ng in return nd help a Sikh on the path to ving mukti , eternal peace is obtained'
1) Piety	- The opportunity for devotion to God, awareness of God.	 follow our own path to God. Everyone has a divine spark within them which unites us all, and this is known as the oneness of humanity. 		It helps Sikhs to show many of the virtues whilst avoidir 5 evils. Tan (Physical Sewa) Man (Mental Sewa	
2) Knowledge	- Knowing about God; learning about and experiencing God	 God is neither Hindu nor Muslim and 'We are all sons of the one God; the 'The Divine Light is within all' 	re is no Hindu and no Muslim'	Using the body to help others e.g. serving in the langar, cleaning shoes or floors	Using the mind and mental skills e.g. reading the GGS, teaching others, inspiring
3) Effort	- Devoting oneself to tuning in with God e.g. through prayer, worship, meditation	 This also means that everyone is equipment important teachings. 	al – this was one of Guru Nanak's most	 	others
4) Grace	- Spiritual blessing given by God (as we can only go so far in developing ourselves)		ality of All sm in the following ways:	Using material wealth to help o income to the sangat or to o Dasve	
5) Truth	- Finding God, the realisation of God. Can only be experienced, not described.	<u>The life of Guru Nanak</u> - Had both Hindu and Muslim friends.	The life of Guru Gobind Singh - Started the Khalsa – both men and	The Sangat: Sikh Re	
The Barriers to Mukti - Sikhs must avoid those things which will stop them from achieving mukti. There are 5 evils (below) bt Sikhs should also guard against: - Haumai (pride and ego) Illusion (inability to see the truth; focus on material things) - Self-centredness (ego, selfishness) Anger An emotion causing someone to act without balance 		 His best friend was a Muslim man called Mardana Emphasised equality between men and women: 'From her, kings are born without woman, there would be no one at all' Taught: There is no Hindu and no Muslim' Introduced the practice of the langar: 'No discrimination must be made while making people sit in rows for eating' 	 women can join. When the first 5 members joined, they wore identical coloured robes to show equality. One key role of the Khalsa is to stand up against inequality Introduced surnames Singh (Lion) and Kaur (princess) to remove inequality shown by the caste system 	Granth Sahib - 'Sat Sangat' means 'True Cc - 'Join the Sat Sangat, the True Lord' - Sikhs may gather together to read the GGS etc - Importance: provides oppor chance to learn from other s	e Congregation, and find the o learn, pray, hold a ceremony, tunities for sewa, gives the
Lust Greed	anger are broken, like a jar of poison	Ine Guru Granth Sahib - The GGS is a collection of hymns and writings from many teachers and saints e.g. the Gurus. - Writers also included Hindus and Muslims, showing the inclusivity of	Sikhism Today - The Langer: free kitchen where everyone is welcome. All sit on the floor together to show all are equal. Food is vegetarian so everyone can eat it.	 Amritdhari Sikhs are those where the second secon	Sahajdhari Sikhs no have been initiated into the daily prayers, wear the 5Ks, take ractise the virtues, be ode of conduct (which includes
Worldly Attachment I	 Placing too much emphasis on material possessions and worldly relationships False pride – being proud of things that were given rather than achieved: 'Why do you take pride in trivial matters?' 	Sikhism. - Guru Gobind Singh declared the GGS 'The Living Guru' and is the 11 th and final teacher for Sikhs. It contains many teachings about equality: 'All beings and creatures are His; He belongs to all'. 'All are made of the same clay'.	 Both men and women take part in worship, reading GGS in the Gurdwara, cooking or serving food etc. 	rules such as no smoking, dri - 5 Ks: Kara (steel bracelet), Ki Kachera (cotton underwear (uncut hair). Amritdhari Sikhs Sahajdhari Sikhs may choose	nking or adultery) irpan (ceremonial sword),), Kanga (wooden comb), Kesh must wear all of the 5 Ks. to wear some. to wear some. the halspage

Sahajdhari Sikhs have not been initiated into the Khals whilst they believe in Waheguru and the Gurus, they do not have to follow the strict rules.



Sikh Practices: Part 1 – Worship and Service



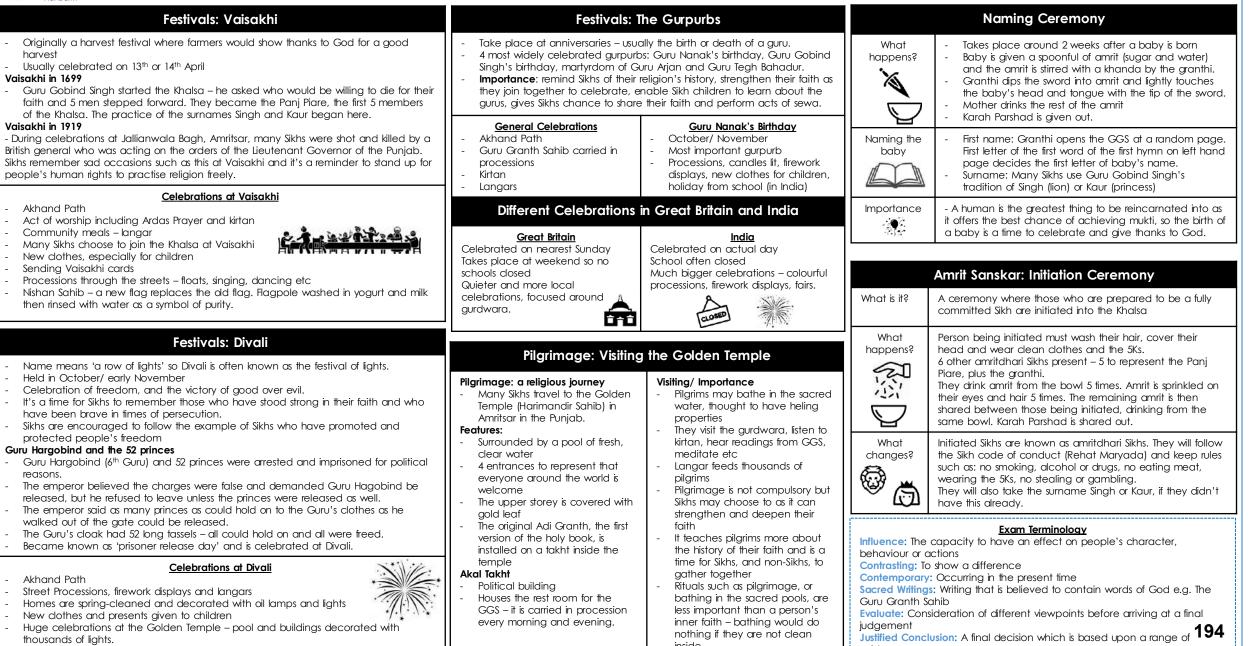
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	The Gurdwara: Religious Features	The Role of Prayer in the Home	The Role and Importance of the Akhand Path		
 The gurdwara is the Sikh place of worship. Translates to 'Door of the Guru' Technically a gurdwara is any place in which the Guru Granth Sahib is installed and treated with proper respect The gurdwara is open to all, no matter their age, race, religion etc. Outside the Gurdwara: Many have a dome and decorations on the outside 		 Sikhs are expected to remember God at all times which includes reciting daily prayers at home. Some Sikhs have a copy of the GGS at home but many have a gutka instead, a prayer book, which is treated with as much respect as the GGS (ideally has its own room, or section of a room, and is kept in a clean doth). 	 What A continuous reading of the Guru Granth Sahib from start to finish – all 1430 pages. Takes approx. 48 hours – male and female Sikhs take shifts to complete it. Karah Parshad is given out at the beginning and the end as a blessing 		
- All will he	Are a flag outside – the Nishan Sahib. This is usually ith the Khanda symbol in blue. Large space with a throne at one end Men and women sit separately so they are not distracted by one	 GGS 305, written by Guru Ram Das, outline how Sikhs should pray: 'One who calls himself a Sikh of the Guru, the True Guru, shall rise in the early morning and meditate on the Lord's Name'. Prayer routine includes waking early, bathing, repeating the japji (a prayer given by Guru Nanak), and repeating other prayers 	When/why - Usually takes place at the start of a festival - Times of joy and sorrow e.g. wedding, funeral, birth of a baby, death of a family member - GGS may be taken to a new home or new business as a blessing		
Darbar Sahib	another, and all sit on the floor so they are lower than the GGS	at different points of the day.	Importance Seen as a great blessing to have the Living Guru recited at an important event – it should not be taken lightly.		
Takht	Throne: represents the GGS being treated like royalty/ a human guru. Seat covered in fine cloth, often surrounded by flowers, space for money and food offerings, and a bowl containing karah parshad	 Nam Japna: Meditating on the Name of God One key responsibility of Sikhs is to meditate on the name of God, known as Nam Japna. 	- If it is to bless a new home or business, the family are expected to listen and take part at some point during the 48 hours to show respect		
Palki	 (a sweet food which is seen as a blessing) Domed structure used to cover the raised area where the GGS is placed. Canopy at the top may be engraved with the word 'Waheguru' or with scripture. 	 This keeps God in mind at all times and helps them to act well in their life. It may involve quietly reciting God's name to oneself, or by saying it out loud in a community recitation. It is the community recitation that is usually known as Nam Japna. Those who have the treasure of the Lord's Name deep within their hearts 	 Showing Respect to the Guru Granth Sahib GGS is known as the Living Guru (as declared by the 10th and final human guru, Guru Gobind Singh) and is treated with as much respect as a human guru would be. Sikhs show respect to the GGS in a number of ways, e.g.: All printed copies have 1430 pages to show they are identical and the words do not change 		
Manji	A small bed on which the GGS is placed during the day.	- the Lord resolves their affairs'			
Chanani	Large canopy made of decorated cloth which is placed over the palki.	The Role of the Gurdwara in the Sikh Community Management and Role	 Sikhs sit on the floor to worship so they are lower than the GGS They do not point their feet towards it They bow before it 		
Langar Hall	The area of the gurdwara where free food is served. All are welcome, everyone sits together on the floor to represent equality. Worship in the Gurdwara	 There are very few paid roles as most Sikhs volunteer as part of sewa. A granthi (or management team) manage and maintain the gurdwara. Many gurdwaras are open 24/7 and all are welcome Besides prayer and worship, other roles for the gurdwara include: langar, meeting/ education rooms to teach Punjabi and Gurmukhi, committee meetings, youth clubs etc. 	 They cover it with a rumalla, a decorated cloth (often made of silk) They wave a chauri over it – a fan made from yak's hair whch would have been waved over honoured teachers to keep them cool and keep flies away At the end of each day it is wrapped in clean cloth and carried in a procession to its rest room, where it is effectively put to bed overnight. The room is called Sach Chand. 		
 Worship can include meditating, listening, singing, reciting, working and serving people – is it true worship as long as God is kept in mind. Worship and adore Him, and you shall be at peace forever'. Worship in the gurdwara may last up to 5 hours but people may come and go. Shoes are removed, heads are covered, hands/ feet washed, before bowing and touching the floor in front of the GGS. Worship services start and end with the Ardas Prayer Kirtan – singing of hymns which nourishes the soul Ragis – musicians who sing ar play accomoniment for kirtan Karah Parshad – given out at the beginning and end of the service, from the same bowl to show equality. Sweet like God's blessings. 		 Granthi: A male or female Sikh, who has been initiated into the Khalsa, who reads the Guru Granth Sahib. They are expected to be of good character and live life according to the Sikh code of conduct They do not have a higher status than other Sikhs but they are highly respected. Granthis arrange and conduct religious services, maintain the gurdwara, leads kirtan (singing hymns), lead an akhand path etc. Most importantly, granthis take care of the Guru Granth Sahib, organizing the ceremony to bring it to and from the rest room each day. 	 Langar as an expression of Sewa Started by Guru Nanak to promote equality in a time of the Caste System Shows equality: all are welcome, all sit on the floor so they are on the same level, food is vegetarian so everyone can eat it. Excellent way for Sikhs to carry out Tan – physical sewa – by cooking, serving, cleaning. Can also show Dhan – material sewa – by donating produce. Both men and women take equal roles in helping in the langar. Many langars run in times of crisis, and to help those in poverty. 193 Guru Gobind Singh: 'Keep the langar ever open'. 		

- -Karah Parshad – given out at the beginning and end of the service, from the same bowl to show equality. Sweet like God's blessings.
- Most importantly, granthis take care of the Guru Granth Sahib, organizing the ceremony to bring it to and from the rest room each day.



Sikh Practices: Part 2 – Festivals and Lifestyle





inside.

evidence.



Christian Beliefs: Part 1 – The Nature of God



Key Words

Monotheistic: A religion which believes in one God Holy: Separate and set apart for a special purpose by God

Omnipotent: All powerful, Almighty

Omnibenevolent: all-loving Just: Fair

Trinity: One God has 3 parts; God the Father, Son and Holy Spirit

Holv Spirit: God's presence in the world

God the Son: Jesus – enables humans to have a special relationship with God

Creation: God bringing the universe into being

The Word: Jesus – as described in the book of John Genesis: The first book in the Bible which has the creation story in it

Incarnation: God in human form – Jesus.

Resurrection: coming back from the dead

Blasphemy: saying or doing something which goes against God

Crucifixion: Roman method of execution where a person is nailed to a cross

Ascension: 40 days after the resurrection when Jesus returned to God in heaven

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Afterlife: What happens when you die

Day of Judgement: God will judge all souls at the end of time

Heaven: Eternal happiness, being in the presence of God

Hell: Eternal suffering, absence of God

Purgatory: Catholic belief in which souls are cleansed in order to enter heaven

Sin: Any action against God

Original Sin: First sin in the world committed by Adam and Eve which means all humans are born with this in them

Salvation: saving the soul from sin and going to heaven thanks to Jesus' sacrifice

Grace: A quality of God which shows to humans that God loves them, which they don't need to earn

Forgiveness: pardoning someone for their wrong doing Atonement: restoring the relationship between people

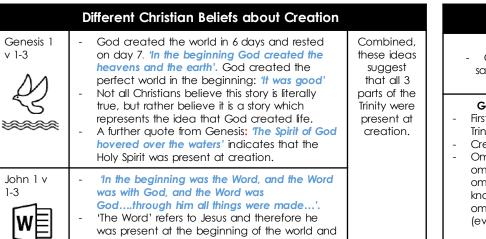
and God through the life, death and resurrection of Jesus

God as omnipotent, loving and just

- Christians believe God is omnipotent: all-powerful. 'Nothing is impossible (I) with God' P.

- God is omnibenevolent: all-loving. Guidelines are given for us to live the best lives we can. Christians should love each other treating everyone with care and respect. 'God so loved the world he gave his one and only Son...

- God has unlimited power and authority, together with complete love, and therefore gives justice in a fair way. Christians should try and bring about fairness in the world.



The Problem of Evil and Suffering: The Inconsistent Triad

- Christians believe God is omnipotent (all-powerful) and omnibenevolent 77 (all-loving). 1
- However, why is there evil and suffering? -
- If God was all powerful, he would be able to stop evil and suffering.
- If he was all loving, he would care enough to stop it. So he would stop it.
- -But he doesn't! So God cannot exist, or is at least not omnipotent or omnibenevolent.
- A Christian defence: God is transcendent (beyond our understanding) and therefore we can trust God to know best – there is a reason for this inconsistency, even if we can't understand it.

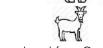
The Oneness of God and the Trinity				
 The Holy Trinity: God the Father, the Son and the Holy Spirit. God is three in one. There are not three Gods, but different forms of the same thing. Each part is fully God, but also these parts of the Trinity are not the same. 				
 God the Father First person of the Trinity. Creator Omnipotent, omnibenevolent, omniscient (all- knowing) and omnipresent (everywhere). 	God the Son - Second person of the Trinity. - Jesus, who became God in human flesh (known as the Incarnation). - Performed miracles, healed the sick and showed kindness to outcasts.	 God the Holy Spirit Third person of the Trinity. God sent the Holy Spirit to influence, guide and sustain all life on earth after Jesus ascended. The unseen power of God. 		

Different Christian Beliefs about the Afterlife

Based on God's judgement Christians believe that people will go to heaven or hell. -

involved in the creation of the world

- Judgement will happen at death or at the day of judgement (Christians vary on their understanding of this)
- The Parable of the Sheep and the Goats shows how people will be judged by God. The sheep represent those who did good actions (therefore going to heaven) and the goats represent those who did bad actions (therefore going to hell)





	Incarnation). - Performed miracles,		life on e
9	healed the sick and showed kindness to outcasts.	-	The uns of God

- Jesus also said, "I am the way the truth and the life, no-one comes to the Father except through me." _
 - So, treating others well and believing in God is important to guarantee a good afterlife. Heaven is seen as being with God and eternal happiness where there is no suffering. Hell is seen as eternal torment or suffering and being absent from God, and where the Devil is.
- Some Christians believe that Heaven is a literal, real place you will go. Other Christians believe it is just being with God, in the same way hell may not be actually real but an absence of God.
- The Bible teaches that there will be a resurrection of the body for all people who go to heaven, though the details of this are debated.
- In the book of Revelation it mentions that people who go to hell will burn in a lake of fire. -
- Roman Catholics believe in a place called purgatory in which your soul goes to be cleansed as not everyone is ready yet to go to heaven. -





Christian Beliefs: Part 2 – Jesus Christ and Salvation



	Jesus' Life: Key Events	
Event	Key Details	Importance/ Influence on Christians
Incarnation: Jesus is God in human form	 'Incarnate' mean 'In the flesh' – Jesus was God in the flesh. Jesus' birth is explained in the Christmas story: the Nativity. Humble birth (in a manger) shows Jesus was fully human, but he was born through the immaculate conception (Mary was a virgin), so he was fully God This means Jesus does not have Original Sin because he was not conceived through sexual relations. 'Before they came together, she was found to be pregnant through the Holy Spirit'. The Word became flesh and made his dwelling among us'. ' 	 Encourages them to celebrate Christmas for its true meaning Helps them in difficult times as they know their role model was human too Allows them to have a personal relationship with God through Jesus
Crucifixion: Jesus' death on the cross	 Crucifixion is remembered on Good Friday. Jesus was arrested (having been betrayed by Judas) and put to death by Pontius Pilate. He was crucified alongside two ariminals. As Jesus was fully human he suffered pain as an ardinary human did. 'Father, into your hands I command my spirit' On the aross Jesus said 'Father forgive them, for they know not what they do'. 	 By accepting Jesus' sacrifice they can be forgiven for sin and go to heaven. Encourages them to follow Jesus' example and forgive others Reminds them to be thankful and remember Jesus (especially on Good Friday) Reminds them that suffering is a part of life and God can understand what it I like for someone to suffer.
Resurrection: Jesus rose from the dead	 Jesus was buried in a tomb and left there until Sunday. Due to it being the Sabbath (on the Saturday), no-one could touch the body until after this. Mary Magdalene returned to the tomb - it was open and empty. An angel appeared and said Jesus had risen from the dead. Evidence of resurrection: he appeared to people including disciples, they saw him eat, Thomas was encouraged to tough Jesus' palms to prove he was not a ghost. 	 Shows Jesus was divine and not just a human so it may strengthen faith Shows the power of good over evil Encourages them to not fear death
Ascension: Jesus went back up to heaven to be with God	 Happened 40 days after the resurrection - Jesus ascended to heaven. He gave the disciples the Great Commission: 'Go and make disciples of all nations, baptizing them in the name of the Father, the Son and the Holy Spirit'. The Holy Spirit was left to guide and comfort people. 	 Will encourage them to call on the Holy Spirit for guidance and comfort Will encourage them to spread the message of Christianity (evangelise)

Geod (Heiy)

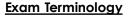
Sin and Salvation

- Sin: anything that goes against God's laws. Separates humans from God.
- Salvation means to be saved from sin (therefore being able to go to heaven)
- Humans are not perfect impossible not to sin
- All humans are born with Original Sin passed down from Adam and Eve.
- This action separated humans from God and brought about death into the world. They were tempted by the serpent (devil) and Christians believe that Christians are tempted in life to do bad things.
- Salvation through Law: humans have free will but should use this to make the right choices using God and Jesus' teachings to guide them, e.g. 10 Commandments.
- Salvation through Grace: being saved by accepting the sacrifice Jesus made on the cross – this showed God's grace: 'For by grace you have been saved through faith'
- Salvation through Spirit: having the Holy Spirit as a guide to accept God's Grace and follow his Law.

The Role of Christ in Salvation: Atonement

Salvation is offered through Jesus, "For the wages of sin is death, but the gift of God is eternal life in Christ Jesus".

- Jesus' death makes up for Original Sin. Humans can receive forgiveness for their sins because of Jesus' death and then receive eternal life.
- His sacrifice provides atonement, which means our relationship with God is restored. This removes the effects of sin and allows humans to get back to God. "He is the atoning sacrifice for our sins and for the sins of the whole world".
- Jesus paid the price for the sin of all mankind through his death and Christians believe if you put your trust in him you can receive eternal life with God.
- Links with Salvation through Grace: salvation is a gift people must choose through belief in Jesus atoning for their sins



Influence: The capacity to have an effect on people's character, behaviour or actions Contrasting: To show a difference Contemporary: Occurring in the present time Sacred Writings: Writing that is believed to contain words of God e.g. The Bible Evaluate: Consideration of different viewpoints before arriving at a final judgement Justified Conclusion: A final decision which is based upon a range of evidence. 196



Christian Practices: Part 1 – Worship and Festivals



	Worship and Festivals					
Practice and Key Words	Details/ C	Importance and Quotations				
Worship: Act of religious honour or devotion	us Liturgical - Takes place in a church and is led by a priest - Formal, set prayers are read out, and the worship follows a set pattern and structure - Formal, set prayers are read out, and the worship - No set prayers, instead people take turns to preach and read from the Bible - A more traditional, and formal form of worship - E.g. Eucharist in the Catholic Church - E.g. Eucharist in the Catholic Church - Service is usually focused around a Bible reading - E.g. Methodist or Baptist services		 Shows gratitude, love and respect to God Could be a way of asking for forgiveness or asking for help Brings comfort and strength Gives time for reflection 'Sing to the Lord, for he has done glorious 			
	Informal - Spontaneous prayers or sharing of thoughts - Community or house churches might meet to eat together and share their faith - Pentecostal Church – 'charismatic' worship. Led by the Holy Spirit and may involve dancing, dapping, calling out, speaking in tongues.	Private - Spending time with God alone or with close friends/ family - May involve prayer, meditation, studying the Bible.	things; let this be known to all the world'.			
Prayer: Communicating with God	Informal - Prayers that are made up by the individual using his or her own words.	 Set Prayers Prayers which have been written down and said many times by many people E.g. The Lord's Prayer: The prayer that Jesus taught the disciples to pray, which includes thanks, asking for forgiveness and asking for guidance. "Our father who art in heaven". 	 Set prayers can bring a sense of unity Prayer brings comfort and builds relationship with God 'Call on me and come and pray to me, and I will listen to you'. 			
Sacraments: An outward sign of inward grace. Eucharist/ Holy Communion	 Roman Catholic Church (Mass) Readings from the Bible Offering of bread and wine brought to the alter Priest says the words of Jesus at the Last Supper, says the Lord's Prayer and gives a sign of peace Congregation come to the alter to receive the communion 	 Orthodox Church (Divine Liturgy) Hymns, prayers, readings from Bible. Priest comes through Royal Doors to chant the gospel. Lord's Prayer said, behind Royal Doors words of Jesus said. Bread divided into four – three consecrated as body and blood and fourth broken into small pieces. Priest gives bread and wine together on a spoon. 	 Jesus started the tradition at The Last Supper (which took place the day before he died) Christians now remember Jesus' death – reminds them of Jesus' sacrifice, and reminds them to forgive others. "This is my body which is for you, do this in remembrance of me" 			
Sacraments: An outward sign of inward grace Baptism	Infant Baptism - Everyone is a descendent of Adam and Eve and therefore carries Original Sin. Baptism washes this away. - - It also welcomes them to the church community. - Infant wears white, Godparents are chosen, font holds the water - E.g. Catholic, Orthodox	 Believer's Baptism (Adult Baptism) Some Christians think children are too young to understand the meaning and therefore don't baptise infants. The person is ad enough to understand the meaning behind what they are doing. This includes a full immersion in a pool to wash away sin and start a new life in Jesus. This is known as being 'born again' E.g. Baptist and Pentecostal churches 	 Brings a person into the Christian family/ community Water symbolises the washing away of sins Jesus was baptised, setting an example for others to follow Jesus also encouraged baptism in the Great Commission: . "Therefore go and make disciples of many nations, baptising them in the name of the father, son and Holy Spirit". 			
Pilgrimage: A special religious journey to a holy site.	 Lourdes (France) Dedicated to Mary as Bernadette believed to have seen visions of Mary in the 19th Century. A spring of water was discovered which had healing powers. Now millions of people have been to drink from the spring of water in the hope of being healed. 	Iona (Island off west coast of Scotland) - Small community set up by St. Columba, an Irish missionary in the 6 th Century - Pilgrimages happen there in dedication to the virgin Mary. - The community in Iona hold daily services in the Church leading a seven mile hike to holy spots – it is a physical and spiritual challenge	 Pilgrimage shows commitment to God and strengthens faith People may go on pilgrimage for healing It brings a sense of community 			
Festivals: Celebrations for religious reasons	 Christmas Remembers the birth of Jesus – his incarnation. It is celebrated on the 25th December. Trees and homes are decorated with nativity scenes. Lights remember Jesus is the light of the world. Carol services happen in Churches with readings from the Bible. Children act out nativity plays and midnight mass takes place on Christmas Eve. 	 Easter Most important festival which celebrates Jesus' resurrection from the dead leading up from holy week. Jesus was arucified on Good Friday and rose on Easter Sunday. Special services take place and processions led by someone carrying a cross. On Easter Sunday, sunrise services take place with hymns which celebrate the resurrection. Easter Eggs are used as a reminder of new life. Paschal candle is lit. 	 Festivals celebrate the most important events of Jesus' life – his birth, death and resurrection. They are a time for believers to come together and celebrate their faith. "I bring you glad tidings that today a 197 is born" 'Christ is risen from the dead'. 			



Christian Practices: Part 2 – The Role of the Church in the Local and Worldwide Community

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Key Words		The Role of the Church in the Local Con	nmunity	Church Growth			
<u>Church:</u> The holy people of God, the body of Christ or a building where Christians worship <u>Agape:</u> compassionate love <u>Mission:</u> A calling where an	Food Banks	 Provide free food (usually non-perishable items) to those in need, usually through donations Some food banks also offer support in seeking employment Example: The Trussell Trust 	The work of food banks and street pastors supports the key Christian message to show love to all. Key words and quotes to support: Agape – compassionate love 'Love your neighbour as you	 Church Growth Estimated to be 2.5 billion Christians in the world Christians are taught to help to grow the church by sharing testimonies (how God has changed their lives), inviting people to meetings, prayer, social events etc. The Great Commission 'Go and make disciples of all nations, baptising them in the name of the Father, the Son and the Holy Spirit' (Jesus' words before he ascended to heaven). 			
individual or group go out and spread the word of God. <u>The Great Commission:</u> Jesus instruction to his followers to go and spread his message "Go and make disciples of many nations"	Street Pastors - Volunteers who patrol streets in urban areas Do not actively preach but show their faith through their actions - Give out flip flops, lollipops and water on nights out, help people to get home safely, offer reassurance and support		love yourself' 'Faith, if not accompanied by action, is dead' 'For I was hungry and you gave me something to eat' (Parable of the Sheep and the Goats)	Mission and Evangelism - Evangelism means to spread the message of Christianity through preaching the Gospel (which means 'Good News') - Example: The Alpha Course. Anyone is welcome to join in with a meal and conversation about the 'Big Questions' of Christianity. • • The aim is to tell people that Jesus is the saviour of the world, in the hope of converting them to Christianity. - Example: The Alpha Course. Anyone is welcome to join in with a meal and conversation about the 'Big Questions' of Christianity.			
Missionary: A person sent on a religious mission to	The	Role of the Worldwide Church: Reconciliation		The Role of the Worldwide Church: World Poverty			
on a religious mission to promote Christianity in a different country through preaching or charity work <u>Evangelism:</u> Spreading the Christian message through preaching the Christian gospels <u>Alpha course:</u> An example of evangelism –trying to tell others about Christianity <u>Convert:</u> Someone who has decided to become committed to a religion and change his or her religious belief.	 Working for Reconciliation Jesus came to earth to restore the relationship between humans and God, so Christians believe they have a responsibility to restore the relationship between themselves and others. Lots of examples of arguing between faiths, even within Christianity e.g. Catholics and Protestants. Irish Churches Peace Project set up to reconcile these denominations Coventry Cathedral – bombed during WW2. Cathedral has now become a centre for reconciliation as Christians wanted to respond with forgiveness, 		Therefore, if you are offering your gift at the altar and there remember that your brother or sister has something against you, leave your gift there in front of the altar. First go and be reconciled to them; then come and offer your gift'. 'Blessed are the persecuted because of righteousness, for theirs is the Kingdom of Heaven'. 'If one part suffers, every part	 Responding to World Poverty Jesus told a rich man to sell everything he had and give it to the poor. Christian Aid: They aim to stop poverty, encourage sustainable development and provide emergency relief in areas such as Africa and the Middle East. Their slogan is 'We believe in life before death' They provide emergency food, shelter, water, sanitation and run a Christian Aid Week to fundraise every year. 'If anyone has material possessions and see a brother or sister in need but has no pity on them, how can the love of God be in that person?' Parable of the Sheep and the Goats 'Go, sell everything you have and give to the poor, and you will have treasure in heaven. Then come, follow me.' 			
Reconciliation: A sacrament in the Catholic		not revenge.	suffers with it' (St Paul, likening members of the Church to	<u>Exam Terminology</u>			
faith, also making up after an argument or disagreement <u>Persecution:</u> hostility or ill- treatment, because of	Responding to Persecution	 Persecution (ill-treatment) happens all over the world – Christians are tortured or even killed for their faith Christians have a responsibility to help those who are persecuted 	different parts of the body) 'Love your neighbour as you love yourself'	Influence: The capacity to have an effect on people's character, behaviour or actions Contrasting: To show a difference Contemporary: Occurring in the present time			
race or religious or political beliefs. <u>Poverty:</u> When people live without having basic human rights such as having enough food, water or shelter	FRI	 They might pray for them, donate to charity ar get involved with charities who work abroad. The Barnabus Fund: send financial support, raise awareness for those persecuted, send spiritual and material support. 	Agape: Compassionate Love	Sacred Writings: Writing that is believed to contain words of God e.g. The Bible Evaluate: Consideration of different viewpoints before arriving at a final judgement Justified Conclusion: A final decision which is based upon a range of evidence.			



Crime and Punishment: Part 1 – Religion, Crime and the Causes of Crime



			Reasons	for Crime		
Cause of Crime		Explanation/ Examples	Christic	in Responses	Sikh Responses	
Poverty and Upbringing	money for what family. Or, they may ho	ay commit a crime as they do not have the they need e.g. they may steal to feed their ave been brought up in an environment where sed ar even expected of them.	reasons and show compassion: 'Love your neighbour as you love yourself 'Blessed are the merciful' - They would also feel a responsibility to he	those who have committed crimes for these "/ Agape, elp those in need so that they had no need to s hungry and you gave me something to eat".	 Sikhs would also show some compassion and understanding: "Show kindness and mercy to all life" Sikhs would also feel a responsibility to help, which is why the service of the langar, an example of sewa, is so important. There should be no need for people to steal food as the langar is always open.: "Keep the langar ever open" Sikhs emphasise the importance of a strong family network in order to raise children in the right way. 	
Mental Illness	which leads peo issues which resu	can cause crime; kleptomania is a condition ople to steal. Mental illness can lead to anger ult in crimes such as assault, and some people urder because of their illness.	 Christians would be compassionate towards those with a mental illness as it is not their fault and they are not always consciously committing crimes: 'Love your neighbour as you love yourself'/ Agape 'Blessed are the merciful' However, justice for the victim is still important, so Christians would support an appropriate punishment coupled with medical support. 		 Sikhs would also show compassion as the criminal was not in full control of their actions: 'Show kindness and mercy to all life'. They would support the criminal getting support and treatment, as well as showing compassion for the victim/ their family. 	
Addiction	Addiction means the body cannot cope without a substance e.g. alcohol or drugs. This can lead to arime if the person is not in control e.g. they are drunk. They may behave in ways they wouldn't usually, or they may drive while intoxicated and cause an accident. They may also steal to fund their habit.		 Whilst illegal drugs are unacceptable, most Christians do not forbid drinking alcohol (Quakers do not drink at all). However, drunkenness is not encouraged, and it is a sin in Catholicism. Addicts would be encouraged to seek support: 'Love your neighbour as you love yourself' / Agape 		 Sikh teachings forbid the use of alcohol/ilegal drugs as they take the mind away from God, and a person is not able to worship. Addicts would be encouraged to seek support: 'Show kindness and mercy to all life' 	
Greed			Christians would not at all accept crime for this reason. One of the 10 Commandments says: 'Do not covet' which means do not be jealous. In the Bible it says: 'you cannot serve both God and money' and 'The love of money is the root of all evil', showing the problems with focusing on material wealth.		Sikhs are encouraged to be Gurmukh – God-centered, and to focus their minds on God and not on money/ possessions. Worldly Attachment is one of the 5 evils/ barriers to mukti, so crime for this reason is totally unacceptable.	
Hate ☆ ≠☆	ate ★≠☆ Hatred is a negative feeling which can lead to violence/ aggression. It might be based on a prejudice e.g. racism.		This is a totally unacceptable reason for arime as 'Human beings were made in God's image' so everyone has the right to be treated fairly.		This is a totally unacceptable reason for crime as 'All are made of the same clay' so all humans should be treated with love and respect.	
Opposition to an uniust law	Opposition to an unjust they felt the law was unfair, e.g. Rosa Parks refusing to give up her sect in 1950s America. In Britain today, au laws are		 In general, Christians are taught to obey the law: Those who refuse to obey the law of the land refuse to obey God'. However, if the law was in opposition to Christian beliefs they would perhaps campaign for it to be changed, ideally without breaking the law themselves. 		Sikhs have historically protested against laws they felt were unfair or in opposition to their beliefs, e.g. the law on motorcycle helmets. However, they would ideally protest to get the law changed rather than break it.	
		Types of Crime		Go	od and Evil Intentions and Action	
Туре	Type of Crime Christian Responses		Sikh Responses	Christian Views:	evil thoughts which lead to evil actions. Avoiding sin and temptation steers	
		'Do not commit murder' Sanctity of Life: 'Human beings were made in God's image'	Sanctity of Life: The Divine Light is within all'	 Christians away from crime. Christians would be more willing to treat out of evil intentions. 	an offender who had good intentions with more mercy than one who acted	
Theft: Stealing something that does not belong to you		'Do not steal'	Rehat Maryada: (code of conduct) 'No Sikh should gamble or commit theft'		evil but that people can be tempted to do wrong and break the law. with ' Original Sin' due to the actions of Adam and Eve, so we are inclined	
Hate Crime: Committing a crime based on prejudice e.g. race, age, sexuality Human beings were made in God's image'., 'There is no Jew nor gentile male nor female, for you are all one in Christ Jesus'		'All are made of the same clay', 'We are all sons of the one God, there is no Hindu and no Muslim'.	 There is no such thing as an evil person but humans do all make mistakes. Having good intentions and obeying the law etc helps Sikhs to build bad karma; those who commit evil actions with evil intentions should face justice in this life but will also face punishment in the next life. 			



Crime and Punishment: Part 2 – Religion and Punishment

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		Aim	ns of Punishment		Forgiveness: To show grace someone for what th	and mercy, and pardoning ey have done wrong.		
Aim	Explanation/ Exan	nples	(Christian and Sikh responses	Christian Views	Sikh Views		
Reformation	Supports the criminal in changin for the better. May involve therapy, education	-	Preferred aim of punishment for Christian: 'Love your neighbour 'Do not take revenge but leave but overcome evil with good'. Sikh: 'Show kindness and mercy	as you love yourself'/ Agape. room for God's punishment do not be overcome by evil	- Christians would still closely linked agree with justly equality.			
	Seeking justice or revenge . The committed a crime, so you dese punished'.	idea that 'you erve to be	should be proportionate to the taught to 'Turn the other cheek'	eye for an eye', but this really meant that a punishment crime. Christians do NOT agree with revenge. Jesus meaning do not retaliate. Je either: 'If someone hits you, do not hit him back. Go	teaching within Christianity.forgiveness, there is G - Forgiveness is not a replacement for punishment; fair			
Deterrence	Putting people off from committie either putting the criminal off from putting society off crime as they punishments they could get.	om re-offending, or			 In his life Jesus was also asked how many times people should forgive. an portain bor longitoring can also be given. The focus of punishme should always be 			
		Treat	ment of Criminals		He replied: 'Not seven times, but seventy-seven times', meaning always	reformation, which is closely linked with forgiveness.		
Type of Punishment	Explanation/ Examples	Strengt	ths and Weaknesses	Christian and Sikh responses	forgive.			
Prison	A secure building where offenders are kept for a period of time set by a judge	reformation e.g. a - Can lead to pool	usually gives opportunity for ounselling, good deterrent r mental health, many people earn poor behaviour or ible in prison.	Christian: Focus should be on reformation: 'Love your neighbour as you love yourself'/ Agape Sikh: Focus should be on reformation: 'Show kindness and mercy to all life'.	 Love your neighbolr ds you love yourself'/ Agape Gee Walker gives Christians a modern role model on the importance of forgiveness. 			
Community Service	An offender contributes to society as a punishment e.g. doing unpaid work in the community	+ Benefits the com reformation - Some see it as a	munity, encourages soft punishment	Christian: Good punishment for minor offences as reformation is encouraged. Encourages stewardship . Sikh: Good punishment for minor offences; will build good karma and could lead to future acts of sewa .				
Corporal Punishment	Punishment of an offender by causing them physical pain e.g. whipping, cutting hands off. Ilegal in the UK.	their family (retribu - Violates human ri reformation, prom	brings justice for the victim/ tion) ghts, does not encourage otes an attitude of revenge assion and forgiveness.	Christian: Disagree – violates sanctity of life and is a revenge-seeking punishment: 'Human beings were made in God's image', 'Do not take revenge but leave room for God's punishment' Sikh: Disagree for same reasons: 'All are made of the same clay', 'If someone hits you, do not hit him back. Go home after kissing his feet'.	Exam Terminology Influence: The capacity to have an effect on people's character, behaviour or actions Contrasting: To show a difference Contemporary: Occurring in the present time Sacred Writings: Writing that is believed to contain words of God e.g. The Bible Evaluate: Consideration of different viewpoints before arriving at a final judgement Justified Conclusion: A final decision which is based upon a range of evidence.			
Capital Punishment/ The Death Penalty	Punishment where an offender is put to death for their crimes e.g. electric chair, hanging, firing squad. Illegal in the UK since the 1960s.	deserve to lose yo utility ; if it benefits their life, perhaps i - No going back if	brings justice e.g. if you kill, you ur life. Supports principle of society for one person to lose t is acceptable. you get the wrong person, ance for reformation, violates	Christian: Disagree – violates sanctity of life. Does not allow for reformation: 'Human beings were made in God's image', Agape etc. Sikh: Disagree for same reasons: 'All are made of the same clay', 'Show kindness and mercy to all life', 'If someone hits you' etc.				



Relationships and Families Part 1: Sex, Marriage and Divorce

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Human Sexuality Heterosexual relationship: a sexual relationship with a member of the opposite sex Homosexual relationship: a sexual relationship with a member of the same sex Homosexual couples can now marry, or convert civil partnerships into marriage (as of 2014) Christian Christian Catholic Church: homosexual people are not sinful, but the sexual act is. Views Therefore they should not have sex, because the Bible recognises it as a sin: Views 'Do not have sexual relations with a man as one does with a woman; that is detestable'. G Church of England: welcomes homosexuals who live in a faithful, committed relationship, but they cannot get married in church. Sikh Views Liberal Christians: Loving, faithful homosexual relationships are just as holy as heterosexual relationships Sikh The Guru Granth Sahib does not mention homosexuality. Some Sikhs may interpret this to mean that the only marriage that should occur is between a man and a woman, as this is the only type of marriage mentioned in the Views GGS. ത Other Sikhs would not view homosexuality as a particular issue of concern. Either way, Sikhs would not discriminate and would ensure homosexuals were treated fairly, in line with their teaches on equality: 'All are made of the same clay', 'Show kindness and mercy to all life'. Sex Before Marriage - In British society, sex before marriage is no longer considered as unacceptable, however some religious views on the issue remain traditional. Ch Vie Christian Many Christians believe that the intimacy of sex requires the commitment of marriage - it should not be a casual, Views temporary pleasure. - In the Bible, St Paul wrote: 'Flee from sexual immorality... whoever sins sexually, sins against his own body... your body is a temple of the Holy Spirit'. Catholic and Anglican church teaches that unmarried people should not have sex. This would therefore make cohabitation (living together) wrong before marriage as well. Liberal Christians accept that for some people, sex is a valid expression of love. Sex before marriage/ living together before marriage could be seen as acceptable. Sik Sikh Views Sex before marriage is not allowed within Sikhism; the Sikh code of conduct (Rehat Maryada) says that anyone who has sexual relations before marriage is not a true Sikh. 'For a moment of sexual pleasure, you shall suffer in pain for millions of days' (GGS) Anything that might lead to lust, one of the 5 evils, should be avoided. This would also make cohabitation before marriage wrong. Sex Outside of Marriage: Adultery Adultery: Having sexual relations with someone who is not your husband or wife Regardless of religion, most people in contemporary society view adultery as wrong because it breaks trust, involves secrecy and lies, can affect children and causes pain to all concerned. Christian Adultery breaks the promises Christian couples make before God during their wedding. One of the 10 Commandments is 'Do not commit adultery'. Views Jesus taught that lust, which could lead to adultery, is also wrong: 1 tell you that anyone who looks at a woman lustfully has already committed adultery with her in his heart'. Adultery is seen as a very serious sin – Sikh marriage is a sacred bond and adultery break that bond, betrays trust Sikh Views and brings shame on the family. It is one of the 4 misdeeds (kurahat) that members of the Khalsa must not commit. The **Kachera** (cotton underwear) is one of the 5 Ks as a reminder to keep the body pure.

The blind fool abandons the wife of his own home, and has an affair with another woman'.

Marriage: A legal union between a man and a woman (or in some countries such as the UK, two people of the same sex), as partners in a relationship. Marriage was one of God's gifts at areation. The Bible teaches: 'A man shall leave his mother and father and be united with his wife, and the two will become one flesh'. Marriage represents a covenant (promise) before God - it is a spiritual bond of trust. 'Husbands, love your wives, just as Christ loved the Church'. The purpose of marriage is to provide a stable, secure environment for family life. Marriage is a spiritual act; an opportunity to become one spirit within two bodies. The ceremony is called 'Anand Karai' which means 'blissful union'. Marriage is a union witnessed by God, shown by the presence of the GGS at the wedding ceremony. The purpose of marriage is companionship and the spiritual development of both partners. They alone are called husband and wife, who have one light in two bodies' GGS. **Divorce and Remarriage** Divorce: Legal ending of a marriage **Remarriage:** When someone marries again while their former spouse (husband or wife) is still glive.

Marriage

hristian iews	 Many churches offer courses to prepare couples for marriage, in the hope of avoiding divorce later. Jesus taught: 'Anyone who divorces his wife and marries another woman commits adultery against her' Christians must balance the need to respect the sanctity of marriage with showing compassion for those in difficult situation e.g. domestic violence. Catholic Church: a civil divorce cannot dissolve a marriage between two baptised people. Catholics can separate but they cannot marry someone else in a Catholic church while their partner is still dive. Catholics can obtain an annulment in certain circumstances if it was never a true marriage. Other Christians see divorce as the lesser of two evils Methodists: accept civil divorce and allow remarriage in a church, as long as the couple take the vows seriously.
kh Views	 Marriage should be for life as the couple are 'one spirit in two bodies' (Guru Amae Das). The Rehat Maryada teaches that in general, no Sikh should marry a second time if their first spouse is still alive. If divorce was looking likely, the Sikh community and the couple's family would support in helping the couple to work through their problems. However, Sikhs reluctantly allow civil divorce – grounds for divorce include adultery, aruelty, desertion, insanity and change of religion. Widowed Sikhs who wish to marry again are encouraged to do so in the gurdwara.
	- Midowed sixing who wash to many again are cheopidged to do so in the guidward.

Annulment: A Catholic Church ruling that a marriage was never valid

Exam Terminology

Influence: The capacity to have an effect on people's character, behaviour or actions Contrasting: To show a difference Contemporary: Occurring in the present time	
Sacred Writings: Writing that is believed to contain words of God e.g. The Bible Evaluate: Consideration of different viewpoints before arriving at a final judgement Religious Traditions: Viewpoints from either a religion or religious denomination Justified Conclusion: A final decision which is based upon a range of evidence	

Rehat Maryada: 'If a woman's husband has died, she may, if she so wishes, finding a match suitable for her, remarry. For a Sikh man... similar ordinance obtains'.



Relationships and Families Part 2: Families and Gender Equality



ACADEMIY		
Part 1 Continued: Contraception and Family Planning	The Nature of Families	
Contraception: the methods used to prevent a pregnancy from taking place. Natural forms include the rhythm method. Artificial forms include: the pill, condom, diaphragm, coil, morning after pill. Family Planning: The practice of controlling how many children couples have and when they have them. Christian - Views - All Christian churches agree that having children is God's greatest gift to a married couple. - They also agree that there may be some circumstances where it is not appropriate to bring	Family: A group of people who are related by blood, marriage or adoption Nuclear Family: A couple and their children regarded as a basic social unit Stepfamily: A family that is formed on the remarriage of a divorced or widowed person and that includes a child or children Extended Family: A family that extends beyond just parents and their children, by including grandparents and other relatives as well Polygamy: The practice or custom of having more than one wife at the same time Bigamy: The offence of marrying someone while already married to another person (illegal in UK). Same-sex parents: People of the same sex who are raising children together	
 children into the world e.g. economic, environmental, physical or psychological reasons. However, churches disagree on how pregnancy should be prevented. Catholics: Artificial contraception goes against God's plans as the purpose of sex is to express love and make new life. These purposes should not be separated. For Catholics, Only the thythm method is suitable for family planning (having sex at times during a woman's cycle where conception is less likely). The Church of England support use of artificial contraception so that couples can plan their families in a healthy and sensible way. Many Christians do not support using the morning after pill because here, the egg may have already been fertilised. Since many Christians believe life begins at conception, preventing a fertilised egg from developing is seen as sinful as it is already a life. Sikh Views 	 own wife and each woman her own husband to avoid sexual immorality. Whilst some Christians disagree with same-sex parents, believing the ideal is for children to have a male and female role model, other Christians believe the most important thing is for a child to be in a loving household. 	
 since this would only be relevant to a married couple. Sikh scriptures give no specific guidance on contraception so Sikh couples choose their own methods. Contraception should not be used to prevent having children altogether, or to cover up an adulterous affair. Some Sikhs would not agree with the morning after pill (reasons as above) and some would also view excessive sex, without allowing children to be conceived, is lust. 	 Sikh Views Family is the essential social unit in Sikh society. Guru Nanak preached that the life of the householder was the highest path to spirituality. The family is where children are trained in sewa (selfless service) and where religious traditions are passed down. Learning the values of the family are needed for prosperity and world peace: 'Gazing upon his family, he blossoms forth like the lotus flower'. Sikhs believe in monogamy; having only one husband or wife). Within Sikh families, men and women are equal but may have different role e.g. the mother usually takes the main role with children for the first few year while the father takes on household duties. Amritchari (baptised) Sikhs and many non-baptised Sikhs would not cut their children's hair. When the children are older, Sikh parents would often support their child in finding a suitable marriage partner, and grown up children would support their elderly parents. Sikhs do not generally approve of same-sex parenting, atthough more liberal Sikhs may support it. 	
Procreation: bringing babies into the world Stability: safety and security, being able to live peacefully Educating children in a faith: bringing up children according to the religious beliefs of the parents.	Gender Equality	
 Christian The idea of family is ingrained within the Christian faith – God reveals himself as a Father, with Jesus as his son, and humankind as his children. Christian parents are expected to be good role models for their children and to teach them moral values. Many Christians make prayer part of the family's daily routine, and some send their children to faith schools or groups run by the church. 	Gender equality: the idea that people should be given the same rights and opportunities regardless of whether they are male or female Gender prejudice: Unfairly judging/ holding biased opinions about an individual or group based on their gender Sexual stereotyping: Having a fixed general idea or image of how men and women will behave Gender discrimination: To act against someone on the bases of their gender	
Parents and children all have responsibilities to each other: - 'Listen to your father who gave you life, and do not despise your mother when she is old' Proverbs. - 'Children are a heritage from the Lord, affspring a reward from him' Sikh Views - Family fife is the highest path to spirituality	 All people have been created equal in the image of God: 'Human beings were made in God's image' Jesus treated women with respect, and the key teaching of 'Love your neighbour' shows that discrimination is wrong. Some traditional Christians believe that men are the head of the family and that women should mainly stay at home and care for children. This may stem from literal interpretation of Bible texts that reflect the time in which they were written: 'Your desire with be for your husband, and he will rule over you' 	
 Family life's the rights pain to spirituality The Sikh community itself is also a family as humans beings are sons and daughters of the same universal father (God). Guru Arjan describes the ideal family where the mother is central to its happiness: 'She is the most noble of all the family. She counsels and advises her hope and desire' Raising a child in the Sikhs faith is very important, according to the Sikh code of conduct: 'If is a Sikh's duty to get his children educated in Sikhism' Rehat Maryada Grandparents would play a large role in a child's upbringing – a married couple often move in with the groom's parents after marriage, and most Sikhs would also not consider putting their elderly parents in a care home, as it would be an honour to look after them themselves. 	 Most Christians today see marriage as an equal partnership where the different gifts of men and women strengthen family life. Sikh Views Guru Nanak spoke up on behalf of women which was unusual at that time. The Gurus changed Indian society as more women began to take part in social, religious and political matters. Within Sikhism, all are treated equally and judged on their actions, not their caste, gender or race. Amritdhari Sikh use the name 'Kaur' which frees them from having to take their husband's name when marrying. Many Sikh women are well educated and encouraged to pursue professional careers. Some Sikh women feel that traditional Indian culture, rather than Skhism itself, has limited their chance for education etc, but Sikhism itself fully promotes equality. 'Without woman, there would be no one at all'. 	



Religion and Life Part 1: The Origins and Value of the Universe



ACADEMY	_		-		· · · · · ·		
		The Origins of	the Universe				
Christian Views	 Creation story found in first book of the Bible: Genesis Christians believe the world originated from a specific act of God and not by chance Genesis 1 v 1: 'In the beginning God created the Heavens and the Earth 	the world in 6 believe the Bik light', and ther - Liberal Christia taken metaph	th Christians: take the creation story days and rested on the 7 th . Also kn oble is the exact, literal word of God re was light'. Ins: believe God did create the wo vorically, so it did not happen exact wn as non-literalists)	own as Literalists as they . 'God said 'Let there be orld but the Bible can be	Can religion and science co-exist? - Fundamentalist views on creation are NOT compatible with science – areation story vs B Bang - Both Liberal and Sikh views on creation coul		
Sikh Views	 The universe was not an accident, but was a deliberate act of God (Waheg Waheguru spoke and his word created everything He willed the whole of creation into existence 'As he commands, so they exist' 	guru)			co-exist with science, however the key conflicting point is that the religious views a the universe was a deliberate creation by C whereas science would argue it was rando		
Scientific Views	 Big Bang Theory: Suggests around 13.8 billion years ago thee was a massive All the matter that formed the universe kept expanding and cooling, forming This was random, not a deliberate act 				🌒 🗮 🕆		
	The Value of the World and the Duty of Humans to Protect it			Use and Abuse of Animals: A	nimal Experimentation		
Christian Views	 The world has value because God made it. Christians would view the world with a sense of awe and wonder – amazem wonderful creation. Christians have a duty to protect the world – stewardship: The Lord took the him in the Garden of Eden to work it and take care of it' However, the Bible also states that 'Human beings were made in God's imore suggesting humans are set apart from God's other creations and have auth world – dominion They may rule over the fish of the sea and the birds of the air' 	What are the issues? - Animals used to be tested on for cosmetics e.g. makeup but this is less common today - Animals are still tested on for medicinal purposes, to advance science/research and	so would not be support - Testing to benefit huma and ultimately human I - Whilst it is not ideal (due	Christian Responses g for cosmetics is not essential and does not show good stewardst uld not be supported. g to benefit human life may be supported as humans have domin ltimately human life is sacred. it is not ideal (due to stewardship), many would see it as accepto e greater good to humans. Sikh Responses			
Sikh Views	 The world has value because God made it. Sikhs would feel a sense of awe and wonder: 'Beholding His wonders, I am value of the earth: The sky, the earth, the trees and the water - all are the Cr Lord' There is a divine spark in all living things that is part of God, and this spark or 	ion and take reation of the	benefit humans	a soul too. - Stewardship is importan also have a negative e - However, a Sikh may no	age animal testing of any kind because animals hand and humans should not abuse their power. It will affect on someone's karma . ot actively protest against testing for research/ s humanely as possible, and with the right intention		
5000	 back to God when released from the cycle of rebirth. For this reason, Sikhs do NOT support the idea of dominion, and do not belie superior creations. 	eve they are	<u>What are the issues?</u> - Many people choose not	Use and Abuse of A - There are no dietary rea	Animals: Food <u>Christian Responses</u> quirements within Christianity – individual choice.		
Use and Abuse of the Environment <u>How do we damage the natural world?</u> - Use of natural resources: overuse of - Both religions would encourage stewardship – taking co			to eat meat, perhaps for ethical, environmental or - The Bible states t you' which sugg		rerything that lives and moves about will be food for imals can be consumed however, Christians would conditions some animals face for the production of		
oil, gas and other resources, defores Pollution: gas emis waste etc.	r non-renewable of the environment by recycling, generati by walking more etc, saving electricity, er	ing less pollution ncouraging ble who support such as	- Vegan: does not eat animals products	 Animals have souls, a Food se 	Sikh Responses vegetarian, especially if they have joined the Khals nd the Guru Granth Sahib forbids killing living being rived in the langar is always vegetarian how kindness and mercy to all life'		



Religion and Life Part 2: The Origins and Value of Human Life



The Origins of Human Life					
Christic - Day 6 of creation: story of Adam and Eve found in first b - Adam was created from the dust of the earth, then Eve - God 'breathed life' into Adam's nostrils - 'Human beings were made in God's image': 'Imago Dei	e from one of Adam's ribs.	Contrasting Christian Views Fundamentalist Christians: take the creation of Adam and Eve to be literally true. Also known as Literalists as they believe the Bible is the exact, literal word of God. This view cannot co-exist with science - Adam and Eve vs Evolution Liberal Christians: believe God was the creator but the theory of evolution explain how it happened, whereas the Bible addresses why. Evolution is the way God designed life to advance and evolve. (Also known as non-literalists) This view can co-exist with science, other thn the idea that evolution was by chance.			
 The 'how' and 'when' humans came to exist is not as im Sikhs do not oppose evolution, but nothing would happe Waheguru oversees the natural process of evolution – it 'Without God, there is nothing at all He is woven into H 	en if it wasn't for hukam (Waheguru's Divine Will) t isn't left to random chance	Scientific Views - Charles Darwin published 'The Origin of Species' in 1859 and outlined the theory of evolution. - Single-celled creatures appeared in the sea which eventually evolved into other species. - Humans evolved over millions of years from other animals on land.			
Sanctity of Life vs Quality of Life	Attitudes Towards Abortion	Attitudes Towards Euthanasia Death and the Afterlife			
 Sanctity of Life is the concept that life is sacred and given by God; both Christians and Sikhs believe in this. For Christians. Sanctity of Life only applies to human life and not the life of animals. Sikhs see animals as more valuable as they believe they have a soul too. 'God breathed life into Adam' supports the Christian view that human life is sacred and symbolises that a part of God lives within all of us - for Christians this is our soul The quality of life refers to the general wellbeing of a person e.g. their health and happiness. When faced with matters of life and death Christians have to consider the quality of a person's life and this can often conflict with the Sanctity of Life Christians and Sikhs would largely view the sanctity of life as more important than the quality of life. 	 Deliberate ending of a pregnancy. England: can take place before 24 weeks and 2 doctors have to agree. Reasons may include: physical or emotional wellbeing of the mother, or potential wellbeing of the foetus or existing children. After 24 weeks: can only take place if there is a serious isk of disability or illness, or if the mother's life is at risk. Pro-life arguments: Against abortion, due to respecting the life of the foetus Pro-choice arguments: May support abortion due to respecting the choice of the mother Christian Views Sanctity of Life – many Christians do not support abortion as the foetus' life is sacred from conception: 'Before 1 formed you in the womb, I knew you' Roman Catholics – do not support abortion at all (only perhaps if the mother's life was as risk): 'Do not commit murder' Liberal Christians – generally oppose it, but it's acceptable in some circumstances e.g. mother's life at risk, rape, child would have severe disability 	 Euthanasia is assisting with the ending of fife for a person who is terminally II or has a degenerative illness. Active euthanasia: life is ended at the request of the patient via lethal hjection or poison (assisted suicide). This is doo known as voluntary euthanasia. Passive euthanasia: whatever is keeping the patient dive is taken away e.g. life support, medication. Their Ilness is then what kills them. This might be nonvoluntary euthanasia, as they may not be able to give consent. Active euthanasia is llegal in UK. It is allowed in some countries e.g. Switzerland Sanctity of Life - many Christians do not support euthanasia as life is still sacred, even if its quality is poor: <i>Human beings were made in God's image</i>. Roman Catholics - do not support euthanasia at all; only God has the fight to end life: <i>Do not commit muder</i>? However, if large doses of painkillers are used to help a person's suffering, and as a result, the person des, this is OK in the yes of the Catholic Church and is called the Doctrine of double effect. The Bible teaches that <i>There is a time to be born and a time to die</i> and therefore people shouldn't meddle in God's plans. Some Christians might argue that humans have been given freewill to make their own choices. Liberal Christians - whilst euchanasia would not be encouraged, it may be acceptable if it will relieve unbearable suffering (arguments for quality of file): Agape, Love your 			
Contemporary: Occurring in the present time Sacred Writings: Writing that is believed to contain words of God e.g. The Bible Evaluate: Consideration of different viewpoints before arriving at a final judgement Religious Traditions: Viewpoints from either a religion or religious denomination Justified Conclusion: A final decision which is based upon a range of evidence	(arguments relating to quality of life): Agape, 'Love your neighbour as you love yourself', 'Blessed are the merciful'. Sikh Views	 neighbour as you love yourself', 'Blessed are the merciful'. Sikh Views Sanctity of Life – Sikhs do not support euthanasia because only God has the right to take life; he areated everyone so their life is sacred: 'He created the creation and watches over it'. Suffering in this life may be a result of bad karma from a previous life, so escaping this suffering with euthanasia would only lead to more suffering in the next life. Helping those who are ill and suffering is a good opportunity for sewa (selfless service): Through selfless service, eternal peace is obtained'. A Sikh would most likely support withdrawing medical provision e.g. life support if there is no chance of recovery. 			



Existence of God: Part 1 – Philosophical arguments for and against the existence of God



The Design Argument (Tele	ological)	The First Cause Argument (Cosmological)		The Argument from Miracles		
Basic Premise and Evidence	Criticisms	Basic Premise and Evidence	Criticisms	Basic Pren	nise and Evidence	Criticisms
 The universe is an intricate, complex design which is too complicated to have come about by chance Therefore, a great design needs a great designer: God. Telos = 'purpose' - the universe has been designed by God with a clear purpose in mind. William Paley's analogy of the Pocketwatch If we found a watch, we would look at its intricacies and realise it has not come about by chance but had a watchmaker who had a clear purpose in mind. The same is true of the natural world - trees, the human eye, birds' wings and fish gills they must have been designed by God. Exac Newton and the flumb Newton argued that the fact that every single human being has a different thumbprint, and the fact that we have opposable thumbs, was enough evidence for God's existence: 'In the absence of any other proof, the thumb alone would convince me of God's existence'. 		 Everything that exists has a cause (everything is contingent - relies on something else for its existence) The universe must, therefore, have a cause too. At some point, there needs to be something eternal, which requires no cause, in order to start the process off This eternal, 'uncaused cause' needs to be an omnipotent being: God. Therefore, God exists. The Bible supports the argument: 'In the beginning God created the heavens and the earth' The Big Bang could still be true - but God caused it! Dominoes The argument can be compared to dominoes - one falls because the previous one fell. At some point, someone had to push the first domino. This can be compared to the need for an uncaused cause to the universe: God. St Thomas Aquinas An infinite sequence of cause and effect is impossible; there needed to be a starting point We can clearly see that the universe exists so it needed a creator to begin with: God. 	 Atheists say the argument contradicts itself; if everything needs a cause, then what caused God? If you say that God is eternal and has always existed, why can't the universe just always have existed too? The Big Bang was a random, spontaneous event, not an action caused by God 	 Events which and cannot l e.g. Jesus turn Events which which occur positive outca survived in Sta September 1 This argument scientific expla must be superi someone/som nature. Only God is ou God exists. Examples: Incarnation ar most importan faith Jesus performe turning water i feeding the 50 Lourdes – man 	ny miracles have been e which have been	 Atheists would argue that 'miracles' are just lucky coincidences There may be a scientific explanation which we just don't know about yet. Miracle healings may be the result of mind over matter, or a misdiagnosis by doctors. Some 'miracles' are fake or exaggerated by people wanting fame or money. Even if miracles do prove that God exits, they show God to be unfair and as having favourites – why doesn't everyone who needs a miracle get one? David Hume Argued that witnesses to miracles are primitive and uneducated, so the evidence is unreliable. (He was writing in the 18th century).
Arguments against God: Evil c	and Suffering	Arguments Against G	od: Science		<u>Exam</u>	<u>Terminology</u>
 Atheists point to the existence of evil and suffering God does not exist. The argument is that Christians believe God to be knowing, all loving etc. Therefore God should be would be aware of it and would care enough the doesn't, so he must not exist. Christian Defence Christians may defend God and say that evil an human free will – Adam and Eve's disobedience into God's perfect world. However, this may explaction) but it does not necessarily explain nature. Christians also defend God with arguments such wouldn't appreciate good. Perhaps suffering is a suffering is an opportunity to show compassion, make the choice to do good over evil. 	e all powerful, all e able to stop evil, o stop it. But he d suffering is the result of e brought evil and suffering plain moral evil (human al evil e.g. earthquakes n as: without evil we a test of faith. Perhaps	 Some atheists and humanists use the developments God. In the past, the arigins of the universe and life on ear assumed it was God. Natural disasters etc were thou Now, science can explain things that people couldr no longer look to religion for answers. Atheists argue a God to fill the gaps of what we do not understand Christian Defence Many Christians see no conflict between science an creation stories literally, so the Big Bang/ Evolution co. God was the force behind it happening as it did. (Fu issue, though, as they believe the creation story to be pope Francis said: The Big Bang does not contradict it When we read the creation story in Genesis we ramagician, with a magic wand but it is not so. He area 	th could not be explained, ght to be God's punishmen 't previously understand, so that there is no need to 'in d religion; many Christians in explain the universe and undamentalist Christians wa e literally true). the divine act of creation; on the risk of imagining that	do not take the life on earth, but build have an rather if requires t God was a	people's character, Contrasting: To show Contemporary: Occu Sacred Writings: Writi contain words of God Evaluate: Considerat before arriving at a fi Religious Traditions: N	a difference urring in the present time ng that is believed to d e.g. The Bible ion of different viewpoints nal judgement /iewpoints from either a enomination A final decision which is



Existence of God: Part 2 – The Nature of the Divine and Revelation

Key Words

Divine: God, Gods or Ultimate Reality

Enlightenment: A state of spiritual awakening and the gaining of a deeper understanding of reality. E.g. Buddhists are seeking how to end suffering and achieve happiness.

General revelation: Indirect revelation: the idea of being able to see something of God through nature and scriptures which are readily available in everyday experience.

Immanent: A characteristic of God: the belief that God is present and involved in the world, (eg through special revelations/miracles).

Impersonal: A characteristic of God: the belief that God is beyond human understandina.

Omniscient: All-knowing: believed by theists to be an attribute of God. Omnipotent: All-powerful; believed by theists to be an attribute of God. **Personal:** A characteristic of God: belief that humans can build relationships with God.

Revelation: When God is revealed to humans; can be special or aeneral.

Special revelation: Direct revelation: God being revealed directly to an individual or aroup through experiences such as visions. Transcendent: A characteristic of God: belief that God is outside space and time.

Ultimate reality: Belief in a supreme and fundamental power in the universe. In Christianity, this ultimate reality is a personal being: one God in three persons. In Hinduism, Brahman is often referred to as the ultimate reality and supreme cosmic power.

Vision: An experience of seeing/experiencing something in the imagination or through a dream.

The Divise / Knowledge of Cod. An Introduction
The Divine/ Knowledge of God – An Introduction

- Every reliaion accepts that there is an ultimate reality that is eternal and unchanging.
- For Christians, their ultimate reality is a personal being. God, who makes himself known in 3 persons; the Father, the Son and the Holy Spirit.
- Some theists say God cannot be known because God is transcendent; beyond human understanding. God cannot be described using the limits of human language.
- Most Christians do accept this, but do think that God can be known through revelations reveal himself to people.
- There are 2 types of revelation; special revelation and general revelation.
- Some experiences can be both general and special revelation.

- when God chooses to
U

Special Revelation: Visions		Nature as a - Links with
Basic Premise and Evidence	Criticisms	of the no
 When people experience God directly in a particular event or direct personal experience. Could be a dream, a vision, a prophecy, a miracle or 'hearing God's call'. Could be experienced alone or with a group of people. The experiences usually have a huge, life-changing impact on people e.g. Nicky Cruz (gang leader turned Christian minister). They are rare, and many believers never experience such events. Visions A form of special revelation which comes in a picture or image form. People may see holy people, angels or hear messages from God Example from the Bible: Saul on the Damascus Road Saul had sworn to wipe out the Christian church. He was travelling with others on the Damascus road to arrest any follower of Jesus. On the way, they were struck by a blinding light, and Saul had a vision of Jesus who spoke to him. Saul's life was changed – he changed his name to Paul (now known as St Paul) and spent the rest of his life committed to Christianity. 	 Special Revelations e.g. visions are subjective, open to interpretation and not open to scientific testing. Many religions have conflicting revelations so some could argue this invalidates them. Visions could be brought about by alcohol ar drugs, ar it could be wishful thinking so someone convinces themselves. It could be a mental or physical illness, or the mind playing tricks. People could also be lying to gain wealth or power, or they may have just been mistaken. 	 Josi a C gives an God is st clever, p The heat proclain Scripture as Religious God act live. E.g. the Jews (OI Jesus (N) All Christ they ma and other When C better up and rece

General Revelation: Nature and S	cripture
Basic Premise and Evidence	Criticisms
Ordinary, every day human experiences that reveal truths about God, General revelation comes to people through nature, a person's reason, their conscience (inner sense of right and wrong), reading scriptures, through worship etc. The experiences are available to everyone, but they do not convince everyone that God is real because they depend on people's interpretation. sture as a way of understanding the Divine Links with the Design Argument: the beauty and order of the natural world gives theists a sense of awe. Just as a painting gives insight into the artist, nature gives an insight into God. God is shown through nature to be creative, artistic, clever, powerful and awesome. The heavens declare the glory of God; the skies proclaim the work of his hands'	 Nature Humanists – the works of nature are not a divine revelation but are special because they can lead to a grater understanding of the world through human observation and science. When a theist and an atheist look at a beautiful landscape, one just sees nature and the other sees God's creation – open to interpretation.
ripture as a way of understanding the Divine.	Scripture
Religious scriptures tell believers what God is like, how God acted in the past and how God wants people to live. E.g. the Bible tells the story of God's covenant with the Jews (Old Testament) and the life and teaching of Jesus (New Testament). All Christians believe the Bible is inspired by God but	 Some argue scriptures cannot reveal anything about God because they are merely their author's opinions and these could be

y may interpret it differently e.g. some take t literally others take a more liberal approach. en Christians read the Bible, they hope to get a ter understanding of the teachings of Christianity receive spiritual strength from God's words.

wrona. Even Christians do not all agree on how

scripture should be interpreted,

- It is difficult to describe the unseen, infinite God within the limits of human language.
- However, God is seen to have a number of characteristics:
- Omnipotent (all powerful) omniscient (all-knowing) and omnibenevolent (all-loving).

Religious thinkers also use the following words to describe different ideas about God:

- Immanent: God is present in the universe and involved with life on earth. God acts in history and influences people.
- Transcendent: God is beyond and outside life on earth and is not limited by the world, time or space. God does not act in the world or intervene in people's lives.
- Personal: God has human characteristics e.g. merciful, compassionate. God loves and cares for every individual and can be communicated with through prayer.
- Impersonal: God does not have human characteristics. God is more like a force or an idea. God is an absolute being who is only understood in terms of itself.

Can God be immanent, transcendent, personal and impersonal?

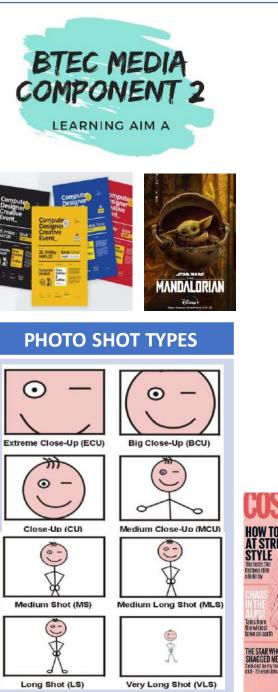
- Many religious believers think that they can experience God and have a personal relationship with him (immanent and personal) but that at the same time, God is the eternal, unlimited creator of the universe (transcendent and impersonal).
- Some religions emphasise one description more than another but others say all descriptions are true, even if they seem contradictory.
- This is because God is a mystery and beyond human understanding.

Christian interpretation:

Different ideas about the Divine: God's Nature

- God's immanence is revealed in Jesus who was God in human form, and in the Holy Spirit who works through Christians today.
- God is also transcendent as he is the creator of the universe, without whom nothing would exist. -
- Christians regard God as personal, a Father who cares for his children and with whom they can 206 have a relationship through prayer.
 - You know when I sit and when I rise; you perceive my thoughts from afar'





PHOTOSHOP TOOLS E 🕨 Move tool Marquee Tool Quick Selection Tool Lasso Tool Q 女。 🖋 Eyedropper Tool Crop & Slice Tool Brush & patch Tool 8.1 Brush & Pencil Tool 1 1 **History Brush Tool** Stamp Tool *1*. 🗖 Eraser Tool Gradient & Paint Bucket Tool ٩ Dodge, Burn & Sponge Tool Blur, Sharpen & Smudge ۰. ø, Τ. Type/text Tool Pen & Point Tool ▶, □, Shape Tool Path & Diract Selection * Zoom Tool Hand & rotate Tool Foreground Color **Background Color** Editing Mode ं Ē Screen Mode

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INSTA FEUDS



GENRE

Steve Neale: "Genres are instances of repetition and difference...difference is absolutely essential to the economy of genre': mere repetition would not attract an audience."

	MAGAZINE CONVENTIONS	Y10 -BTEC MEDIA - COMPONENT 2
Masthead	The title of a newspaper or magazine as it appears across the first page, front cover, or titl e page.	BTEC MEDIA COMPONENT 2
Sell line	On the cover of the magazine - This is used to attract the audience's attention by using words such as 'EXCLUSIVE' and 'FREE'.	LEARNING AIM B
House style	The overall design concept applied to a magazine, website, or media company. Can include the use of particular fonts, colours, layouts, images, logos, etc. This is designed to differentiate the product or company from its competitors and is often part of an overall campaign.	Masthead – Name of the magazine. Original Typography Sufficient Marketine Ma
Cover lines	Short statements on the cover of the magazine that describe the articles inside. Their purpose is to entice the reader into picking up and/or buying the magazine.	Sell line / Price
Pull Quotes	In the double page spread A brief, attention- catching quotation taken from the main text of an article and used as a subheading or graphic feature.	Cover lines Cover line
Mode of address	The way a media product 'speaks' to it's audience. The use of language that appeals to the target audience	Main Image Anchorage Anchorage
Main image	Takes up a large amount of the front cover this image catches the attention of the reader, links to main article and often uses celebrity appeal	Direct address (Image/text) Graphics 208



Y10 -BTEC MEDIA - COMPONENT 2

AUDIENCE KEY TERMS RECAP

Primary audience	The audience that the media producer has in mind for consuming the text. This is the audience they intend to target. Your target audience is teenagers
VALS profiles	These define an audience by how they think and by considering their values, attitudes and lifestyle (VALs). People can be classed as 1 of the following: The Aspirer -seeks status The Explorer - seeks discovery The Mainstreamers -seeks security The Reformer - seeks enlightenment The Resigned - seeks to survive The Struggler - seeks to escape The Succeeder - seeks control
Socio Demographic groups	 Group Description and examples. A Higher managerial, Professional e.g. Chief executive, senior civil servant, surgeon B Intermediate managerial, professional e.g. bank manager, teacher C1 Supervisory, junior managerial e.g. shop floor supervisor, sales person C2 Skilled manual workers e.g. electrician D Semi-skilled and unskilled manual workers e.g. assembly line worker, refuse collector E Casual labourers, pensioners, unemployed e.g. pensioners without private pensions and anyone living on benefits

PRODUCTION PROCESS

Pre-production: Planning and research

Production: Creation of content such as article/ images/ graphics

Post production: Editing and putting together considering space and layout choices



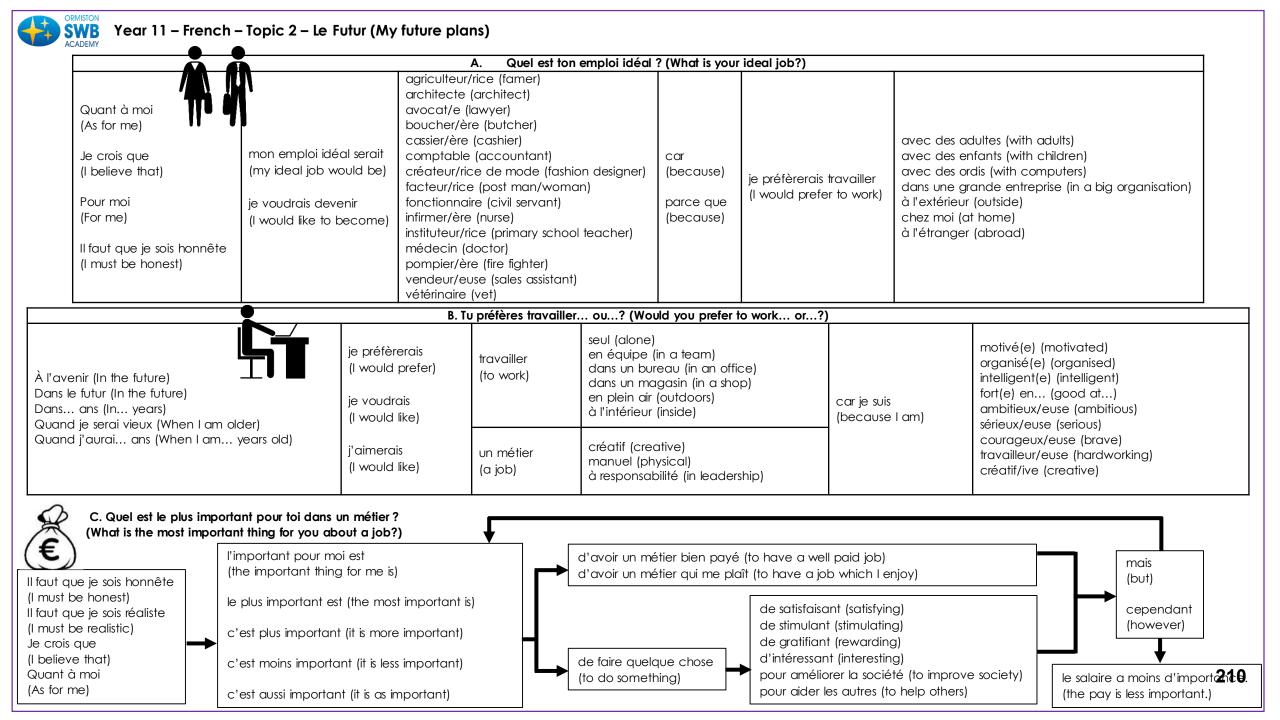
Fashion in the FAST LAWE

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LEARNING OUTCOMES AND EVIDENCE REQUIRED

A - Develop media production skills and techniques	Idea generation Photography work shop Photoshop experimentation and practice demonstrating skills development
B - Apply media pre-production and production skills and techniques	Idea generation Target audience profile Typography choices Masthead creation Article writing Thumbnail drawings Photography plan Photography workshop Questionnaires and results Production log of skills development creating magazine
C - Review own progress and development of skills and practices	Report to assess your finished work and practices. 209



Year 11 – French – Topic 2 – Le Fi D. Quelles sont tes qualités personnelle (What are you personal qualities?) Je crois que (I believe that) Quant à moi (As for me) Il faut que je sois honnête (I must be honest) E. A par	s? de bien or de très mo de très tra de très ca qui est tou qui est tou (who is alv qui peut p qui peut n qui aime v	plans) ganisé (well organised) otivé (very motivated) vailleur (very hardworking Ime (very calm) ujours prêt à apprendre (r ujours prêt a vivre de nou vays ready for new expe parler un autre langue (w n'entendre bien avec de voyager (who likes to trav	who is always read velles expériences riences) ho can speak ano es jeunes (who can vel)	ther language) get along with young		er) (I am not someone)
Avant de continuer mes études (Before continuin Après avoir terminé mes examens (After finishing Après avoir quitté le collège (After leaving school Plus tard (Later) A l'avenir (In the future) Dans le futur (In the future) Dans ans (In years) Quand je serai vieux (When I am older) Quand j'aurai ans (When I am,,, years old)	il faut que je sois honné (I must be honest) il faut que je sois réalist (I must be realistic) il faut que je sois heure (I must be happy)	nête ste donc (therefore) j'ai l'intention de/d' (l intend to) je vais (l'm going to) faire du bénévolat (do some voluntary work) faire le tour du monde (travel around the world aller à l'université (go to university) avoir des enfants (have children) fonder une famille (start a family) commencer un apprentissage (start an apprentissage (start an apprentissage)				
F. Quel travail est-ce que Quand j'étais petit/e (When I was little) Quand j'étais plus jeune (When I was younger) Quand j'avais ans (When I was years old) Avant (Before) A l'école primaire (At primary school)	je voulais travo (I wanted to v ça m'intéresso	ailler	dans l'audiovisue dans les médias (dans l'informatiqu dans l'hôtellerie e dans les arts et la dans le commerc dans le sport et le dans la médecine	I (in audiovisual media in media) Je (in ICT) It la restauration (in car culture (in arts and cul ce (in business) Ioisirs (in sport and leis e et la santé (in medec) tering) Iture) sure)	N'oubliez pas! Par exemple (For example) Comme (Like) Bien que je sache que (Although I know) Ce n'est pas facile (It's not easy) Je suis sans voix! (I'm speechless!) 211

SWB Year 11 – French – Topic 1 – A	MES EIU							
		A Quelle est ta	matière préférée	e? (What is your	favourite subje	ct?)	1	
J'adore (I love) Je déteste (I hate) Ma matière préférée est (My favourite subject is) J'étudie (I study) Je peux étudier (I can study) Je dois étudier (I must study)		le commerce (business studies) la sociologie (sociology) l'étudie de médias (media studies) l'économie (economics) les sciences (science) les langues (languages) les matières obligatoires (core subjects) les matières facultatives (option subjects) les arts ménagers (home economics)			je je car (because) puisque (because) pourtant (however) c'		je suis fort/e en (I am good at) je suis faible (I am not good at it) je ne suis pas doué (I'm not very talented) le prof est impatient (the teacher is impatient) le prof nous fait rire (the teacher makes us laugh) le prof nous critique (the teacher criticises us) c'est facile (it's easy) c'est utile (it's useful) il y a trop de devoirs (there's too much homework)	
		B. Qu'est-ce que tu faisais	à l'école prima	ire ? (What did ya	ou do at prima	y school?)		
(when I was younger,) Quand j'étais à l'école primaire j'adorai		ais (I studied) ais étudier (I had to study) ais (I loved) estais (I hated) estais (I hated)		(geography) y) h) igieuse (RE) sience)	assez (quite) un peu (a little) c'était très (very) (it was) trop (too) super (super) vraiment (really)		a little)) oer)	cool (cool) amusant (fun) intéressant (interesting) difficile (difficult) utile (useful) facile (easy) génial (great) nul (rubbish) ennuyeux (boring)
		C. Fais-moi une descriptio	on de ton collèg	e. (Give me a de	scription of you	ur school.)		
Mon collège s'ap	opelle	My school is called	Mon collège est mode				ménagé.	My school is modern/old/well equipped.
I ('est lin college mixte/lin internat/lin pensionnat I				Mon collège n' bien aménagé	collège n'est pas moderne/ vieux/ aménagé.			My school is not modern/old/ well equipped.
Il y a élèves et profs.				Les bâtiments sont modernes/vieux.			The buildings are modern/old.	
Les cours commencent à heures et finisse à heures.	ent	· · · · ·		Il y a une salle de sport/ une bibliothèque/ des labos.			There is/There are a sport hall/ 212 a library/ labs.	

GRMISTON SWB ACADEMY Year 1	1 – French –	Topic 1 – Mes	études (My Stu	dies)						N'oubliez pas!	
Lundi (Monday) Mardi (Tuesday) Mercredi (Wednesd Jeudi (Thursday) Vendredi (Friday)	avl à_	' une journée typ _ heures o'clock)	ique. (Talk abou les cours com les cours finisse J'étudie (I st c'est la recrée c'est le déjeur il y a des clubs	mencent (les ent (lessons fi udy) e (it's break ti ner (it's lunch	ssons start) nish) ime)				Comme (lik Tu rigoles ! (e (for example) e) You're joking!) nemar ! (What a nightm	nare!)
			E. Comment es	t le règleme	nt dans ton collège ? (What are t	he rules	s like in your schoo	ol?)			
Dans mon collège (In my school)	il faut faire il faut porte il ne faut p il ne faut p il est interd il est interd il est interd	ses devoirs (you er l'uniforme sco as manquer les as tricher pendo it de porter des it d'utiliser son p it de harceler d'	ortable en classe autres élèves (it	omework) vear your uni not skip lesso you must not quillage (you e (it is forbidd is forbidden	ns) cheat in a test) u must not wear jewellery or make len to use your mobile in class) to harrass other students)	.,	et je trouve ça (and I find it)	assez (qu un peu (a très (very trop (too) super (su vraiment	a little)) 	raisonnable (reasonna logique (logical) juste (fair) ridicule (ridiculous) frustrant (fristrating) trop strict (too strict) injuste (unfair)	able)
	Il est interd	it de sortir de l'e	cole pendant le	dejeuner (if	is forbidden to leave school at lu	nch)		une perte	e de temps (c	waste of time)	
	F. Quelle	s sont les différe	nces entre les co	ollèges en Ar	ngleterre et en France ? (What ar	e the di	ifferences betwee	n school in	Eng. and Fra	nce?)	
Les élèves achètent Avant de quitter le	eurs propres v cent avant 8h ces durent 2 r nt pas assez d tous les équi collège les él	rêtements (stude 130 (lessons start mois (the summe de progrès redo ipements (stude èves passent le	ents wear their ov before 8.30) er holidays last 2 r ublent (students nts buy all their e orevet (before le	wn clthes) months) who don't m quipment) aving schoo	akre progress retake the year) I they take their exams) school on Saturdays)	(so I p donc,	, je préfère le systè prefer the French s , je préfère le systè prefer the British sys	ystem) eme britani			
L.	AN AN		G. Parle	-moi un peu	d'une visite scolaire. (Talk to me	about	a school trip.)				
L'année dernière (la Récemment (recen	, ,	je suis allé (I w on est allé (we	ent to) e went to)	et (and)	on s'est fait des nouveaux amis on a amélioré ses compétence on a habité chez une famille d on a visité un nouveau pays (w on a apprécié nos différences, on a voyagé en (we travelle on a acheté des souvenirs (we	es [°] en lo l'une cu ve visite /similari d by)	angues (we improv ulture différente (v ed a new country) tés (we appreciat	ved our lar we lived wi	th a family fro		213

Year 11 BTEC Sport – Unit 3: Applying the Principles of Personal Training

Unit 3: Learning Aim A Design a Personal Fitness Training ProgrammePersonal information• Personal goals (SMARTER)• Aims• Objectives• Lifestyle• Physical Activity Questionnaire• Medical History questionnaire• Attitude and Motivation towards Exercise	 Unit 3: Learning Aim C Implement (do) <u>a Personal Fitness Training Programme</u> Safely implement the programme Take part in the training programme to the best of your ability. Wear the correct training gear. Conduct the training programme safely Complete and record your outcomes for every session accurately.
Unit 3: Learning Aim A Design a Personal Fitness Training Programme Programme Design• Use of personal information• Selection of appropriate training method/activity• Safe and creative design• Application of FIIT• Application of SPIRRAV• Include warm-up/cool-down• Discuss HR training zones• Include RPE	Unit 3: Learning Aim C Implement (do) a Personal Fitness Training ProgrammeSession training diary• Include date/time/duration/ location of training• Aims/objectives met?• Type of training done• How FITT was adapted from last time• Log achievements and progress• List resources used• How progressive overload was used• HR and RPE recorded
Unit 3: Learning Aim B Musculoskeletal and cardiorespiratory training systems and the effects on the body when training • Location of the major muscles and bones • Structure and function of the 4 synovial joints • Short term effects of training on these systems • Structures of the cardiovascular	Unit 3: Learning Aim C Implement (do)a Personal Fitness Training ProgrammeMeasures for success• Intrinsic/extrinsic motivation• Benefits of motivation/self- confidence• Motivation for training and feedback on how the sessions felt• Adaptations• Achievements against aims/ objectives/goals/targets

and respiratory systems.

Keyword	Definition
SMARTER Targets/Goals	Specific, Measureable, Achievable, Realistic, Time-related, Exciting, Recorded
Training Methods	Flexibility training (static/ballistic/PNF) Strength, muscular endurance and power training (Circuit training, Free weights, plyometrics) Aerobic endurance training (Continuous, Fartlek, Interval, Circuit) Speed training (Hollow, Acceleration, Interval)
FITT	Frequency, Intensity, Time, Type
SPIRRAV	Specificity, Progressive overload, Individual needs, Rest and recovery, Reversibility, Adaptation, Variation
Borg RPE	Rate of Perceived exertion
Musculoskeletal system	The muscular and skeletal systems combined.
Cardiorespiratory system	The cardiovascular and respiratory systems combined.
Progressive overload	In order to progress, training needs to be demanding enough to cause the body to adapt, improving performance.
Intrinsic motivation	Motivation that comes from yourself: enjoyment, fun, feeling good.
Extrinsic motivation	Motivation that comes from external factors: rewards, money, medals.
Adaptations	Changes
Psychological	To do with your thoughts, feelings, beliefs and values

Unit 3: Learning Aim D Review your Personal Fitness Training Programme

Review Programme

Discuss short term psychological effects

- After every session
- Evidence of your adaptations
- Strengths
- Areas for Development
- Recommendations for improving future training and performance



Year 11 BTEC Sport – Unit 6 Leading Sports Activities

Unit 6: Learning Aim A Successful Sports leaders and their attributes Types of sports leaders:

- Coaches
- Fitness instructors
- School/college coaches
- National/international coaches
- Amateur coaches

Attributes

- Skills
 - Communication
 - Organisation of equipment
 - Knowledge
- Advanced skills
 - Activity structure
 - Target setting
 - Use of language
 - Evaluation
- Qualities
 - Appearance
 - Enthusiasm
 - Confidence
- Additional qualities
 - Leadership style
 - Motivation
 - Humour
 - Personality

Responsibilities

- Core responsibilities
 - Professional conduct
 - Health and safety
 - Equality
- Wider responsibilities
 - Insurance
 - Child protection
 - Legal obligations
 - Ethics and values
 - Rules and regulations

Unit 6: Learning Aim B The planning of sports activities Select your sport/activity

- Individual sports
- Team sports
- Fitness session

Considerations for planning

- Participants
- Aims and objectives
- Resources
- Warm up
 - Pulse raiser
 - Mobilise
 - Stretch
- Main component of activity
- Safe activities
- Cool down
 - Pulse lower
 - Stretch
- Health and safety guidelines
- Risk assessment

Unit 6: Learning Aim B The leading of sports activities

- Lead
- Demonstration of attributes
- Completion of core and wider responsibilities

Measures of success

- Coverage of planned components
- Meeting set aims and objectives
- Is it organised and safe?

Definition Keyword Attributes A quality or feature of a person's character Qualities A distinctive attribute or characteristic displayed by someone The manner and approach you take when leading Leadership style Responsibilities The things that are required when taking on a role The right to be equal, in terms of rights and opportunities Equality Something that provides protection against accident or injury Insurance Child Protection Protecting children from violence, abuse, neglect and exploitation Legal obligations Things you have to do by law Ethics and values Your beliefs and values Activities to mobilise the joints such as knees, elbows, hips and Mobilise shoulders

Unit 6: Learning Aim C Review the planning and leading of sports activities Review

- Feedback for review
- Methods
- Strengths and areas for improvement

Targets for Development

- SMARTER targets
- Development plan
 - Aims and objectives
 - Goals
 - SMARTER targets
 - Activities and opportunities
 - Barriers



Year 11 BTEC Travel & Tourism – Unit 1: The UK Travel and Tourism Sector

Unit 1: Learning Aim A Understand the UK Travel & Tourism Sector and it's importance to the UK economy

Types of tourism

- Domestic
- Outbound
- Inbound
- Types of travel
 - Leisure travel: day trips, short breaks, holidays, visiting friends & relatives (VFR), staycations, special events.
 - Business Travel: MICE Meetings, Incentives, Conferences, Events
 - Specialist Travel: adventure, health, education, cultural heritage, dark tourism, voluntary work, conservation, eco-tourism.
- Principles of sustainable tourism
 - Definition
 - Principles: tourism attempting to make a low impact on the environment and local culture, whilst helping to generate future employment for local people
 - Benefits: Environmental protection, reduced energy consumption, reduced waste, links with the local community, competitive edge, image, cost savings.
- The importance of the T&T industry on the economy
 - Employment: direct and indirect
- Gross Domestic Product (GDP)
 - a measure of all goods and services over a specific time, tourist numbers and the value of tourism spend:
 - Inbound tourism the number of inbound tourists, type of trip, length of stay, overnight trips, type of accommodation, and their associated spend
 - Domestic tourism the number of domestic tourist, type of trip, length of stay, overnight trips, type of accommodation and their associated spend
 - Understanding the economic contribution of tourism in different regions of the UK
 - identifying visitor expenditure, number of trips taken.
- Economic Multiplier Effect
- Infrastructure Development tourism growth/decline.

FREEDOM		
CONCEPTS	PASSPORTIES THAT HALLYSIS	
SALIS PROHIMAN	ADVENTURE State	
FESTIVAL	PLANT PLANT	

Key Terms	Definition
Domestic tourism	Taking holidays and trips in your own country.
Outbound tourism	Travelling to a different country for a holiday or visit.
Inbound tourism	Visitors from overseas coming into the country for a holiday.
Sustainable tourism	Meeting the needs of present tourists and local communities while protecting the natural, historical and cultural environment for the future.
Direct employment	These are jobs created directly by the T&T industry. Examples may include pilots and travel agents.
Indirect employment	These are jobs that are created to support the T&T industry. Examples may include hotel builders and cleaners of hotels.
GDP	This stands for Gross Domestic Product. It is the value of a country's economy.
Economic Multiplier Effect	This is the direct spending by tourists, circulated throughout the economy on indirect products and services. It also has to be considered that the total value of goods and services (GDP) and the way in which growth/investment in one area tends to lead to growth and new jobs in another (the multiplier effect).
Infrastructure	The basic physical and organisation structure of facilities e.g. buildings, roads, power supplies
Serviced accommodation	These are places to stay that have full furniture and facilities.
Non-serviced accommodation	These are places to stay where the tourist provides their own towels, cutlery and other basic materials.
Ancillary Organisations	These are companies that offer services that help the T&T industry but aren't directly involved. E.g. a cleaner in a hotel.
Common Ownership	This is where the same company owns lots of different businesses.
Vertical integration	This is where a business at one point of the chain of distribution purchases or acquires a business at a higher or lower level in the chain
Horizontal integration	This is where businesses at the same level in the chain of distributi 276 merge together or are purchased by another business

Unit 1: Learning Aim B Know about the industries, and key organisations, within the travel and tourism sector, their roles and interrelationships

- Tour operators:
 - types domestic, outbound, inbound
 - role to assemble and operate component parts of holidays as a package for retail travel agents or direct sales
 - products and services provided package holidays, accommodation, travel, transfers, excursions, other services.
- Travel agents:
 - types multiples, independent, online
 - role to provide expert advice and guidance, arranging and booking trips, excursions and package holidays or individual components
 - products and/or services provided information; foreign exchange; sales and booking service for packaged holidays, accommodation, flights, transfers and ground transport, excursions, tickets, insurance, ancillary services.
- Online Travel Services
- Passenger transport
 - types road, rail, air, sea, regional, national, global
 - role to provide transport from one location to selected destination, safety
 - products and/or services provided transport, hospitality, entertainment, information and advice.
- Accommodation
 - types hotels, motels, guest houses/bed and breakfast, apartments, youth hostels, caravans, chalets, camping, holiday cottages, holiday parks)
 - role to provide a range of accommodation options, services and facilities
 - products and/or services:
 - serviced (room, concierge, restaurant, bar, housekeeping, leisure, conferences)
 - non-serviced (room, self-catering facilities)
 - youth hostels (dormitory/shared rooms, private rooms, self-catering facilities, information and advice, social areas, meals)
 - camping, caravan sites, holiday parks (tent and mobile caravan pitches, static caravan hire, lodge and chalet accommodation, entertainment and hospitality, sports and recreation)
 - location (rural, urban).
- Visitor attractions:
 - type natural, purpose-built, heritage
 - role to provide entertainment, education, recreation, fun, hospitality, other visitor facilities
 - products and/or services information and interpretation, rides, exhibits, events, tours and guides, educational talks, hospitality and catering, souvenirs.
- Arts and entertainment:
 - types special events, festivals, theatre
 - role to entertain, inform
 - products and/or services live entertainment, music and dance, hospitality and catering, merchandise.
- Tourism development and promotion:
 - types national and regional tourism agencies, tourist and visitor information centres, national and local government departments
 - role encourage visitors, increase tourism revenue, promote special events, market and promote the UK, its regions and destinations, provide information, advice and guidance
 - products and/or services information, literature, souvenirs and merchandise, visitor centres, industry representation.









Unit 1: Learning Aim B Know about the industries, and key organisations, within the travel and tourism sector, their roles and interrelationships

- Trade and professional bodies:
 - types Civil Aviation Authority (CAA), Office of Rail Regulation (ORR), the Association of British Travel Agents (ABTA), the Association of Independent Tour Operators (AITO), Ukinbound
 - role regulation and protection, member representation, liaison with government
 - products and/or services dealing with consumer complaints/arbitration, advice and support, representation, insurance, repatriation, licensing.
- Ancillary organisations:
 - types car hire, travel insurance, airport services, event booking, product comparison providers
 - role supporting services for tourists and travellers
 - products and/or services car hire, insurance, parking, lounges, information and advice.
- Private:
 - characteristics organisations owned or controlled by private individuals or shareholders
 - functions sales of goods and services to make a profit, maximise sales revenue, increase market share, support members.
- Public:
 - characteristics funded and sometimes owned by central and local government
 - functions to provide a service, regulation, to educate, promote and/or inform.
- Voluntary:
 - characteristics independent organisations; funded by membership, donations, grants, sales of products and services
 - functions to provide a service, provide/sell products, support members, promote a particular cause, educate and inform.
- Common ownership,
 - corporate groups (vertical integration where a business at one point of the chain of distribution purchases or acquires a business at a higher or lower level of the chain of distribution/horizontal integration where businesses at the same level in the chain of distribution merge together or are purchased by another business)
- Commercial partnerships (separate organisations working together for a common venture).
- Advantages/Disadvantages for organisations of corporate groups/common ownership/commercial partnerships including:
 - marketing and promotion
 - increase sales and income
 - provide good customer care
 - economies of scale
 - shared resources
 - wider customer base.

size of operation

less customer choice loss of personalised customer care inflexibility



Unit 1: Learning Aim C Understand the role of consumer technology in the travel and tourism industry

- Airports (self-service check-in, security)
- Visitor attractions (multi-media, interpretation, online bookings)
- Accommodation (entertainment, communication, online bookings)
- Mobile applications (communication of information, booking, comparison)
- Electronic and mobile ticketing (e-tickets, m-tickets)
- Websites (reviews, booking, virtual tours).

